

Electronic Supplementary Material

**A colorimetric detection of Pb^{2+} by using sodium thiosulfate and hexadecyl thimethyl ammonium
bromide modified gold nanoparticles**

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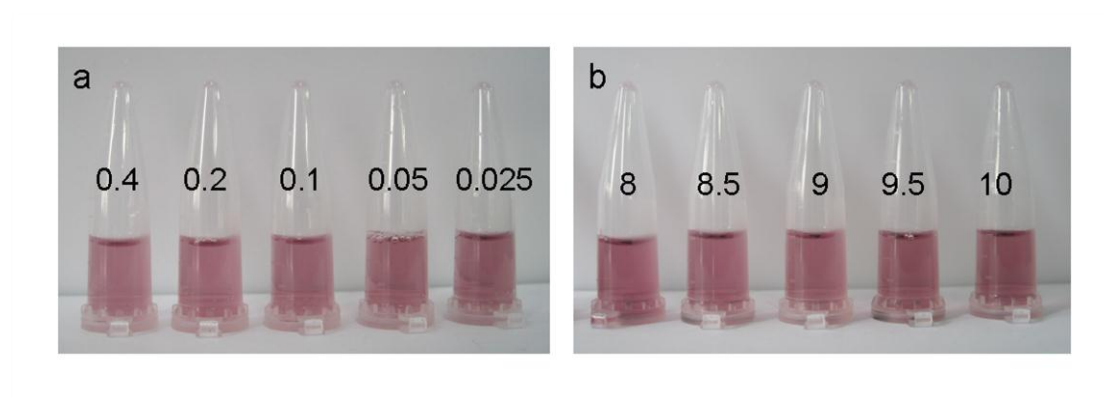


Fig. S1 (a) Effect of the concentration of $\text{Na}_2\text{S}_2\text{O}_3$ on the color of Au NPs solutions in the presence of $10\ \mu\text{M}\ \text{Pb}^{2+}$ ($[\text{Na}_2\text{S}_2\text{O}_3]=0.4, 0.2, 0.1, 0.05, 0.025\ \text{M}$) (b) Effect of pH on the color of Au NPs solutions in the presence of $3\ \mu\text{M}\ \text{Pb}^{2+}$ (pH=8, 8.5, 9, 9.5, 10).

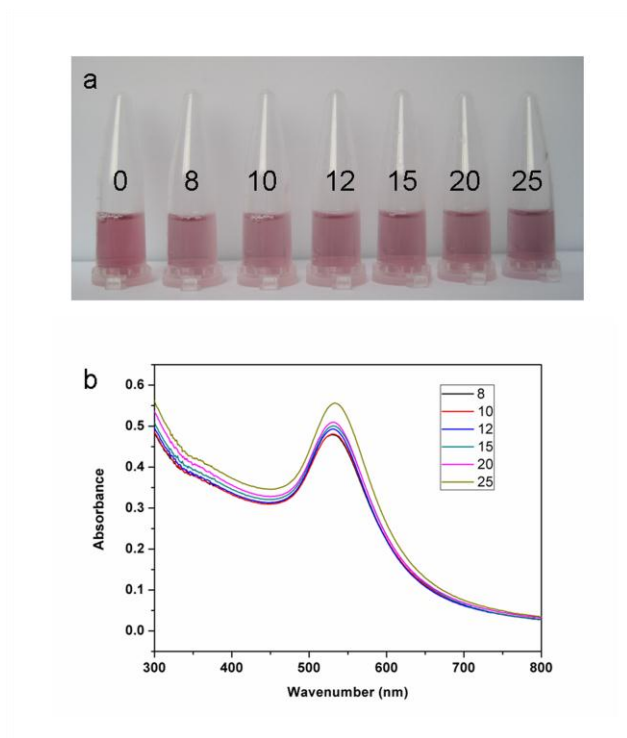


Fig. S2 (a) The corresponding photo images (b) UV-vis absorption spectra of samples with different concentrations of Pb^{2+} (8, 10, 12, 15, 20, 25 μM).

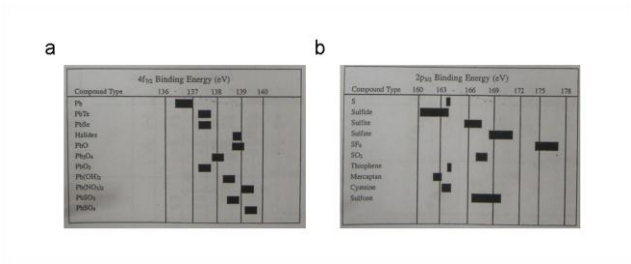


Fig. S3 XPS data of Pb (a) and S (b).

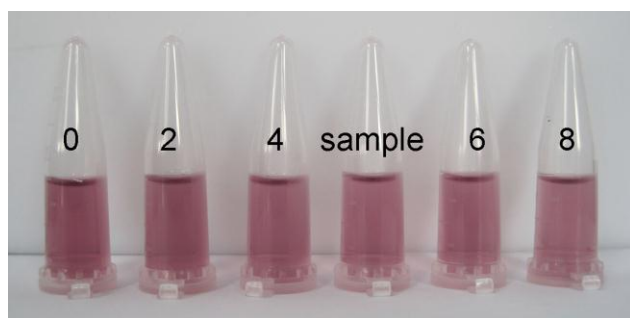


Fig. S4 The photo images of CTAB-Au solutions in the presence of Pb^{2+} with different concentrations (blank, 2 μM , 4 μM , sample, 6 μM , 8 μM).