

## Supporting information

### Diorganotin(IV) 2-pyridyl selenolates: Synthesis, structures and their utility as molecular precursors for the preparation of tin selenide nanocrystals and thin films

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#### Figure Captions

- Fig. S1**  $^{77}\text{Se}\{\text{H}\}$ NMR spectrum of  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})\text{Cl}]$  (**7**).
- Fig. S2**  $^{119}\text{Sn}\{\text{H}\}$ NMR spectrum of  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})\text{Cl}]$  (**7**).
- Fig. S3**  $^{77}\text{Se}\{\text{H}\}$ NMR spectrum of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**).
- Fig. S4**  $^{119}\text{Sn}\{\text{H}\}$ NMR spectrum of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**).
- Fig. S5** TG curves of a)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**1**), b)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**2**) and c)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})\text{Cl}]$  (**7**).
- Fig. S6** TG curves of a)  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) and b)  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**).
- Fig. S7** TG curves of a)  $[{}^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**5**) and b)  $[{}^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})\text{Cl}]$  (**8**).
- Fig. S8** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**1**) in 3 ml of oleylamine at 215 °C for 15 min.
- Fig. S9** a) XRD pattern, b) EDX spectrum and c) SEM image of  $\text{SnSe}_2$  obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) in 3 ml of oleylamine at 215 °C for 25 min.
- Fig. S10** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) in 6 ml of oleylamine at 215 °C for 25 min.
- Fig. S11** a) XRD pattern and b) EDX spectrum of SnSe obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**) in 3 ml of oleylamine at 215 °C for 25 min.
- Fig. S12** a) XRD pattern and b) EDX spectrum of  $\text{SnSe}_2$  obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**) in 6 ml of oleylamine at 215 °C for 25 min.
- Fig. S13** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of  $[{}^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**5**) in 3 ml of oleylamine at 215 °C for 25 min.
- Fig. S14** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of  $[{}^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**5**) in 6 ml of oleylamine at 215 °C for 25 min.

**Fig. S15** a) XRD pattern and b) EDX spectrum of SnSe obtained by the AACVD of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) at 490 °C for 3 h on glass substrate.

**Fig. S16** a) XRD pattern, b) EDX spectrum and c) SEM of SnSe deposited by the AACVD of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) at 490 °C for 3 h on silicon substrate.

**.Fig. S17** a) EDX spectrum and b) SEM of SnSe obtained by the AACVD of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) at 530 °C for 3 h on glass substrate.

**Fig. S18** a) XRD pattern, b) EDX spectrum and c) SEM of SnSe obtained by AACVD of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) at 530 °C for 3 h on silicon substrate.

**Fig. S19** a) TEM image and b) SAED pattern of SnSe obtained by thermolysis of [Me<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**1**) in 3 ml oleylamine at 215 °C for 25 min.

**Fig. S20** a) TEM image and b) SAED pattern of SnSe<sub>2</sub> obtained by thermolysis of [Et<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**3**) in 3 ml oleylamine at 215 °C for 25 min.

**Fig. S21** a) TEM image and b) SAED pattern of SnSe obtained by thermolysis of [Et<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**3**) in 6 ml oleylamine at 215 °C for 25 min.

**Fig. S22** a), c) TEM image and b), d) SAED pattern of SnSe obtained by thermolysis of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) in 3 ml oleylamine at 215 °C for 25 min.

**Fig. S23** a) TEM image and b) SAED pattern of SnSe<sub>2</sub> obtained by thermolysis of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) in 6 ml oleylamine at 215 °C for 25 min.

**Table S1.** Experimental details and results of tin selenide nanostructures obtained from the complexes, **1** and **3-5**.

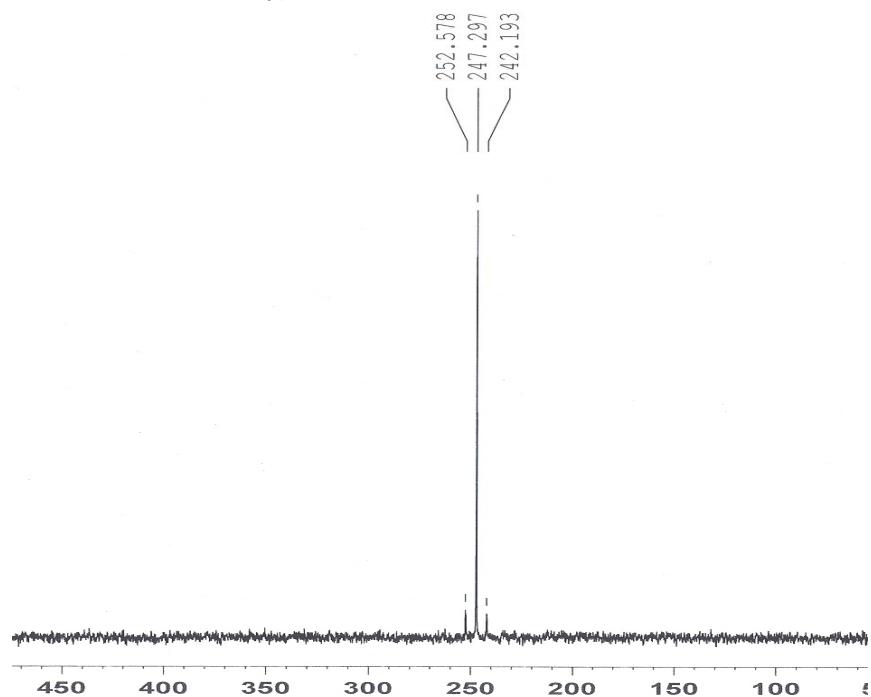
**Table S2.** Lattice parameters of tin selenide nanostructures calculated by using eq 1.

**Table S3.** Experimental details and results of tin selenide thin films deposited by AACVD of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**).

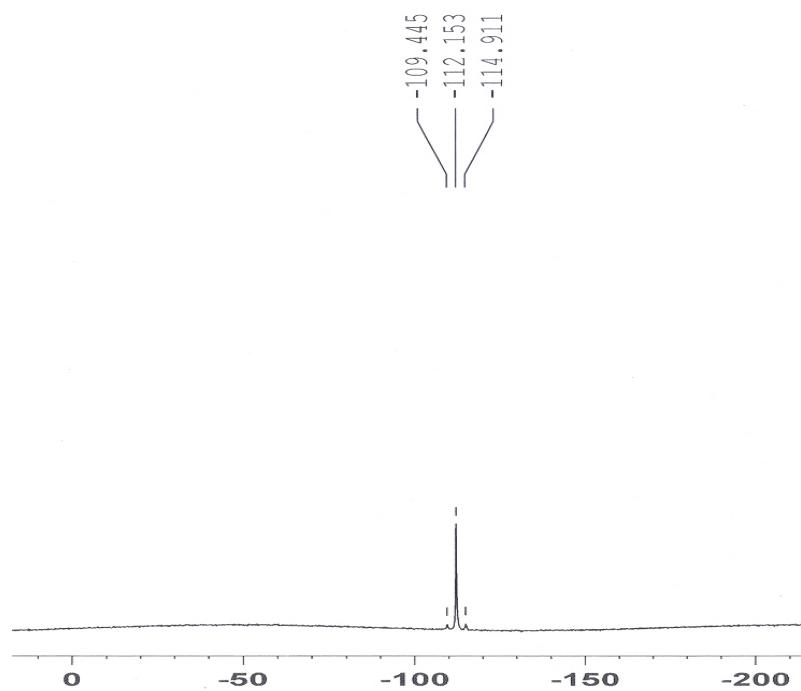
Crystal Information File of [Me<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**1**)

Crystal Information File of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**)

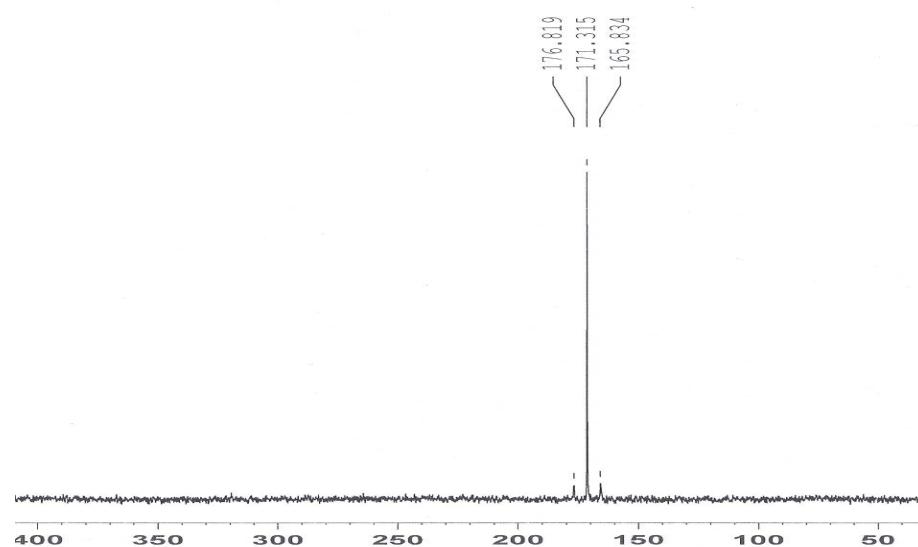
Crystal Information File of [Me<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)Cl] (**7**)



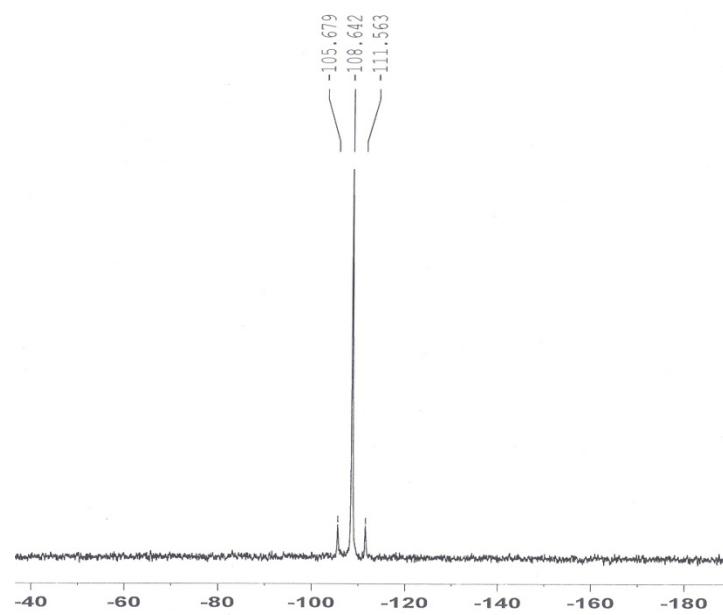
**Fig. S1**  $^{77}\text{Se}\{\text{H}\}$  NMR spectrum of  $[\text{Me}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_3(\text{Me}-3)\text{N})\text{Cl}]$  (7)



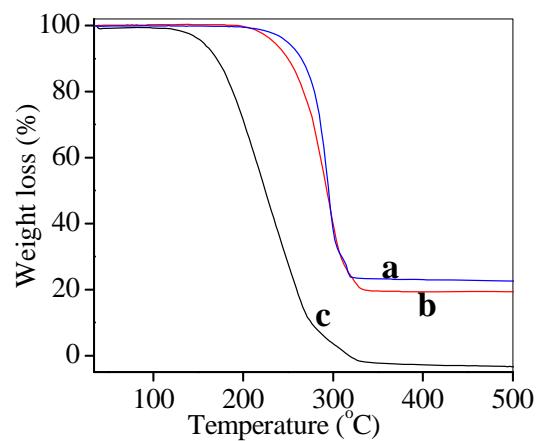
**Fig. S2**  $^{119}\text{Sn}\{\text{H}\}$  NMR spectrum of  $[\text{Me}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_3(\text{Me}-3)\text{N})\text{Cl}]$  (7)



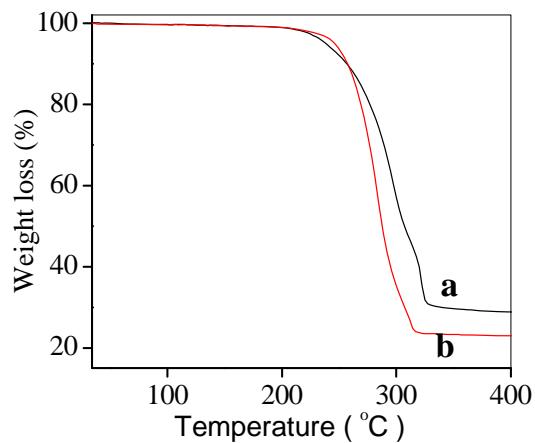
**Fig. S3**  $^{77}\text{Se}\{\text{H}\}$ NMR spectrum of  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2]$  (**3**)



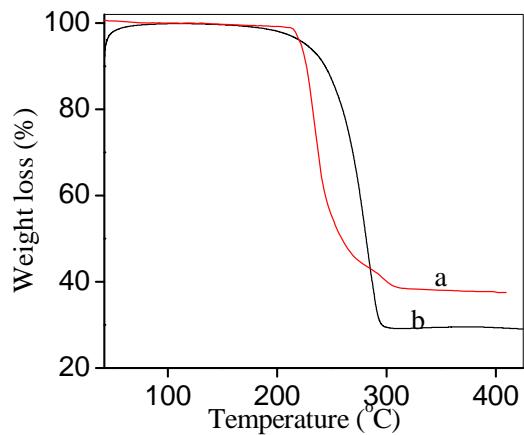
**Fig. S4**  $^{119}\text{Sn}\{\text{H}\}$ NMR spectrum of  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2]$  (**3**)



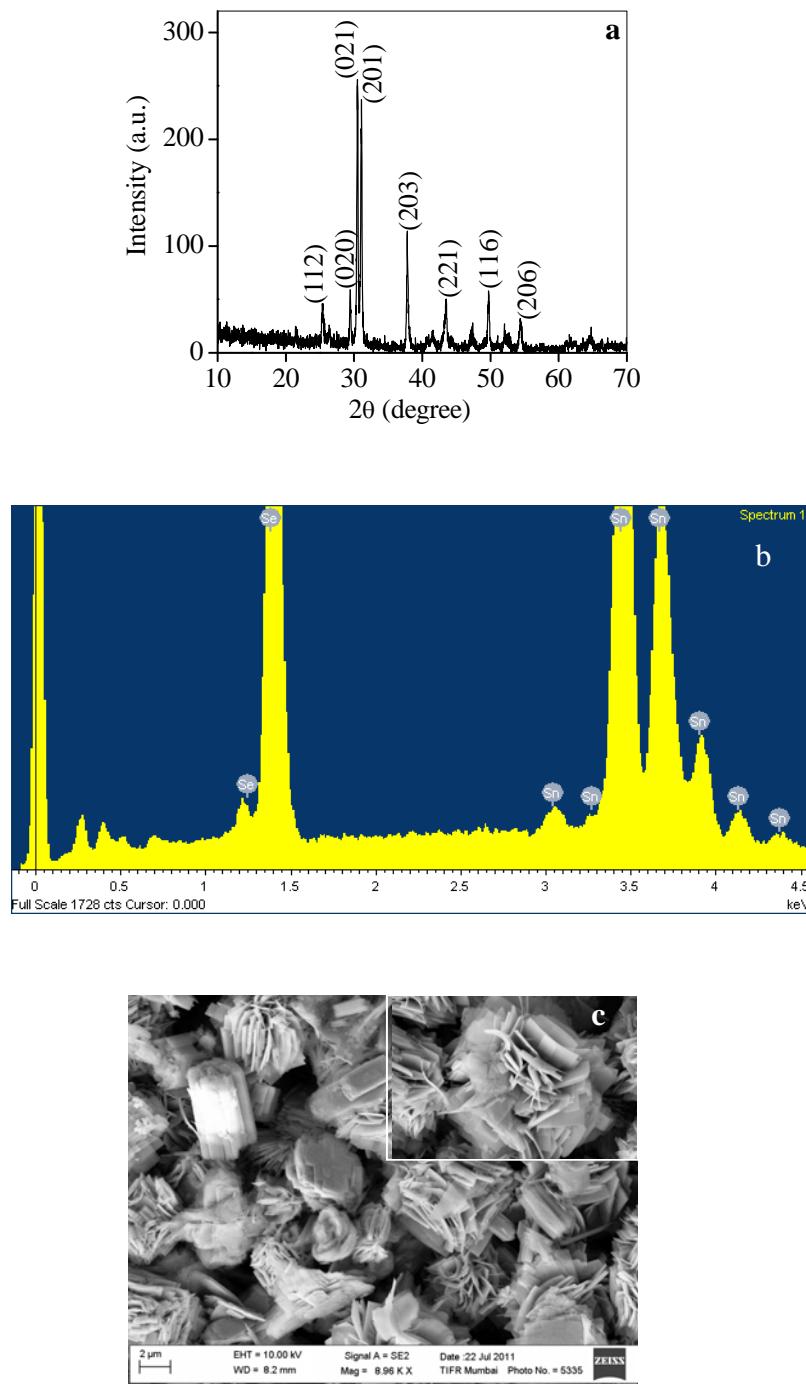
**Fig. S5** TG curves of a)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**1**), b)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me-3})\text{N})_2]$  (**2**) and c)  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_3(\text{Me-3})\text{N})\text{Cl}]$  (**7**).



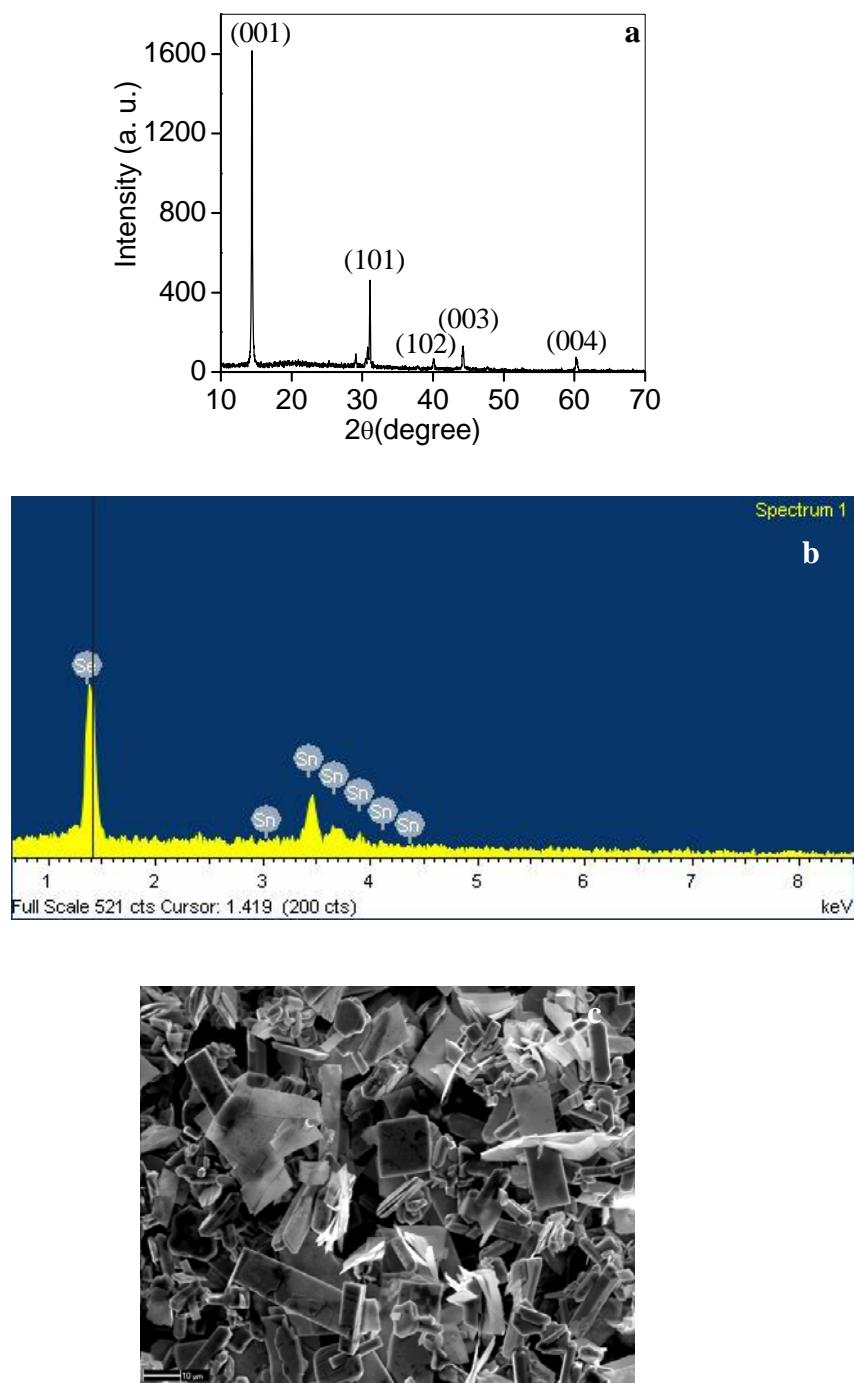
**Fig. S6** TG curves of a)  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2]$  (**3**) and b)  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**).



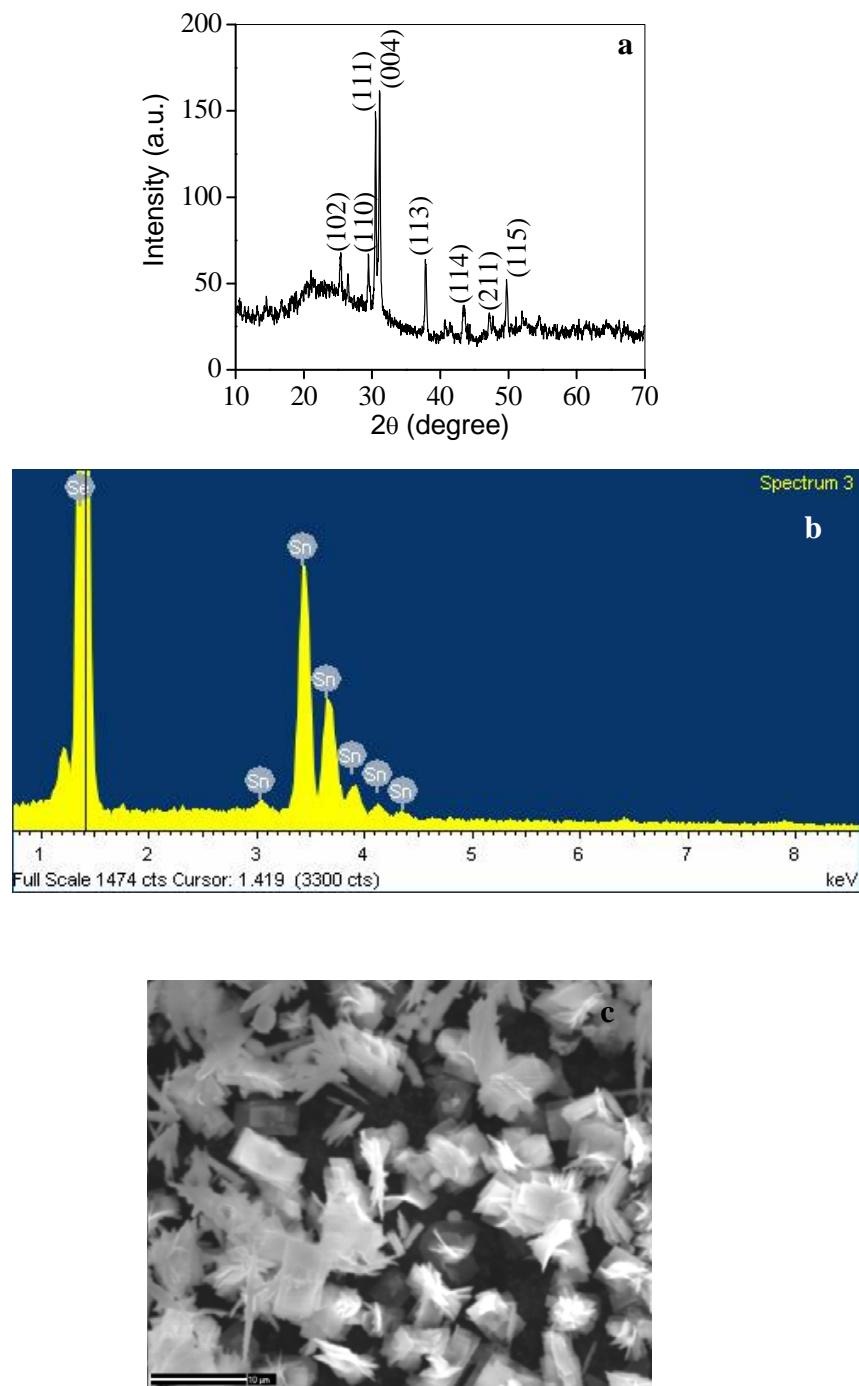
**Fig. S7** TG curves of a)  $[\text{tBu}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2]$  (**5**) and b)  $[\text{tBu}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})\text{Cl}]$  (**8**).



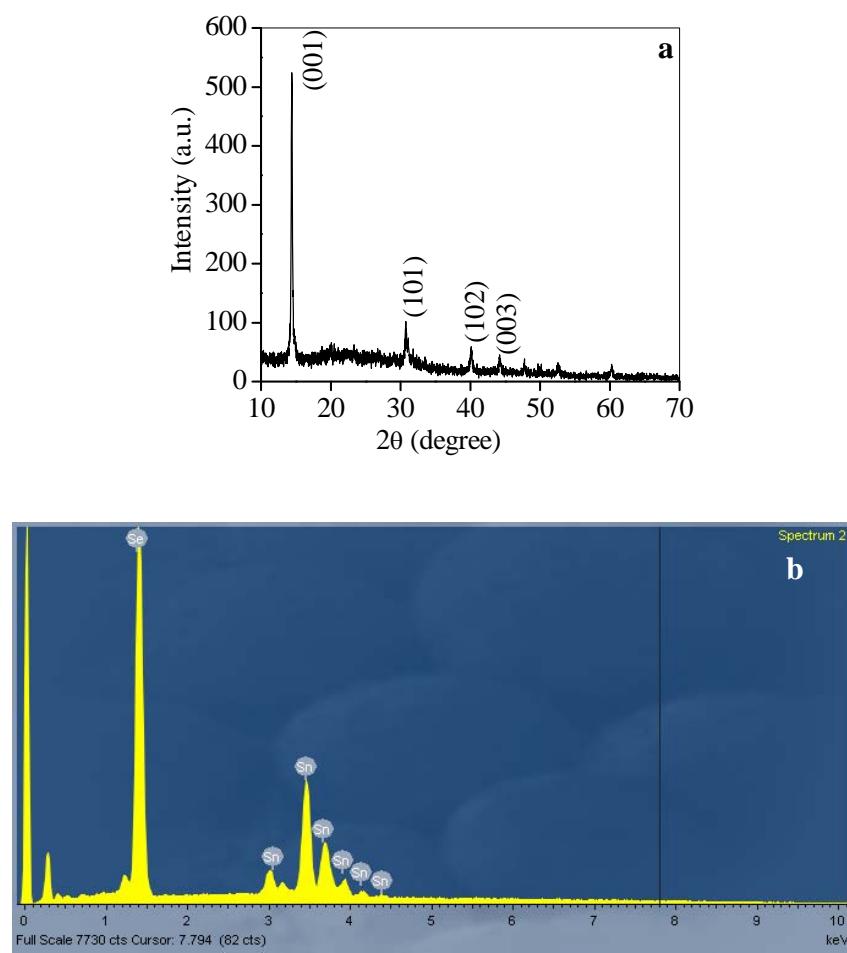
**Fig. S8** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by the thermolysis of  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**1**) in 3 ml of oleylamine at 215 °C for 15 min.



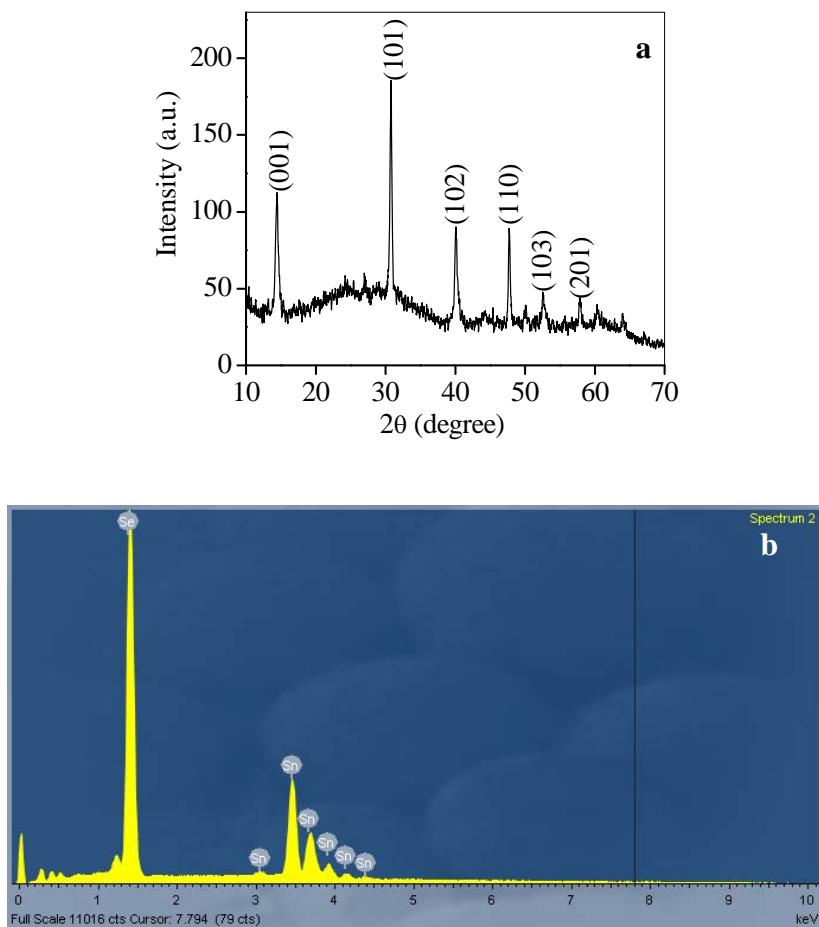
**Fig. S9** a) XRD pattern, b) EDX spectrum and c) SEM image of  $\text{SnSe}_2$  obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) in 3 ml of oleylamine at  $215^\circ\text{C}$  for 25 min.



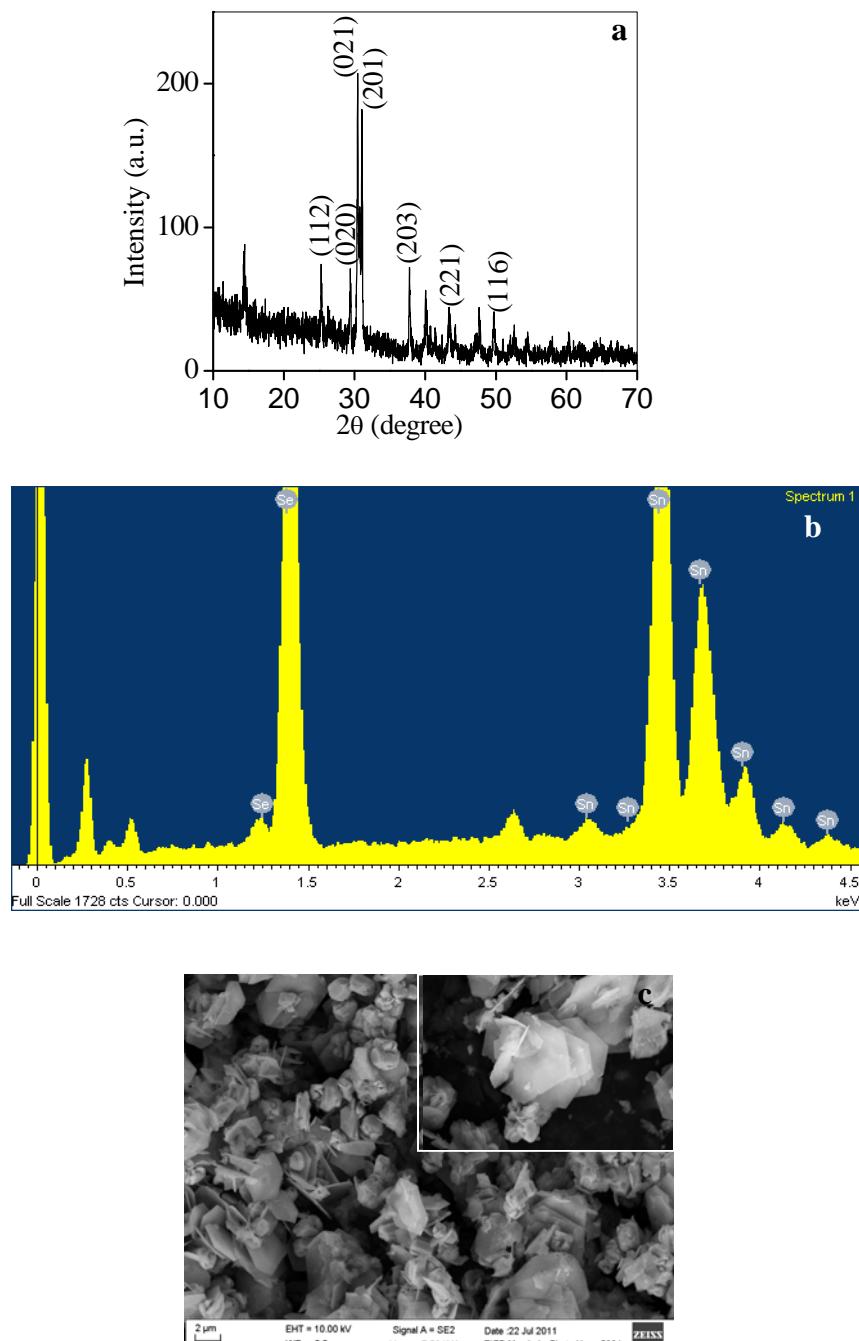
**Fig. S10** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) in 6 ml of oleylamine at  $215^\circ\text{C}$  for 25 min.



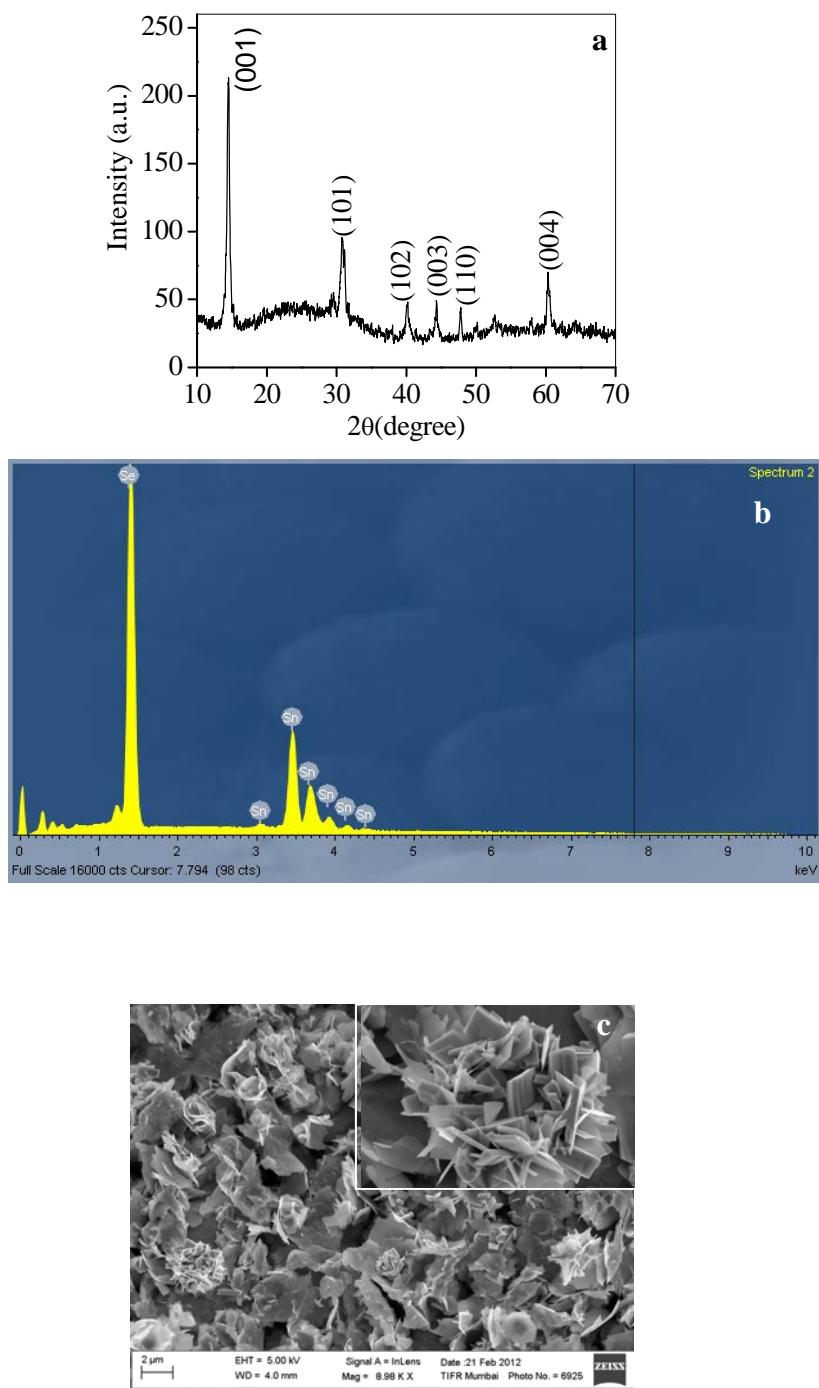
**Fig. S11** a) XRD pattern and b) EDX of  $\text{SnSe}_2$  obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**) in 3 ml of oleylamine at  $215^\circ\text{C}$  for 25 min.



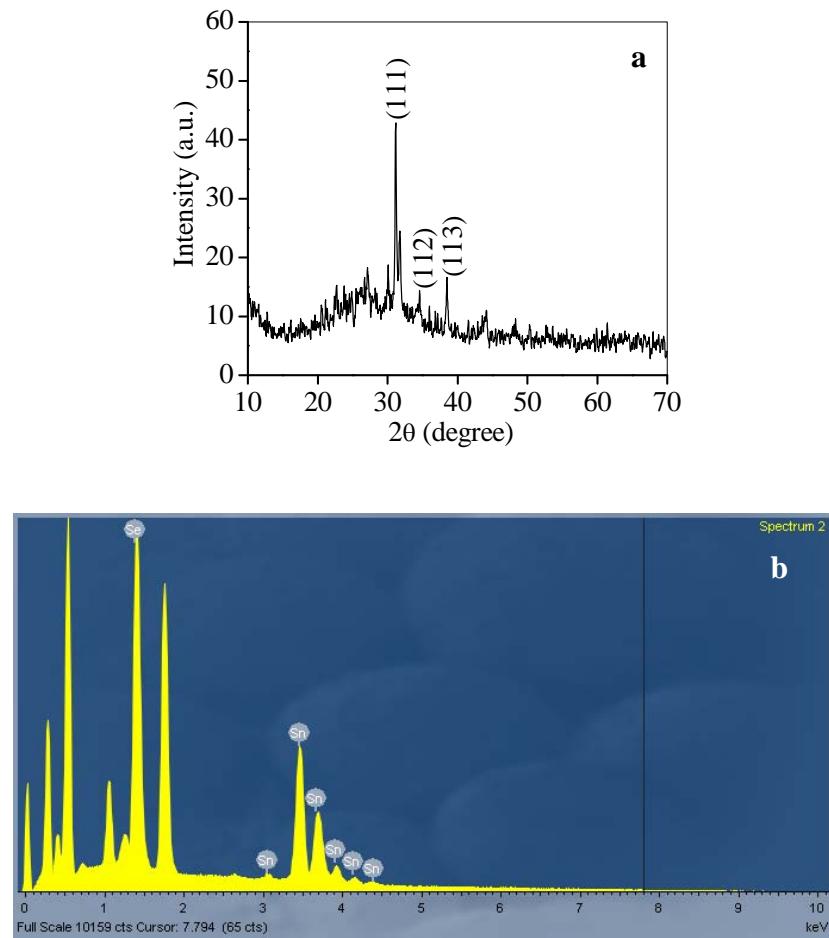
**Fig. S12** a) XRD pattern and b) EDX spectrum of  $\text{SnSe}_2$  obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_3(\text{Me}-3)\text{N})_2]$  (**4**) in 6 ml of oleylamine at  $215^\circ\text{C}$  for 25 min.



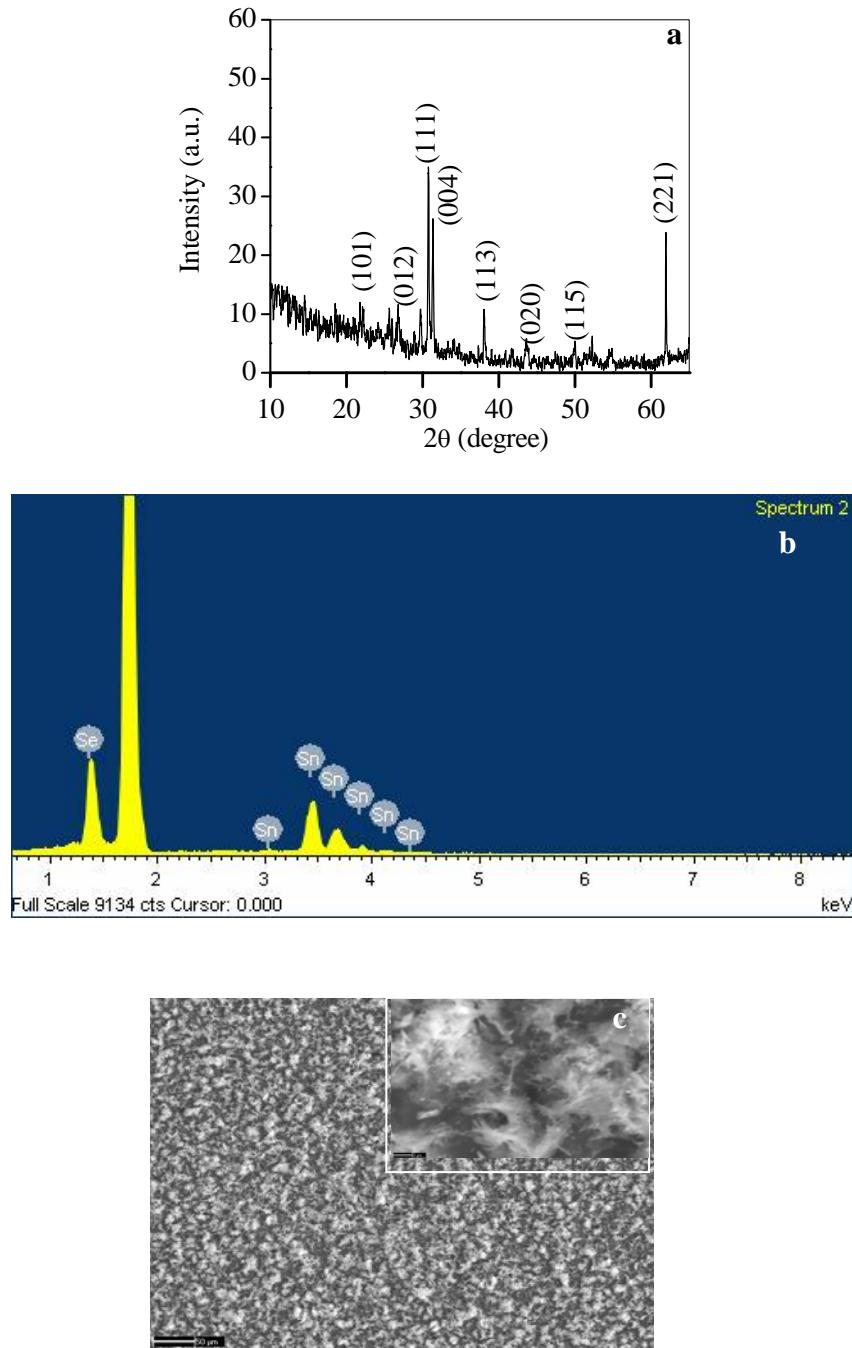
**Fig. S13** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of [ $^t\text{Bu}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2$ ] (**5**) in 3 ml of oleylamine at 215 °C for 25 min.



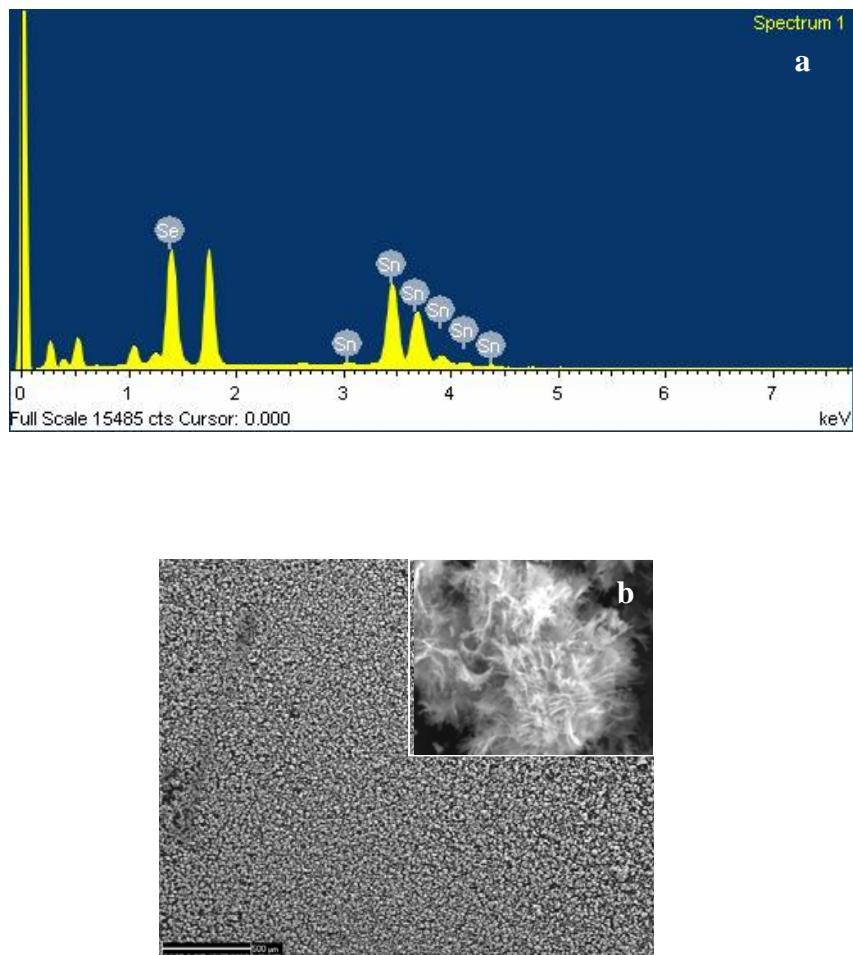
**Fig. S14** a) XRD pattern, b) EDX spectrum and c) SEM image of SnSe obtained by thermolysis of [ $^t\text{Bu}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2$ ] (**5**) in 6 ml of oleylamine at 215 °C for 25 min.



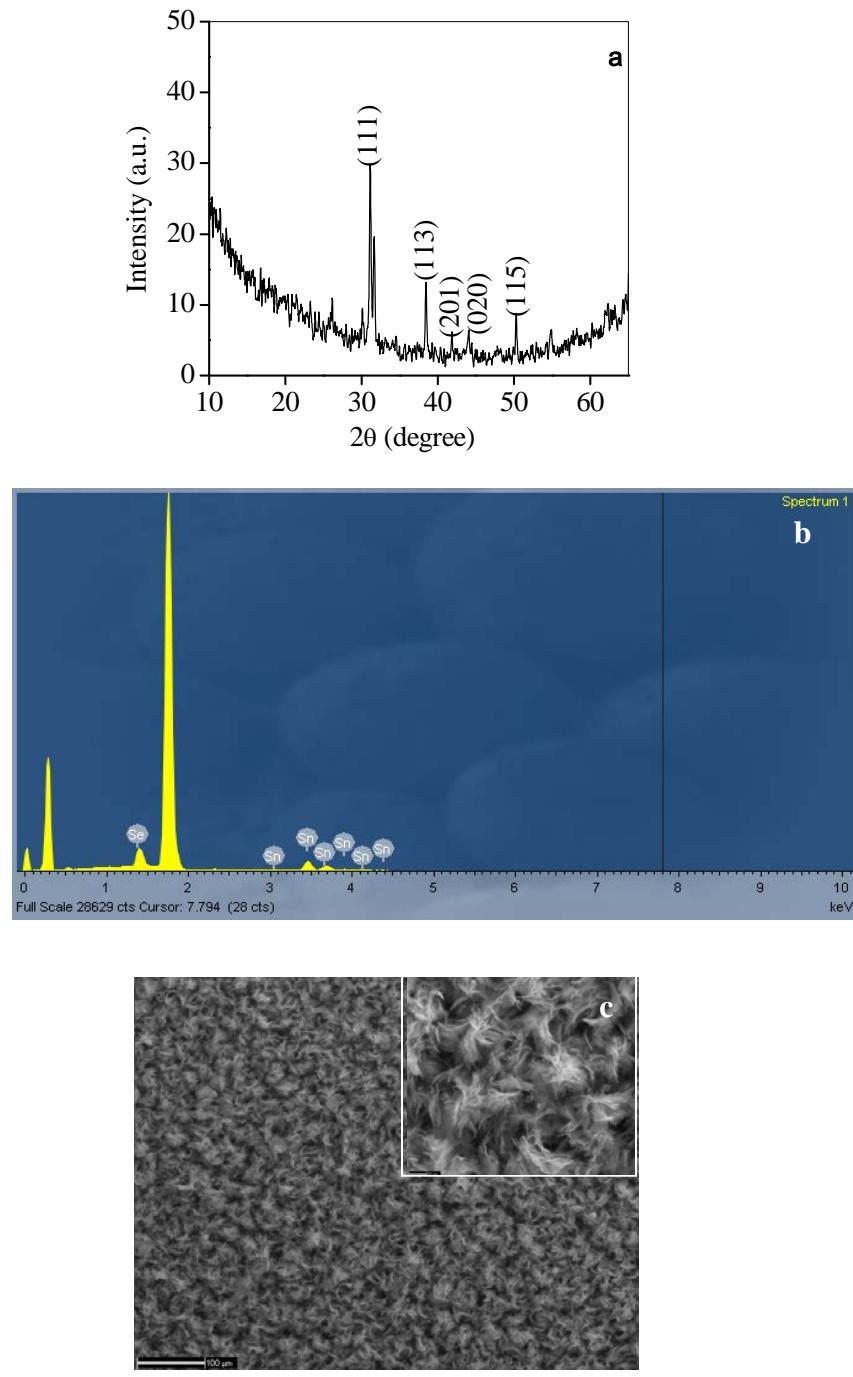
**Fig. S15** a) XRD pattern and b) EDX spectrum of SnSe deposited by AACVD of  $[^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**5**) at 490 °C for 3 h on glass substrate.



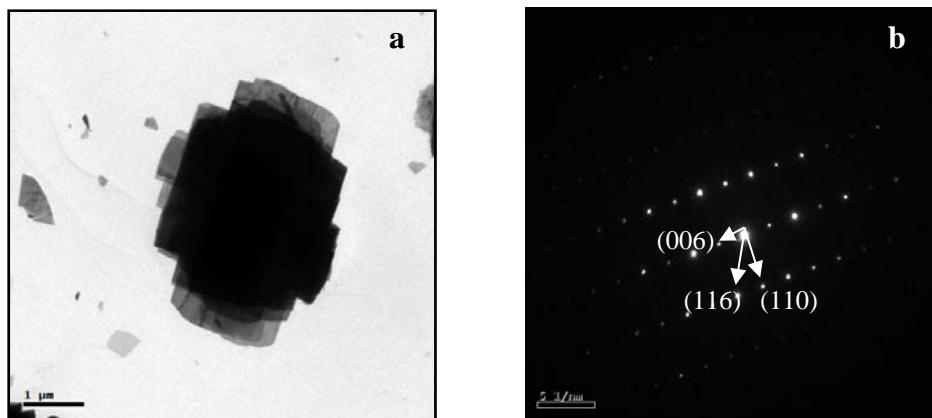
**Fig. S16** a) XRD pattern, b) EDX spectrum and c) SEM of SnSe deposited by the AACVD of [ $^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2$ ] (**5**) at 490 °C for 3 h on silicon substrate.



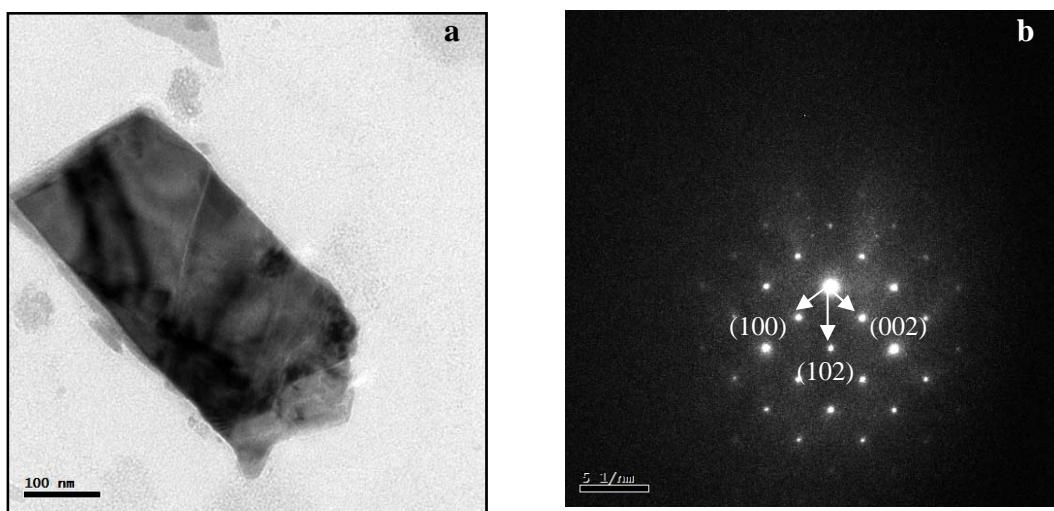
**Fig. S17** a) EDX spectrum and b) SEM of SnSe deposited by AACVD of  $[^t\text{Bu}_2\text{Sn}(\text{Se}-\text{C}_5\text{H}_4\text{N})_2]$  (**5**) at  $530^\circ\text{C}$  for 3 h on glass substrate.



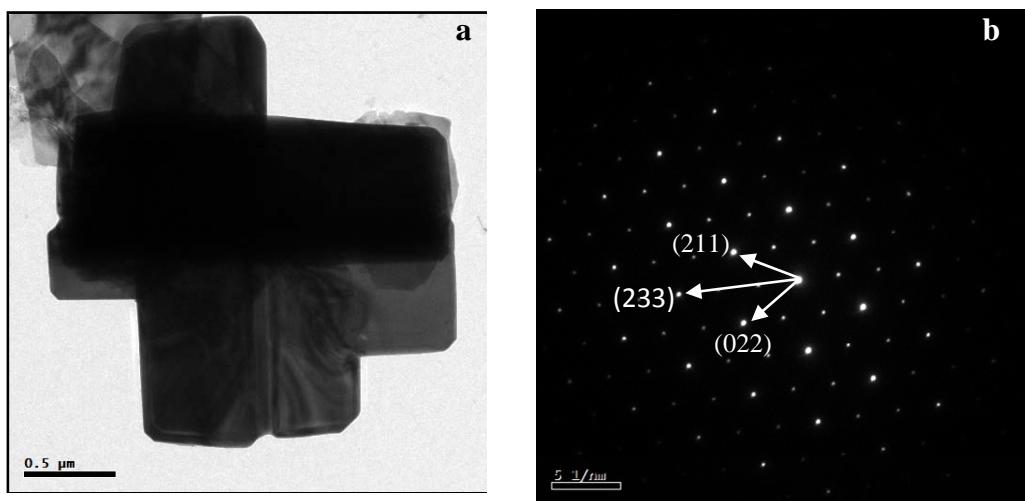
**Fig. S18** a) XRD pattern, b) EDX spectrum and c) SEM of SnSe obtained by AACVD of  $[^t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**5**) at 530 °C for 3 h on silicon substrate.



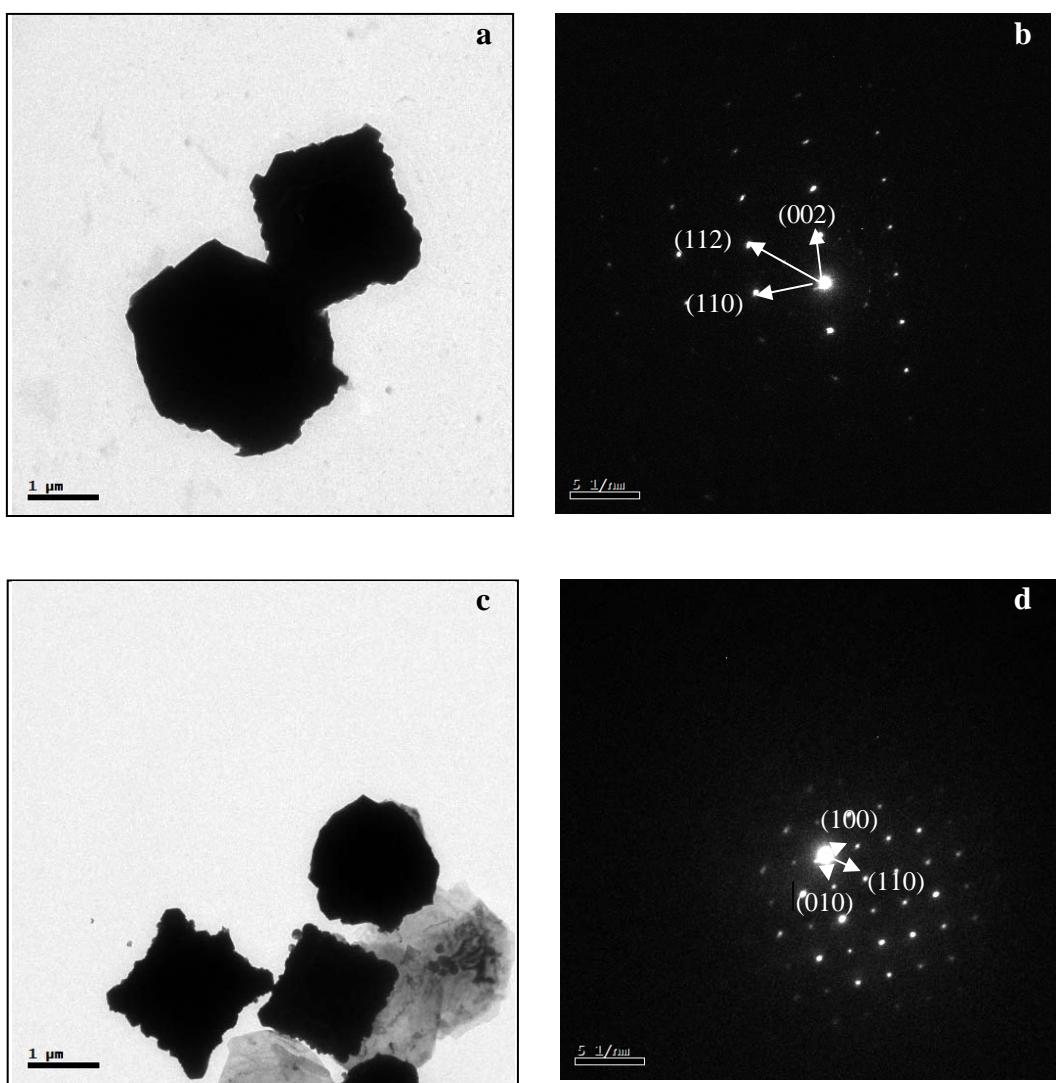
**Fig. S19** a) TEM image and b) SAED pattern of SnSe obtained by thermolysis of  $[\text{Me}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**1**) in 3 ml oleylamine at  $215^\circ\text{C}$  for 25 min.



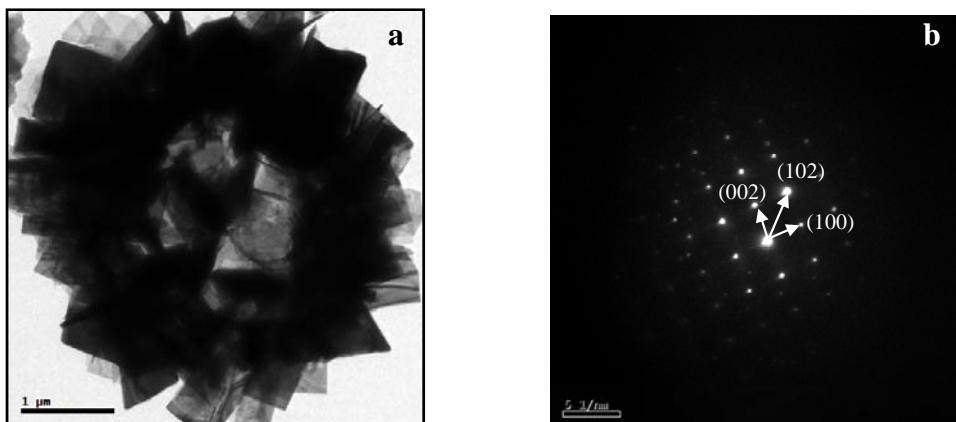
**Fig. S20** a) TEM image and b) SAED pattern of SnSe<sub>2</sub> obtained by thermolysis of [Et<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**3**) in 3 ml oleylamine at 215 °C for 25 min.



**Fig. S21** a) TEM image and b) SAED pattern of SnSe obtained by thermolysis of  $[\text{Et}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2]$  (**3**) in 6 ml oleylamine at 215 °C for 25 min.



**Fig. S22** a), c) TEM images and SAED patterns of b) hexagonal and d) square like SnSe obtained by thermolysis of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) in 3 ml oleylamine at 215 °C for 25 min.



**Fig. S23** a) TEM image and b) SAED pattern of SnSe<sub>2</sub> obtained by thermolysis of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (**5**) in 6 ml oleylamine at 215 °C for 25 min.

**Table S1.** Experimental details and results of tin selenide nanostructures obtained from the complexes, **1** and **3-5**.

Exp. No.	Precursor	Amoun t of OA taken in the flask (ml)	Amount of OA/CH <sub>2</sub> Cl <sub>2</sub> taken with the precursor (ml)		React ion durati on (min utes)	EDX Se/Sn ratio	Phase/composition/JCPD S No.	Morphology as revealed by SEM
			OA	CH <sub>2</sub> Cl <sub>2</sub>				
1	[Me <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] ( <b>1</b> )	3	—	2	15	0.87	Ortho/SnSe/81-0013	Bundles of rectangular sheets
3	[Et <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] ( <b>3</b> )	3	—	2	25	2.10	Hex/SnSe <sub>2</sub> /23-0602	Rectangular bars
4	[Et <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] ( <b>3</b> )	3	3	—	25	0.99	Ortho/SnSe/72-1460	Regular bars
5	[Et <sub>2</sub> Sn{Se-C <sub>5</sub> H <sub>3</sub> (3-Me)N} <sub>2</sub> ] ( <b>4</b> )	3	—	2	25	1.52	Hex/SnSe <sub>2</sub> /23-0602	Irregular hexagonal sheets
6	[Et <sub>2</sub> Sn{Se-C <sub>5</sub> H <sub>3</sub> (3-Me)N} <sub>2</sub> ] ( <b>4</b> )	3	3	—	25	1.74	Hex/SnSe <sub>2</sub> /23-0602	Regular hexagons
5	[ <sup>t</sup> Bu <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] ( <b>5</b> )	3	—	2	25	0.96	Mixture of Ortho/SnSe/81-0013 and Hex/SnSe <sub>2</sub> /23-0602	Hexagon sheets and square like structures
6	[ <sup>t</sup> Bu <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] ( <b>5</b> )	3	3	—	25	1.50	Hex/SnSe <sub>2</sub> /23-0602	Rectangular sheets

**Table S2.** Lattice parameters of tin selenide nanostructures calculated by using eq 1.

Exp. No.	Precursor	Amount of OA taken in the flask (ml)	Amount of OA/CH <sub>2</sub> Cl <sub>2</sub> taken with the precursor (ml)		Reaction duration (minutes )	JCPDS No.	Lattice parameter Calculated by eq. 1 Reported in JCPDS )		
			OA	CH <sub>2</sub> Cl <sub>2</sub>			a	b	c
1	[Me <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] (1)	3	—	2	15	81-0013	5.905(8) (5.928)	6.020(6) (5.970)	12.406(9) (12.28)
2	[Et <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] (3)	3	—	2	25	23-0602	3.810 (3.810)	3.810 (3.810)	6.133(1) (6.140)
3	[Et <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] (3)	3	3	—	25	72-1460	4.424(10) (4.460)	4.149(8) (4.190)	11.491(11) (11.570)
4	[Et <sub>2</sub> Sn{Se-C <sub>5</sub> H <sub>3</sub> (3-Me)N} <sub>2</sub> ] (4)	3	—	2	25	23-0602	3.810 (3.810)	3.810 (3.810)	6.137(4) (6.140)
5	[Et <sub>2</sub> Sn{Se-C <sub>5</sub> H <sub>3</sub> (3-Me)N} <sub>2</sub> ] (4)	3	3	—	25	23-0602	3.810 (3.810)	3.810 (3.810)	6.141(7) (6.140)
6	[ <sup>t</sup> Bu <sub>2</sub> Sn(Se-C <sub>5</sub> H <sub>4</sub> N) <sub>2</sub> ] (5)	3	3	—	25	23-0602	3.810 (3.810)	3.810 (3.810)	6.131(9) (6.140)

**Table S3.** Experimental details and results of tin selenide thin films deposited by AACVD of [ $t\text{Bu}_2\text{Sn}(\text{Se-C}_5\text{H}_4\text{N})_2$ ] (**5**).

Exp. No.	Substrate on which thin film was deposited by AACVD	Substra te tempera ture (°C)	Reacti on durati on (in hour)	JCPDS No.	Lattice parameter calculated by eq 1. (Reported in JCPDS File)		
					a	b	c
1	Si	490	3	72-1460	4.445(10) (4.460)	4.218(3) (4.190)	11.584(6) (11.570)
2	Glass	490	3	72-1460	4.528(8) (4.460)	4.173(5) (4.190)	11.483(5) (11.570)
3	Si	530	3	72-1460	4.513(25) (4.460)	4.194(13) (4.190)	11.509(19) (11.570)
4	glass	530	3	72-1460	4.520(7) (4.460)	4.177(10) (4.190)	11.523(8) (11.570)

## Crystal Information File of [Me<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (1)

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C7 C 0.5033(8) 1.030(2) 1.2936(18) 0.051(5) Uani 1 1 d . . .  
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C10 C 0.5296(7) 1.181(3) 1.485(2) 0.073(7) Uani 1 1 d . . .  
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N2 N 0.4998(6) 1.2318(18) 1.4077(17) 0.063(5) Uani 1 1 d . . .  
C11 C 0.4108(6) 1.400(2) 1.5185(17) 0.066(6) Uani 1 1 d . . .  
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H11C H 0.4279 1.4285 1.5889 0.099 Uiso 1 1 calc R . . .  
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H12B H 0.5098 1.6349 1.2830 0.137 Uiso 1 1 calc R . . .  
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C14 C 0.2597(7) 0.748(2) 1.974(2) 0.076(3) Uani 1 1 d . . .  
H14 H 0.2723 0.7018 2.0400 0.091 Uiso 1 1 calc R . . .  
C15 C 0.2272(6) 0.684(2) 1.903(2) 0.076(3) Uani 1 1 d . . .  
H15 H 0.2186 0.5933 1.9206 0.091 Uiso 1 1 calc R . . .  
C16 C 0.2076(7) 0.747(3) 1.809(2) 0.076(3) Uani 1 1 d . . .  
H16 H 0.1843 0.7032 1.7659 0.091 Uiso 1 1 calc R . . .  
C17 C 0.2216(9) 0.875(3) 1.778(2) 0.090(8) Uani 1 1 d . . .  
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N3 N 0.2529(6) 0.9444(17) 1.8440(19) 0.072(5) Uani 1 1 d . . .  
C18 C 0.3497(7) 1.487(2) 1.865(3) 0.076(3) Uani 1 1 d . . .  
C19 C 0.3721(7) 1.616(2) 1.869(2) 0.073(6) Uani 1 1 d . . .  
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C20 C 0.3639(8) 1.713(2) 1.776(2) 0.070(7) Uani 1 1 d . . .  
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H23B H 0.2315 1.3220 1.9574 0.118 Uiso 1 1 calc R . . .  
H23C H 0.2129 1.1921 1.8870 0.118 Uiso 1 1 calc R . . .  
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Se2 Se 0.44429(7) 1.2472(2) 1.21052(19) 0.0664(7) Uani 1 1 d . . .  
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C22 0.069(16) 0.11(2) 0.037(13) 0.041(14) -0.023(11) -0.027(16)  
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Sn1 0.0542(8) 0.0595(9) 0.0545(8) 0.0000(9) 0.0034(10) -0.0019(7)  
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are estimated using the full covariance matrix. The cell esds are taken  
into account individually in the estimation of esds in distances, angles  
and torsion angles; correlations between esds in cell parameters are only

used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.  
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C5 N1 1.35(2) . ?
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C6 N2 1.30(2) . ?
C6 C7 1.31(3) . ?
C6 Se2 1.94(2) . ?
C7 C8 1.39(3) . ?
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C19 C20 1.41(3) . ?
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C20 H20 0.9300 . ?  
C21 C22 1.28(3) . ?  
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C23 H23C 0.9600 . ?  
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C24 H24C 0.9600 . ?  
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Sn1 C12 H12B 109.5 . . ?  
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## Crystal Information File of [<sup>t</sup>Bu<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>4</sub>N)<sub>2</sub>] (5)

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'-x, y, -z'
'x+1/2, y+1/2, z'
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    F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc. and is
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N1 N 0.6795(7) 0.1900(15) 0.2318(11) 0.047(3) Uani 1 1 d . . .
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C2 C 0.7966(8) -0.024(2) 0.3876(13) 0.056(3) Uani 1 1 d . . .  
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H7B H 0.4419 0.2208 -0.3281 0.105 Uiso 1 1 calc R . . .  
H7C H 0.4976 0.0348 -0.3027 0.105 Uiso 1 1 calc R . . .  
C4 C 0.8315(10) 0.296(2) 0.4199(15) 0.073(5) Uani 1 1 d . . .  
H4 H 0.8726 0.3923 0.4689 0.088 Uiso 1 1 calc R . . .  
C3 C 0.8571(9) 0.118(3) 0.4579(14) 0.072(5) Uani 1 1 d . . .  
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C6 C 0.5649(7) 0.2139(17) -0.1341(11) 0.041(3) Uani 1 1 d . . .  
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C7 0.084(9) 0.092(11) 0.045(7) -0.012(7) 0.040(7) -0.029(8)  
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C3 0.046(6) 0.093(14) 0.066(8) -0.024(11) 0.016(6) -0.011(9)  
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Se1 Sn1 C6 C9 -101.1(8) 2_655 . . . . ?
Se1 Sn1 C6 C9 -6.8(9) . . . . ?
C6 Sn1 C6 C8 6.0(8) 2_655 . . . . ?
Se1 Sn1 C6 C8 139.4(9) 2_655 . . . . ?
Se1 Sn1 C6 C8 -126.3(9) . . . . ?
C6 Sn1 C6 C7 -115.9(9) 2_655 . . . . ?
Se1 Sn1 C6 C7 17.5(9) 2_655 . . . . ?
Se1 Sn1 C6 C7 111.8(9) . . . . ?
C1 N1 C5 C4 -1(2) . . . . ?
C3 C4 C5 N1 1(2) . . . . ?

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## Crystal Information File of [Me<sub>2</sub>Sn(Se-C<sub>5</sub>H<sub>3</sub>(Me-3)N)Cl] (7)

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'C8 H12 Cl N Se Sn'
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'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'H' 'H' 0.0000 0.0000
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'N' 'N' 0.0061 0.0033
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Cl' 'Cl' 0.1484 0.1585
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
'Se' 'Se' -0.0929 2.2259
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'Sn' 'Sn' -0.6537 1.4246
'International Tables Vol C Tables 4.2.6.8 and 6.1.1.4'
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\_symmetry\_Int\_Tables\_number 14

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'x, y, z'  
'-x, y+1/2, -z+1/2'  
'-x, -y, -z'  
'x, -y-1/2, z-1/2'

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\_cell\_length\_b 9.916(3)  
\_cell\_length\_c 13.060(3)  
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\_cell\_angle\_beta 112.521(19)  
\_cell\_angle\_gamma 90.00  
\_cell\_volume 1213.6(5)  
\_cell\_formula\_units\_Z 4  
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\_cell\_measurement\_reflns\_used 25  
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\_exptl\_absorpt\_process\_details '(North, Phillips & Mathews, 1968)'  
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_diffrn_reflns_av_sigmaI/netI   0.1806
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_diffrn_reflns_limit_l_min      -17
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Refinement of F^2^ against ALL reflections. The weighted R-factor wR and
goodness of fit S are based on F^2^, conventional R-factors R are based
on F, with F set to zero for negative F^2^. The threshold expression of
F^2^ > 2sigma(F^2^) is used only for calculating R-factors(gt) etc. and is
not relevant to the choice of reflections for refinement. R-factors based
on F^2^ are statistically about twice as large as those based on F, and R-
factors based on ALL data will be even larger.
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_atom_sites_solution_secondary   difmap
_atom_sites_solution_hydrogens  geom
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_refine_ls_extinction_method    none
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_refine_ls_number_reflns         2797
_refine_ls_number_parameters     112
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Se1 Se 0.40581(11) 0.14351(12) 1.03607(9) 0.0639(4) Uani 1 1 d . . .
Cl1 Cl 0.1570(3) -0.0925(3) 0.8331(3) 0.1002(12) Uani 1 1 d . . .
N1 N 0.2475(9) 0.3680(9) 0.9567(7) 0.058(2) Uani 1 1 d . . .
C1 C 0.3696(11) 0.3327(11) 1.0409(9) 0.055(3) Uani 1 1 d . . .
C7 C -0.0208(10) 0.1644(11) 0.9051(8) 0.076(4) Uani 1 1 d . . .
H7A H -0.0833 0.2316 0.8584 0.114 Uiso 1 1 calc R . .
H7B H 0.0047 0.1888 0.9814 0.114 Uiso 1 1 calc R . .
H7C H -0.0686 0.0787 0.8914 0.114 Uiso 1 1 calc R . .
C5 C 0.2082(14) 0.4967(14) 0.9504(11) 0.085(4) Uani 1 1 d . . .
H5 H 0.1228 0.5218 0.8941 0.101 Uiso 1 1 calc R . .
C3 C 0.4137(14) 0.5557(13) 1.1090(10) 0.078(4) Uani 1 1 d . . .
H3 H 0.4681 0.6199 1.1594 0.094 Uiso 1 1 calc R . .
C2 C 0.4584(12) 0.4229(11) 1.1202(10) 0.061(3) Uani 1 1 d . . .
C4 C 0.2866(16) 0.5946(13) 1.0223(13) 0.091(4) Uani 1 1 d . . .
H4 H 0.2566 0.6840 1.0140 0.109 Uiso 1 1 calc R . .
C8 C 0.1828(10) 0.2155(13) 0.7207(8) 0.084(4) Uani 1 1 d . . .
H8A H 0.1686 0.1399 0.6717 0.126 Uiso 1 1 calc R . .
H8B H 0.2758 0.2531 0.7369 0.126 Uiso 1 1 calc R . .
H8C H 0.1115 0.2827 0.6860 0.126 Uiso 1 1 calc R . .
C6 C 0.5886(11) 0.3765(11) 1.2128(8) 0.075(4) Uani 1 1 d . . .
H6A H 0.6548 0.4500 1.2383 0.113 Uiso 1 1 calc R . .
H6B H 0.6318 0.3044 1.1878 0.113 Uiso 1 1 calc R . .
H6C H 0.5635 0.3451 1.2725 0.113 Uiso 1 1 calc R . .

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Se1 0.0587(7) 0.0508(7) 0.0681(8) -0.0012(7) 0.0086(6) 0.0088(7)
Cl1 0.097(3) 0.071(2) 0.116(3) -0.025(2) 0.022(2) -0.005(2)
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N1 0.068(6) 0.045(6) 0.059(6) 0.004(5) 0.024(5) 0.014(5)  
C1 0.062(7) 0.054(8) 0.056(7) 0.010(6) 0.032(6) 0.008(6)  
C7 0.056(7) 0.112(10) 0.067(7) 0.000(8) 0.032(6) -0.011(7)  
C5 0.114(12) 0.072(10) 0.079(10) 0.009(9) 0.050(9) 0.023(9)  
C3 0.102(10) 0.062(10) 0.077(9) -0.006(8) 0.042(8) 0.000(8)  
C2 0.069(8) 0.037(7) 0.080(8) -0.016(7) 0.031(7) -0.007(6)  
C4 0.121(13) 0.039(8) 0.121(12) -0.001(9) 0.056(10) 0.018(8)  
C8 0.073(8) 0.132(12) 0.058(7) -0.004(8) 0.037(6) -0.004(8)  
C6 0.069(8) 0.078(9) 0.067(8) -0.006(7) 0.013(6) -0.019(7)

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All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.  
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Sn1 C7 2.120(9) . ?  
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Sn1 Cl1 2.457(3) . ?  
Sn1 Se1 2.5667(14) . ?  
Se1 Cl1 1.917(11) . ?  
N1 C5 1.330(13) . ?  
N1 C1 1.350(12) . ?  
C1 C2 1.404(13) . ?  
C7 H7A 0.9600 . ?  
C7 H7B 0.9600 . ?  
C7 H7C 0.9600 . ?  
C5 C4 1.374(15) . ?  
C5 H5 0.9300 . ?  
C3 C2 1.382(14) . ?  
C3 C4 1.405(15) . ?  
C3 H3 0.9300 . ?  
C2 C6 1.482(13) . ?  
C4 H4 0.9300 . ?  
C8 H8A 0.9600 . ?  
C8 H8B 0.9600 . ?  
C8 H8C 0.9600 . ?  
C6 H6A 0.9600 . ?  
C6 H6B 0.9600 . ?  
C6 H6C 0.9600 . ?  
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C8 Sn1 N1 91.7(4) . . ?  
C7 Sn1 N1 90.9(4) . . ?  
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C7 Sn1 Cl1 97.7(3) . . ?  
N1 Sn1 Cl1 160.7(2) . . ?  
C8 Sn1 Se1 113.8(3) . . ?  
C7 Sn1 Se1 117.0(3) . . ?  
N1 Sn1 Se1 65.6(2) . . ?  
Cl1 Sn1 Se1 95.12(9) . . ?  
C1 Se1 Sn1 82.4(3) . . ?  
C5 N1 C1 117.3(11) . . ?  
C5 N1 Sn1 141.4(9) . . ?  
C1 N1 Sn1 101.1(7) . . ?  
N1 C1 C2 124.4(10) . . ?  
N1 C1 Se1 110.8(8) . . ?  
C2 C1 Se1 124.8(8) . . ?  
Sn1 C7 H7A 109.5 . . ?  
Sn1 C7 H7B 109.5 . . ?  
H7A C7 H7B 109.5 . . ?  
Sn1 C7 H7C 109.5 . . ?  
H7A C7 H7C 109.5 . . ?  
H7B C7 H7C 109.5 . . ?  
N1 C5 C4 123.7(12) . . ?  
N1 C5 H5 118.2 . . ?  
C4 C5 H5 118.2 . . ?  
C2 C3 C4 120.6(12) . . ?  
C2 C3 H3 119.7 . . ?  
C4 C3 H3 119.7 . . ?  
C3 C2 C1 115.9(11) . . ?  
C3 C2 C6 122.4(11) . . ?  
C1 C2 C6 121.6(10) . . ?  
C5 C4 C3 118.0(12) . . ?  
C5 C4 H4 121.0 . . ?  
C3 C4 H4 121.0 . . ?  
Sn1 C8 H8A 109.5 . . ?  
Sn1 C8 H8B 109.5 . . ?  
H8A C8 H8B 109.5 . . ?  
Sn1 C8 H8C 109.5 . . ?  
H8A C8 H8C 109.5 . . ?  
H8B C8 H8C 109.5 . . ?  
C2 C6 H6A 109.5 . . ?  
C2 C6 H6B 109.5 . . ?  
H6A C6 H6B 109.5 . . ?  
C2 C6 H6C 109.5 . . ?  
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H6B C6 H6C 109.5 . . ?  
  
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C7 Sn1 Se1 C1 -76.5(4) . . . . ?  
N1 Sn1 Se1 C1 1.3(3) . . . . ?  
C11 Sn1 Se1 C1 -177.8(3) . . . . ?  
C8 Sn1 N1 C5 67.1(12) . . . . ?  
C7 Sn1 N1 C5 -57.9(12) . . . . ?  
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Se1 Sn1 N1 C5 -177.4(12) . . . . ?  
C8 Sn1 N1 C1 -117.3(6) . . . . ?  
C7 Sn1 N1 C1 117.6(6) . . . . ?  
C11 Sn1 N1 C1 0.8(10) . . . . ?  
Se1 Sn1 N1 C1 -1.9(5) . . . . ?  
C5 N1 C1 C2 -1.4(15) . . . . ?  
Sn1 N1 C1 C2 -178.2(8) . . . . ?  
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Sn1 N1 C1 Se1 2.5(6) . . . . ?  
Sn1 Se1 C1 N1 -2.3(6) . . . . ?  
Sn1 Se1 C1 C2 178.4(9) . . . . ?  
C1 N1 C5 C4 1.9(17) . . . . ?  
Sn1 N1 C5 C4 177.0(9) . . . . ?  
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C4 C3 C2 C6 -177.5(10) . . . . ?  
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Se1 C1 C2 C3 179.8(8) . . . . ?  
N1 C1 C2 C6 177.9(9) . . . . ?  
Se1 C1 C2 C6 -2.9(14) . . . . ?  
N1 C5 C4 C3 -1.7(19) . . . . ?  
C2 C3 C4 C5 0.8(18) . . . . ?  
  
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\_diffrn\_reflns\_theta\_full 27.54  
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