

## Supplementary material

### Synthesis and crystal structure of two synthetic oxofluoride framework compounds – $\text{Co}_2\text{TeO}_3\text{F}_2$ and $\text{Co}_2\text{SeO}_3\text{F}_2$

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**Table S1-1**

Atomic coordinates and Equivalent Isotropic Displacement Parameters ( $\text{\AA}^2$ ) for  $\text{Co}_2\text{TeF}_2\text{O}_3$ .

Atom	<i>X</i>	<i>Y</i>	<i>z</i>	<i>U</i> (eq)
Te(1)	0.69889(7)	$\frac{1}{4}$	0.18293(9)	0.00634(16)
Co(1)	0.58538(10)	0.56678(7)	0.24427(13)	0.0070(2)
F(1)	0.6619(4)	0.4566(3)	0.5749(6)	0.0119(7)
O(1)	0.6487(7)	$\frac{3}{4}$	0.3931(11)	0.0108(12)
O(2)	0.5410(5)	0.3830(4)	0.0861(7)	0.0082(8)

**Table S1-2**

Atomic coordinates and Equivalent Isotropic Displacement Parameters ( $\text{\AA}^2$ ) for  $\text{Co}_2\text{SeF}_2\text{O}_3$ .

Atom	<i>X</i>	<i>Y</i>	<i>z</i>	<i>U</i> (eq)
Se(1)	0.68383(15)	$\frac{1}{4}$	0.1481(2)	0.0111(3)
Co(1)	0.59148(13)	0.56454(10)	0.2482(2)	0.0077(3)
F(1)	0.6600(5)	0.4535(5)	0.5746(7)	0.0138(10)
O(1)	0.6645(10)	$\frac{3}{4}$	0.3949(11)	0.0097(17)
O(2)	0.5419(6)	0.3786(5)	0.0770(9)	0.0070(12)

*Note.* *U*(eq) is defined as one third of the trace of the orthogonalized  $U_{ij}$  tensor.

**Table S2-1**

Selected Bond Lengths (Å) and Angles (°) for Co<sub>2</sub>TeF<sub>2</sub>O<sub>3</sub>.

Te1—O2 <sup>i</sup>	1.860(4)	Co1—F1 <sup>ii</sup>	2.084(3)
Te1—O2	1.860(4)	Co1—F1	2.157(3)
Te1—O1 <sup>ii</sup>	1.904(6)	F1—Co1 <sup>iv</sup>	2.075(3)
Co1—O2 <sup>iii</sup>	2.049(4)	F1—Co1 <sup>v</sup>	2.084(3)
Co1—F1 <sup>iv</sup>	2.075(3)	O1—Te1 <sup>v</sup>	1.904(6)
Co1—O2	2.078(4)	O1—Co1 <sup>vi</sup>	2.081(3)
Co1—O1	2.081(3)	O2—Co1 <sup>iii</sup>	
O2 <sup>i</sup> —Te1—O2	93.59(17)	F1 <sup>iv</sup> —Co1—F1	78.18(12)
O2 <sup>i</sup> —Te1—O1 <sup>ii</sup>	98.48(11)	O2—Co1—F1	84.25(14)
O2—Te1—O1 <sup>ii</sup>	98.48(11)	O1—Co1—F1	95.76(9)
O2 <sup>iii</sup> —Co1—F1 <sup>iv</sup>	91.33(14)	F1 <sup>ii</sup> —Co1—F1	93.22(12)
O2 <sup>iii</sup> —Co1—O2	78.97(15)	Co1 <sup>iv</sup> —F1—Co1 <sup>v</sup>	125.10(15)
F1 <sup>iv</sup> —Co1—O2	86.80(14)	Co1 <sup>iv</sup> —F1—Co1	101.82(13)
O2 <sup>iii</sup> —Co1—O1	101.66(12)	Co1 <sup>v</sup> —F1—Co1	130.10(15)
F1 <sup>iv</sup> —Co1—O1	97.21(9)	Te1 <sup>v</sup> —O1—Co1	116.02(2)
O2—Co1—O1	175.91(12)	Te1 <sup>v</sup> —O1—Co1 <sup>vi</sup>	116.02(2)
O2 <sup>iii</sup> —Co1—F1 <sup>ii</sup>	93.89(14)	Co1—O1—Co1 <sup>vi</sup>	127.70(3)
F1 <sup>iv</sup> —Co1—F1 <sup>ii</sup>	166.69(12)	Te1—O2—Co1 <sup>iii</sup>	134.68(21)
O2—Co1—F1 <sup>ii</sup>	82.19(14)	Te1—O2—Co1	116.54(20)
O1—Co1—F1 <sup>ii</sup>	93.73(9)	Co1 <sup>iii</sup> —O2—Co1	101.03(16)
O2 <sup>iii</sup> —Co1—F1	160.69(14)		

**Table S2-2**

Selected Bond Lengths (Å) and Angles (°) for Co<sub>2</sub>SeF<sub>2</sub>O<sub>3</sub>.

Se1—O2 <sup>i</sup>	1.692(5)	Co1—F1 <sup>ii</sup>	2.039(4)
Se1—O2	1.692(5)	Co1—F1	2.130(4)
Se1—O1 <sup>ii</sup>	1.747(7)	F1—Co1 <sup>iv</sup>	2.067(4)
Co1—O2 <sup>iii</sup>	2.073(5)	F1—Co1 <sup>v</sup>	2.039(4)
Co1—F1 <sup>iv</sup>	2.067(4)	O1—Se1 <sup>v</sup>	1.747(7)
Co1—O2	2.105(5)	O1—Co1 <sup>vi</sup>	2.083(3)
Co1—O1	2.083(3)	O2—Co1 <sup>iii</sup>	2.073(5)
O2 <sup>i</sup> —Se1—O2	98.96(23)	F1 <sup>iv</sup> —Co1—F1	77.54(15)
O2 <sup>i</sup> —Se1—O1 <sup>ii</sup>	102.10(16)	O2—Co1—F1	86.39(19)
O2—Se1—O1 <sup>ii</sup>	102.10(16)	O1—Co1—F1	95.46(13)
O2 <sup>iii</sup> —Co1—F1 <sup>iv</sup>	89.80(17)	F1 <sup>ii</sup> —Co1—F1	96.96(15)
O2 <sup>iii</sup> —Co1—O2	78.24(19)	Co1 <sup>iv</sup> —F1—Co1 <sup>v</sup>	124.45(18)
F1 <sup>iv</sup> —Co1—O2	88.38(18)	Co1 <sup>iv</sup> —F1—Co1	102.46(17)
O2 <sup>iii</sup> —Co1—O1	101.06(14)	Co1 <sup>v</sup> —F1—Co1	128.86(19)
F1 <sup>iv</sup> —Co1—O1	97.45(13)	Se1 <sup>v</sup> —O1—Co1	116.96(3)
O2—Co1—O1	174.14(15)	Se1 <sup>v</sup> —O1—Co1 <sup>vi</sup>	116.96(3)
O2 <sup>iii</sup> —Co1—F1 <sup>ii</sup>	93.16(17)	Co1—O1—Co1 <sup>vi</sup>	125.85(4)
F1 <sup>iv</sup> —Co1—F1 <sup>ii</sup>	169.91(15)	Se1—O2—Co1 <sup>iii</sup>	133.05(27)
O2—Co1—F1 <sup>ii</sup>	82.80(18)	Se1—O2—Co1	117.99(26)
O1—Co1—F1 <sup>ii</sup>	91.45(14)	Co1 <sup>iii</sup> —O2—Co1	101.76(21)
O2 <sup>iii</sup> —Co1—F1	160.39(18)		

*Note.* Symmetry transformations used to generate equivalent atoms:

- (i) x, 0.5-y, z; (ii) 1.5-x, 1-y, -0.5+z; (iii) 1-x, 1-y, -z; (iv) 1-x, 1-y, 1-z;  
 (v) 1.5-x, 1-y, 0.5+z; (vi) x, 1.5-y, z.

**Table S3-1**

Results from Bond Valence Sum (BVS) calculations of  $\text{Co}_2\text{TeF}_2\text{O}_3$ .

<u>Atoms</u>	<u>Bonding distance</u>	<u>Bond valence</u>	<u>Atoms</u>	<u>Bonding distance</u>	<u>Bond valence</u>
Te1—O1 <sup>ii</sup>	1.904(5)	1.2	O1—Co1	2.081(3)	0.35
Te1—O2	1.860(3)	1.4	O1—Co1 <sup>vi</sup>	2.081(3)	0.35
Te1—O2 <sup>i</sup>	1.860(3)	1.4	O1—Te1 <sup>v</sup>	1.904(5)	1.20
		<b>4.0</b>			<b>1.9</b>
Co1—O1	2.081(3)	0.35	O2—Co1	2.049(4)	0.35
Co1—O2 <sup>iii</sup>	2.049(4)	0.38	O2—Co1 <sup>iii</sup>	2.049(4)	0.35
Co1—O2	2.078(3)	0.35	O2—Te1	1.860(3)	1.30
Co1—F1 <sup>iv</sup>	2.075(3)	0.31			<b>2.0</b>
Co1—F1 <sup>ii</sup>	2.084(3)	0.30	F1—Co1 <sup>iv</sup>	2.075(3)	0.30
Co1—F1	2.157(3)	0.25	F1—Co1 <sup>v</sup>	2.084(3)	0.30
		<b>1.9</b>	F1—Co1	2.157(3)	0.25
					<b>0.85</b>

**Table S3-2**

Results from Bond Valence Sum (BVS) calculations of  $\text{Co}_2\text{SeF}_2\text{O}_3$ .

<u>Atoms</u>	<u>Bonding distance</u>	<u>Bond valence</u>	<u>Atoms</u>	<u>Bonding distance</u>	<u>Bond valence</u>
Se1—O1 <sup>ii</sup>	1.747(7)	1.2	O1—Co1	2.083(3)	0.35
Se1—O2	1.692(5)	1.2	O1—Co1 <sup>vi</sup>	2.083(3)	0.36
Se1—O2 <sup>i</sup>	1.692(5)	1.3	O1—Se1 <sup>v</sup>	1.747(7)	1.19
		<b>3.7</b>			<b>1.9</b>
Co1—O1	2.083(3)	0.35	O2—Co1	2.105(5)	0.33
Co1—O2 <sup>iii</sup>	2.073(5)	0.36	O2—Co1 <sup>iii</sup>	2.073(5)	0.36
Co1—O2	2.105(5)	0.33	O2—Se1	1.692(5)	1.38
Co1—F1 <sup>iv</sup>	2.067(4)	0.27			<b>2.0</b>
Co1—F1 <sup>ii</sup>	2.039(4)	0.34	F1—Co1 <sup>iv</sup>	2.067(4)	0.32
Co1—F1	2.130(4)	0.32	F1—Co1 <sup>v</sup>	2.039(4)	0.34
		<b>1.97</b>	F1—Co1	2.130(4)	0.27
					<b>0.93</b>

*Note.* All the calculation were made using following  $R_0$  values: 1.977 for Te—O bonds <sup>[A]</sup>, 1.811 for Se—O bonds <sup>[A]</sup>, 1.692 for Co—O bonds <sup>[A]</sup> and 1.640 for Co—F bonds <sup>[B]</sup>.

Symmetry transformations used to generate equivalent atoms:

- (i)  $x, 0.5-y, z$ ; (ii)  $1.5-x, 1-y, -0.5+z$ ; (iii)  $1-x, 1-y, -z$ ; (iv)  $1-x, 1-y, 1-z$ ;  
 (v)  $1.5-x, 1-y, 0.5+z$ ; (vi)  $x, 1.5-y, z$ .

### References

- [A] I.D. Brown and D. Altermatt, *Acta Cryst.* B41 (1985) 244-247.  
[B] N.E. Brese and M. O'Keeffe, *Acta Cryst.* B47 (1991) 192-197.