

Electronic Supplementary Information

for

A Triangular Prismatic Hexanuclear Iridium(III) Complex Bridged by Flavin Analogues Showing Reversible Redox Processes

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Table S1. X-ray crystallographic data for **1•PF₆**.

formula	C ₉₀ H ₉₉ Cl ₆ F ₁₈ Ir ₆ N ₁₂ O ₉ P ₃
fw	3293.78
crystal system	trigonal
space group	R ₃ c
T (K)	123
a (Å)	16.0783(6)
c (Å)	74.528(6)
V (Å ³)	16685.1(15)
Z	6
no. of reflections measured	31593
no. of observations	2214
no. of parameters refined	211
R1 ^a	0.0965
R _w ^b	0.2415 (I > 2.0σ(I)) ^c
GOF	1.092
CCDC no.	888985

^a R1 = Σ ||Fo| - |Fc|| / Σ |Fo|. ^b R_w = [Σ (w (F₀² - F_c²)²) / Σ w(F₀²)²]^{1/2}. ^c w = 1 / [σ²(F_θ²) + (0.1361P)² + 1691.4005P], where P = (Max(F_θ², 0) + 2F_c²) / 3

Table S2. Selected bond lengths (\AA) and angles (deg) of **1**

Ir1 – O1	2.14(4)	N1 – Ir1 – O1	72.5(11)
Ir1 – N1	2.17(3)	N1 – Ir1 – N3	87.1(14)
Ir2 – N2	1.98(3)	O1 – Ir1 – N3	82.7(13)
Ir2 – O2	2.22(4)	Ir1 – O1 – C12	119 (3)
Ir2 – O3	2.13(4)	Ir1 – N1 – C11	111.7(17)
Ir1 – N3	2.16(2)	Ir1 – N3 – C13	118.1(18)
O1 – C12	1.23(4)	Ir1 – N3 – C14	121 (2)
N1 – C11	1.35(4)	O1 – C12 – C11	116 (3)
N1 – C20	1.35(4)	N1 – C11 – C12	116(2)
N2 – C12	1.39(5)	N2 – Ir2 – O2	60.6(12)
N2 – C13	1.39(4)	N2 – Ir2 – O3	91.0(13)
O2 – C13	1.10(5)	Ir2 – N2 – C13	92.3(19)
N3 – C13	1.39(5)	Ir2 – O2 – C13	89(3)
N3 – C14	1.39(5)	O2 – C13 – N2	118(3)
N4 – C14	1.35(4)	O2 – C13 – N3	122(3)
N4 – C15	1.35(4)		
C11 – C12	1.39(5)		

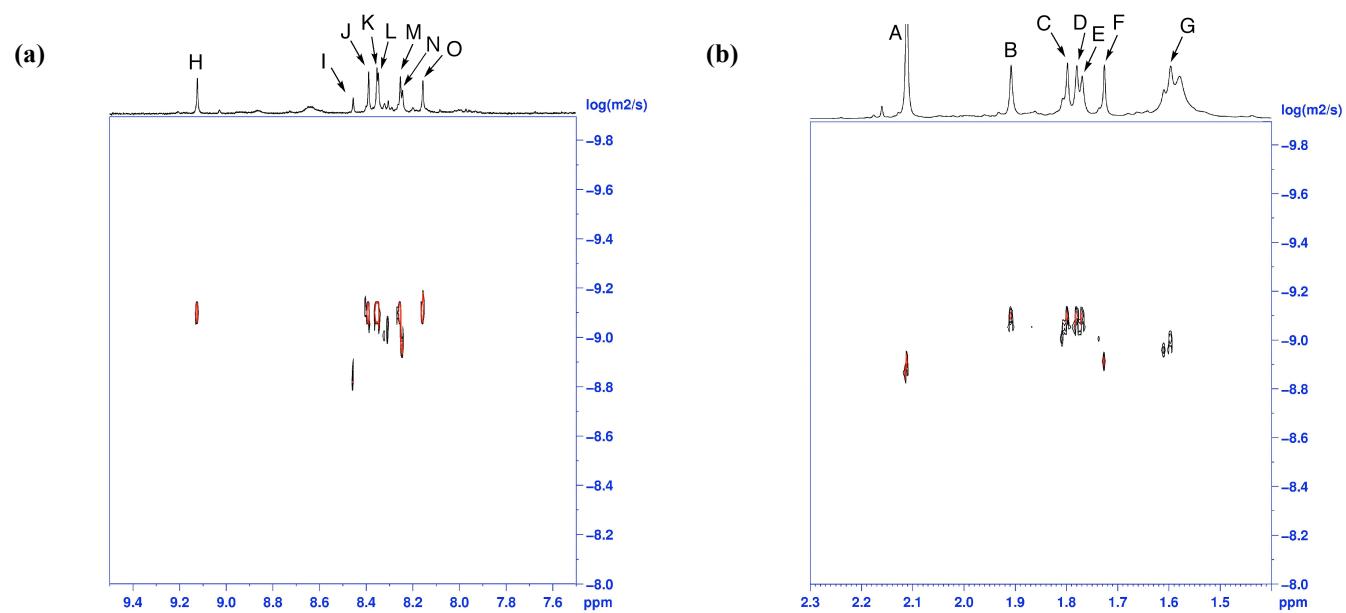


Fig. S1 DOSY spectra of **1** (2.8 mM) at room temperature in CD_2Cl_2 after 48 h since making sample: (a) Low magnetic field; (b) high magnetic field.

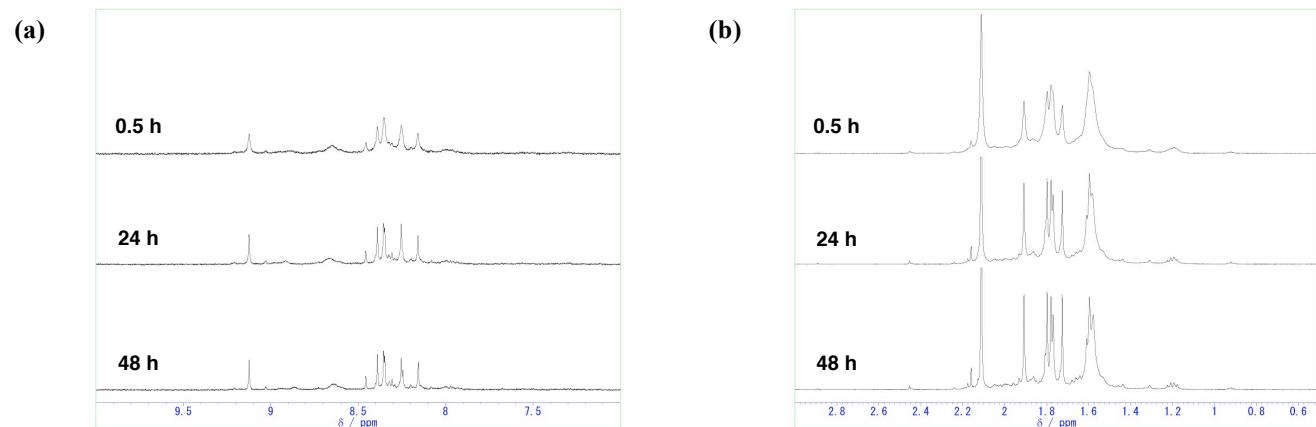


Fig. S2 ^1H NMR spectra of **1** (2.8 mM) at room temperature in CD_2Cl_2 after 0.5 h, 24 h and 48 h: (a) Low magnetic field; (b) high magnetic field.

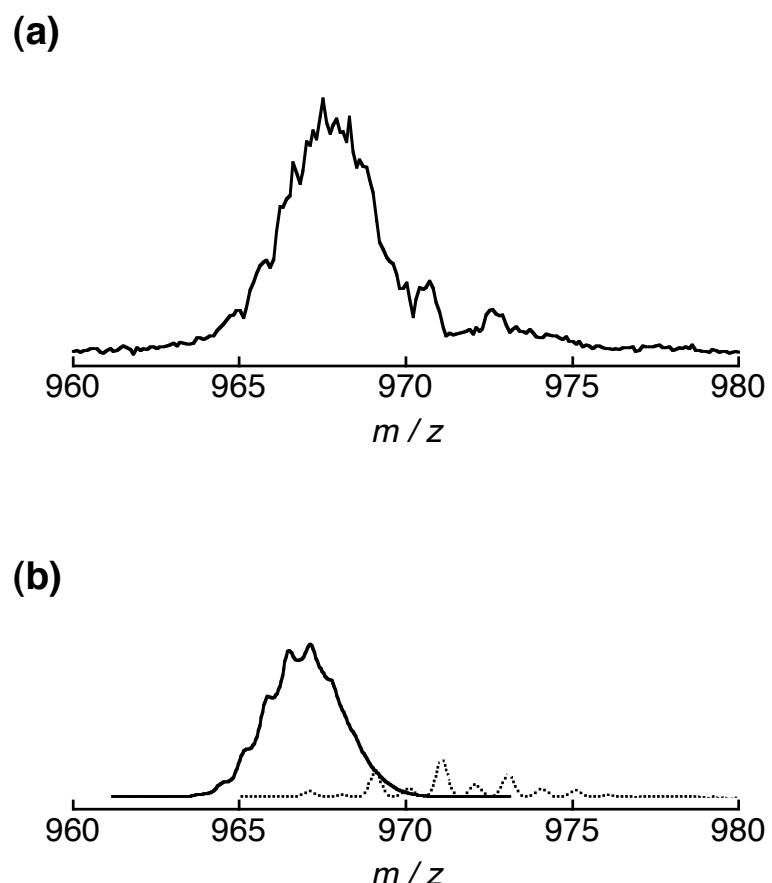


Fig. S3 (a) ESI-MS spectrum of **1** in CH_3OH at room temperature. (b) Calculated mass number of $[\{\text{Ir}(\text{Cp}^*)\}_6(\text{alloCl}_2)_3(\text{CH}_3\text{O})_3]^{3+}$ (black line) and $[\{\text{Ir}(\text{Cp}^*)\}_2(\text{alloCl}_2)\text{Cl}]^+$ (dotted line).

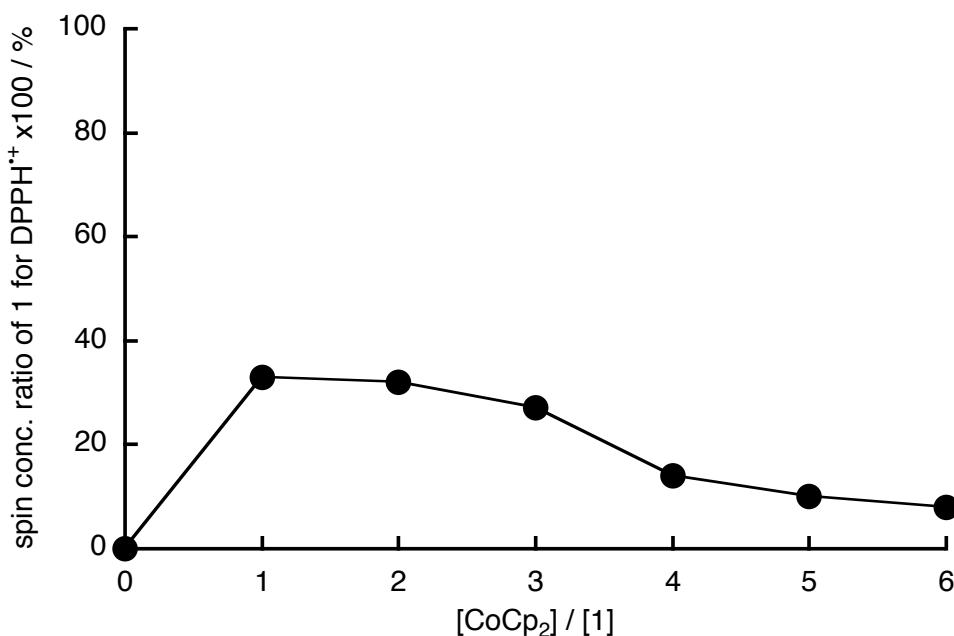


Fig. S4 Change of spin concentration of reduced species of **1** relative to the ratio of $[\text{CoCp}_2]/[\mathbf{1}]$. The spin concentration was represented by the percentage of double integration of the signals due to the reduced species of **1** relative to that of the same concentration of DPPH (2,2-diphenyl-1-picrylhydrazyl).

References

- [1] A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648; C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B*, 1988, **37**, 785.
- [2] W. J. Hehre, L. Radom, P. v. R. Schleyer and J. A. Pople, *Ab Initio Molecular Orbital Theory*, Wiley, New York, 1986.
- [3] M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, J. T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez and J. A. Pople, *Gaussian 03, Revision C.02*, Gaussian, Inc., Wallingford CT, 2004.