

Supporting information for:

Synthesis and Characterization of Tantalum Silsesquioxane Complexes

Pascal Guillo,^{a,b} Meg E. Fasulo,^a Michael I. Lipschutz^a and T. Don Tilley^{*a,b}

^a Department of Chemistry, University of California, Berkeley, CA 94720-1461, USA.

^b Chemical Sciences Division, Lawrence Berkeley National Laboratory, 1 Cyclotron Road, Berkeley, CA 94720, USA.

E-mail: tdtilley@berkeley.edu

Index	Page
General considerations	2
Synthesis of 2	2
Synthesis of 3	3
Synthesis of 4	3
Synthesis of 5	3
¹ H NMR spectrum of 5 in C ₆ D ₆ at 25 °C (Fig. S1)	4
Crystal data and structure refinement for 2 (Table S1)	5
Crystal data and structure refinement for 3 (Table S2)	6
Crystal data and structure refinement for 4 ·C ₆ H ₆ (Table S3)	7
Crystal data and structure refinement for 5 (Table S4)	8
Crystal data and structure refinement for 5 (CH ₃ CN)·CH ₃ CN (Table S5)	9

General considerations

All manipulations of air sensitive compounds were conducted under a nitrogen atmosphere using standard Schlenk techniques or using a nitrogen atmosphere glovebox. Solvents were stored in PTFE valved flasks after drying using Vacuum Atmosphere solvent purification systems or by distillation under nitrogen from appropriate drying agents. C₆D₆ was purchased from Cambridge Isotope Laboratories, dried over Na/K alloy, and then degassed by several freeze-pump-thaw cycles. NMR spectra were recorded on Bruker spectrometers at room temperature unless otherwise noted. Spectra were referenced internally by solvent peaks for ¹H and ¹³C{¹H} NMR, and tetramethylsilane for ²⁹Si-¹H HMBC experiments. X-ray analyses were carried out at UC Berkeley CHEXRAY crystallographic facility. Measurements of **2-5** and **5(CH₃CN)** were made on an APEX CCD area detector with Mo K α radiation (λ = 0.71069 Å) monochromated with QUAZAR multilayer mirrors. Elemental analyses were performed by the College of Chemistry Microanalytical Laboratory at the University of California, Berkeley.

Cp*TaCl₄¹ and LiB(C₆F₅)₄·2Et₂O² were prepared according to literature methods. Incompletely condensed silsesquioxane **1** was purchased from Hybrid Plastics Inc. and dried overnight under vacuum at 50 °C prior to use. All other chemicals were purchased from commercial sources and used without further purification.

Synthesis of **2**

Toluene (20 mL) was added to a solid mixture of **1** (0.300 g, 0.38 mmol) and TaCp*Cl₄ (0.174 g, 0.38 mmol). Stirring at 90 °C for 12 h afforded a slightly yellow solution, which was evaporated under reduced pressure. The residue was redissolved in 1 mL of toluene. Acetonitrile was added dropwise to the yellow solution to precipitate **2** and this mixture was kept at -30 °C for 12 h to achieve complete precipitation. After filtration, washing with cold acetonitrile and drying under vacuum, **2** (0.340 g, 79%) was obtained as a white powder. Crystals suitable for X-ray diffraction were obtained by cooling a saturated solution of **2** in hexanes at -30 °C.

Anal. Calcd for C₃₈H₇₈ClO₁₂Si₇Ta: C, 40.04; H, 6.90. Found: C, 40.25; H, 6.89. ¹H NMR (600 MHz, C₆D₆, 25 °C) δ = 2.23-2.15 (m, 7H), 2.10 (s, 15H), 1.19 (d, 18H, 6.6 Hz), 1.12 (d, 18H, 6.6 Hz), 1.09 (d, 6H, 6.6 Hz), 0.95 (d, 6H, 6.6 Hz), 0.92 (d, 6H, 7.2 Hz), 0.84 (d, 2H, 7.2 Hz). ¹³C{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = 125.40, 25.94, 25.67, 25.57, 24.29, 24.07, 23.02, 22.91, 11.74. ²⁹Si{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -63.3, -66.9.

Synthesis of 3

To 0.200 g (0.175 mmol) of **2** in 5 mL of diethyl ether at -30 °C, was added methyllithium solution (1.6 M in diethyl ether) (132 µL, 0.211 mmol) via syringe. The resulting solution was stirred at room temperature for 2 h. After filtration to remove LiCl, the solution was evaporated under reduced pressure. The residue was dissolved in 2 mL of toluene and filtered through a plug of celite. Acetonitrile was then added dropwise to the yellow solution to precipitate **3** and this mixture was kept at -30 °C for 12 h to achieve complete precipitation. After filtration and drying under vacuum, **3** (0.185 g, 94%) was obtained as a white powder. Crystals suitable for X-ray diffraction were obtained by cooling a saturated solution of **3** in pentane at -30 °C.

Anal. Calcd for C₃₉H₈₁O₁₂Si₇Ta: C, 41.84; H, 7.29. Found: C, 41.54; H, 7.40. ¹H NMR (600 MHz, C₆D₆, 25 °C) δ = 2.22-2.10 (m, 7H), 1.91 (s, 15H), 1.20 (d, 18H, 6.6 Hz), 1.13 (d, 18H, 6.6 Hz), 1.11 (d, 6H, 7.8 Hz), 0.91 (d, 6H, 6.6 Hz), 0.88 (d, 6H, 6.6 Hz), 1.20 (d, 2H, 7.2 Hz), 0.71 (s, 3H). ¹³C{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = 122.51, 53.56, 26.78, 26.47, 26.30, 25.24, 24.91, 24.85, 23.97, 23.68, 11.87. ²⁹Si{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -65.0, -67.3, -67.8.

Synthesis of 4

To 0.100 g (0.088 µmol) of **2** in 3 mL of toluene, was added silver trifluoromethanesulfonate (0.023 g, 0.088 µmol) and the mixture was stirred at room temperature for 2 h. After filtration to remove AgCl, the filtrate was evaporated under vacuum and analytically pure **4** (0.110 g, 100%) was obtained as a white powder. Crystals suitable for X-ray diffraction were obtained as colorless needles after slow evaporation of a solution of **4** in benzene at 20 °C.

Anal. Calcd for C₃₉H₇₈F₃O₁₅SSi₇Ta: C, 37.36; H, 6.27. Found: C, 37.22; H, 6.11. ¹H NMR (600 MHz, C₆D₆, 25 °C) δ = 2.22-2.06 (m, 7H), 2.12 (s, 15H), 1.17 (d, 18H, 6.6 Hz), 1.11 (d, 18H, 7.2 Hz), 1.08 (d, 6 H, 6.6 Hz), 1.02 (d, 6H, 7.2 Hz), 0.97 (d, 7.2 Hz), 0.81 (d, 2H, 6.6 Hz). ¹³C{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = 127.96, 26.54, 26.35, 26.27, 25.12, 24.86, 24.79, 23.50, 23.36, 12.10. ²⁹Si{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -63.1, -66.8, -67.4. ¹⁹F{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -75.6.

Synthesis of 5

To 0.300 g (0.263 mmol) of **2** in 5 mL of dichloromethane, Li[B(C₆F₅)₄]⁻·2Et₂O (0.252 g, 0.302 mmol) was added and the mixture was stirred at room temperature for 3 h. After

filtration to remove LiCl, the filtrate was evaporated under vacuum. The complex **5** was then crystallized from acetonitrile at -30 °C. Crystals of **5**(CH₃CN) suitable for X-ray diffraction were obtained by this method. After removal of the solvent and drying under vacuum, **5** (400 mg, 85%) was obtained as a white powder. Crystals of **5** suitable for X-ray diffraction were obtained by diffusion of pentane into a solution of **5** in toluene at -30 °C.

Anal. Calcd for C₆₂H₇₈BF₂₀O₁₂Si₇Ta: C, 41.75; H, 4.41. Found: C, 41.87; H, 4.74. ¹H NMR (600 MHz, C₆D₆, 25 °C) δ = 1.99-1.90 (m, 7H), 1.89 (s, 15H), 1.04 (d, 18H, 6.6 Hz), 0.98 (m, 24H), 0.79 (d, 6H, 7.2 Hz), 0.73 (d, 6H, 7.2 Hz), 0.71 (d, 2H, 7.2 Hz). ¹³C{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = 129.10, 26.00, 25.94, 24.62, 24.55, 23.07, 22.73, 22.37, 10.83. ²⁹Si{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -68.4, -67.8, -59.7. ¹⁹F{¹H} NMR (600 MHz, C₆D₆, 25 °C) δ = -130.7, -161.9, -165.6.

Fig. S1 ¹H NMR spectrum of **5** in C₆D₆ at 25 °C.

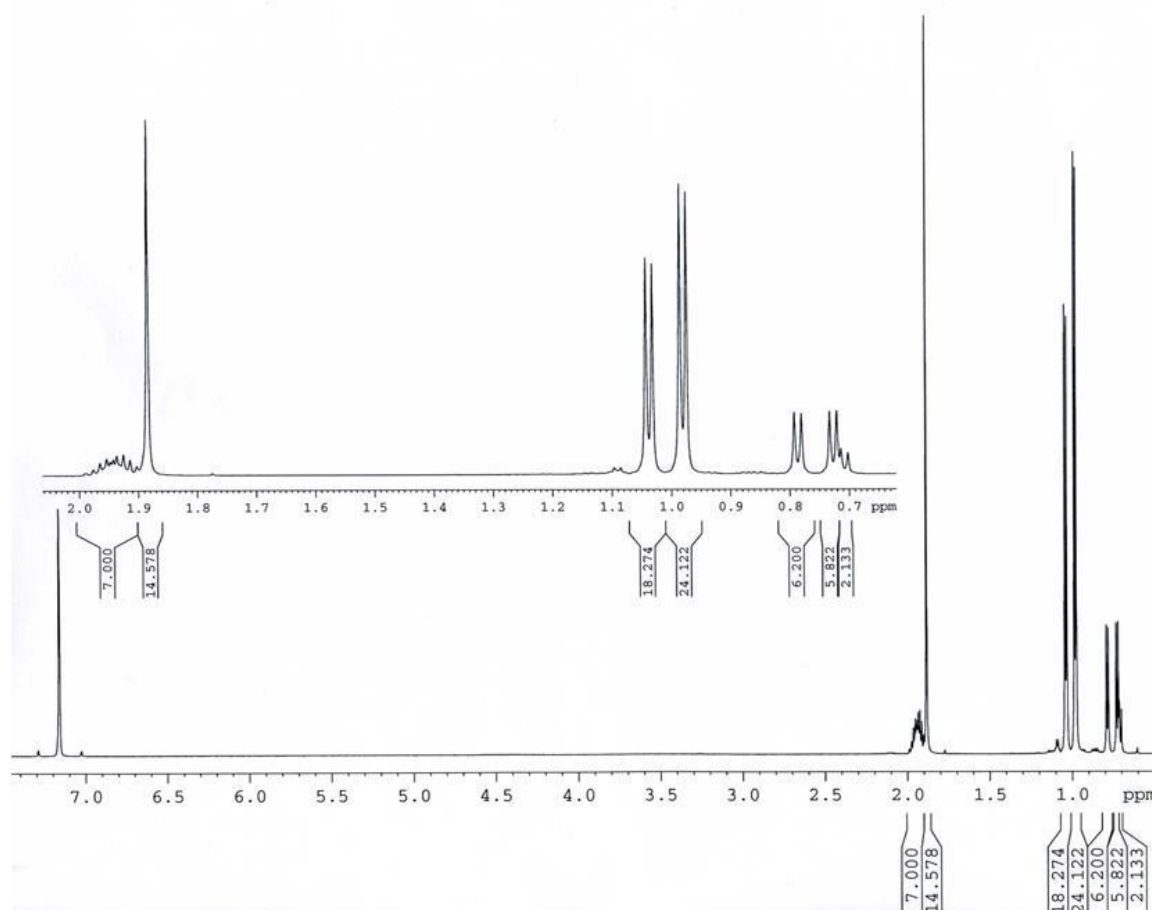


Table S1 Crystal data and structure refinement for **2**

Empirical formula	C38 H78 Cl O12 Si7 Ta		
Formula weight	1140.04		
Temperature	100(2) K		
Wavelength	0.71073 Å		
Crystal system	triclinic		
Space group	P -1		
Unit cell dimensions	a = 14.690 Å	α = 105.25°	
	b = 19.058 Å	β = 107.43°	
	c = 21.339 Å	γ = 100.37°	
Volume	5277.2 Å ³		
Z	4		
Density (calculated)	1.336 Mg/m ³		
Absorption coefficient	2.340 mm ⁻¹		
F(000)	2048		
Crystal size	0.14 x 0.14 x 0.10 mm ³		
Theta range for data collection	1.51 to 25.40°.		
Index ranges	-17<=h<=17, -22<=k<=22, -25<=l<=25		
Reflections collected	59433		
Independent reflections	19091 [R(int) = 0.0229]		
Completeness to theta = 25.40°	98.2 %		
Max. and min. transmission	0.7997 and 0.7353		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	19091 / 0 / 1101		
Goodness-of-fit on F ²	1.137		
Final R indices [I>2sigma(I)]	R1 = 0.0286, wR2 = 0.0895		
R indices (all data)	R1 = 0.0367, wR2 = 0.1117		
Largest diff. peak and hole	1.336 and -1.844 e. Å ⁻³		

Table S2 Crystal data and structure refinement for **3**

Empirical formula	C ₃₉ H ₈₁ O ₁₂ Si ₇ Ta	
Formula weight	1119.62	
Temperature	100(2) K	
Wavelength	0.71073 Å	
Crystal system	Triclinic	
Space group	P -1	
Unit cell dimensions	a = 10.4766(7) Å b = 11.7685(7) Å c = 22.1260(15) Å	$\alpha = 78.832(3)^\circ$ $\beta = 79.710(3)^\circ$ $\gamma = 79.243(3)^\circ$
Volume	2600.4(3) Å ³	
Z	2	
Density (calculated)	1.430 Mg/m ³	
Absorption coefficient	2.327 mm ⁻¹	
F(000)	1164	
Crystal size	0.80 x 0.10 x 0.10 mm ³	
Theta range for data collection	1.79 to 25.41°.	
Index ranges	-12 ≤ h ≤ 12, -14 ≤ k ≤ 14, -26 ≤ l ≤ 26	
Reflections collected	67291	
Independent reflections	9559 [R(int) = 0.0326]	
Completeness to theta = 25.41°	99.5 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.8024 and 0.2603	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	9559 / 0 / 552	
Goodness-of-fit on F ²	1.066	
Final R indices [I > 2σ(I)]	R1 = 0.0231, wR2 = 0.0559	
R indices (all data)	R1 = 0.0249, wR2 = 0.0575	
Largest diff. peak and hole	1.608 and -1.660 e.Å ⁻³	

Table S3 Crystal data and structure refinement for **4·C₆H₆**

Empirical formula	C ₄₅ H ₈₄ F ₃ O ₁₅ S Si ₇ Ta	
Formula weight	1331.78	
Temperature	100(2) K	
Wavelength	0.71069 Å	
Crystal system	Monoclinic	
Space group	P 2 ₁ /n	
Unit cell dimensions	a = 13.252(5) Å	α = 90.000(5)°
	b = 10.146(5) Å	β = 93.033(5)°
	c = 44.599(5) Å	γ = 90.000(5)°
Volume	5988(4) Å ³	
Z	4	
Density (calculated)	1.477 Mg/m ³	
Absorption coefficient	2.078 mm ⁻¹	
F(000)	2752	
Crystal size	0.10 x 0.03 x 0.03 mm ³	
Theta range for data collection	1.58 to 25.71°	
Index ranges	-16 ≤ h ≤ 13, -9 ≤ k ≤ 12, -53 ≤ l ≤ 54	
Reflections collected	40832	
Independent reflections	11160 [R(int) = 0.0660]	
Completeness to theta = 25.71°	97.7 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9394 and 0.8167	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	11160 / 12 / 668	
Goodness-of-fit on F ²	1.330	
Final R indices [I > 2σ(I)]	R1 = 0.0905, wR2 = 0.1828	
R indices (all data)	R1 = 0.1167, wR2 = 0.1898	
Largest diff. peak and hole	2.594 and -6.870 e.Å ⁻³	

Table S4 Crystal data and structure refinement for 5

Empirical formula	C62 H78 B F20 O12 Si7 Ta		
Formula weight	1782.12		
Temperature	100(2) K		
Wavelength	0.71073Å		
Crystal system	Triclinic		
Space group	P -1		
Unit cell dimensions	a = 13.3982(17)Å	α= 97.778(7)°.	
	b = 29.185(4)Å	β= 90.888(8)°.	
	c = 40.772(5)Å	γ = 92.306(8)°.	
Volume	15780(3)Å ³		
Z	8		
Density (calculated)	1.500 Mg/m ³		
Absorption coefficient	1.600 mm ⁻¹		
F(000)	7204		
Crystal size	0.14 x 0.08 x 0.02 mm ³		
Theta range for data collection	1.41 to 31.75 °.		
Index ranges	-19<=h<=17, -34<=k<=39, -56<=l<=56		
Reflections collected	307718		
Independent reflections	84272 [R(int) = 0.0702]		
Completeness to theta = 25.00°	100 %		
Absorption correction	None		
Max. and min. transmission	0.9687 and 0.8070		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	84272 / 6 / 3785		
Goodness-of-fit on F ²	1.067		
Final R indices [I>2sigma(I)]	R1 = 0.0693, wR2 = 0.1650		
R indices (all data)	R1 = 0.1111, wR2 = 0.1800		
Largest diff. peak and hole	5.407 and -2.960 e.Å ⁻³		

Table S5 Crystal data and structure refinement for 5(CH₃CN)·CH₃CN

Empirical formula	C ₆₆ H ₈₄ B F ₂₀ N ₂ O ₁₂ Si ₇ Ta	
Formula weight	1865.75	
Temperature	100(2) K	
Wavelength	0.71069 Å	
Crystal system	Triclinic	
Space group	P -1	
Unit cell dimensions	a = 14.324(5) Å	α = 102.447(5)°
	b = 14.832(5) Å	β = 90.694(5)°
	c = 19.356(5) Å	γ = 94.252(5)°
Volume	4003(2) Å ³	
Z	2	
Density (calculated)	1.548 Mg/m ³	
Absorption coefficient	1.581 mm ⁻¹	
F(000)	1892	
Crystal size	0.05 x 0.05 x 0.05 mm ³	
Theta range for data collection	1.41 to 23.30°.	
Index ranges	-15 ≤ h ≤ 15, -16 ≤ k ≤ 16, -21 ≤ l ≤ 21	
Reflections collected	26149	
Independent reflections	11281 [R(int) = 0.0539]	
Completeness to theta = 25.27°	95.8 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9253 and 0.9253	
Refinement method	Full-matrix least-squares on F ²	
Data / restraints / parameters	13907 / 30 / 1003	
Goodness-of-fit on F ²	1.072	
Final R indices [I > 2σ(I)]	R1 = 0.0830, wR2 = 0.2084	
R indices (all data)	R1 = 0.1189, wR2 = 0.2439	
Largest diff. peak and hole	4.209 and -2.517 e.Å ⁻³	

References

- ¹ R. D. Sanner, S. T. Carter and W. J. Bruton Jr., *J. Organomet. Chem.* **1982**, 240, 157.
- ² J. B. Lambert, S. Zhang and S. M. Ciro, *Organometallics* **1994**, 13, 2430.