

## **Electronic Supplementary Information**

for

### **Sterically encumbered pyrrolyl ligands and their incorporation into the cycloheptatrienyl zirconium coordination sphere**

Markus Kreye, Andreas Glöckner, Constantin G. Daniliuc, Matthias Freytag, Peter G. Jones,  
Matthias Tamm and Marc D. Walter\*

Institut für Anorganische und Analytische Chemie, Technische Universität Braunschweig,  
Hagenring 30, 38106 Braunschweig (Germany), Fax: +49 531-391-5309  
E-mail: [mwalter@tu-bs.de](mailto:mwalter@tu-bs.de)

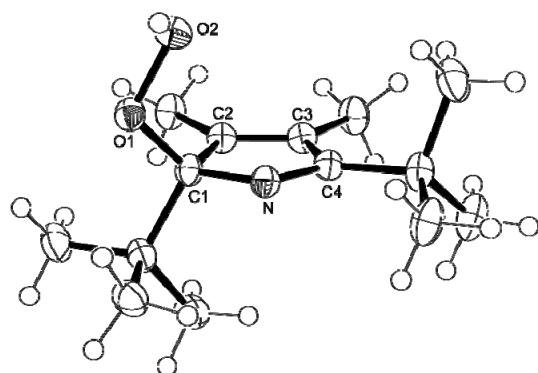
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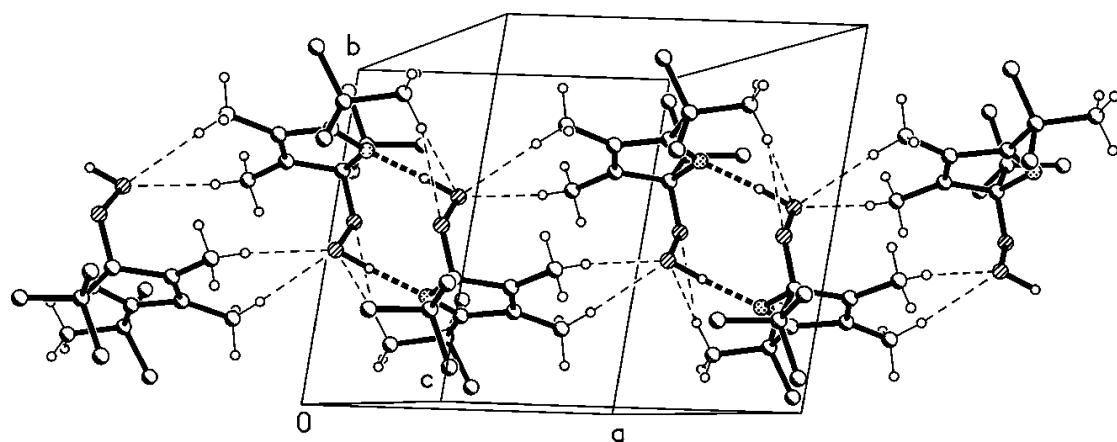
## 1. Additional molecular structures

### Molecular structure of the hydroperoxide of 3

Special features: The OH hydrogen was refined freely.



**Figure S1.** ORTEP diagram with thermal displacement parameters drawn at 50% probability.



**Figure S2.** Section of the crystal lattice showing classical and "weak" hydrogen bonds (thick and thin dashed lines respectively).

### Crystal data and structure refinement

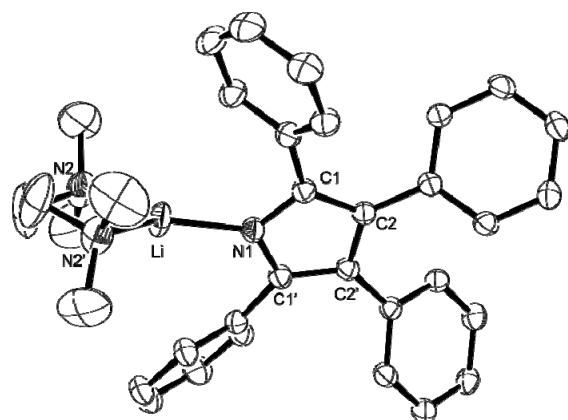
Empirical formula	C <sub>14</sub> H <sub>25</sub> NO <sub>2</sub>
Formula weight	239.35
Temperature	100(2) K
Wavelength	1.54184 Å

Crystal system	Triclinic
Space group	$P\bar{1}$
Unit cell dimensions	$a = 8.7383(8)$ Å $\alpha = 81.310(7)^\circ$ $b = 9.5314(8)$ Å $\beta = 63.418(8)^\circ$ $c = 9.7570(8)$ Å $\gamma = 82.113(7)^\circ$
Volume	$716.17(11)$ Å <sup>3</sup>
Z	2
Density (calculated)	1.110 Mg/m <sup>3</sup>
Absorption coefficient	0.574 mm <sup>-1</sup>
F(000)	264
Crystal size	0.25 x 0.15 x 0.10 mm <sup>3</sup>
Theta range for data collection	4.71 to 76.14°
Index ranges	-10<=h<=10, -11<=k<=11, -12<=l<=12
Reflections collected	30721
Independent reflections	2965 [R(int) = 0.0284]
Completeness to theta = 75.00°	99.9 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9448 and 0.8698
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	2965 / 0 / 166
Goodness-of-fit on F <sup>2</sup>	1.033
Final R indices [I>2sigma(I)]	R1 = 0.0384, wR2 = 0.1031
R indices (all data)	R1 = 0.0406, wR2 = 0.1052
Largest diff. peak and hole	0.282 and -0.238 e.Å <sup>-3</sup>

### Molecular structure of (tmeda)Li(Pyr<sup>Ph4</sup>)

A suspension of 2,3,4,5-tetraphenyl-1*H*-pyrrol<sup>[1]</sup> (3.0 g, 8.08 mmol) in a pentane:toluene mixture (1:1) (40 mL) was cooled to 0°C and *n*-BuLi (5.1 mL, 1.6M in hexane) was added. The reaction mixture was warmed to ambient temperature and TMEDA (1 mL) was added and the suspension

was stirred at room temperature overnight. The solution was filtered and dried under dynamic vacuum to yield a colourless powder (3.89 g, 8.06 mmol, 99%). (*tmeda*)Li(Pyr<sup>Ph<sub>4</sub></sup>) can be crystallized from a saturated toluene solution at -30 °C. The compound displays crystallographic twofold symmetry and crystallizes with two molecules of toluene per molecule, which are each disordered over an inversion centre.



**Figure S3.** ORTEP diagram with thermal displacement parameters drawn at 50% probability, hydrogen atoms and solvent molecules are omitted for clarity.

### Crystal data and structure refinement

Empirical formula	C <sub>48</sub> H <sub>52</sub> LiN <sub>3</sub>
Formula weight	677.87
Temperature	100(2) K
Wavelength	1.54184 Å
Crystal system	monoclinic
Space group	C2/c
Unit cell dimensions	a = 20.6834(4) Å      α = 90° b = 17.2623(2) Å      β = 110.377(2)° c = 11.8317(2) Å      γ = 90°
Volume	3960.07(11) Å <sup>3</sup>
Z	4
Density (calculated)	1.137 Mg/m <sup>3</sup>

Absorption coefficient	0.492 mm <sup>-1</sup>
F(000)	1456
Crystal size	0.07 x 0.07 x 0.06 mm <sup>3</sup>
Theta range for data collection	3.43 to 76.08°
Index ranges	-24<=h<=25, -21<=k<=21, -14<=l<=14
Reflections collected	26512
Independent reflections	4114 [R(int) = 0.0250]
Completeness to theta = 75.00°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.65569
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	4114 / 98 / 277
Goodness-of-fit on F <sup>2</sup>	1.063
Final R indices [I>2sigma(I)]	R1 = 0.0504, wR2 = 0.1309
R indices (all data)	R1 = 0.0540, wR2 = 0.1341
Largest diff. peak and hole	0.448 and -0.291 e.Å <sup>-3</sup>

## 2. Cone angles

The cone angles are measured according to a recent report.<sup>[2]</sup> Values used for the determination of  $\Theta$  or  $\Omega$  are highlighted in yellow.

### $[(\eta^7\text{-C}_7\text{H}_7)\text{Zr}(\eta^5\text{-Pyr}^{t\text{Bu}_2})] \text{ (6)}$

#### Centroid-Zr-H

N	29.77		
H9	42.70		
H10	42.33		

Theta: 111.92

#### *t*Bu on C8

1. Me	82.12	
	70.65	
	66.01	
2. Me	72.10	
	63.80	
	52.94	
3. Me	45.48	
	35.94	
	30.82	

#### *t*Bu on C11

1. Me	82.87	
	70.95	
	67.03	
2. Me	72.11	
	64.86	
	53.13	
3. Me	46.35	
	36.74	
	31.48	

#### C(alpha)-C(ipso)-H

##### *t*Bu on C8

1. Me	50.64		
	49.52		
	25.25		
2. Me	51.07		

Omega: 102.65

	48.66
	25.20
3. Me	52.32
	51.60
	27.15
<hr/>	
tBu on C11	
1. Me	51.64
	47.77
	25.10
2. Me	52.20
	51.29
	26.88
3. Me	50.09
	49.76
	25.03

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$[(\eta^7\text{-C}_7\text{H}_7)\text{Zr}(\eta^5\text{-Pyr}^{t\text{Bu}_2\text{Me}_2})]$  (7)

Centroid-Zr-H

N	30.29	Theta:	124.42
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tBu on C8

1. Me	75.96
	67.92
	58.10
2. Me	79.33
	69.00
	61.04
3. Me	44.74
	32.86
	32.17
Me on C9	60.72
	52.51
	41.23

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Me on C10

61.17
49.82
41.62

---

tBu on C11

1. Me	79.55
	68.24
	61.23
2. Me	76.30
	68.26
	58.49
3. Me	45.01
	33.04
	32.47

C(alpha)-C(ipso)-H

tBu on C8		For tBu	
1. Me	49.37	Omega:	101.90
	46.67		
	23.31	For Me	
2. Me	50.15	Omega:	53.44
	49.73		
	25.11		
3. Me	52.95		
	51.76		
	27.57		

Me on C9

26.73
26.72
26.74

Me on C10

26.71
26.71
26.70

tBu on C11

1. Me	51.27
	48.37
	25.19
2. Me	49.31
	46.64
	23.32
3. Me	52.65
	51.68
	27.45

**[( $\eta^7$ -C<sub>7</sub>H<sub>7</sub>)Zr( $\eta^5$ -Pyr<sup>tBu<sub>3</sub></sup>)] (8)**

These values are based on a calculated molecular structure (see below).

Centroid-Zr-H

N	29.55		
H10	43.84	Theta:	127.75

*t*Bu on C8

1. Me	82.58
	69.31
	62.58
2. Me	77.17
	67.27
	57.74
3.Me	45.98
	33.64
	31.4

*t*Bu on C9

1. Me	80.61
	70.99
	68.8
2. Me	63.23
	56.85
	42.86
3. Me	44
	40.05
	26.75

*t*Bu on C10

1. Me	82.8
	69.44
	64.42
2. Me	74.39
	64.97
	53.65
3.Me	45.16
	34.73
	29.54

C(alpha)-C(ipso)-H

*t*Bu on C8

1. Me	52.84	Omega:	104.32
	50.78		
	24.53		
2. Me	48.78		
	47.04		
	20.71		
3. Me	54.2		
	53.02		
	26.49		
<hr/>			
tBu on C9			
1. Me	49.71		
	48.2		
	21.79		
2. Me	51.58		
	51.35		
	24.25		
3. Me	53.87		
	50.71		
	25.31		
<hr/>			
tBu on C11			
1. Me	51.53		
	51.25		
	24.02		
2. Me	52.48		
	49.65		
	23.76		
3. Me	54.46		
	53.94		
	26.86		

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**[( $\eta^7$ -C<sub>7</sub>H<sub>7</sub>)Zr( $\eta^5$ -Im<sup>tBu<sub>3</sub></sup>)] (9)**

Centroid-Zr-H		Theta:	
N1	29.00		
N2	29.11		117.91

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**tBu on C8**

1. Me	73.41
	63.36

		54.57
2. Me		78
		66.06
		61.05
3.Me		41.98
		32.44
		28.16
<hr/>		
tBu on C9		
1. Me		80.9
		69.73
		69.08
2. Me		65.98
		59.53
		46.73
3. Me		44.21
		38.15
		28.59
<hr/>		
tBu on C10		
1. Me		77.77
		66.48
		60.84
2. Me		72.32
		63.46
		54.81
3.Me		42.29
		31.32
		29.6
<hr/>		

C(alpha)-C(ipso)-H

			Omega:	101.88
tBu on C8				
1. Me		50.91		
		47.59		
		24.51		
2. Me		50.78		
		49.05		
		25.03		
3.Me		54.01		
		52.01		
		28.2		
<hr/>				
tBu on C9				
1. Me		48.14		

	46.65
	22.74
2. Me	50.61
	50.39
	25.79
3. Me	51.87
	50.31
	26.45
<hr/>	
tBu on C10	
1. Me	47.3
	46.47
	22.08
2. Me	51.5
	48.88
	25.59
3. Me	53.36
	51.41
	27.71

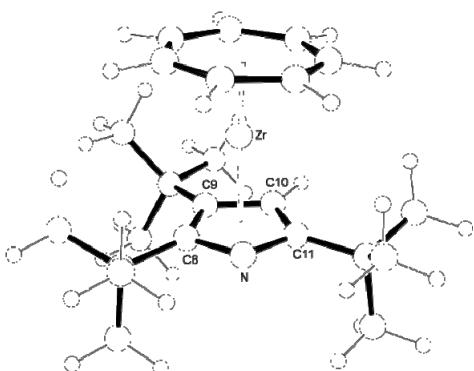
### 3. Computational details

All computations were performed using the density functional method M06 as implemented in the Gaussian09 program.<sup>[3]</sup> For all main-group elements (C, H and N) the all-electron triple-z basis set (6-311G\*\*) was used,<sup>[4]</sup> whereas for zirconium a small-core relativistic ECP together with the corresponding double-z valence basis set was employed (Stuttgart RSC 1997 ECP).<sup>[5]</sup>

#### Energies of the optimized structures

Compound	E(0 K) <sup>a</sup> [Ha]	H(298 K) <sup>b</sup> [Ha]	G(298 K) <sup>b</sup> [Ha]
[Cp <sup>tBu<sup>2</sup></sup> ] <sup>-</sup>	-507.457974	-507.442228	-507.498594
[Pyr <sup>tBu<sup>2</sup></sup> ] <sup>-</sup>	-523.541060	-523.525492	-523.581666
[Cp <sup>tBu<sup>3</sup></sup> ] <sup>-</sup>	-664.504258	-664.483676	-664.548774
[Pyr <sup>tBu<sup>3</sup></sup> ] <sup>-</sup>	-680.588493	-680.567830	-680.634315
[Im <sup>tBu<sup>3</sup></sup> ] <sup>-</sup>	-696.672281	-696.651709	-696.717735
[(η <sup>7</sup> -C <sub>7</sub> H <sub>7</sub> )Zr(η <sup>5</sup> -Cp <sup>tBu<sup>2</sup></sup> )]	-825.219538	-825.196222	-825.269813
[(η <sup>7</sup> -C <sub>7</sub> H <sub>7</sub> )Zr(η <sup>5</sup> -Pyr <sup>tBu<sup>2</sup></sup> )] ( <b>6</b> )	-841.283342	-841.260272	-841.332909
[(η <sup>7</sup> -C <sub>7</sub> H <sub>7</sub> )Zr(η <sup>5</sup> -Cp <sup>tBu<sup>3</sup></sup> )]	-982.262952	-982.234451	-982.318294
[(η <sup>7</sup> -C <sub>7</sub> H <sub>7</sub> )Zr(η <sup>5</sup> -Pyr <sup>tBu<sup>3</sup></sup> )] ( <b>8</b> )	-998.331440	-998.302969	-998.385832
[(η <sup>7</sup> -C <sub>7</sub> H <sub>7</sub> )Zr(η <sup>5</sup> -Im <sup>tBu<sup>3</sup></sup> )] ( <b>9</b> )	-1014.395807	-1014.367815	-1014.449099

<sup>a</sup>DFT energy incl. ZPE. <sup>b</sup>standard conditions T = 298.15 K and p = 1 atm.



**Figure S4.** Calculated molecular structure of **8**. Selected bond lengths (Å): Zr-N 2.390, Zr-C8 2.476, Zr-C9 2.615, Zr-C10 2.580, Zr-C11 2.481.

## 4. References

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M. Dolg, H. Stoll, H. Preuss and R. M. Pitzer, *J. Phys. Chem.*, 1993, **97**, 5852.