Thomas K. Nielsen Supporting Information

Methods to Stabilize and Destabilize Ammonium Borohydride

Thomas K. Nielsen,¹ Abhi Karkamkar,² Mark Bowden,² Flemming Besenbacher,³ Torben R. Jensen,¹ Tom Autrey^{2*}

¹Center for Energy Materials, Interdisciplinary Nanoscience Center (iNano) and Department of Chemistry, Aarhus University, DK-8000 Aarhus, Denmark.

²Pacific Northwest National Laboratory, Richland, Washington 99354.

³ Interdisciplinary Nanoscience Center (iNANO) and Department of Physics and Astronomy, Aarhus University, DK-8000 Aarhus C, Denmark.

* Corresponding author: tom.autrey@pnnl.gov (Autrey, S. Thomas)



Figure S1. Differential scanning calorimetry (DSC) of sample DADB-HT and DADB-LT. Samples were heated from -30 °C to 190 °C (heating rate 1 °C/min).

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Details of crystal structure refinement

Two identical and symmetric NH₃BH₂NH₃ fragments were refined in the unit cell. The N-H and B-H distances were fixed at 0.85 and 1.01 Å respectively from the corresponding values in the high temperature structure (refined from neutron data). The N-B-N angle refined to $107.6(9)^\circ$; all other angles were set to the tetrahedral angle of 109.47° where possible. The positions of these fragments were defined by the boron atoms and refined to x = 0.1389(6), y = 0.4310(18), z = 0.7683(14) and x = 0.3995(5), y = 0.4888(12), z = 0.784(2) respectively. The rotational orientation of each fragment was refined independently.

The BH₄ fragments were treated as regular tetrahedra with B-H distances of 1.16 Å, again from the high temperature structure. The positions and orientations of both were also refined independently, leading to boron positions of x = 0.0105(10), y = 0.7694(18), z = 0.508(2) and x = 0.7750(9), y = 0.2627(17), z = 0.0050(18). The orientations and therefore H positions should be regarded as approximate at best given the weak scattering power of hydrogen.

The resulting coordinates of all atoms are given in the following table. All atoms belong to Wyckoff site 8c and have occupancy of 1.

| Atom | x/a | y/b | z/c |
|------|--------|--------|--------|
| B1 | 0.1389 | 0.4310 | 0.7683 |
| HB11 | 0.123 | 0.365 | 0.696 |
| HB12 | 0.112 | 0.512 | 0.774 |
| B2 | 0.3995 | 0.4888 | 0.784 |
| HB21 | 0.379 | 0.538 | 0.704 |
| HB22 | 0.384 | 0.524 | 0.876 |
| B3 | 0.0105 | 0.7694 | 0.508 |
| HB31 | 0.039 | 0.860 | 0.471 |
| HB32 | 0.039 | 0.692 | 0.568 |
| HB33 | -0.010 | 0.717 | 0.411 |
| HB34 | -0.026 | 0.809 | 0.581 |
| B4 | 0.7750 | 0.2627 | 0.0050 |
| HB41 | 0.781 | 0.163 | 0.068 |
| HB42 | 0.816 | 0.330 | 0.019 |
| HB43 | 0.734 | 0.321 | 0.046 |
| HB44 | 0.769 | 0.238 | -0.113 |
| N1 | 0.1417 | 0.3593 | 0.9150 |
| HN11 | 0.165 | 0.290 | 0.910 |
| HN12 | 0.155 | 0.416 | 0.977 |
| HN13 | 0.107 | 0.332 | 0.939 |
| N2 | 0.2020 | 0.4785 | 0.7266 |
| HN21 | 0.201 | 0.518 | 0.645 |
| HN22 | 0.215 | 0.535 | 0.789 |
| HN23 | 0.225 | 0.408 | 0.723 |
| N3 | 0.3867 | 0.3325 | 0.7721 |
| HN31 | 0.400 | 0.302 | 0.693 |
| HN32 | 0.404 | 0.290 | 0.841 |
| HN33 | 0.349 | 0.319 | 0.776 |
| N4 | 0.4676 | 0.5114 | 0.7759 |
| HN41 | 0.475 | 0.598 | 0.782 |
| HN42 | 0.484 | 0.469 | 0.845 |
| HN43 | 0.481 | 0.480 | 0.697 |

Table S1 Crystallographic coordinates from refinement against synchrotron data.



Figure S2. Observed and calculated synchrotron diffraction patterns for low temperature DADB ($\lambda = 0.75338$ Å). The calculated pattern from the high temperature phase is shown in blue, corresponding to 6.4 wt% of the sample; all other peaks belong to the low temperature phase. The broad background arises from the glass capillary used to contain the sample.



Figure S3. (001) projections of low temperature (A) and high temperature (B) DADB structures.