## Supporting information for: Structural Properties of *Trans* Hydrido-Hydroxo M(H)(OH)(NH<sub>2</sub>CMe<sub>2</sub>CMe<sub>2</sub>NH<sub>2</sub>)(PPh<sub>3</sub>)<sub>2</sub> (M = Ru, Os) Complexes and their Proton Exchange Behaviour with Water in Solution

Demyan E. Prokopchuk, Alba Collado, Alan J. Lough, and Robert H. Morris\*

Department of Chemistry, University of Toronto, 80 St. George Street, Toronto, ON, M5S3H6,

Canada

E-mail: rmorris@chem.utoronto.ca

<sup>\*</sup>To whom correspondence should be addressed

Compounds	3Ru	3Os
Empirical formula	$C_{42}H_{50}N_2O_2P_2Ru$	$C_{42}H_{50}N_2O_2P_2O_5$
Formula mass	777.85	866.98
Temperature (K)	150(2)	150(1)
Wavelength (Å)	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic
Space group	$P2_{1}/n$	$P2_{1}/n$
a (Å)	13.9422(4)	13.9210(3)
b (Å)	15.7847(2)	15.8226(4)
c (Å)	16.9683(4)	16.9605(6)
$\alpha$ (°)	90	90
<b>β</b> (°)	94.2440(9)	94.033(2)
$\gamma$ (°)	90	90
Volume ( $Å^3$ )	3724.0(2)	3726.5(2)
Z	4	4
Density (calculated, g/cm <sup>3</sup> )	1.387	1.545
Absorption Coefficient (mm <sup>-1</sup> )	0.545	3.546
F(000)	1624	1752
Crystal Size (mm <sup>3</sup> )	0.14  imes 0.12  imes 0.06	0.10  imes 0.10  imes 0.10
Theta range for data collection ( $^{\circ}$ )	2.73–27.56	2.57–27.55
Reflections collected	20367	34052
Independent reflections	8073 [R(int) = 0.0473]	8502 [R(int) = 0.0843]
Completeness to $\theta = 25.00^{\circ}$ (%)	98.0	98.9
Absorption correction	Multi-scan	Multi-scan
Max. and min. transmission	0.969, 0.882	0.712, 0.577
Largest diff. peak and hole (e $Å^3$ )	1.355, -1.163	3.258, -1.037
Refinement method	Full-matrix least-squares on $F^2$	Full-matrix least-squares on $F^2$
Data/restraints/parameters	8073/0/449	8502/0/449
Goodness-of-fit on F <sup>2</sup>	1.040	1.054
Final <i>R</i> indices $[I > 2\sigma(I)]^{a}$	$R_1 = 0.0461$	$R_1 = 0.0428$
<i>R</i> indices (all data) $^{b}$	$wR_2 = 0.1268$	$wR_2 = 0.0974$

Table 1: X-Ray crystal structure and refinement data for complexes **3Ru** and **3Os**.

$${}^{a}R_{1} = \frac{\Sigma(F_{O} - F_{C})}{\Sigma(F_{O})}$$
$${}^{b}wR_{2} = \sqrt{\frac{\Sigma(w(F_{O}^{2} - F_{C}^{2})^{2})}{\Sigma(w(F_{O}^{2})^{2})}}$$



Figure 1: <sup>1</sup>H NMR spectrum (400 MHz) of  $1Ru + H_2O$  at -60 °C in toluene- $d_8$ .



Figure 2: <sup>1</sup>H NMR spectrum (400 MHz) of  $1Ru + 2H_2O$  at -80 °C in toluene- $d_8$ .



Figure 3: <sup>1</sup>H NMR spectrum (400 MHz) of  $10s + H_2O$  at -40 °C in toluene- $d_8$ .



Figure 4: <sup>1</sup>H NMR spectrum (400 MHz) of  $10s + 2H_2O$  at -60 °C in toluene- $d_8$ .