Supporting Information

Zn(II), Cd(II) and Cu(II) complexes of 2,5–bis{N–(2,6–diisopropyl phenyl)iminomethyl}pyrrole: synthesis, structures and their high catalytic activity for efficient cyclic carbonate synthesis from epoxides and CO₂ at atmospheric pressure

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Parameters	3a	5a
Empirical formula	C ₆₀ H ₇₆ N ₆ Zn	$C_{61}H_{80}CdN_6O$
Formula weight	946.64	1025.71
Crystal System	Monoclinic	Orthorhombic
Space group	P2(1)/n	Pnma
a (Å)	10.559	19.145
b (Å)	21.306	21.378
c (Å)	24.365	14.170
α (deg)	90	90
β (deg)	97.43	90
γ (deg)	90	90
Volume $(\text{\AA})^3$	5435.2	5799.7
Temperature (K)	100(2)	100(2)
Ζ	4	4
$\mu \text{ mm}^{-1}$	0.494	0.419
D (calcd.) (g/cm^3)	1.157	1.175
F(000)	2032	2176
Crystal size (mm)	$0.16 \times 0.12 \times 0.10$	$0.22 \times 0.18 \times 0.14$
θ range (deg)	1.27 to 25.05	1.72 to 25.07
Index ranges	-12<=h<=12	-22<=h<=22
	-21<=k<=25	-25<=k<=25
	-29<=l<=26	-16<=l<=16
Reflections collected	24621 / 9522	54127 / 5296
/ unique	$R_{int} = 0.0807$	$R_{int} = 0.0534$
Data completeness	98.9 %	99.9 %
Transmission	0.9522 / 0.9251	0.9436/0.9134
(max/min)		
Data / restraints /	9522 / 0 / 652	5296 / 0 / 366
parameters		
Goodness-of-fit on F ²	1.075	1.161
Final R indices	$R_1 = 0.0720$	$R_1 = 0.0440$
[I>2sigma(I)]	$wR_2 = 0.1685$	$wR_2 = 0.0984$
R indices (all data)	$R_1 = 0.0946$	$R_1 = 0.0472$
	$wR_2 = 0.1825$	$wR_2 = 0.1003$
Largest diff. peak /	0.749/-0.399	0.583 / -0.459
hole, e Å ⁻³		

Table S1 Crystallographic data for complexes 3a, 3b, 5a and 5b

Crystal Structure of 3a



Fig. S1 ORTEP of molecular structure of **3a**. Thermal ellipsoids are shown at 30% probability levels. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Zn1-N2 = 1.971(3), Zn1-N1 = 2.097(3), Zn1-N5 = 1.996(3), Zn1-N4 = 2.054(3), N2-Zn1-N1 = 82.55(12), N4-Zn1-N1 = 116.93(12), N5-Zn1-N4 = 84.00(11), N2-Zn1-N5 = 111.04(12), N5-Zn1-N1 = 117.46(12), N2-Zn1-N4 = 147.48(12), C9-C8-C7 = 133.2(4), C10-C11-C12 = 128.2(4), C7-N1-C6 = 117.7(3), C12-N3-C13 = 119.5(3), C42-N6-C43 = 118.1(3), C40-C41-C42 = 128.6(3), C39-C38-C37 = 132.5(3), C37-N4-C36 = 116.7(3).



Fig. S2 ORTEP of molecular structure of **3b**. Thermal ellipsoids are shown at 30% probability levels. Hydrogen atoms are omitted for clarity.

Crystal Structure of 5a



Fig. S3 ORTEP of molecular structure of **5a**. Thermal ellipsoids are shown at 30% probability levels. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cd1–N2 = 2.271(3), N5–Cd1 = 2.716(3), Cd1–N4 = 2.149(3), N6–Cd1 = 2.716(3), Cd1–N1 = 2.275(3), Cd1–O1 = 2.336(3), N2–Cd1–N1 = 76.37(10), N1–Cd1–O1 = 85.08(11), N4–Cd1–O1 = 94.68(11), N4–Cd1–N2 = 103.87(10), N4–Cd1–N5 = 67.62, N4–Cd1–N6 = 67.62, N5–Cd1–N6 = 135.09, N6–Cd1–O1 = 90.12, N6–Cd1–N1 = 112.38, N6–Cd1–N2 = 96.86, C7–C6–C5 = 128.97, C8–C9–C10 = 127.1(3), C5–N1–C4 = 116.4(3), C10–N3–C11 = 128.5(3), C29–C28–C27 = 132.4(2), C27–N5–C26 = 118.0(2), N4–Cd1–N1 = 179.76(10), N2–Cd1–O1 = 161.45(11).

Crystal Structure of 5b



Fig. S4 ORTEP of molecular structure of **5b**. Thermal ellipsoids are shown at 30% probability levels. Hydrogen atoms are omitted for clarity.





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Element Name	Element %	Ret. Time
Nitrogen	8. 75	0. 78
Carbon	76. 32	1, 16
Hydrogen	7.96	3. 74

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Element Name	Element %	Ret. Time
Nitrogen	8.36	0. 74
arbon	72. 45	1. 13
Hydrogen	7. 78	3.83

S12

(By



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HRMS — 8

LCMS-2010A DATA REPORT SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD

LCMS-2010A DATA REPORT SCHOOL OF CHEMISTRY UNIVERSITY OF HYDERABAD

190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 ppm

-155.100

-19.374

S31

HRMS — 13c

S43

HRMS — 13f

Display Report						
Analysis Info Analysis Name Method Gample Name Comment	D:\Data\2012\Dr.K.I tune_low_Pos.m KPK-V1-DCM-ACN	Muralidharan\FEB\KPK-V1	.d	Acquisition Date Operator Instrument	2/7/2012 3:38 UOH maXis	57 PM 10138
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1.25-						
1.00-						
0.75-						
0.50-						
0.25						
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0.00	80	100	120	140	160	<u> </u>