Supporting Information

Selective recognition of sulphate in Cu(II) assisted 1D polymer of a simple pentafluorophenyl substituted pyridyl-urea via second sphere coordination

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Figure 1S. Single crystal X-ray structure of L^1

Parameters	L
Empirical formula	C12 H6 F5 N3 O
Formula weight	303.20
Crystal system	Orthorhombic
Space group	Pbca
<i>a</i> (Å)	12.7261(15)
b (Å)	10.2498(12)
c (Å)	18.184(2)
a (deg)	90.00
β (deg)	90.00
γ (deg)	90.00
$V(\dot{A}^3)$	2371.9(5)
Ζ	8
$d_{\rm calc}$ (g/cm ³)	1.765
Crystal size (mm ³)	0.40x0.20x0.06
Diffractometer	Smart CCD
<i>F</i> (000)	1216
μ MoKa (mm ⁻¹)	0.165
<i>T</i> (K)	150(2)
θ max	27.31
Observed Reflections	2654
Parameters refined	190
$\mathbf{R}_1; \mathbf{W}\mathbf{R}_2$	0.0407; 0.1558
GOF (F2)	1.038

Table 1S. Crystallographic table for L^1

Parameters	Complex 1
Empirical formula	C54 H38 Cu F20 N14 O10 S
Formula weight	1518.58
Crystal system	Triclinic
Space group	P-1
a (Å)	11.7823(17)
b (Å)	14.103(2)
<i>c</i> (Å)	23.069(3)
a (deg)	95.942(3)
β (deg)	104.696(3)
γ (deg)	111.866(3)
V (Å ³)	3356.3(8)
Ζ	2
$d_{\rm calc}$ (g/cm ³)	1.503
Crystal size (mm ³)	1.00x0.08x0.06
Diffractometer	Smart CCD
F(000)	1530
μ MoKa (mm ⁻¹)	0.476
<i>T</i> (K)	150(2)
θ max	25.66
Observed Reflections	12616
Parameters refined	878
$\mathbf{R}_1; \mathbf{W} \mathbf{R}_2$	0.0917; 0.2477
GOF (F2)	1.162

Table 2S. Crystallographic table for complex 1

D-HA	d(H-A) Å	d(D-A) Å	<dha (□)<="" th=""></dha>
N5-H5 O1	2.11	2.93	157
N6-H6 O3	2.13	2.97	166
N7-H7 O1	2.36	2.97	128
N7-H7 O2	2.49	3.24	146
N8-H8 O2	1.99	2.83	167
N9-H9 O3	2.16	2.87	141
N10-H10 O1	2.04	2.88	168
N11-H11 O4	1.90	2.76	178
N12-H12 O3	2.15	2.93	152
C3-H3 O10	2.37	3.07	132
C27-H27 O8	2.49	3.14	127
C38-H38 F16	2.43	3.20	140
C2-H2 F14	2.54	3.24	133
C45-H45AF18	2.47	3.29	143

Table 3S. Hydrogen bonding table for complex 1

Symmetry codes: 1+x, y, z; x,1+y,z; x,-1+y,z; 2-x,1-y,1-z



Figure 2S. C-F^{...}H-C interactions in complex 1

Table 4S. Metal-Ligand bond distances in complex 1

Atoms	Distance (Å)
Cu1 -O5	2.5027
Cu1 -N1	2.0536
Cu1 -N2	2.0458
Cu2 -N3	2.0538
Cu2 -N4	2.0536
Cu1 -O5a	2.5027
Cu1 -N1a	2.0536
Cu1 -N2a	2.0458
Cu2 -N4b	2.0536
Cu2 -N3b	2.0538

Parameters	Complex 2
Empirical formula	C48 H32 Cu F20 N14 O14
Formula weight	1472.42
Crystal system	Monoclinic
Space group	C2/c
a (Å)	32.1757(18)
b (Å)	8.1355(5)
c (Å)	25.9525(15)
a (deg)	90.00
β (deg)	119.3470(10)
γ (deg)	90.00
$V(\dot{A}^3)$	5921.6(6)
Ζ	4
d_{calc} (g/cm ³)	1.652
Crystal size (mm ³)	0.08x0.06x0.04
Diffractometer	Smart CCD
<i>F</i> (000)	2956.0
μ MoK α (mm ⁻¹)	0.507
<i>T</i> (K)	150
θ max	26.880
Observed Reflections	6334
Parameters refined	447
$\mathbf{R}_1; \mathbf{W} \mathbf{R}_2$	0.0411; 0.1178
GOF (F2)	1.012

Table 5S. Crystallographic table for complex 2

D-H A	d(H-A) Å	d(D-A) Å	<dha< b=""> (□)</dha<>
N1-H1O6	2.42	3.020(4)	128
N2-H2O7	2.10	2.903(3)	156
N5-H5O5	2.14	2.960(3)	158
N6-H6O6	2.08	2.923(4)	169
O13-H13O11	2.06	2.779(3)	149
O14-H14AO11	2.09	2.824(3)	151
С9-Н9О7	2.54	3.315(3)	141
C14-H14O5	2.57	3.332(4)	140
C16-H16F4	2.53	3.109(4)	120

Table 6S.	. Hydrogen	bonding table	for complex 2
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Symmetry codes: 1/2-x,-1/2+y,1/2-z; -x,-1+y,1/2-z; -x,2-y,-z

Figure 3S.	Scattered	plot of	complex	2
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Atoms	Distance (Å)
Cu1 -O13	2.2809(1)
Cu1 -O14	2.4167(1)
Cu1 -N3	2.0554
Cu1 -N4	2.0499
Cu1 -N3a	2.0554
Cu1 -N4a	2.0499

Table 7S. Metal-Ligand bond distances in complex $\mathbf{2}$

Parameters	Complex 3
Empirical formula	C48 H48 Co F15 N13 O12 S
Formula weight	1374.98
crystal system	Monoclinic
Space group	P2(1)/c
a (Å)	17.833(2)
b (Å)	12.8082(17)
<i>c</i> (Å)	26.372(3)
a (deg)	90.00
β (deg)	105.751(3)
γ (deg)	90.00
V(Å ³)	5797.4(13)
Z	4
d_{calc} (g/cm ³)	1.575
Crystal size (mm ³)	1.00x0.08x0.06
Diffractometer	Smart CCD
F(000)	2804
μ MoKa (mm ⁻¹)	0.451
<i>T</i> (K)	150(2)
θ max	25.00
Observed Reflections	10204
Parameters refined	827
$R_1; WR_2$	0.0349; 0.0927
GOF (F2)	0.985

Table 8S. Crystallographic table for complex 3

D-HA	d(H-A) Å	d(D-A) Å	<dha (□)<="" th=""></dha>
N7-H7O8	2.04	2.840(2)	155
N8-H8O9	1.99	2.807(2)	158
N9-H9O7	2.18	2.897(2)	141
N10-H10O7	2.17	2.946(2)	150
N10-H10O9	2.36	3.110(2)	146
N11-H1107	1.98	2.811(2)	163
N12-H12O11	2.03	2.826(2)	153
C1-H1F9	2.48	3.384(2)	164
C2-H2O8	2.49	3.236(3)	138
C14-H14F5	2.53	3.184(3)	127
С33-Н33О6	2.57	3.316(2)	137
C38-H38CO11	2.57	3.473(3)	158
C44-H44CO3	2.59	3.489(3)	156

Table 9S. Hydrogen bonding table for complex 3

Symmetry codes: x,-1+y,z; 1-x,1-y,-z; -x,1-y,-z; -x,2-y,-z; x,1+y,z; -1+x,y,z; 1-x,-1/2+y,1/2-

Z

Parameters	Complex 4		
Empirical formula	C48 H48 F15 N13 Ni O12 S		
Formula weight	1374.76		
crystal system	Monoclinic		
Space group	P2(1)/c		
a (Å)	17.8094(11)		
b (Å)	12.7781(8)		
<i>c</i> (Å)	26.5467(17)		
a (deg)	90.00		
β (deg)	105.688(2)		
γ (deg)	90.00		
$V(\dot{A}^3)$	5816.2(6)		
Z	4		
d_{calc} (g/cm ³)	1.570		
Crystal size (mm ³)	0.06x0.04x0.02		
Diffractometer	Smart CCD		
F(000)	2808		
μ MoKa (mm ⁻¹)	0.488		
<i>T</i> (K)	150(2)		
θ max	22.41		
Observed Reflections	7513		
Parameters refined	809		
$\mathbf{R}_1; \mathbf{W} \mathbf{R}_2$	0.0522; 0.1518		
GOF (F2)	1.045		

Table 10S. Crystallographic table for complex 4

D-HA	d(H-A) Å	d(D-A) Å	<dha (□)<="" th=""></dha>
N(4)H(4)O(4)	2.06	2.8657	155
N(5)H(5A)O(2)	1.99	2.8104	160
N(6)H(6)O(3)	1.97	2.8089	164
N(7)H(7)O(11)	2.05	2.8514	155
N(8)H(8)O(3)	2.21	2.9403	143
N(9)H(9)O(2)	2.33	3.0875	147
N(9)H(9)O(3)	2.22	2.9930	149
C(3)H(3)O(4)	2.53	3.2764	138
C(26)H(26)F(10)	2.52	3.1771	128

Table 11S. Hydrogen bonding table for complex 4

Symmetry codes:	1-x,-y,-z;	-x,-y,-z; -	1+x,y,z;	-x,1-y,-z; x,	l+y,z
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Parameters	Complex 5		
Empirical formula	C27 H31 Cu N7 O9 S		
Formula weight	693.19		
crystal system	Monoclinic		
Space group	C2/c		
<i>a</i> (Å)	32.699(5)		
b (Å)	10.4948(17)		
c (Å)	23.118(4)		
a (deg)	90.00		
β (deg)	129.583(2)		
γ (deg)	90.00		
$V(\dot{A}^3)$	6114.2(16)		
Z	8		
d_{calc} (g/cm ³)	1.506		
Crystal size (mm ³)	1.00x0.08x0.06		
Diffractometer	Smart CCD		
F(000)	2872		
μ MoKa (mm ⁻¹)	0.847		
<i>T</i> (K)	150(2)		
θ max	25.00		
Observed Reflections	5388		
Parameters refined	410		
$\mathbf{R}_1; \mathbf{W} \mathbf{R}_2$	0.0403; 0.1075		
GOF (F2)	S = 1.041		

Table 12SCrystallographic table for complex 5

D-HA	d(H-A) Å	d(D-A) Å	<dha (□)<="" th=""></dha>
N3-H3O5	2.07	2.771(5)	139
N4-H4O5	2.20	2.922(4)	141
N5-H5O4	2.08	2.931(3)	171
N6-H6O7	2.13	2.966(4)	165
O8-H8AO2	1.97	2.736(3)	155
C3-H3AO1	2.42	3.251(5)	148
C4-H4AO7	2.49	3.276(4)	142

Table 13S. Hydrogen bonding table for complex 5

Symmetry codes: -x,y,1/2-z; 1/2-x,3/2-y,1-z; x,-1+y,z

Figure 4S. Comparison of PXRD pattern of complex **3**: (Blue) simulated and (Red) experimental



Figure 5S. Comparison of PXRD pattern of complex **4**: (Blue) simulated and (Red) experimental



Figure 6S. Comparison of PXRD pattern of complex 5: (Blue) simulated and (Red) experimental







Figure 8S. PXRD pattern of CuBr₂ complex of L¹







Figure 10S. ¹H-NMR spectrum of L¹



Figure 11S. ¹³C-NMR spectrum of L¹







Figure 138^{.1}H-NMR spectrum of L^2



Figure 14S. ¹³C-NMR spectrum of L²





Figure 15S. FT-IR spectrum of complex 1



Figure 16S. FT-IR spectrum of complex 2



Figure 17S. FT-IR spectrum of complex 3

Peak at 1113 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .



Figure 18S. FT-IR spectrum of complex 4



Figure 19S. FT-IR spectrum of complex 5

Star sign at 1097 cm⁻¹ (symmetric stretching frequency of SO_4^{2-}) indicate the presence of SO_4^{2-} .



Figure 20S. FT-IR spectrum of $CuBr_2$ complex of L^1



Figure 21S. FT-IR spectrum of $CuCl_2$ complex of L^1