

Electronic Supporting Information

Novel tetracarboxylatoplatinum(IV) complexes as carboplatin prodrugs

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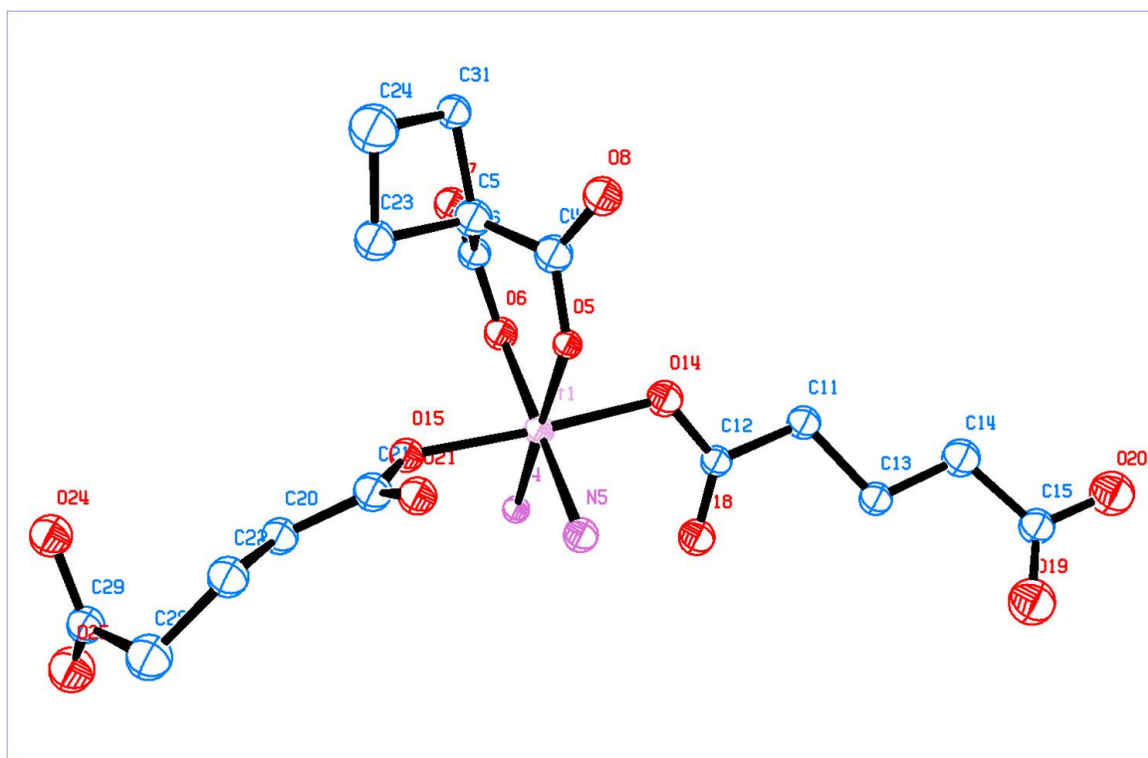


Fig. S1. ORTEP diagram of **4** displaying thermal ellipsoids at 55% probability level.

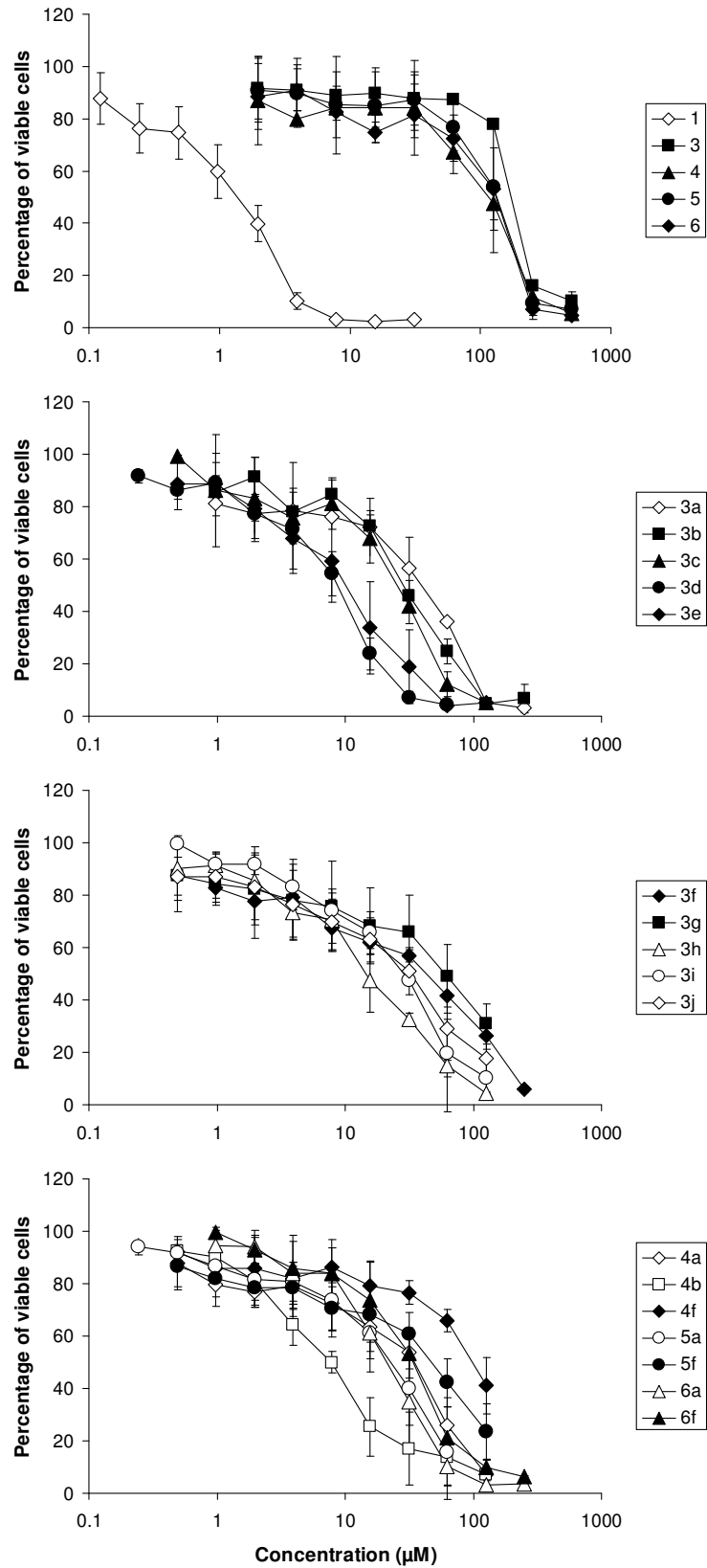


Fig. S2. Concentration–effect curves (means $\pm$ standard deviations) of investigated compounds in CH1 cells (MTT assay, exposure time 96 h).

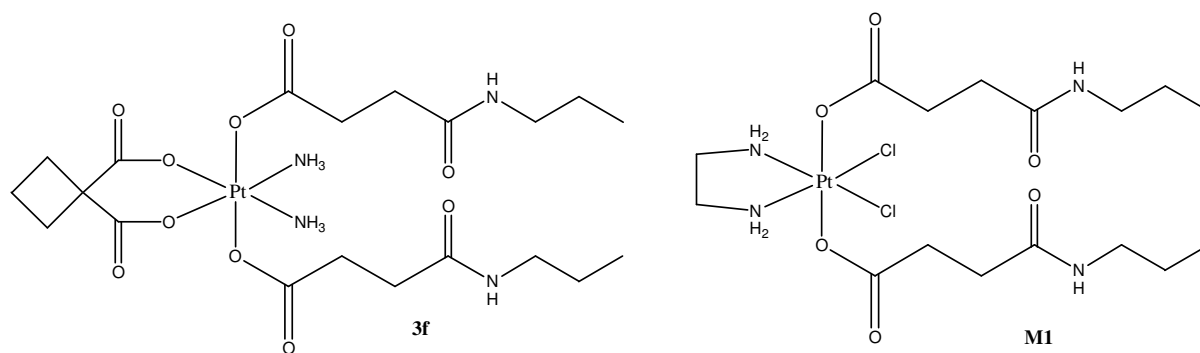


Fig. S3. Chemical structures of complexes **3f** and **M1**

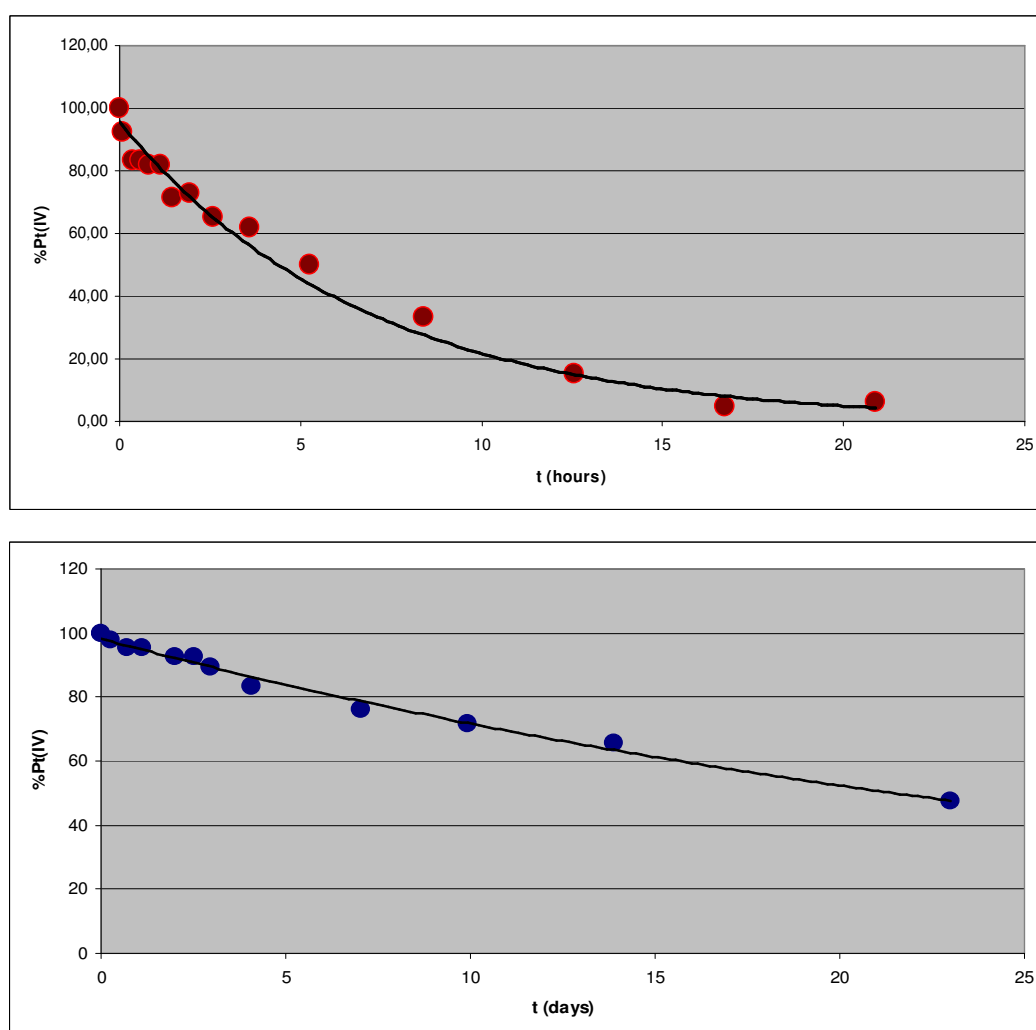


Fig. S4. Time dependent reduction of **M1** (top) and **3f** (bottom) in the presence of ascorbic acid; ambient temperature, pD =7.4, 1 mM complex, 50 mM phosphate buffer, 25 mM ascorbic acid.

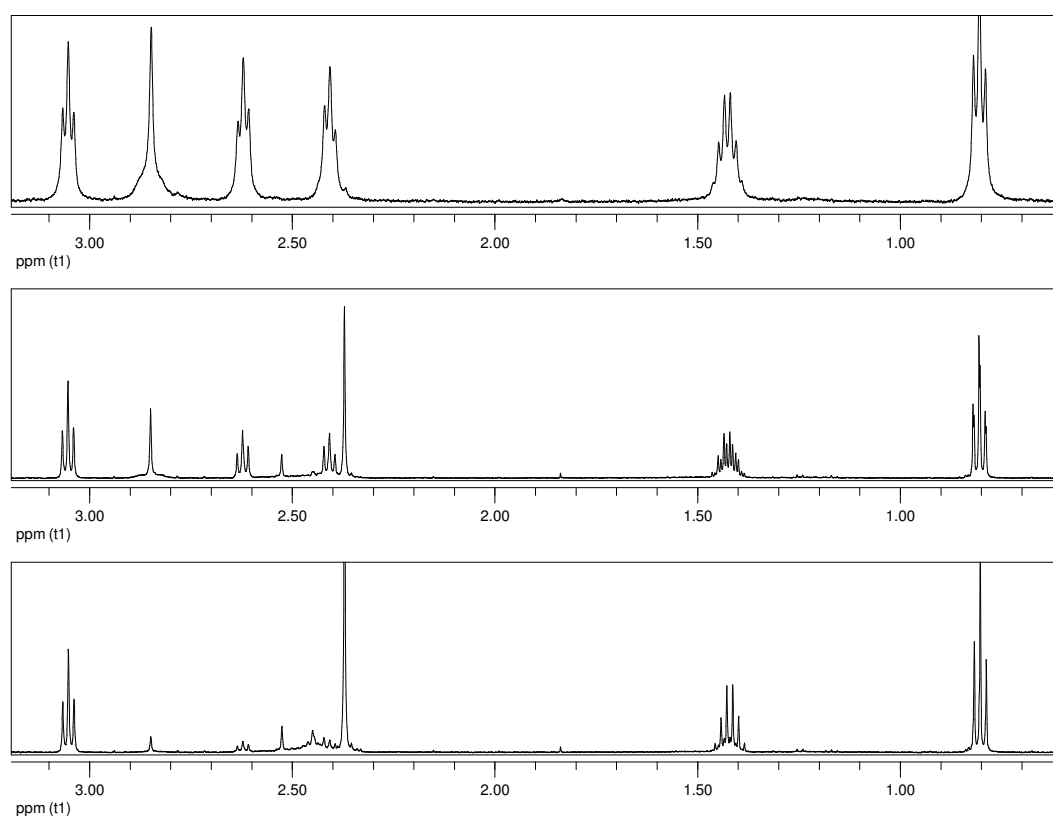


Fig. S5. ¹H NMR spectra of complex **M1** after addition of ascorbic acid (top – immediately, middle – after 5 hours, bottom – after 17 h); ambient temperature, pD = 7.4, 1 mM complex, 50 mM phosphate buffer, 25 mM ascorbic acid.

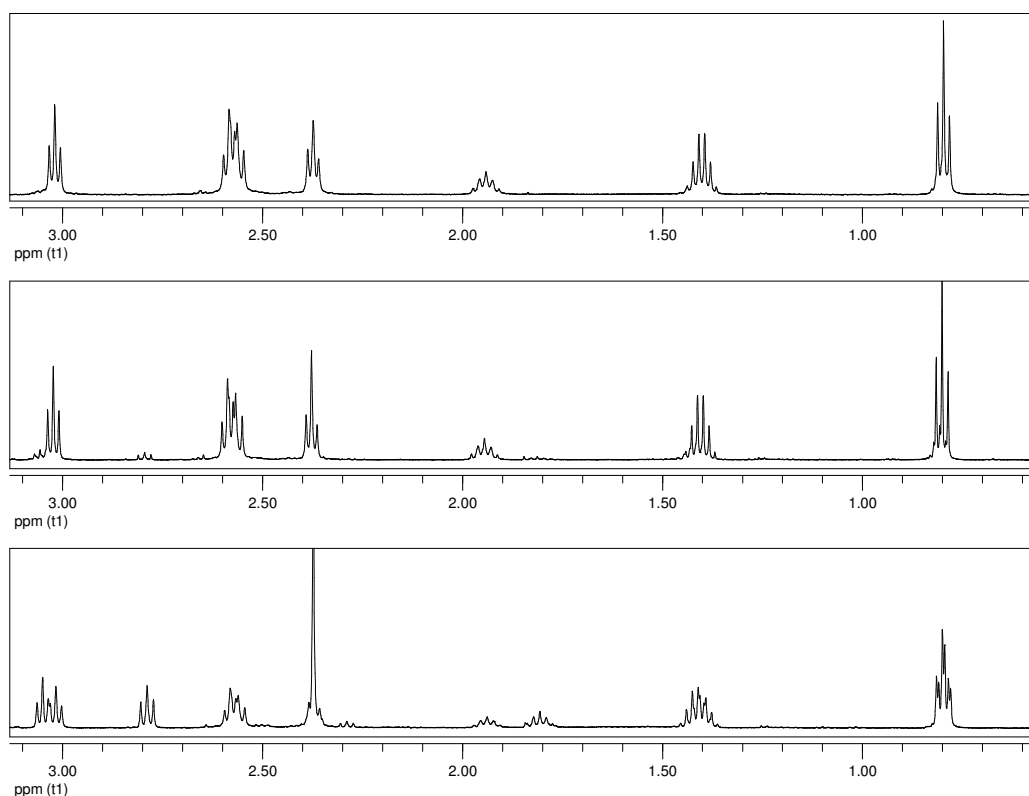


Fig. S6. ^1H NMR spectra of complex **3f** after addition of ascorbic acid (top – immediately, middle – after 3 days, bottom – after 23 days); ambient temperature, pD = 7.4, 1 mM complex, 50 mM phosphate buffer, 25 mM ascorbic acid

Table S1. Comparison of redox potentials, halve life times of reduction by ascorbic acid and cytotoxicity for complexes **M1** and **3f**.

compound	E_p (V)	$t_{1/2}$	IC_{50} (CH1), μM	IC_{50} (SW480), μM
M1	-0.60	5 h	2.3 ± 1.1^a	31 ± 15^a
3f	-0.68	21 d	44 ± 8	>500

^a data, taken from ref. ¹

¹ M.R. Reithofer, S.M. Valiahdi, M.A. Jakupc, V.B. Arion, A. Egger, M. Galanski, B.K. Keppler, *J. Med. Chem.*, 2007, **50**, 6692–6699.