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Bimetallic core-shell nanocomposites using weak reducing agent and their transformation to alloy nanostructures

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Figure S1. Characterization of Ag nanoparticles prepared using AgNO₃ as metal precursor and *dry* THF as solvent: a) UV-visible spectrum; b) TEM BF image; c) SAED pattern; d) powder XRD pattern; e) XPS spectrum

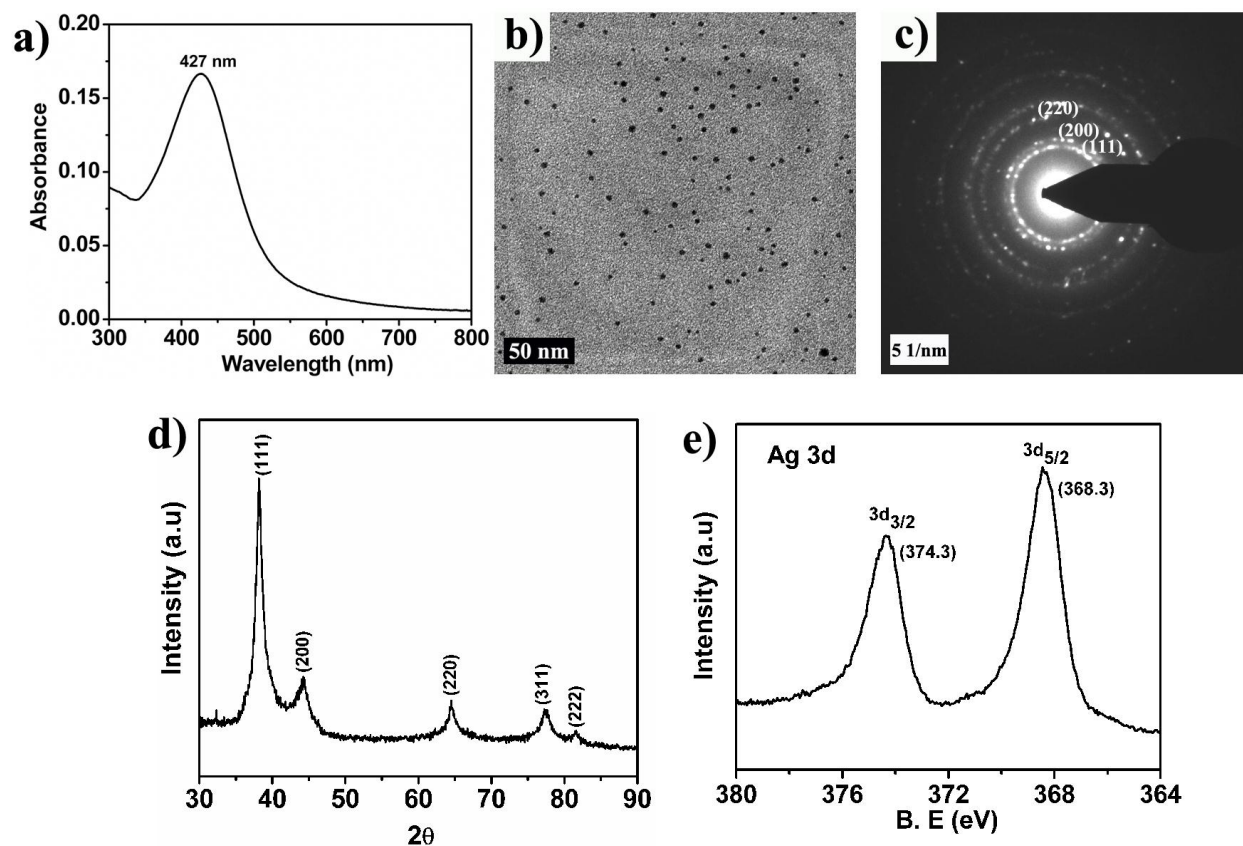


Figure S2. Characterization of Ag nanoparticles prepared using AgNO_3 as metal precursor and *wet* THF as solvent: a) UV-visible spectrum; b) TEM BF image; c) SAED pattern

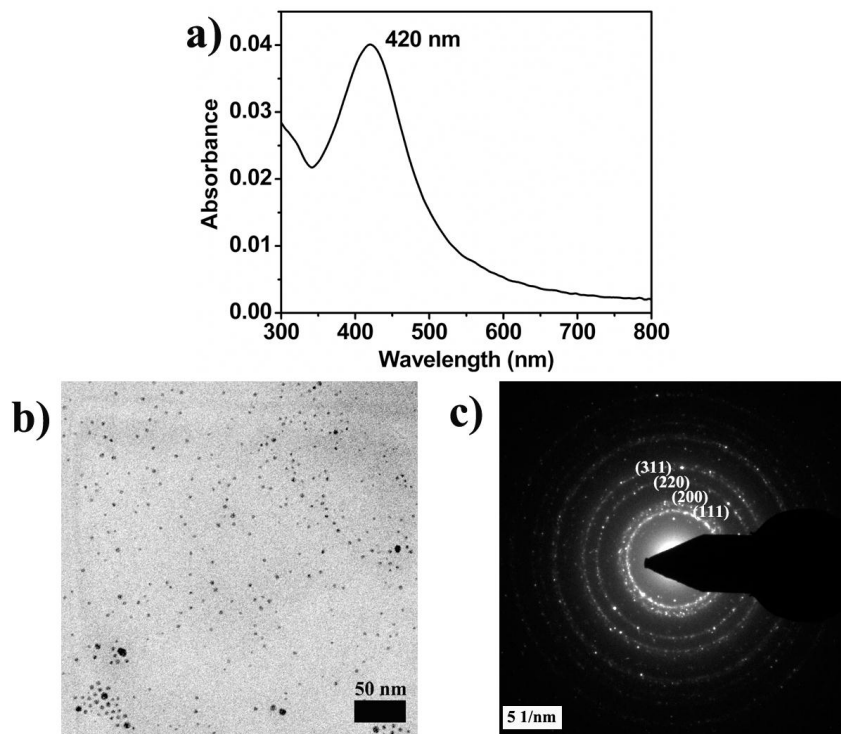


Figure S3. Characterization of Ag nanoparticles prepared using AgOCOCH_3 as metal precursor and *dry* THF as solvent: a) UV-visible spectrum; b) TEM BF image; c) HRTEM image

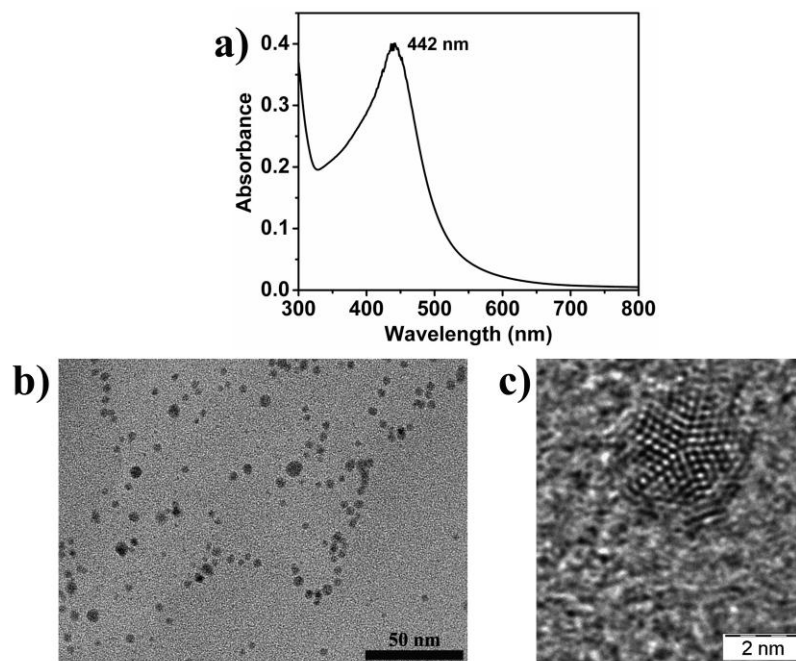


Figure S4. XPS spectrum of Au(I) intermediate (binding energy value $4f_{7/2}$ 85.2 and $4f_{5/2}$ 88.9 eV)

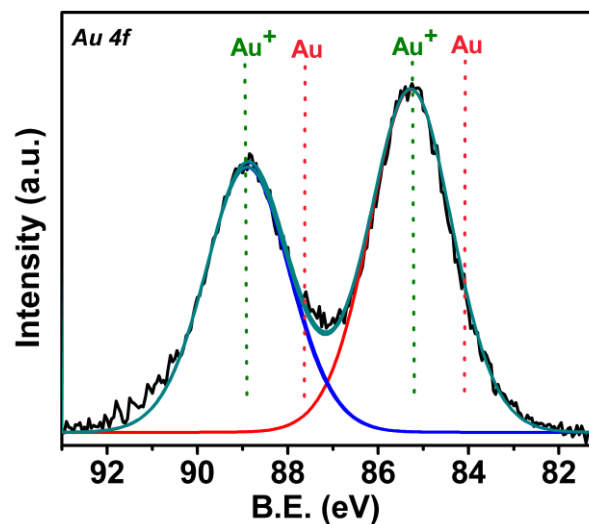


Figure S5. Characterization of Au nanoparticles: a) UV-visible spectrum; b) TEM BF image; c) SAED pattern; d) powder XRD pattern; e) XPS spectrum

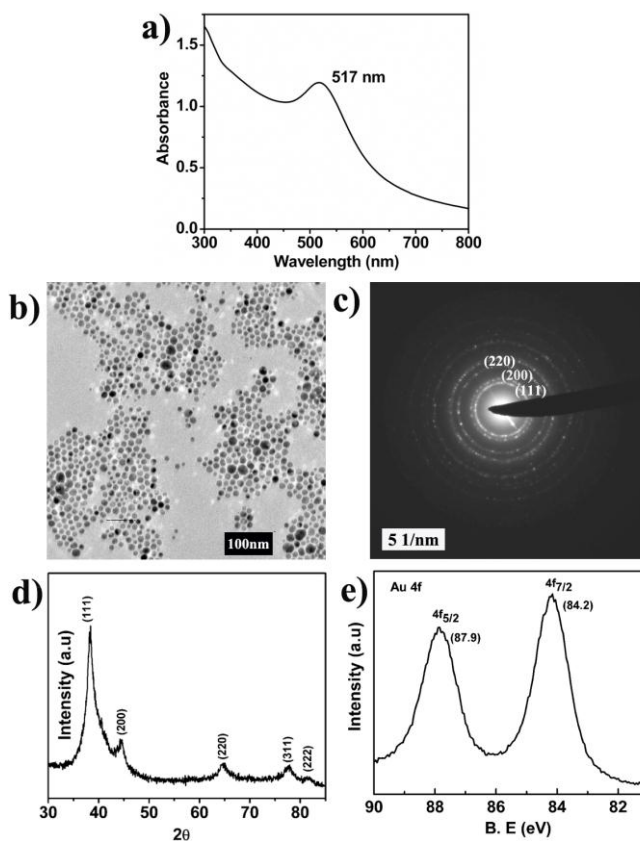


Figure S6. Characterization of Pd nanoparticles: a) UV-visible spectrum; b) TEM BF image; c) HRTEM image; d) FFT pattern generated from Fig S5c; e) powder XRD pattern; f) XPS spectrum

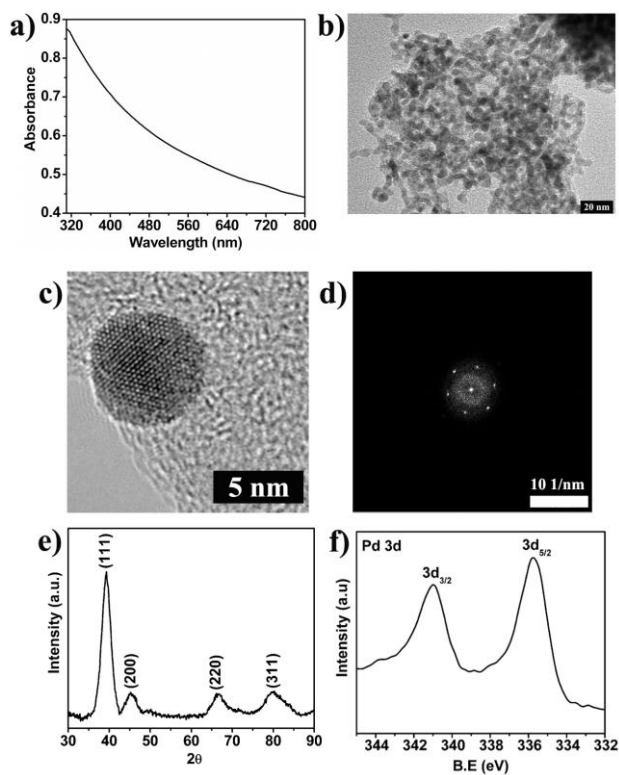


Figure S7. The TEM BF image of Au seeds isolated during the preparation of Au@Ag core-shell nanoparticles

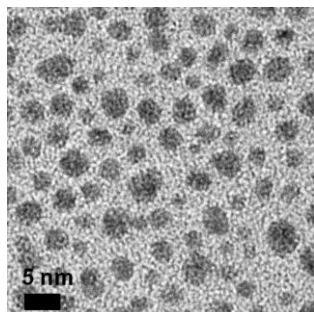


Figure S8. a) HRTEM image; b) powder XRD pattern; and XPS spectra c) Ag 3d and d) Au 4f of Au@Ag core-shell nanoparticles

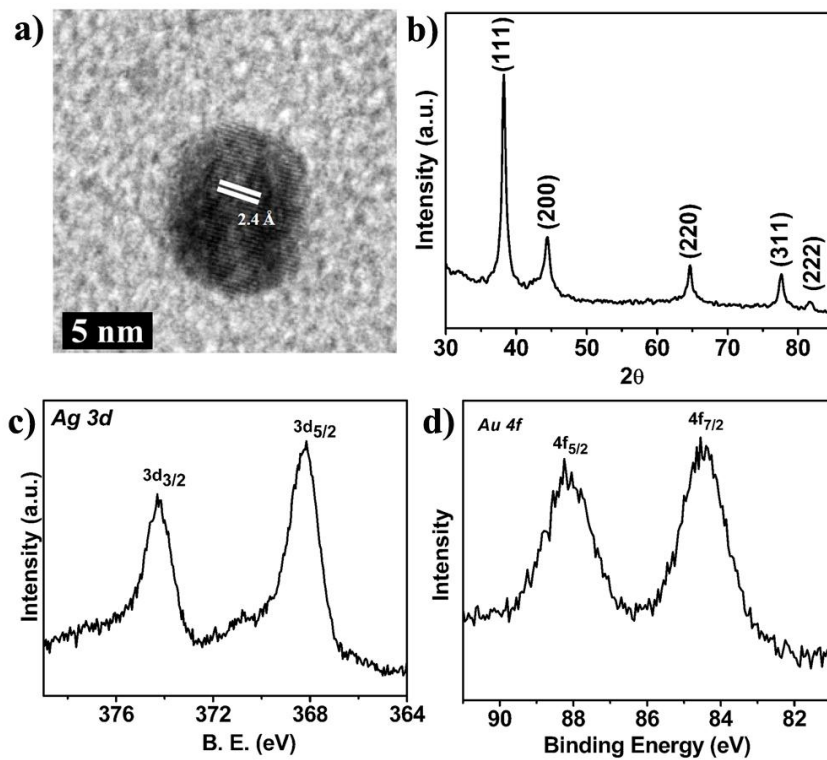


Figure S9. a) HRTEM image; b) powder XRD pattern of Ag@Au core-shell nanoparticles

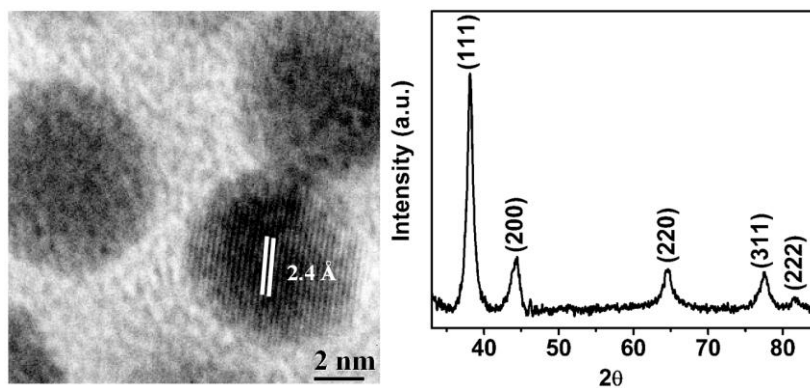


Figure S10. Characterization of Au-Ag alloy nanoparticles prepared from Au@Ag core-shell nanoparticles: a) TEM BF image; b) SAED pattern; c) powder XRD pattern; XPS spectra d) Ag 3d and e) Au 4f; f) HAADF image marked with analyzed particles; g) the corresponding EDS spectrum

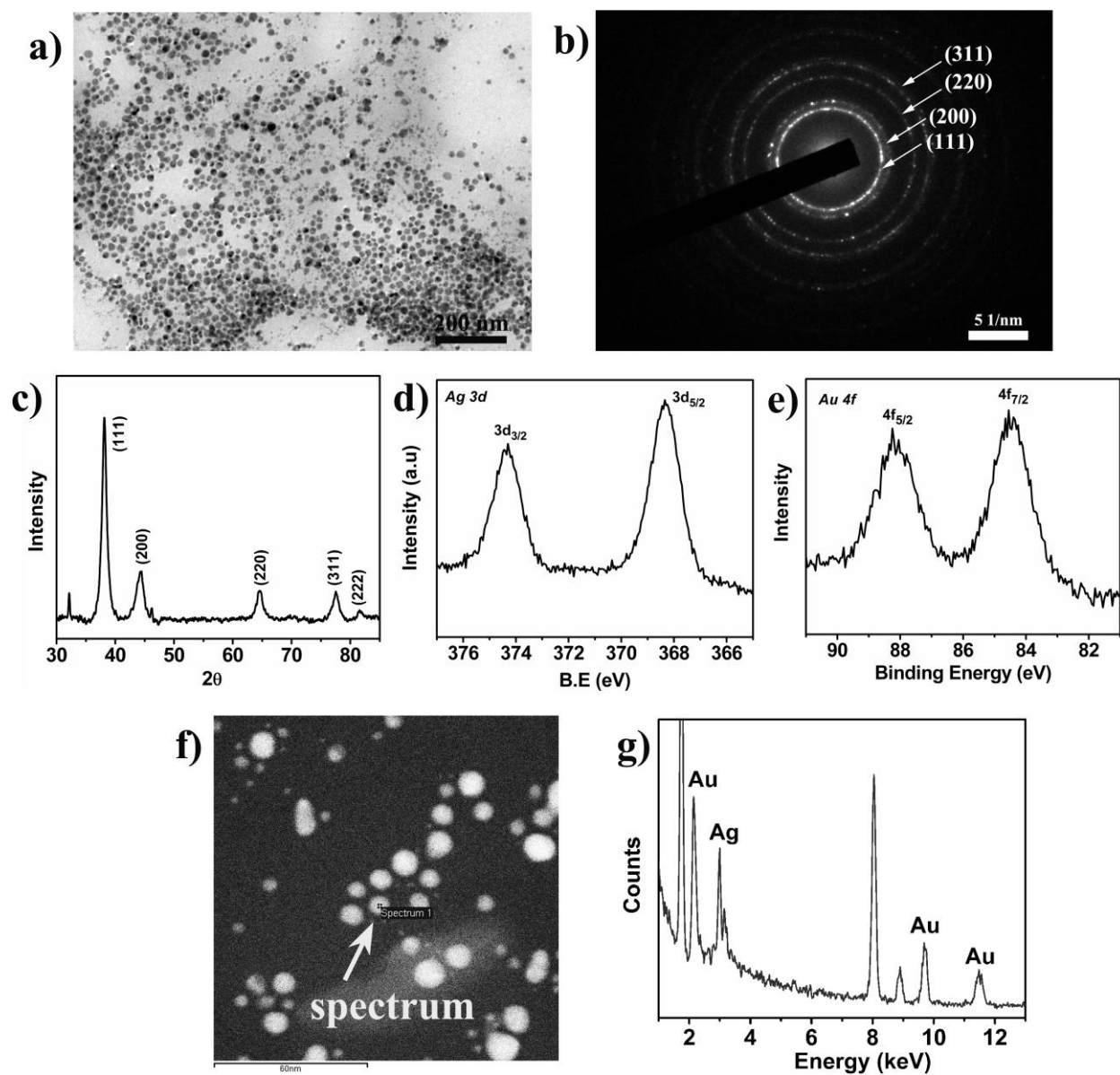


Figure S11. Characterization of Au-Ag alloy nanoparticles prepared from Ag@Au core-shell nanoparticles: a) TEM BF image; b) HRTEM image; c) powder XRD pattern; XPS spectra d) Ag 3d and e) Au 4f; f) HAADF image marked with analyzed particles; g) the corresponding EDS spectrum

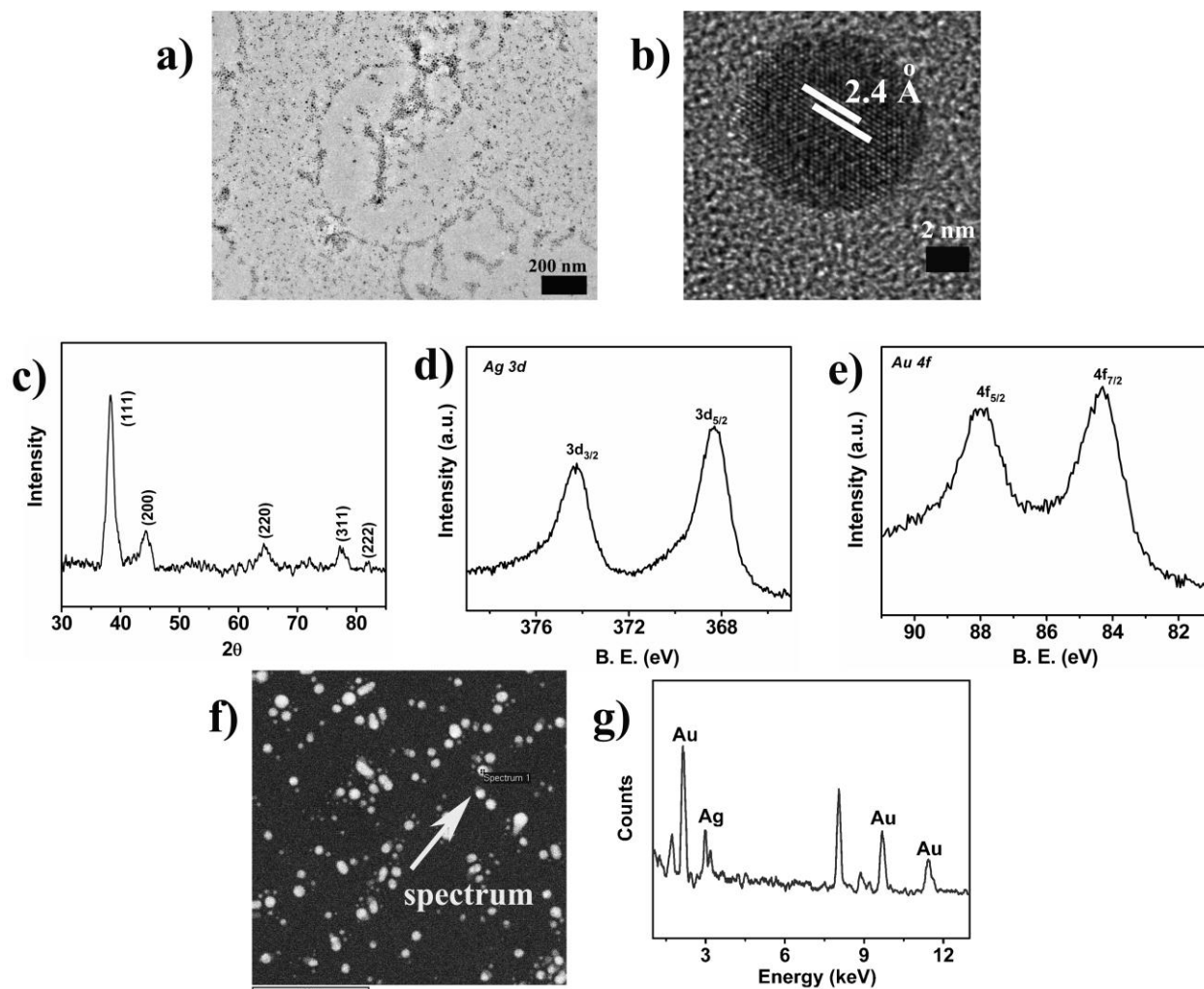


Figure S12. Characterization of Ag seeds isolated during the preparation of Ag@Pd core-shell nanoparticles: a) UV-visible spectrum; b) TEM BF image; c) HRTEM image; XPS spectra d) Ag 3d (368 eV and 374 eV) and e) Pd 3d (340 eV and 346 eV).

In the XPS spectra we noted that a small intense peak at higher binding energy (370.4 eV and 376.3 eV) along with the peak due to Ag(0). This could be attributed to the oxidation of Ag nanoparticles during sample preparation. In a similar manner a small peak appeared in the lower binding energy (336 eV and 341.3 eV) in Pd 3d spectrum which could be assigned to Pd(0). The origin of the Pd(0) could be due to the reduction of Pd(II) ions by TMAB in the solid state during sample preparation.

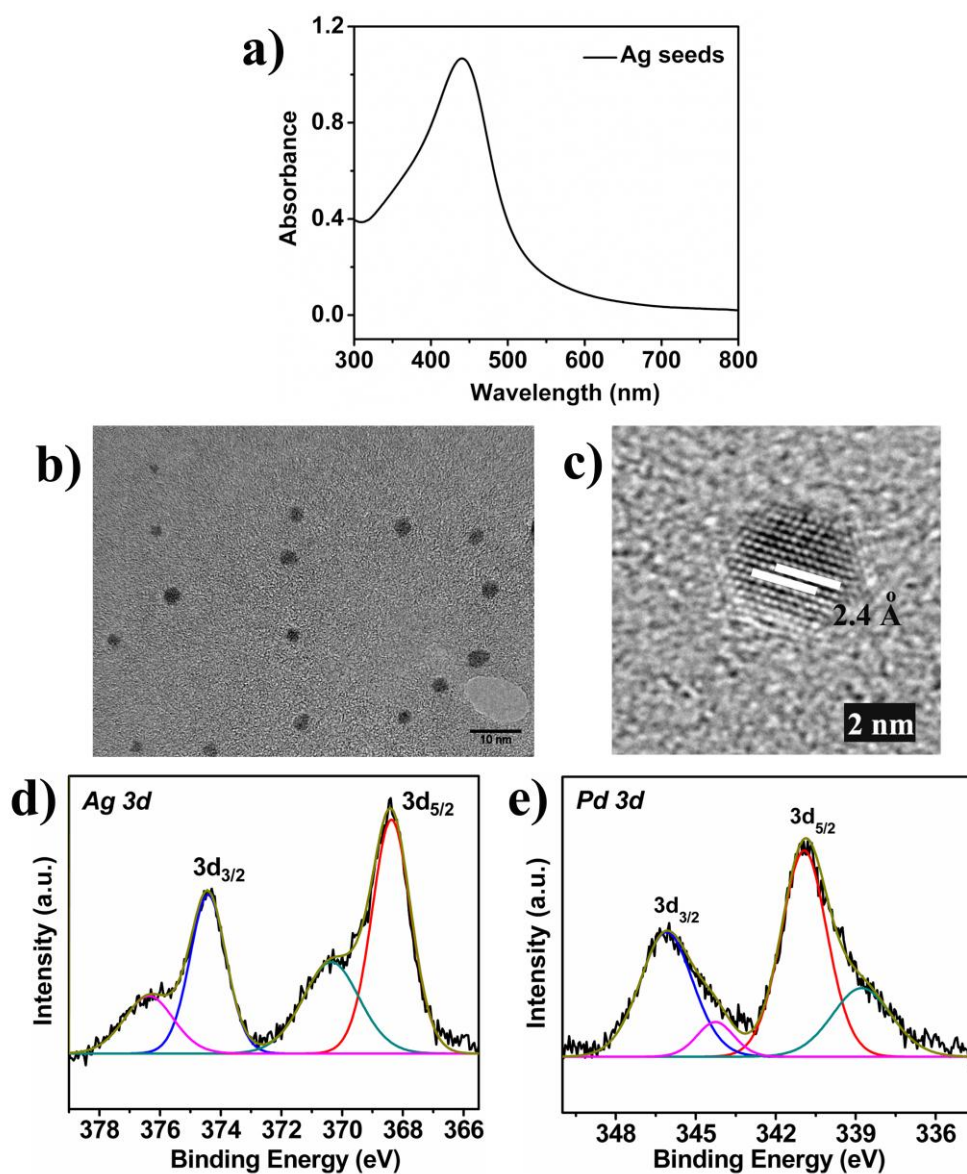


Figure S13. Powder XRD pattern of Ag@Pd core-shell nanoparticles

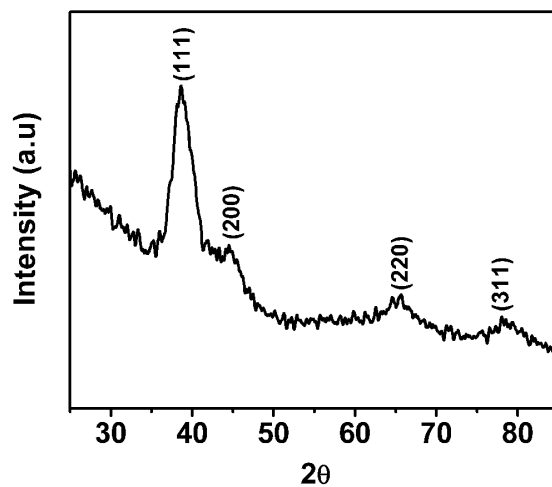


Figure S14. a) HAADF image marked with the individual particle which was analyzed; b) the EDS spectrum, of Ag-Pd alloy nanoparticles

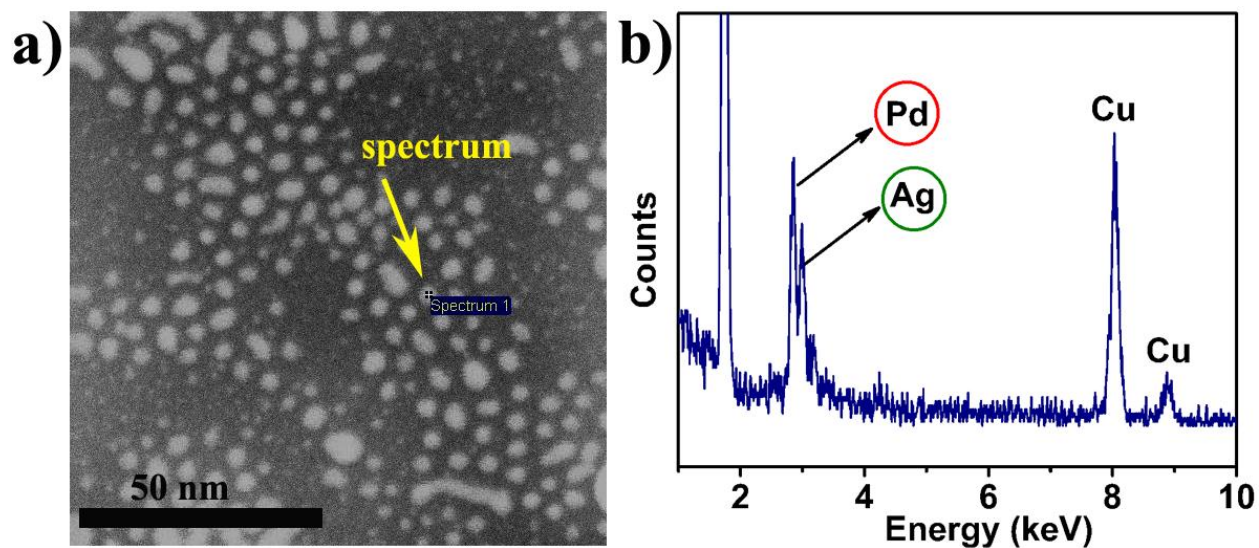


Figure S15. Detection of peroxide in *wet* THF

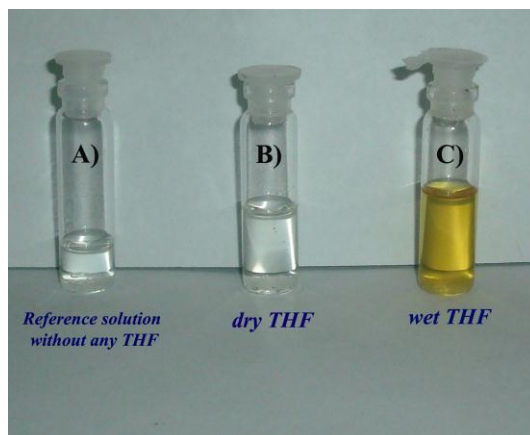


Figure S16. Transformation of Au@Ag core-shell to alloy nanoparticles at 40 °C. UV-visible absorption spectra recorded at different time intervals a) 20 h – black trace; b) 50 h – red trace; c) 90 h – blue trace

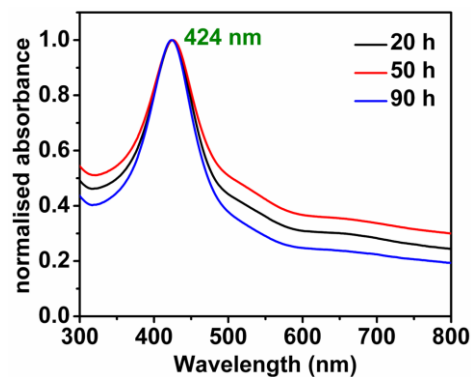


Figure S17. Ag@Au core-shell nanoparticles obtained from a reaction carried out in *wet* THF (no peroxide/dioxygen). a) UV-visible absorption spectrum; b) TEM BF image; c) HRTEM image; d) HAADF image; e) HAADF image showing the analyzed particle marked; f) EDS spectrum obtained from the particles marked in Fig S17e

