

Supporting Information

A Comparison of Sensitised Ln(III) Emission with Pyridine- and Pyrazine-2,6-Dicarboxylates – Part II

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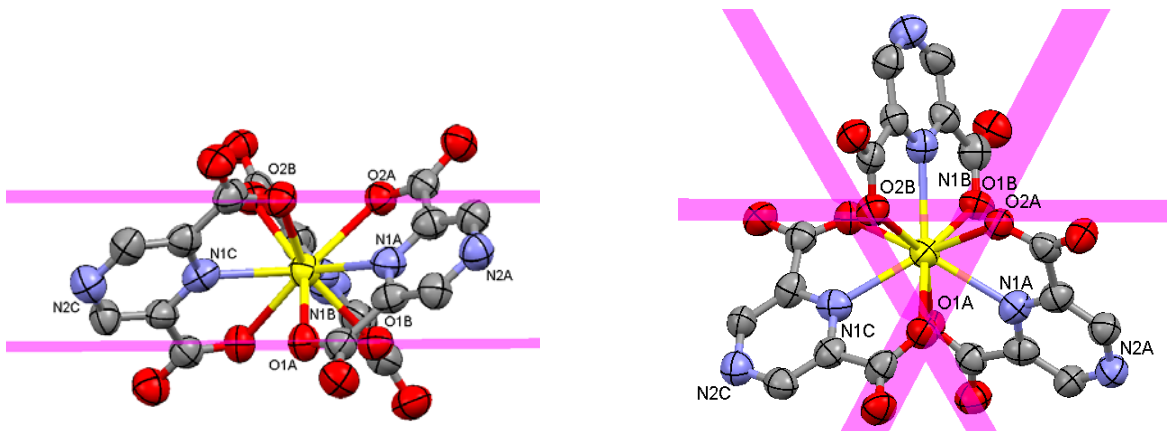


Figure S1. Additional view of X-ray crystal structure for $(\text{Me}_4\text{N})_3[\text{Eu}(\text{PYZ})_3]\cdot 4\text{H}_2\text{O}$ showing (a) upper and lower basal planes defined by the $\{\text{O1A}, \text{O1B}, \text{O1C}\}$ and $\{\text{O2A}, \text{O2B}, \text{O2C}\}$ atoms (left) and three lateral planes defined by $\{\text{O2A}, \text{O1B}, \text{O2B}, \text{O1C}\}$, $\{\text{O1A}, \text{O2A}, \text{O1B}, \text{O2C}\}$ and $\{\text{O1A}, \text{O2B}, \text{O1C}, \text{O2C}\}$ (right) (C, grey; O, red; N, blue; Eu, yellow; 50% probability ellipsoids). Me_4N^+ cations, solvent water molecules, and H atoms omitted for clarity.

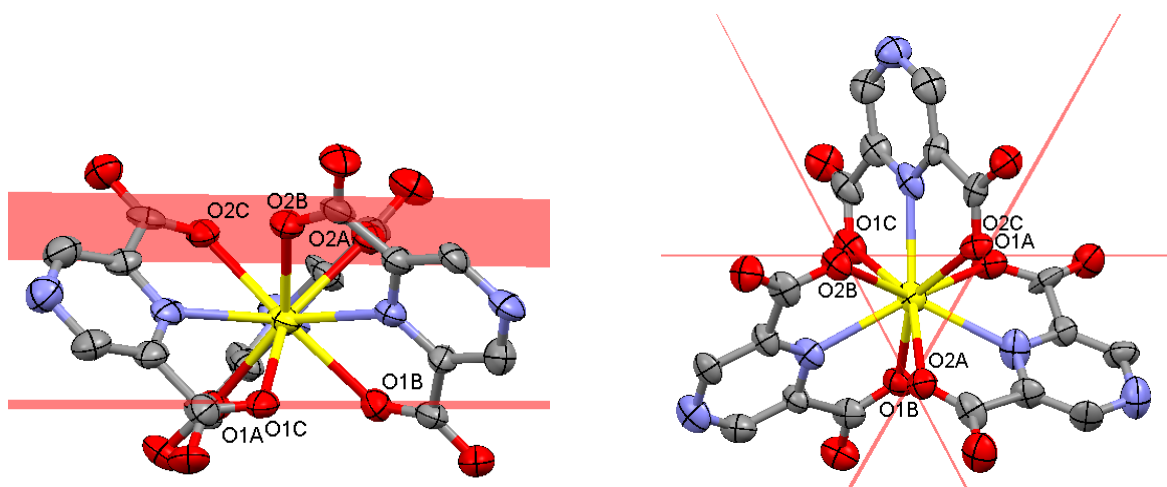


Figure S2. Additional view of X-ray crystal structure for $(\text{Me}_4\text{N})_3[\text{Eu}(\text{PYZ})_3]\cdot 5\text{MeOH}$ showing (a) upper and lower basal planes defined by the $\{\text{O1A}, \text{O1B}, \text{O1C}\}$ and $\{\text{O2A}, \text{O2B}, \text{O2C}\}$ atoms (left) and three lateral planes defined by $\{\text{O2A}, \text{O1B}, \text{O2B}, \text{O1C}\}$, $\{\text{O1A}, \text{O2A}, \text{O1B}, \text{O2C}\}$ and $\{\text{O1A}, \text{O2B}, \text{O1C}, \text{O2C}\}$ (right) (C, grey; O, red; N, blue; Eu, yellow; 50% probability ellipsoids). Me_4N^+ cations, solvent water molecules, and H atoms omitted for clarity.

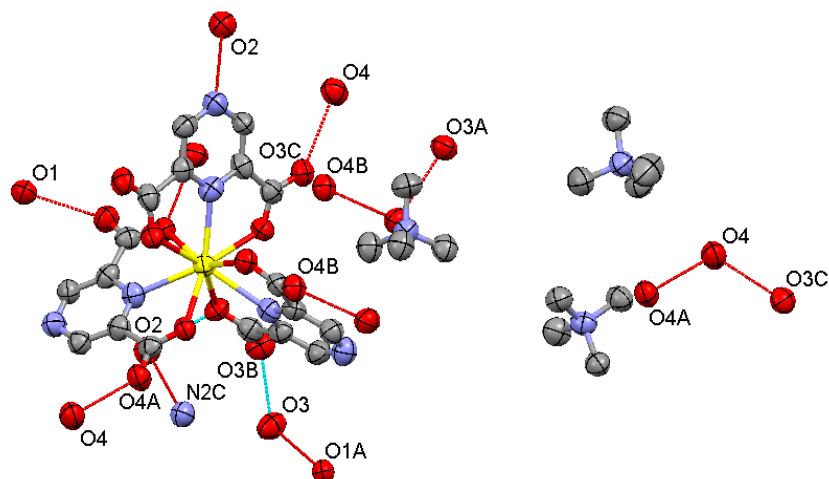


Figure S3. Additional view of X-ray crystal structure for $(\text{Me}_4\text{N})_3[\text{Eu}(\text{PYZ})_3]\cdot 4\text{H}_2\text{O}$ showing intermolecular H-bonding interactions (C, grey; O, red; N, blue; Eu, yellow; 50% probability ellipsoids). H atoms omitted for clarity.

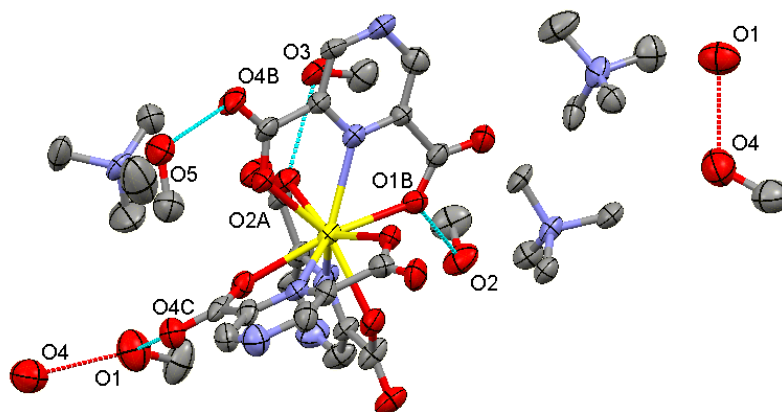


Figure S4. Additional view of X-ray crystal structure for $(\text{Me}_4\text{N})_3[\text{Eu}(\text{PYZ})_3]\cdot 5\text{MeOH}$ showing intermolecular H-bonding interactions (C, grey; O, red; N, blue; Eu, yellow; 50% probability ellipsoids). H atoms omitted for clarity.

Table S1. List of hydrogen bonding interaction distances and angles for the (Me₄N)₃[Eu(PYZ)₃].4H₂O and (Me₄N)₃[Eu(PYZ)₃].5MeOH complexes.

(Me₄N)₃[Eu(PYZ)₃].4H₂O						
Donor (D)	H Atom (H)	Acceptor (A)	D-H (Å)	H---A (Å)	D---A (Å)	D-H---A (°)
O1	H1	O4B	0.83(9)	1.98(10)	2.805(12)	173(12)
O1	H2	O3A	0.83 (9)	2.03 (9)	2.818 (12)	158 (10)
O2	H3	O1B	0.85 (4)	2.05 (7)	2.855 (12)	158 (13)
O2	H4	N2C	0.84 (11)	2.23 (11)	3.010 (15)	154 (11)
O3	H5	O1A	0.85 (11)	2.11 (12)	2.860 (12)	147 (10)
O3	H6	O3B	0.84 (3)	2.05 (5)	2.820 (13)	152 (11)
O4	H7	O4A	0.85 (10)	1.91 (10)	2.743 (12)	167 (6)
O4	H8	O3C	0.86 (13)	1.95 (13)	2.800 (12)	174 (-1)
(Me₄N)₃[Eu(PYZ)₃].5MeOH						
Donor (D)	H Atom (H)	Acceptor (A)	D-H (Å)	H---A (Å)	D---A (Å)	D-H---A (°)
O1	H1	O4C	0.82	1.85	2.673 (18)	179
O2	H2	O1B	0.82	1.94	2.718 (16)	157
O5	H5	O4B	0.82	1.95	2.759 (18)	167

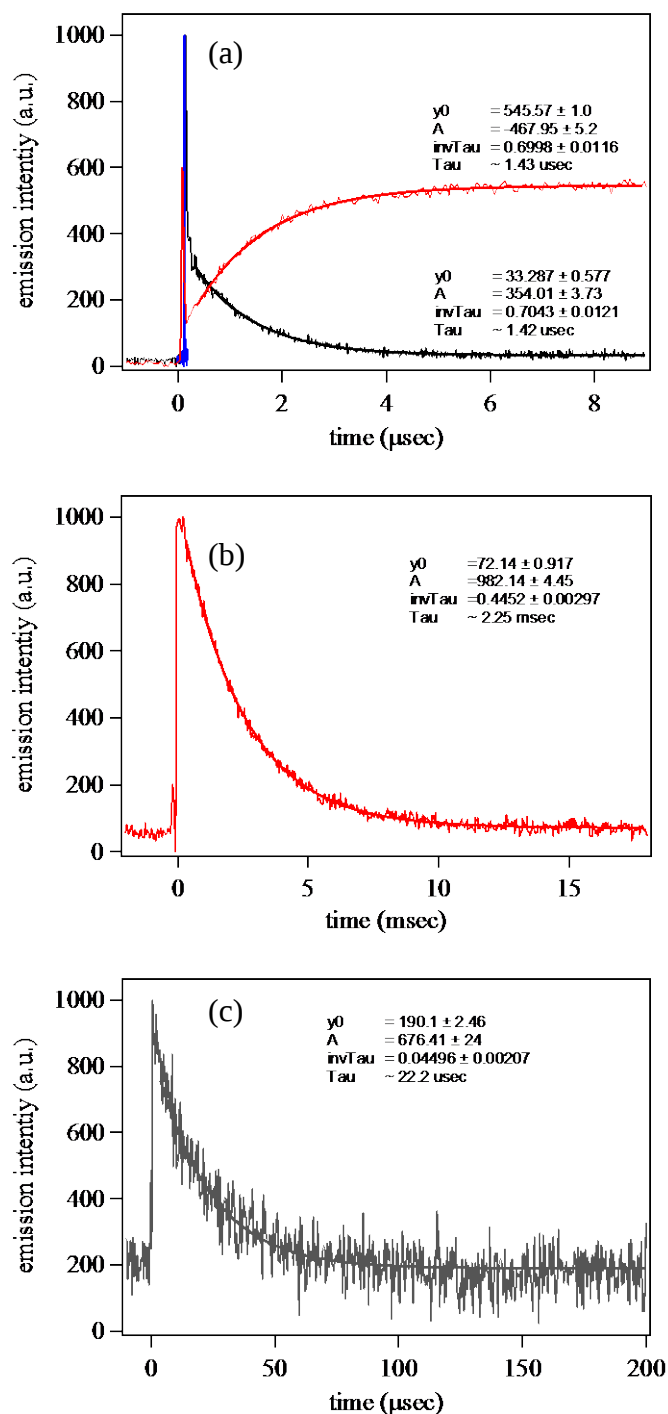
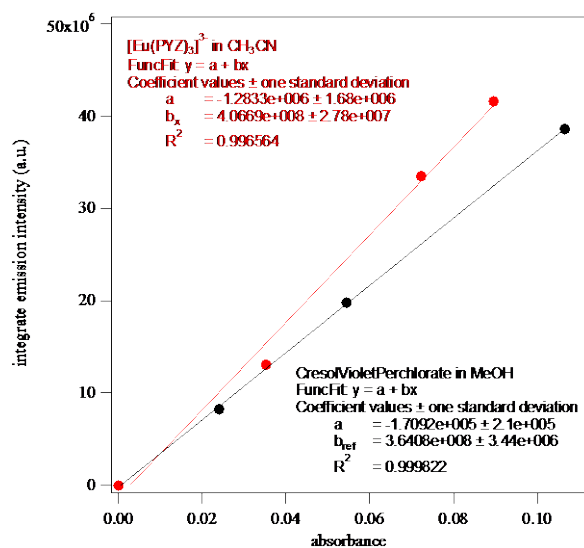


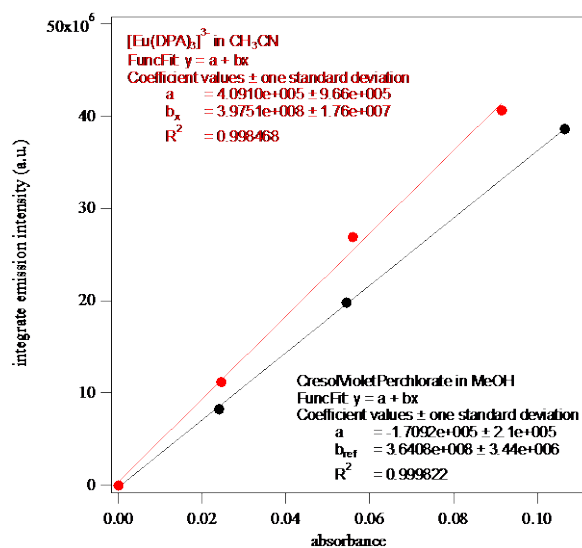
Figure S5. (a) Nanosecond laser setup instrument response function (IRF, blue, FWHM ~ 8 ns), observed decay of 5D_1 excited state (black, $\lambda_{\text{emis}} = 538$ nm, $\lambda_{\text{ex}} = 280$ nm) and corresponding rise (red, $\lambda_{\text{emis}} = 615$ nm, $\lambda_{\text{ex}} = 280$ nm) of the 5D_0 excited state for the $[\text{Eu}(\text{DPA})_3]^{3-}$ complex. (b) Long lived decay of the 5D_0 excited state for the $[\text{Eu}(\text{DPA})_3]^{3-}$ complex. (c) Decay of the $^2F_{5/2}$ excited state for the $[\text{Yb}(\text{DPA})_3]^{3-}$ complex in the NIR region (grey, $\lambda_{\text{emis}} = 980$ nm, $\lambda_{\text{ex}} = 266$ nm).



$$\Phi_x = \Phi_{\text{ref}} \times (b_x/b_{\text{ref}}) \times (\eta_x^2/\eta_{\text{ref}}^2)$$

$$= 0.54 \times (4.0669\text{e}8/3.6408\text{e}8) \times (1.3441^2/1.3280^2)$$

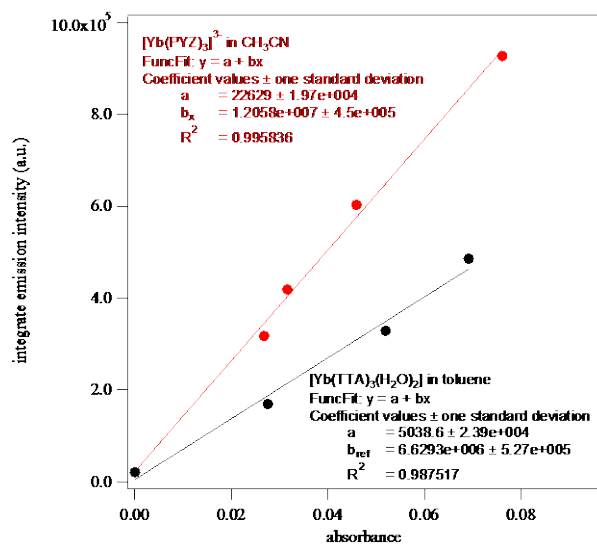
$$= 0.6179 (\sim 61.8\%)$$



$$\Phi_x = \Phi_{\text{ref}} \times (b_x/b_{\text{ref}}) \times (\eta_x^2/\eta_{\text{ref}}^2)$$

$$= 0.54 \times (3.9751\text{e}8/3.6408\text{e}8) \times (1.3441^2/1.3280^2)$$

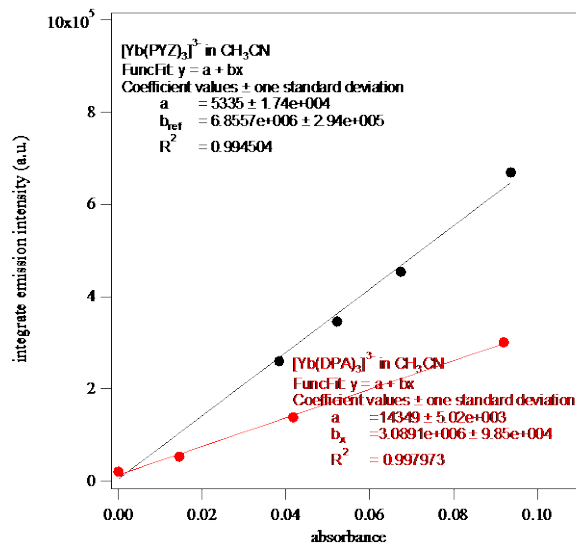
$$= 0.6039 (\sim 60.4\%)$$



$$\Phi_x = \Phi_{\text{ref}} \times (b_x/b_{\text{ref}}) \times (\eta_x^2/\eta_{\text{ref}}^2)$$

$$= 0.0035 \times (1.2058\text{e}7/6.6293\text{e}6) \times (1.3441^2/1.4969^2)$$

$$= 0.00513 (\sim 0.51\%)$$



$$\Phi_x = \Phi_{\text{ref}} \times (b_x/b_{\text{ref}}) \times (\eta_x^2/\eta_{\text{ref}}^2)$$

$$= 0.00513 \times (3.0891\text{e}6/6.8557\text{e}6) \times (1.3441^2/1.3441^2)$$

$$= 0.00231 (\sim 0.23\%)$$

Figure S6. Absorption versus integrated emission plots for; **(a)** Relative quantum yield determination of [Eu(PYZ)₃]³⁻ complex in CH₃CN ($\eta = 1.3441$) vs cresol violet perchlorate in MeOH ($\eta = 1.3280$) ($\lambda_{\text{ex}} = 280$ nm, $\Phi_{\text{ref}} = 54\%$). **(b)** Relative quantum yield determination of [Eu(DPA)₃]³⁻ complex in CH₃CN ($\eta = 1.3441$) vs cresol violet perchlorate in MeOH ($\eta = 1.3280$) ($\lambda_{\text{ex}} = 280$ nm, $\Phi_{\text{ref}} = 54\%$). **(c)** Relative quantum yield determination of [Yb(PYZ)₃]³⁻ complex in CH₃CN ($\eta = 1.3441$) vs [Yb(TTA)₃(H₂O)₂] in toluene ($\eta = 1.4969$) ($\lambda_{\text{ex}} = 330$ nm, $\Phi_{\text{ref}} = 0.35\%$). **(d)** Relative quantum yield determination of [Yb(DPA)₃]³⁻ complex in CH₃CN ($\eta = 1.3441$) vs [Yb(PYZ)₃]³⁻ complex in CH₃CN ($\eta = 1.3441$) ($\lambda_{\text{ex}} = 280$ nm, $\Phi_{\text{ref}} = 0.513\%$).

Table S2A. Final Coordinates for [Y(PYZ)₃]³⁻ from TD-DFT (B3LYP/LANL2DZ) calculations.

Atom	x	y	z
C,	-2.1823453111,	-1.7767712063,	1.8598399613
C,	2.4973817968,	1.777519627 3,	-1.4090623041
O,	3.4480786399,	2.5994457189,	-1.6320036277
O,	1.8210653884,	1.5880567431,	-0.3152366942
O,	-2.7190455969,	-2.5987257094,	2.6756011709
O,	-0.9571125619,	-2.5971973144,	-3.6949012092
O,	3.6763560309,	-2.6001755785,	1.0184252305
C,	-0.0279030132,	1.7763927589,	2.8682175624
C,	-2.4700760163,	1.7771265434,	-1.4581042052
C,	1.1772285517,	0.856310572,	3.0722957791
C,	2.0719979864,	0.8581227405,	-2.5555467421
O,	-0.6378474624,	1.5874749008,	1.7359738435
O,	-0.3091600054,	2.5983407221,	3.803329282
O,	-3.1394037028,	2.5981243927,	-2.1702418538
C,	1.9899410482,	0.851178475,	4.2243378228
O,	-1.1845033713,	1.588274276,	-1.4200636387
C,	2.6630244663,	0.8540514194,	-3.8354657147
C,	1.2374616613,	-0.8512691519,	-4.5029839087
C,	0.6336473458,	-0.8562188456,	-3.229024874
N,	1.0664319693,	0.0007308411,	-2.2802070984
H,	3.4692411719,	1.5442179236,	-4.0558724931
H,	0.8901085314,	-1.541161537,	-5.2633979659
C,	-3.2494738643,	0.8570497112,	-0.5164402127
C,	-4.6533944617,	0.8523213646,	-0.3878226292
C,	-4.5173850669,	-0.8529090095,	1.180482607
C,	-3.1122648463,	-0.8572264312,	1.0658913828
N,	-2.5076066306,	-0.0000016253,	0.2163843701
H,	-5.0017864203,	-1.5429426574,	1.8617267826
H,	-5.247726902,	1.5423511771,	-0.9755811122
C,	3.2796826242,	-0.8546573516,	3.3231221542
C,	2.478662491,	-0.858295984,	2.162911813
N,	1.4411485668,	-0.0005994767,	2.0633085436
H,	1.7777902114,	1.5410984526,	5.032951902
H,	4.1114141955,	-1.5451758197,	3.4023120706
Y,	0.0000641026,	0.0002223802,	-0.0003171163
N,	3.0433487305,	-0.002079891,	4.3590154611
N,	-5.2964239594,	-0.000412712,	0.4578780524
N,	2.2526426275,	0.0016584884,	-4.8156732785

Table S2B. Summary of excitation energies and oscillator strengths for $[\text{Y}(\text{PYZ})_3]^{3-}$ from TD-DFT (B3LYP/LANL2DZ) calculations.

Excited State 1: Triplet-A 2.8630 eV 433.06 nm f=0.0000

112 ->136	0.1044	2%
124 ->136	-0.12343	3%
128 ->134	0.1753	6%
128 ->135	0.10487	2%
128 ->136	0.13093	3%
129 ->134	0.1055	2%
129 ->135	-0.17204	6%
129 ->136	0.2178	9%
131 ->139	0.1266	3%
132 ->134	0.23187	11%
132 ->135	0.36658	27%

Excited State 2: Triplet-A 2.8631 eV 433.04 nm f=0.0000

111 ->136	0.1032	2%
123 ->136	0.12241	3%
128 ->134	-0.10387	2%
128 ->135	0.17667	6%
128 ->136	0.21625	9%
129 ->134	0.17681	6%
129 ->135	0.10336	2%
129 ->136	-0.13179	3%
130 ->138	-0.12286	3%
132 ->134	0.36401	27%
132 ->135	-0.23134	11%

Excited State 3: Triplet-A 2.8790 eV 430.66 nm f=0.0000

110 ->136	0.10114	2%
111 ->134	0.12866	3%
112 ->135	0.12894	3%
123 ->134	0.16017	5%
124 ->135	-0.16002	5%
128 ->134	0.31673	20%
129 ->135	0.31595	20%
130 ->137	0.1169	3%
132 ->136	0.3687	27%

Excited State 4: Triplet-A 3.0125 eV 411.57 nm f=0.0000

128 ->137	0.14754	4%
129 ->138	-0.12495	3%
130 ->134	0.37966	29%
131 ->135	0.39797	32%
132 ->139	0.17024	6%
133 ->136	-0.27247	15%

Excited State 5: Triplet-A 3.0155 eV 411.15 nm f=0.0000

127 ->139	0.10172	2%
128 ->137	-0.10887	2%

129 ->139	-0.16829	6%
130 ->134	-0.24344	12%
130 ->135	-0.10071	2%
131 ->134	-0.1115	2%
131 ->135	0.20628	9%
131 ->136	-0.28457	16%
132 ->137	-0.11866	3%
132 ->138	0.12502	3%
133 ->134	0.13182	3%
133 ->135	0.35992	26%

Excited State 1: Singlet-A 3.2084 eV 386.44 nm f=0.0012

128 ->136	0.1273	3%
130 ->134	-0.23155	11%
131 ->135	0.33546	23%
131 ->136	-0.24696	12%
132 ->134	0.23475	11%
133 ->135	0.36267	26%

Excited State 2: Singlet-A 3.2086 eV 386.41 nm f=0.0011

129 ->136	-0.12833	3%
130 ->135	-0.28404	16%
130 ->136	-0.24025	12%
131 ->134	-0.31368	20%
132 ->135	-0.23249	11%
133 ->134	0.32596	21%

Excited State 3: Singlet-A 3.2091 eV 386.35 nm f=0.0001

130 ->134	0.45157	41%
131 ->135	0.36588	27%
133 ->136	-0.28955	17%

Excited State 4: Singlet-A 3.2235 eV 384.63 nm f=0.0016

128 ->134	0.21736	9%
129 ->135	-0.21535	9%
129 ->136	0.20101	8%
130 ->136	-0.19971	8%
131 ->139	0.11783	3%
132 ->135	0.40538	33%
133 ->134	0.22791	10%

Excited State 5: Singlet-A 3.2237 eV 384.61 nm f=0.0016

128 ->135	0.21927	10%
128 ->136	0.20057	8%
129 ->134	0.22055	10%
130 ->138	-0.11159	2%
131 ->136	0.19778	8%
132 ->134	0.40315	33%
133 ->135	-0.23093	11%

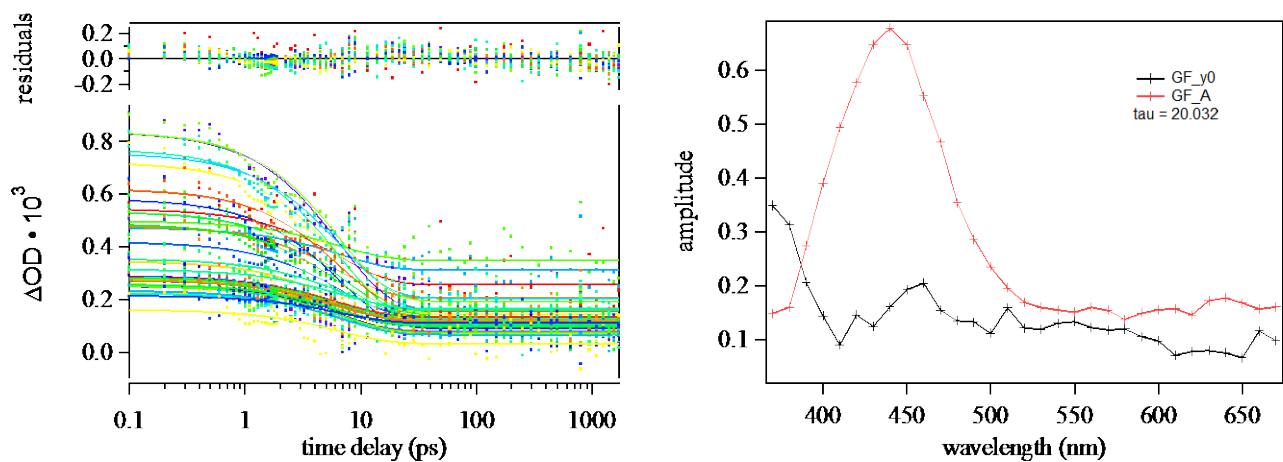


Figure S7. Results of global fitting observed TA spectra across all wavelengths for H₂DPA to an equation of the form; $\Delta OD(t) = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S₁ state lifetime, and y₀ the observed residual TA spectrum at long time delays (> 1.7 ns).

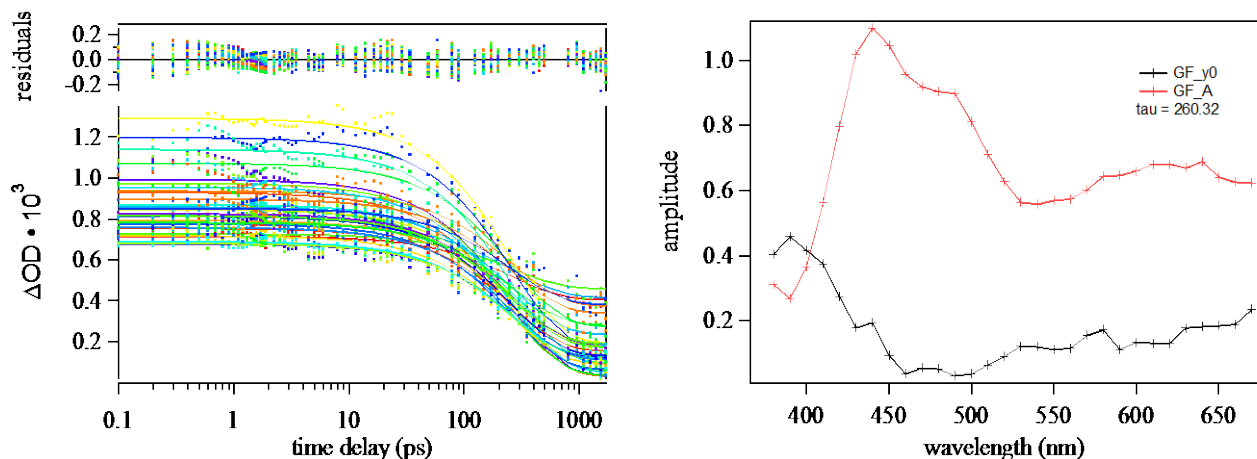


Figure S8. Results of global fitting observed TA spectra across all wavelengths for H₂PYZ to an equation of the form; $\Delta OD(t) = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S₁ state lifetime, and y₀ the observed residual TA spectrum at long time delays (> 1.7 ns).

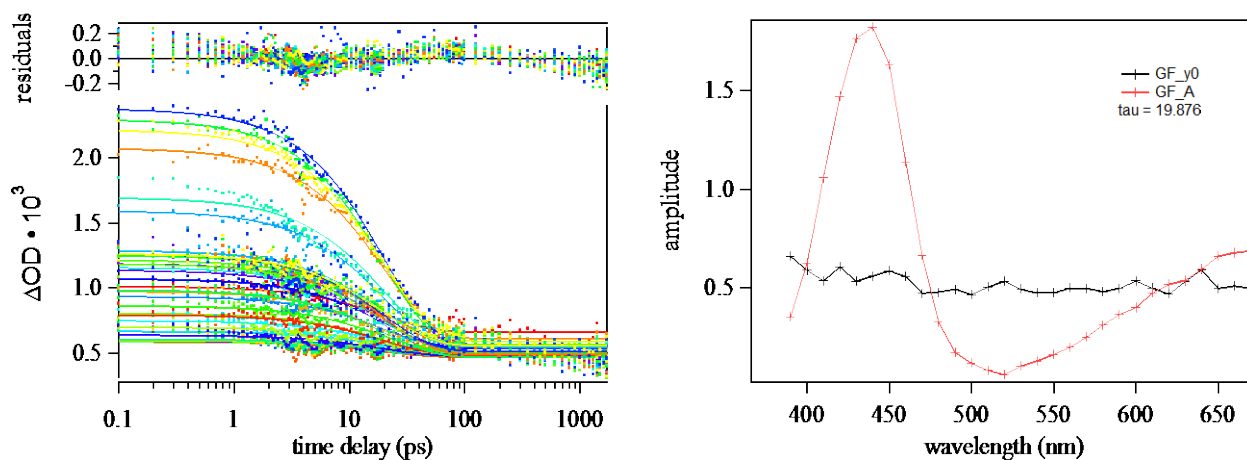


Figure S9. Results of global fitting observed TA spectra across all wavelengths for $[\text{Gd}(\text{DPA})_3]^{3-}$ to an equation of the form; $\Delta OD(t) = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S_1 state lifetime, and y_0 the observed residual TA spectrum at long time delays (> 1.7 ns).

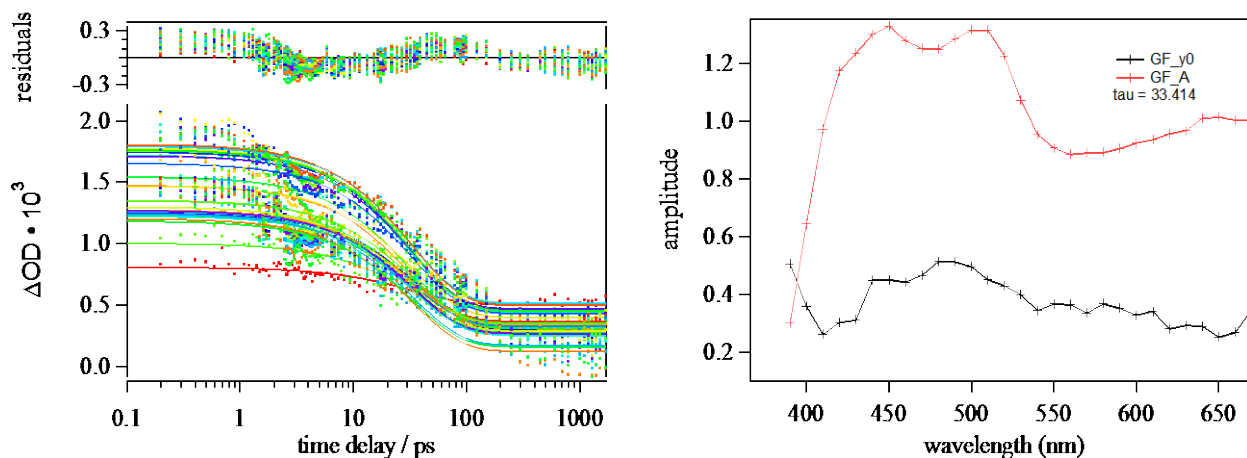


Figure S10. Results of global fitting observed TA spectra across all wavelengths for $[\text{Gd}(\text{PYZ})_3]^{3-}$ to an equation of the form; $\Delta OD(t) = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S_1 state lifetime, and y_0 the observed residual TA spectrum at long time delays (> 1.7 ns).

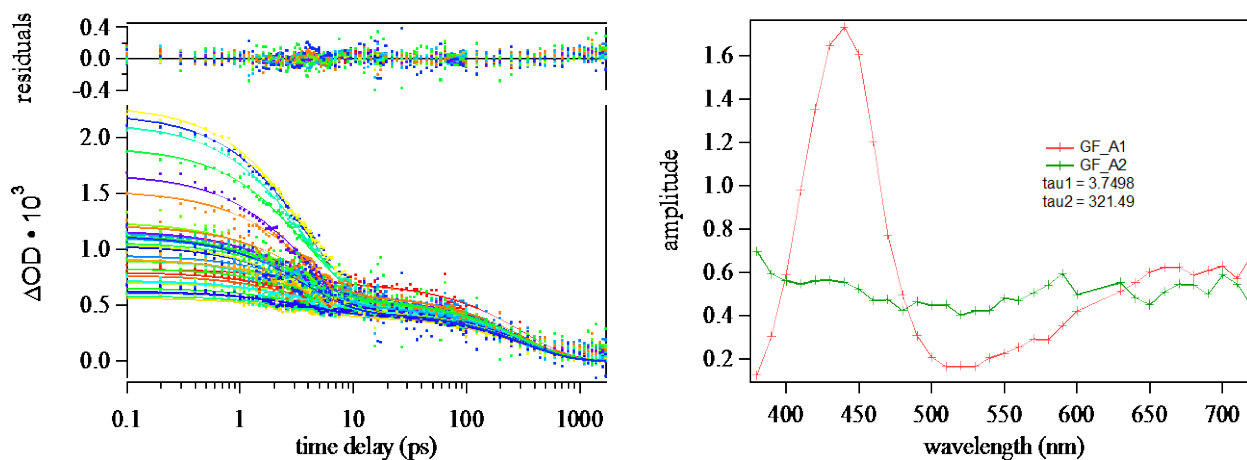


Figure S11. Results of global fitting observed TA spectra across all wavelengths for $[\text{Eu}(\text{DPA})_3]^{3-}$ to an equation of the form; $\Delta\text{OD}(t) = A_1 \exp^{-1/\tau_1 \times t} + A_2 \exp^{-1/\tau_2 \times t}$, where τ_1 represents the S_1 state lifetime, and τ_2 represents the T_1 state lifetime.

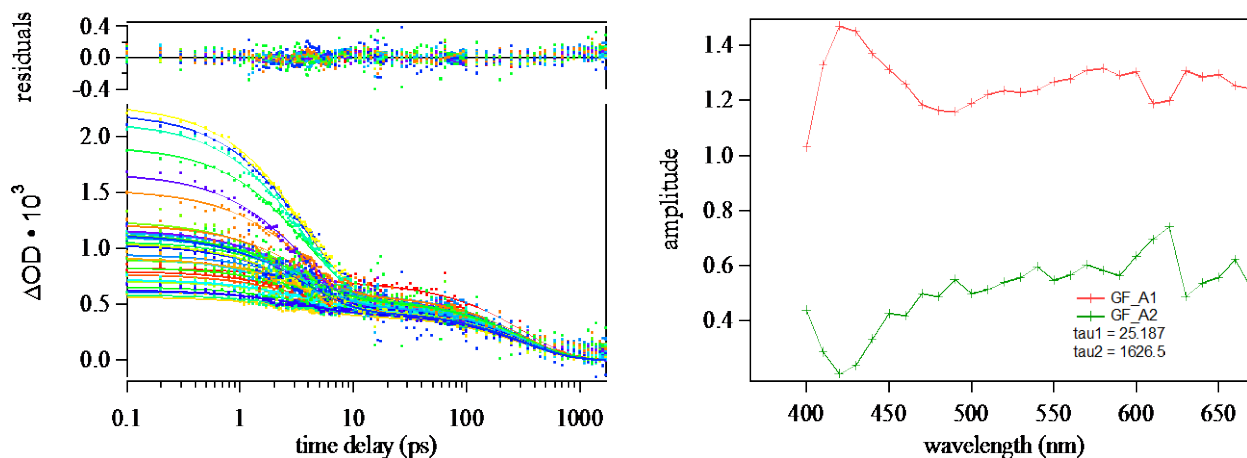


Figure S12. Results of global fitting observed TA spectra across all wavelengths for $[\text{Eu}(\text{PYZ})_3]^{3-}$ to an equation of the form; $\Delta\text{OD}(t) = A_1 \exp^{-1/\tau_1 \times t} + A_2 \exp^{-1/\tau_2 \times t}$, where τ_1 represents the S_1 state lifetime, and τ_2 represents the T_1 state lifetime.

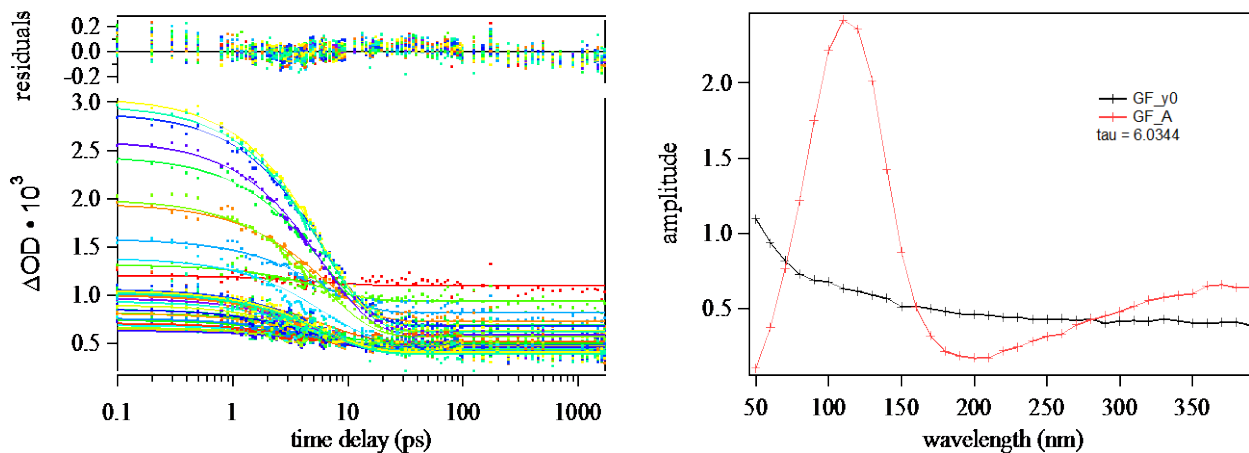


Figure S13. Results of global fitting observed TA spectra across all wavelengths for $[\text{Yb}(\text{DPA})_3]^{3-}$ to an equation of the form; $\Delta\text{OD}_{(t)} = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S_1 state lifetime, and y_0 the observed residual TA spectrum at long time delays (> 1.7 ns).

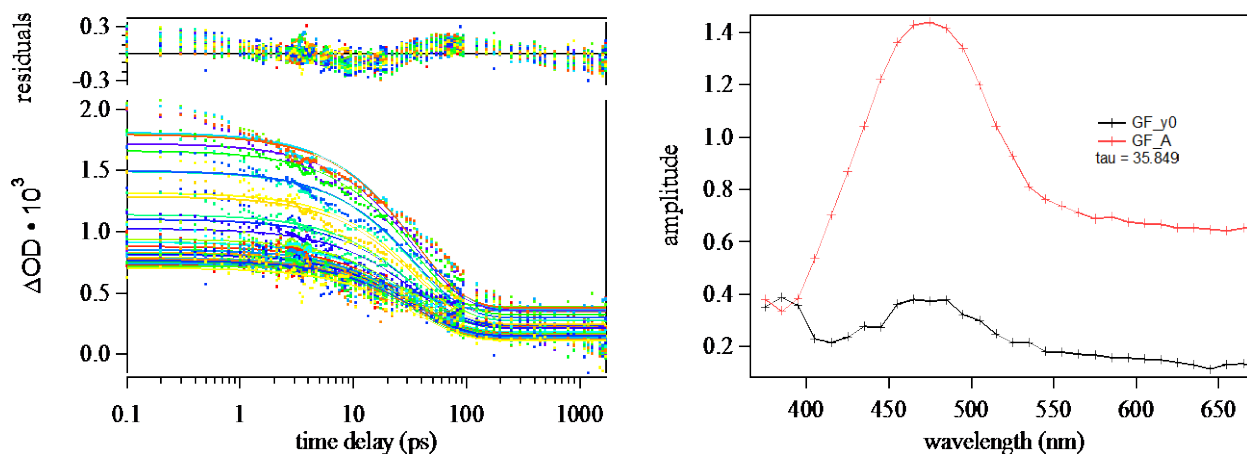


Figure S14. Results of global fitting observed TA spectra across all wavelengths for $[\text{Yb}(\text{PYZ})_3]^{3-}$ to an equation of the form; $\Delta\text{OD}_{(t)} = y_0 + A \exp^{-1/\tau \times t}$, where tau represents the S_1 state lifetime, and y_0 the observed residual TA spectrum at long time delays (> 1.7 ns).