

Electronic Supplementary Information (ESI) for

Discrete spherical hexadecavanadates incorporating a bromide with oxidative bromination activity

Naohiro Kato^a and Yoshihito Hayashi*^a

^aDepartment of Chemistry, Graduate School of Natural Science and Technology, Kanazawa University, Kakuma, Kanazawa, Japan, 920-1192; Fax: +81-76-264-5742; Tel: +81-76-264-5695; E-mail: hayashi@se.kanazawa-u.ac.jp.

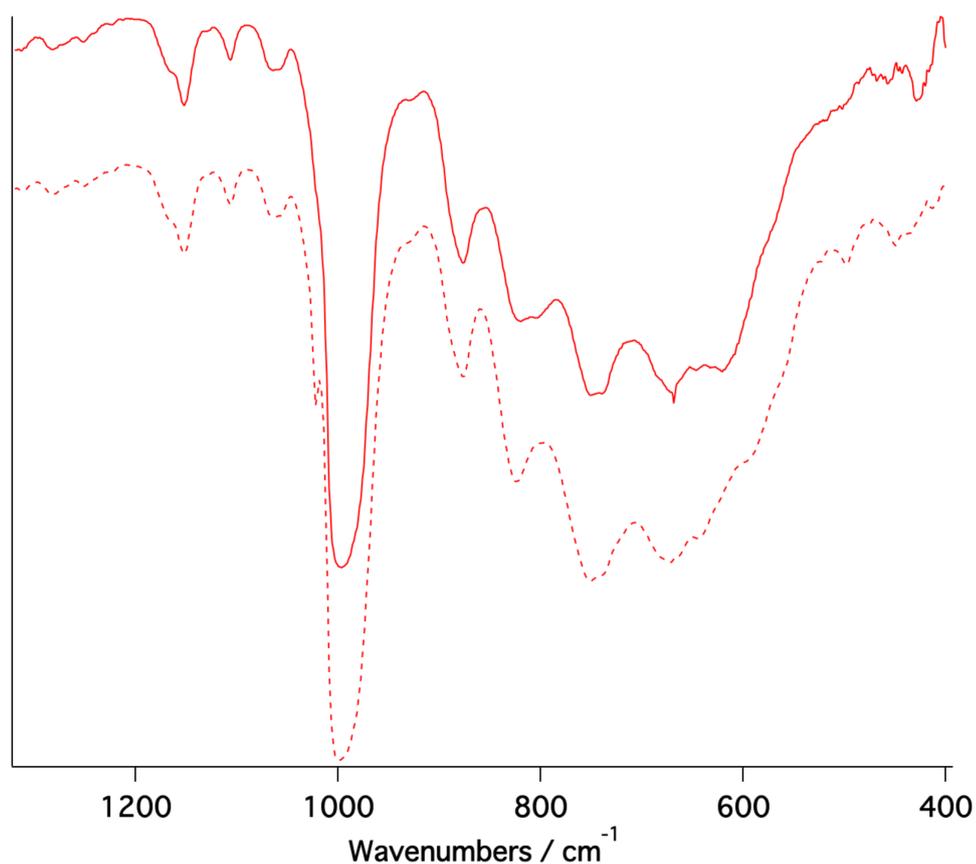


Figure S1. FT-IR spectra of cluster 1 (red line) and cluster 2 (dotted line).

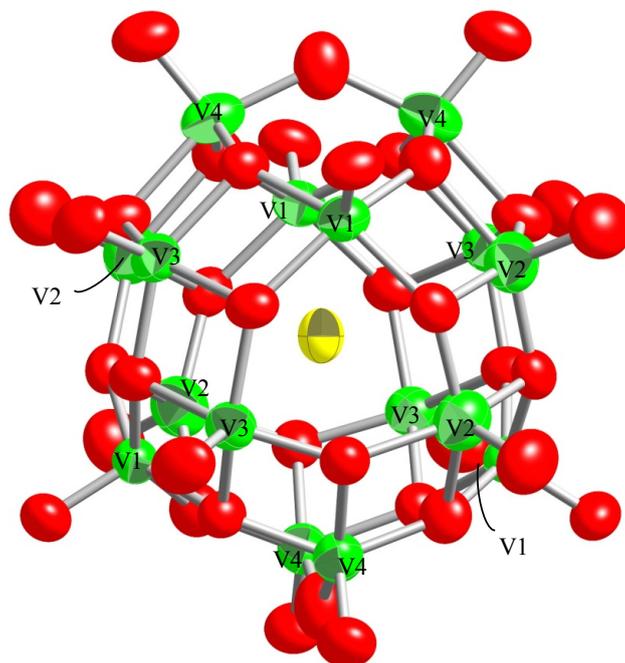


Figure S2. ORTEP drawing of cluster **1** with thermal ellipsoids drawn at the 50% probability level. Yellow, green and red ellipsoids represent chlorine, vanadium and oxygen atoms, respectively.

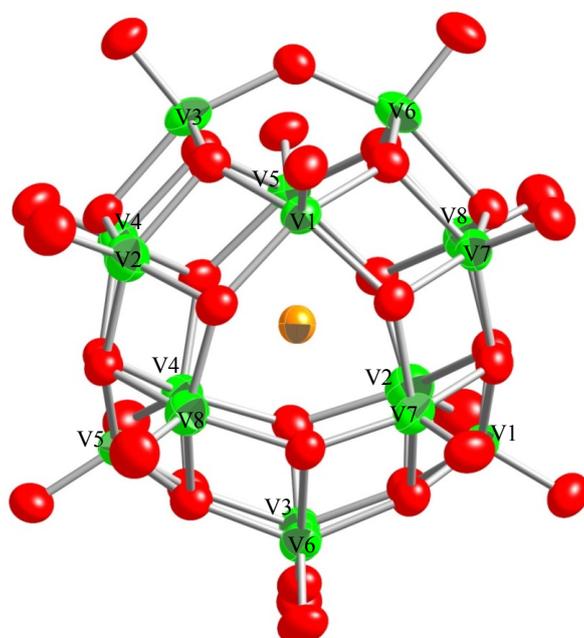


Figure S3. ORTEP drawing of cluster **2** with thermal ellipsoids drawn at the 50% probability level. Brown, green and red ellipsoids represent Bromine, vanadium and oxygen atoms, respectively.

Table S1. The distance between vanadium and halogen atoms. Numberings are based on the ORTEP views in Figure S2 and S3.

	V···Cl (Å)	V···Br (Å)
V1	3.385	3.425
V2	3.429	3.459
V3	3.433	3.700
V4	3.683	3.451
V5		3.433
V6		3.706
V7		3.464
V8		3.482
Average	3.483	3.515

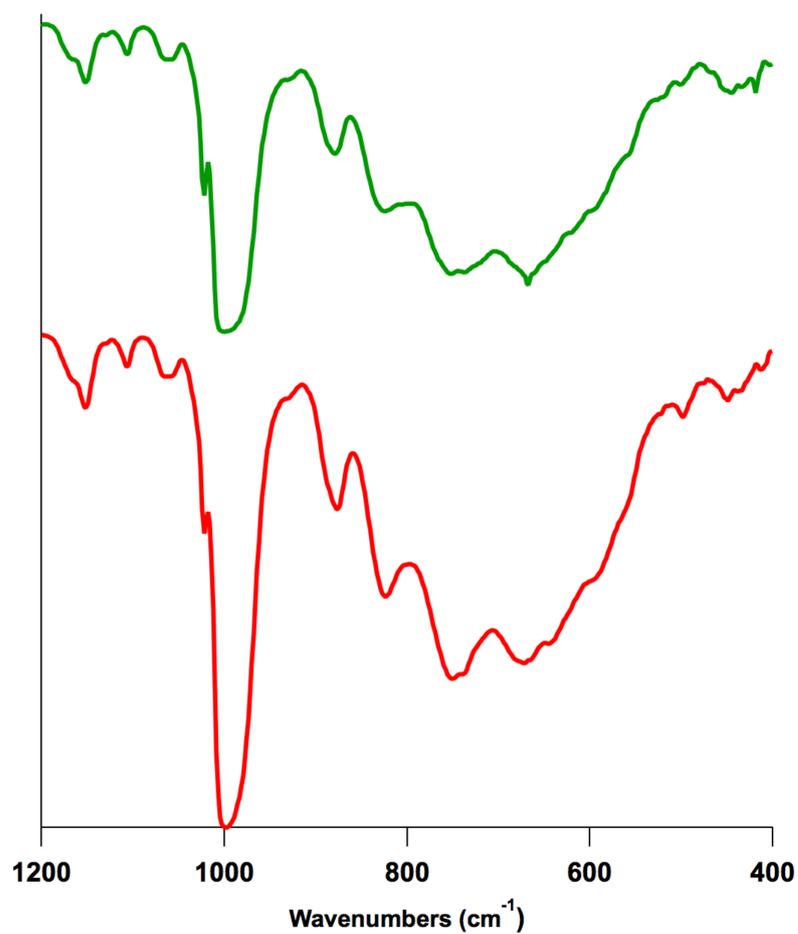


Figure S4. FT-IR spectra of cluster **2** (red line) and recovered **2** after oxidation reaction of *p*-cresol (green line).