

Supporting Information

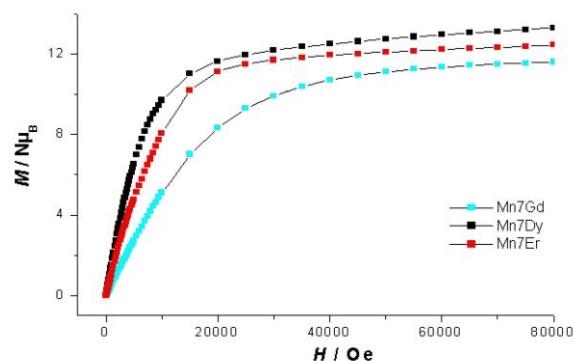


Figure S1. Plots of $M/N\mu_B$ vs H for complexes **1**- **3** at 2K. Solid lines are eye guides.

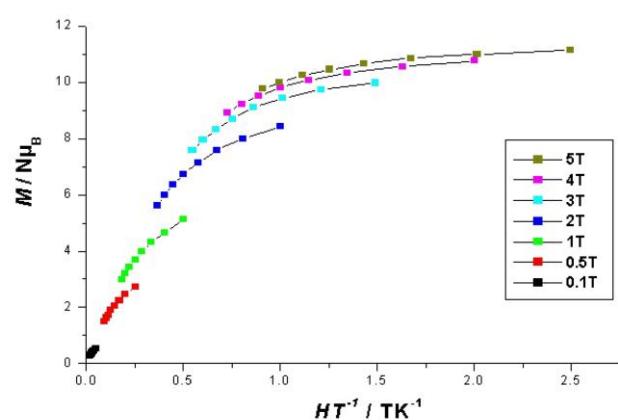


Figure S2. Plots of $M/N\mu_B$ vs H/T for complex **1** (Mn₇Gd) at the indicated applied fields. Solid lines are eye guides.

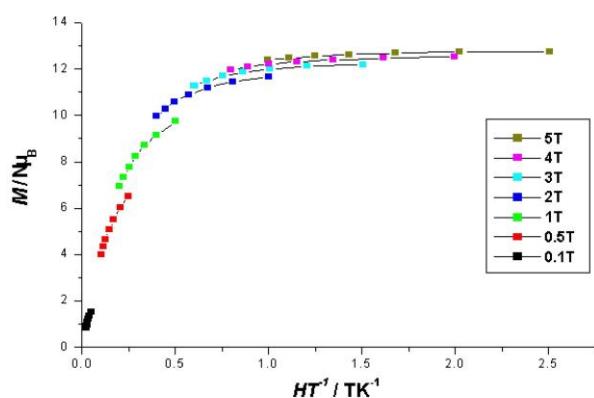


Figure S3. Plots of $M/N\mu_B$ vs H/T for complex **2** (Mn₇Dy) at the indicated applied fields. Solid lines are eye guides.

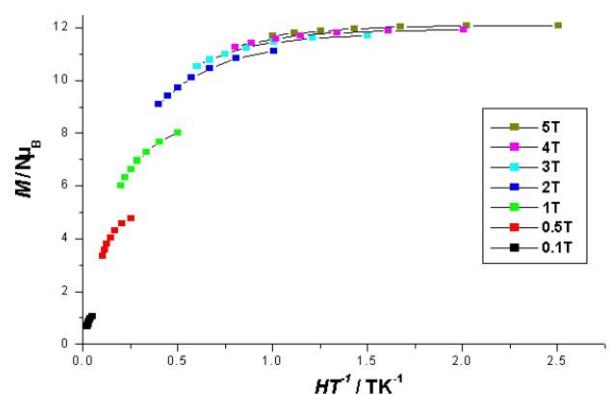


Figure S4. Plots of $M/N\mu_B$ vs H/T for complex **3** (Mn_7Er) at the indicated applied fields. Solid lines are eye guides.

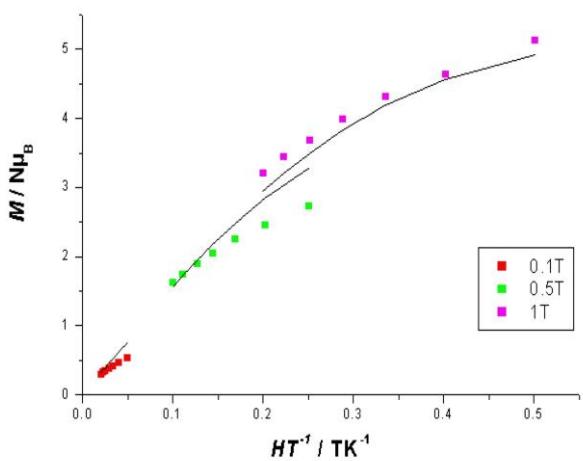


Figure S5. Plots of $M/N\mu_B$ vs H/T for complex **1** (Mn_7Gd) at the indicated applied fields. Solid lines represent the best fits of the data.

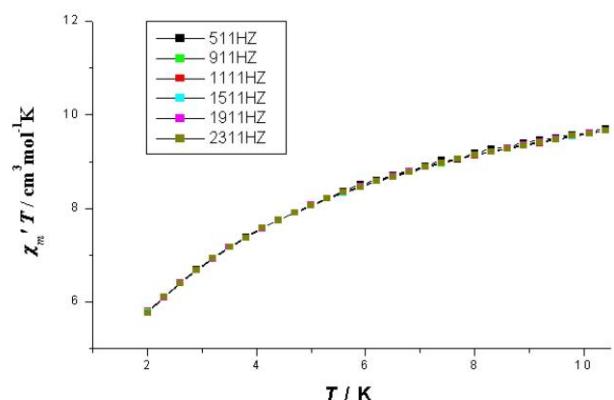


Figure S6. In-phase (χ'_m) (as $\chi'_m T$) ac susceptibility signals of complex **1** in a 3.0 G field oscillating at the indicated frequencies.

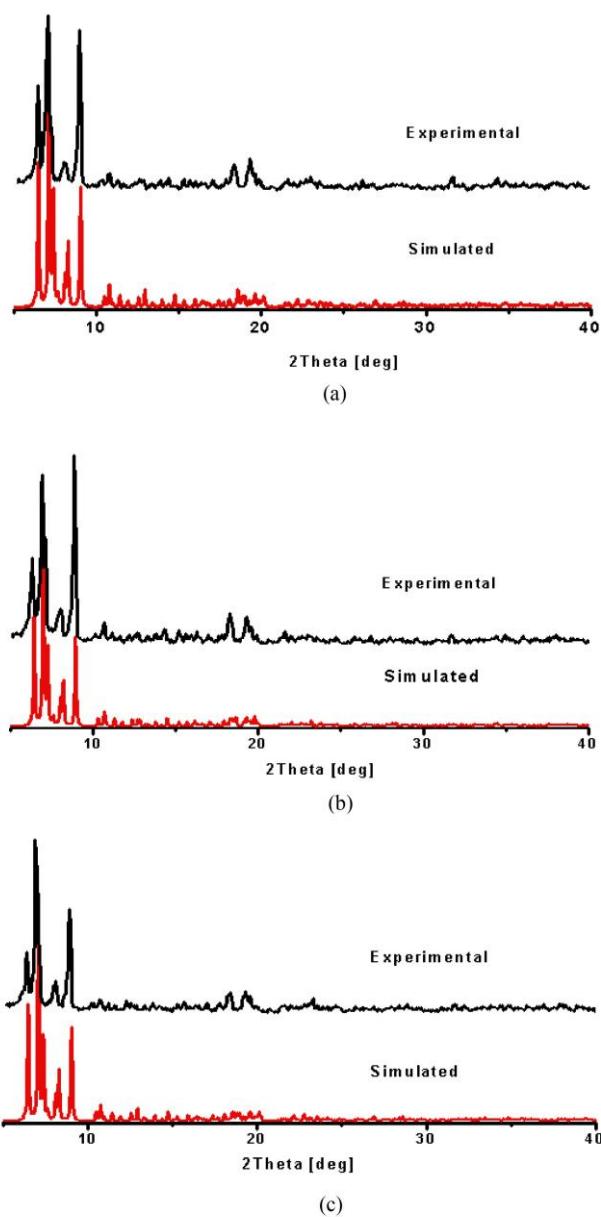


Figure S7. XRPD patters for complexes 1-3

Table S1. Bond-Valence Sums for the O Atoms of Complex 3^a.

atom	BVS	assignment	group	atom	BVS	assignment	group
O1	1.82	RO ⁻	Piv ⁻	O16	1.81	RO ⁻	Piv ⁻
O2	1.68	RO ⁻	Piv ⁻	O17	1.69	RO ⁻	Mdea ²⁻
O3	1.8	RO ⁻	Piv ⁻	O18	1.86	RO ⁻	Mdea ²⁻
O4	1.71	RO ⁻	Piv ⁻	O19	1.94	RO ⁻	Piv ⁻
O5	1.93	RO ⁻	Mdea ²⁻	O20	1.83	RO ⁻	Piv ⁻
O6	1.98	RO ⁻	Mdea ²⁻	O21	2.24	RO ⁻	Mdea ²⁻
O7	1.93	RO ⁻	Piv ⁻	O22	2.23	RO ⁻	Mdea ²⁻
O8	1.49	RO ⁻	Piv ⁻	O23	1.96	RO ⁻	Piv ⁻

O9	1.9	RO ⁻	Piv ⁻	O24	1.75	RO ⁻	Piv ⁻
O10	1.93	RO ⁻	Piv ⁻	O25	1.11	OH	OH
O11	1.85	RO ⁻	Piv ⁻	O26	0.98	OH	OH
O12	1.88	RO ⁻	Piv ⁻	O27	2.02	O ²⁻	O ²⁻
O13	1.87	RO ⁻	MdeaH ⁻	O28	0.99	OH	OH
O14	1.23	ROH	MdeaH ⁻	O29	2.03	O ²⁻	O ²⁻
O15	2.06	RO ⁻	Piv ⁻				

^a BVS values for O atoms of RO⁻ and ROH groups are typically 1.8–2.0 and 1.0–1.2, respectively, but can be affected by hydrogen bonding.

Table S2. Bond-Valence Sums for the Mn^a Atoms of Complex 3.

atom	Mn ²⁺	Mn ³⁺	Mn ⁴⁺
Mn1	3.2	<u>2.95</u>	2.92
Mn2	3.38	<u>3.12</u>	3.06
Mn3	<u>1.91</u>	1.76	1.73
Mn4	3.40	<u>3.14</u>	3.08
Mn5	3.38	<u>3.11</u>	3.06
Mn6	3.23	<u>2.98</u>	2.92
Mn7	3.34	<u>3.08</u>	3.02

^a The underlined value is the closest to the charge for which it was calculated, and the nearest whole number can be taken as the oxidation state of the atom.