Electronic Supplementary Information

Position-Selective Intramolecular Aromatic C-H Bond Activation of 1,2,3-Triazol-5-ylidene (tzNHC)

Ligands in (p-Cymene)Ruthenium(II) Complexes

Kenichi Ogata, Sayuri Inomata and Shin-ichi Fukuzawa*

Department of Applied Chemistry, Institute of Science and Engineering, Chuo University

1-13-27 Kasuga, Bunkyo-ku, Tokyo 112-8551, Japan

Contents

1.	Experimental procedures	S2-S6
	1.1 General procedures	S2
	1.2 Synthesis of triazole and triazolium salt	S2-S3
	1.3 Synthesis of ruthenium NHC complexes	S4-S6
2.	X-ray Crystallographic data	S7
3.	References	S 7

1. Experimental Procedures

1.1 General Procedures

All manipulations involving air- and moisture-sensitive organometallic compounds were carried out under a nitrogen atmosphere by using a standard Schlenk tube or high vacuum techniques. CH_2Cl_2 was distilled over calcium hydride under nitrogen atmosphere prior to use. THF was distilled over sodium-benzophenone under nitrogen atmosphere prior to use. 2,6-dimethyl phenyl azide,¹ 2-methyl phenyl azido,² **1a**,³ **1d**⁴ were prepared according to literature procedures. Other reagents employed in this research were commercially available and used without further purification. Column chromatography was performed on silica gel (spherical, 60 µm) and alumina (spherical, 75 µm). The ¹H and ¹³C NMR spectra recorded with a Varian Mercury-300 or Varian 400-MR spectrometer at ambient temperature. The chemical shifts were reported in δ units downfield from the internal reference (Me₄Si). All coupling constants were recorded in Hz. Elementary analysis were measured on a Perkin-Elmer 2400 series II CHN analyzer.

1.2 Synthesis of triazole and triazolium salt

1.2.1 Preparation of the 1-(2,6-xylyl)-4-phenyl-1,2,3-triazole.



The mixture of Phenylacetylene (0.37 mL, 3.3 mmol), 2,6-xylylazido (450 mg, 3.1 mmol), sodium ascorbate (1450 mg, 3.6 mmol), anhydrous copper sulfate (579 mg, 3.6 mmol), H₂O (5.0 mL) and *tert*-BuOH (20 mL) were charged in a 100 mL round-bottomed flask. After stirring at room temperature for 43 h under air, the mixture was extracted with ethyl acetate, washed with conc. NH₄Cl aq. dried over MgSO₄, and the volatile was removed under reduced pressure. The residual solid was purified by silica gel chromatograph (eluted with hexane first, then with ethyl acetate) to yield the 1-(2,6-xylyl)-4-phenyl-1,2,3-triazole as a white solid (721 mg, 2.89 mmol, 93%). ¹H NMR (300 MHz, CDCl₃) δ 7.94 (d, *J* = 7.0 Hz, 2H, Ar), 7.87 (s, 1H, triazol C=CH), 7.5 (m, 2H, Ar), 7.4-7.3 (m, 2H, Ar) 7.21 (d, *J* = 7.7 Hz, 2H, Ar), 2.07 (s, 6H, xylyl-CH₃). ¹³C NMR (75 MHz, CDCl₃) δ 147.6, 135.8, 135.4, 130.3, 130.0, 128.9, 128.4, 128.3, 125.7, 121.3 (s, Ar, triazole C=C), 17.3 (s, xylyl-CH₃). Anal. Calcd for C₁₆H₁₅N₃: C, 77.08; H, 6.06; N, 16.85. Found: C, 77.34; H, 6.04; N, 16.96%

1.2.2 Preparation of the 1-(2-methyl phenyl)-4-phenyl-1,2,3-triazole.



The mixture of Phenylacetylene (0.85 mL, 7.7 mmol), 2-methyl phenyl azido (1.02 g, 7.7 mmol), sodium ascorbate (153 mg, 0.77 mmol), copper sulfate pentahydrate (96.3 mg, 0.39 mmol), H_2O (11.0 mL) and *tert*-BuOH (11.0 mL) were charged in a 100 mL round-bottomed flask. After stirring at room temperature for overnight under air, the mixture was extracted with ethyl acetate, washed with conc. NH₄Cl aq. dried over MgSO₄, and the volatile was

removed under reduced pressure. The residual solid was purified by silica gel chromatograph (eluted with hexane first then with ethyl acetate) to yield the compound as a white solid (1.51 g, 6.42 mmol, 83%). The compound was prepared by modified method for previous report which have been synthesized this compound.² Spectroscopic data also was compared to data for previous report.⁵

1.2.3 Preparation of the triazolium salt 1b



The mixture of 1-(2-methyl phenyl)-4-phenyl-1,2,3-triazole (186 mg, 0.79 mmol), Me₃OBF₄ (159 mg, 1.03 mmol) and CH₂Cl₂ (10 mL) were charged in a 50 mL Schlenk tube. After stirring at room temperature for overnight, the reaction was quenched with MeOH, and the volatile was removed under reduced pressure. The residual solid was washed with Et₂O and dried to yield **1b** as a white solid (252 mg, 2.89 mmol, 95%). ¹H NMR (300 MHz, CDCl₃) δ 8.35 (s, 1H, triazolium CH), 7.8-7.7 (m, 3H, Ar), 7.6 (m, 3H), 7.6-7.5 (m, 1H, Ar), 7.43 (dd, *J* = 6.3, 6.3 Hz, 2H, Ph), 4.39 (s, 3H, N-CH₃), 2.38 (s, 3H, *o*-tolyl-CH₃). ¹³C NMR (75 MHz, (CD₃)₂SO) δ 142.7 (triazolium CH), 134.2, 133.4, 132.1, 131.7, 130.4, 129.5, 129.5, 127.6, 126.2, 122.6 (s, Ar, triazole), 39.0 (s, N-CH₃), 17.0 (s, xylyl-CH₃).Anal. Calcd for C₁₆H₁₆N₃: C, 57.00; H, 4.78; N, 12.46. Found: C, 57.08; H, 4.75; N, 12.54%

1.2.4 Preparation of the triazolium salt 1c.



The mixture of 1-xylyl-4-phenyl 1,2,3-triazole (125 mg, 0.50 mmol), Me₃OBF₄ (99.7 mg, 0.67 mmol) and CH₂Cl₂ 10 mL were charged in a 50 mL Schlenk tube. After stirring at room temperature for overnight, the reaction was quenched with MeOH, and the volatile was removed under reduced pressure. The residual solid was washed with Et₂O and dried to yield **1c** as a white solid (161 mg, 0.46 mmol, 92%). ¹H NMR (300 MHz, CDCl₃) δ 8.33 (s, 1H, triazolium CH), 7.81 (dd, *J* = 6.7, 3.1Hz, 2H, Ar), 7.7-7.6 (m, 3H, Ar), 7.48 (d, *J* = 8.0 Hz, 1H, Ar), 7.3 (m, 2H, Ar), 4.47 (s, 3H, N-CH₃), 2.22 (s, 6H, xylyl-CH₃). ¹³C NMR (100 MHz, (CD₃)₂SO) δ 143.2 (s, triazolium CH), 134.9, 133.7, 132.0, 131.7, 130.9, 129.7, 129.4, 129.1, 122.7 (s, Ar, triazole), 39.1 (s, N-CH₃), 16.9 (s, xylyl-CH₃). Anal. Calcd for C₁₇H₁₈N₃: C, 58.15; H, 5.17; N, 11.97. Found: C, 58.14; H, 5.06; N, 11.94%

Electronic Supplementary Material (ESI) for Dalton Transactions This journal is © The Royal Society of Chemistry 2013

1.3 Synthesis of ruthenium NHC complexes

1.3.1 Synthesis of ruthenium complex 2a.



The mixture of a triazolium salt 1a (171 mg, 0.53 mmol), Ag₂O (74.3 mg, 0.32 mmol), Me₄NCl (69.7 mg, 0.64 mmol) and CH₂Cl₂/CH₃CN (1:1, 20 mL) were charged in a 50 mL Schlenk tube. After stirring at room temperature for 2 h, the volatile was removed under reduced pressure. The residual solid was extracted with CH₂Cl₂ and filtered, and the volatile was removed under reduced pressure. To this residue, [RuCl₂(*p*-cymene)]₂ (162 mg, 0.26 mmol) and CH₂Cl₂15 mL were added. After stirring for 3h at room temperature, the volatile was concentrated in reduced pressure and purified by alumina column chromatography (CH₂Cl₂/acetone = 9/1) to yield **2a** as a yellow solid (142) mg, 0.28 mmol, 53%). Single crystals suitable for X-ray diffraction analysis were obtained by recrystallization from CH₂Cl₂ and Et₂O. ¹H NMR (400 MHz, CDCl₃) δ 8.24 (dd, J = 7.4, 1.1 Hz, 1H, Ar), 7.9 (m, 2H, Ar), 7.6 (m, 4H, Ar), 7.12 (ddd, J = 7.4, 7.3, 1.4 Hz, 1H, Ar), 7.02 (ddd, J = 7.6, 7.5, 1.4Hz, 1H, Ar), 5.26 (d, J = 7.0 Hz, *p*-cymene ring), 5.25 (d, J = 7.0 Hz, 1H, *p*-cymene ring), 4.83 (d, J = 5.8 Hz, 1H, *p*-cymene ring), 4.80 (d, J = 5.8Hz, 1H, p-cymene ring), 4.11 (s, 3H, N-CH₃), 2.14 (sep, 1H, J = 6.9 Hz, p-cymene-CH(CH₃)₂), 1.94 (s, 3H, *p*-cymene-CH₃), 0.81 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂), 0.81 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂). ¹³C NMR (100 MHz, CDCl₃) δ 173.0 (s, Ru-C(Ar)), 166.3 (s, carbene carbon), 145.7 144.9, 142.0, 130.6, 129.7, 128.9, 128.7, 127.4, 122.3, 113.8 (s, Ph, triazole), 102.1, 99.6, 89.7, 89.6, 88.4, 84.3 (s, p-cymene ring), 37.0, 30.9, 23.1, 21.5, 18.9 (s, p-cymene-CH₃, p-cymene-CH(CH₃)₂, N-CH₃). Anal. Calcd for C₂₅H₂₆ClN₃Ru: C, 59.46; H, 5.19; N, 8.32. Found: C, 59.42; H, 5.11; N, 8.28%

1.3.2 Synthesis of ruthenium complexes 2b.



Complex 2b was prepared from triazolium salt **1b** (80.7 mg, 0.24 mmol), Ag₂O (33.9 mg, 0.15 mmol), Me₄NCl (31.6 mg, 0.29 mmol) and [RuCl₂(*p*-cymene)]₂ (73.9 mg, 0.12 mmol) in the same manner as that for **2a** except for purification method. Complex **2b** was isolated by recrystallization from CH₂Cl₂/Et₂O as an orange solid (50.3 mg, 0.10 mmol, 40%). Single crystals suitable for X-ray diffraction analysis were obtained by recrystallization from 1,2-C₂H₄Cl₂ and Et₂O. ¹H NMR (300 MHz, CDCl₃) δ 8.13 (d, *J* = 6.9 Hz, 1H, *o*-tolyl), 7.9 (m, 2H, Ph), 7.6 (m, 3H,

Ph), 6.98 (dd, 1H, J = 7.4, 7.4 Hz, *o*-tolyl), 6.80 (d, J = 7.2 Hz, 1H, *o*-tolyl), 5.21 (s, 2H, *p*-cymene ring), 4.80 (d, J = 5.8 Hz, 1H, *p*-cymene ring), 4.11 (s, 3H, N-CH₃), 2.62 (s, 3H, *o*-tolyl-CH₃), 2.12 (sep, J = 6.9 Hz, 1H, *p*-cymene-CH(CH₃)₂), 1.92 (s, 3H, *p*-cymene-CH₃), 0.80 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂), 0.73 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂). ¹³C NMR (75 MHz, CDCl₃) δ 174.2 (s, Ru-C(Ar)), 167.6 (s, carbene carbon), 144.5, 143.6, 139.7, 130.8, 129.6, 128.9, 128.8, 126.6, 126.1, 125.9 (s, Ar, triazole), 102.1, 99.5, 90.1, 88.6, 84.7 (s, *p*-cymene ring), 37.1, 30.8, 23.0, 21.5, 20.8, 18.9 (s, *p*-cymene-CH(CH₃)₂, *p*-cymene-CH(CH₃)₂, N-CH₃, *o*-tolyl-CH₃). Anal. Calcd for C₂₆H₂₈ClN₃Ru-0.25CH₂Cl₂: C, 58.36; H, 5.32; N, 7.78. Found: C, 58.46; H, 5.21; N, 7.88%

1.3.3 Synthesis of ruthenium complexes 3c.



Complex **3c** was prepared from triazolium salt **1c** (231 mg, 0.66 mmol), Ag₂O (92.6 mg, 0.40 mmol), Me₄NCl (86.6 mg, 0.79 mmol) and [RuCl₂(*p*-cymene)]₂ (203 mg, 0.33 mmol) in the same manner as that for **2b**. Complex **3c** was isolated as an orange solid (300 mg, 0.53 mmol, 80%). Single crystals suitable for X-ray diffraction analysis were obtained by recrystallization from CH₂Cl₂ and Et₂O. ¹H NMR (300 MHz, CDCl₃) δ 8.0 (m, 2H, Ar), 7.6 (m, 3H, Ar), 7.23 (t, *J* = 7.5 Hz, 1H, Ar), 7.05 (d, *J* = 7.5 Hz, 2H, Ar), 5.13 (d, *J* = 5.9 Hz, 2H, *p*-cymene ring), 4.22 (d, 2H, *J* = 5.9 Hz, *p*-cymene ring), 3.97 (s, 3H, N-CH₃), 2.85 (sep, *J* = 6.9 Hz, 1H, *p*-cymene-CH(CH₃)₂), 1.80 (s, 3H, *p*-cymene-CH₃), 1.16 (d, *J* = 6.9 Hz, 6H, *p*-cymene-CH(CH₃)₂). ¹³C NMR NMR (300 MHz, CDCl₃) δ 162.4 (s, carbene carbon), 147.7, 140.1, 135.4, 131.8, 130.1, 129.8, 129.3, 128.6, 127.3 (s, Ar, triazole), 106.6, 97.0, 88.6, 80.4 (*p*-cymene ring), 37.6, 30.3, 22.4, 18.5, 18.4 (s, *p*-cymene-CH(CH₃)₂, *p*-cymene-CH(CH₃)₂, *N*-CH₃, xylyl). Anal. Calcd for Calcd for C₂₇H₃₁Cl₂N₃Ru·1/3CH₂Cl₂: C, 54.91; H, 5.34; N, 7.03. Found: C, 54.63; H, 5.27; N, 6.99%

1.3.4 Synthesis of ruthenium complexes 3d.



Complex **3d** was prepared from **1d** (186 mg, 0.55 mmol), Ag₂O (77.3 mg, 0.33 mmol), Me₄NCl (72.3 mg, 0.66 mmol) and [RuCl₂(*p*-cymene)]₂ (169 mg, 0.28 mmol) in the same manner as that for **2b**. Complex **3d** was isolated as an orange solid (297 mg, 0.53 mmol, 97%). ¹H NMR (300 MHz, CDCl₃) δ 7.72 (dd, *J* = 6.5, 3.1 Hz, 2H, Ar), 7.5 (m, 3H, Ar), 7.4-7.3 (m, 5H, Ar), 6.28 (s, 2H, Bn-CH₂), 5.13 (d, *J* = 6.0 Hz, 2H, *p*-cymene ring), 4.74 (d, *J* = 6.0

Hz, 2H, *p*-cymene ring), 3.74 (s, 3H, N-CH₃), 2.47 (sep, J = 6.9 Hz, 1H, *p*-cymene-CH(CH₃)₂), 1.59 (s, 3H, *p*-cymene-CH₃), 1.07 (d, J = 6.9 Hz, 6H, *p*-cymene-CH(CH₃)₂). ¹³C NMR (75MHz, CDCl₃) δ 161.3 (s, carbene carbon), 148.5, 136.4, 131.9, 129.7, 128.6, 128.5, 127.9 (s, Ar, triazole), 105.5, 96.4, 84.7, 83.2 (s, *p*-cymene ring), 57.1 (s, Bn-CH₂), 37.1, 30.3, 22.4, 17.9 (s, *p*-cymene-CH₃, *p*-cymene-CH(CH₃)₂, *p*-cymene-CH(CH₃)₂, N-CH₃). Anal. Calcd for Calcd for C₂₆H₂₉Cl₂N₃Ru·0.2CH₂Cl₂: C, 54.97; H, 5.18; N, 7.34. Found: C, 55.26; H, 5.25; N, 7.49%

1.3.5 Synthesis of ruthenium complexes 4c.



The mixture of the ruthenium complex **3c** (89.5 mg, 0.16 mmol), K_2CO_3 (35.0 mg, 0.25 mmol) and THF 5 mL were charged in a 20 mL Schlenk tube. After stirring at reflux for 6 h, the volatile was removed under reduced pressure. The residual solid was extracted with CH₂Cl₂ and filtered. The volatile was removed under reduced pressure and the residue was recrystallized (CH₂Cl₂/Et₂O) to yield **4c** as a yellow solid (61.6 mg, 0.12 mmol, 75 %). ¹H NMR (300 MHz, CDCl₃) δ 8.27 (d, J = 7.4, Hz, 1H, Ph), 7.47 (dd, J = 7.6, 7.6 Hz, 1H, Ph), 7.4-7.3 (m, 3H, xylyl), 7.06 (dd, J = 7.4, 7.2 Hz, 1H, Ph), 6.96 (dd, J = 7.6, 7.6 Hz, 1H, Ph), 5.53 (d, J = 6.0 Hz, 1H, *p*-cymene ring), 4.91 (d, 1H, J = 6.0 Hz, *p*-cymene ring), 4.55 (d, 1H, J = 5.5 Hz, *p*-cymene ring), 4.28 (d, 1H, J = 5.5 Hz, *p*-cymene ring), 4.27 (s, 3H, N-CH₃), 2.39 (s, 3H, *p*-cymene-CH₃), 2.30 (sep, J = 6.9 Hz, 1H, *p*-cymene-CH(CH₃)₂), 2.10 (s, 3H, xylyl-CH₃), 1.82 (s, 3H, xylyl-CH₃), 0.92 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂), 0.52 (d, J = 6.9 Hz, 3H, *p*-cymene-CH(CH₃)₂). ¹³C NMR NMR (300 MHz CDCl₃) δ 180.8, 178.4 (s, Ru-C(Ar), carbene carbon), 152.9 (s, triazle), 142.2, 138.1, 137.9, 136.5, 135.2, 130.2, 128.9, 127.8, 127.4, 121.8, 120.8 (s, Ar), 100.9, 99.8, 93.6, 90.8, 89.5, 75.9 (s, *p*-cymene ring), 37.3, 30.6, 23.9, 20.7, 18.9, 18.4, 17.6 (s, *p*-cymene-CH(CH₃)₂), *p*-cymene-CH(CH₃)₂, *x*ylyl, N-CH₃). HR-MS (ESI) calcd for C₂₇H₃₀ClN₃Ru [M+Na]+ 556.1069, found 556.1068.

2. X-ray Crystallographic data

Crystal data for **2a**: $C_{25}H_{26}ClN_3Ru$, M = 505.02, monoclinic, a = 7.9014(19), b = 23.660(6), c = 11.737(3) Å, V = 2187.0(9) Å³, T = 123 K, space group P2₁/n (no. 14), Z = 4, μ (Mo–K α) = 8.562 cm⁻¹, 16427 measured reflections, 4990 unique reflections. R1 = 0.0303, wR2 = 0.0703, for 4550 reflections. For **2b**·CH₂Cl₂: $C_{28}H_{32}Cl_3N_3Ru$, M = 618.01, monoclinic, a = 29.7166(6), b = 11.3579(2), c = 21.1834(4) Å, V = 5507(2) Å³, T = 198 K, space group C2/c (no. 15), Z = 8, μ (Mo–K α) = 8.822 cm⁻¹, 25744 measured reflections, 6277 unique reflections. R1 = 0.0365, wR2 = 0.0917, for 5904 reflections. For **3c**·2CH₂Cl₂: $C_{29}H_{35}Cl_6N_3Ru$, M = 739.40, monoclinic, a = 22.430(5), b = 13.091(3), c = 23.693(3) Å, V = 6449(3) Å³, T = 123 K, space group P2₁/n (no. 14), Z = 8, μ (Mo–K α) = 10.070 cm⁻¹, 48952 measured reflections, 14715 unique reflections. R1 = 0.0252, for 11198 reflections. For **4c**·CH₂Cl₂: $C_{28}H_{32}Cl_3N_3Ru$, M = 618.01, monoclinic, a = 10.8748(3), b = 9.4047(2), c = 26.8586(6) Å, V = 2739.6(2) Å³, T = 198 K, space group P2₁/n (no. 14), Z = 4, μ (Mo–K α) = 8.867 cm⁻¹, 25313 measured reflections, 6266 unique reflections. R1 = 0.0302, wR2 = 0.0541, for 5599 reflections. CCDC numbers: 901766 (**2a**), 901767 (**2b**), 901768 (**3c**) and 901934 (**4c**).

3. References

- 1. L. P. Spencer, R. Altwer, P. Wei, L. Gelmini, J. Gauld, D. W. Stephan, Organometallics, 2003, 22, 3841.
- H. Cheng, J. Wan, M.-I. Lin, Y. Liu, X. Lu, J. Liu, Y. Xu, J. Chen, Z. Tu, Y.-S. E. Cheng, K. Ding, J. Med. Chem. 2012, 55, 2144.
- 3. T. Nakamura, T. Terashima, K. Ogata, S.-i. Fukuzawa, Org. Lett., 2011, 13, 620.
- 4. K. J. Kilpin, U. D. D. Paul, A.-L. Lee, J. D. Crowly, Chem. Commun., 2011, 47, 328.
- 5. C. Tao, X. Cui, J. Li, A.-X. Liu, L. Liu, Q.-X. Guo, Terahedron Letters, 2007, 48, 3525.