

Electronic Supplementary Information

Quick and selective synthesis of $\text{Li}_6[\alpha\text{-P}_2\text{W}_{18}\text{O}_{62}]\cdot 28\text{H}_2\text{O}$ soluble in various organic solvents

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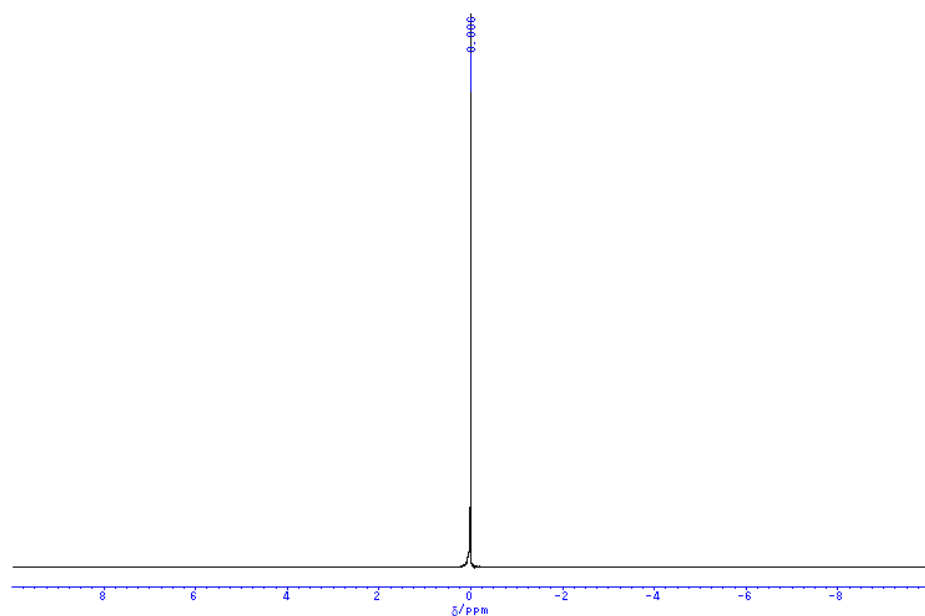


Figure S1. The ^7Li -NMR spectrum of the compound measured with deoxidized 0.3 M lithium chloride methanol solution as reference. There is only singlet peak at 0.006 ppm which signifies presence of solvated lithium ion.

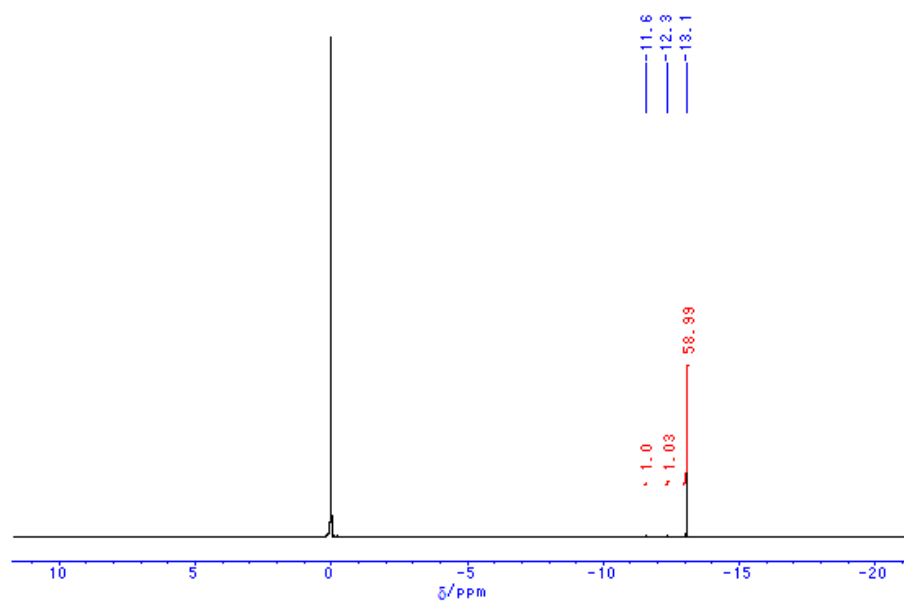


Figure S2. The ^{31}P -NMR spectrum of the product material; $[\alpha\text{-P}_2\text{W}_{18}\text{O}_{62}]^{6-}$ (-13.1 ppm) and $[\beta\text{-P}_2\text{W}_{18}\text{O}_{62}]^{6-}$ (-11.6 and -12.3 ppm). The integration ratio for $[\alpha\text{-P}_2\text{W}_{18}\text{O}_{62}]^{6-}$ and $[\beta\text{-P}_2\text{W}_{18}\text{O}_{62}]^{6-}$ was approximately 30 : 1.

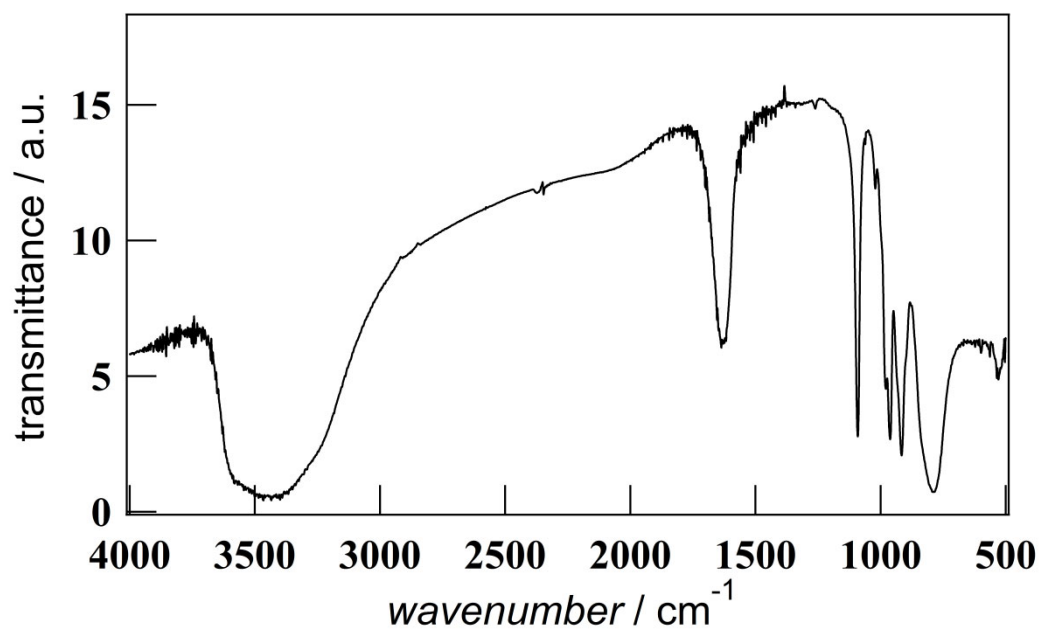


Figure S3. The IR spectrum (KBr) of $\text{Li}_6[\alpha\text{-P}_2\text{W}_{18}\text{O}_{62}] \cdot 28\text{H}_2\text{O}$.

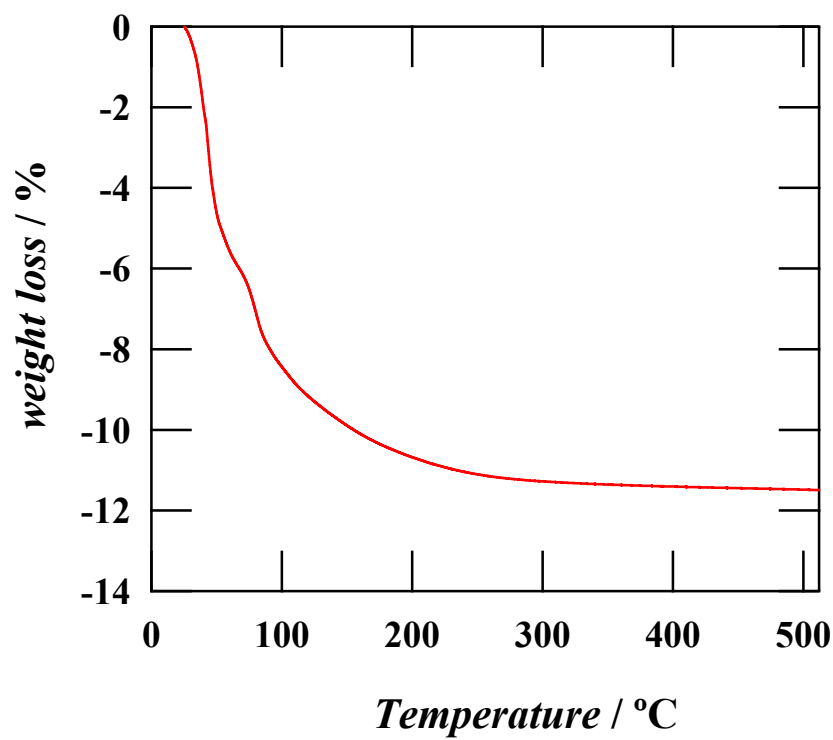


Figure S4. TG curve of $\text{Li}_6[\alpha\text{-P}_2\text{W}_{18}\text{O}_{62}] \cdot 28\text{H}_2\text{O}$.

Table S1. The comparison of this work and previous total experimental steps, time, yield and purity for the synthesis of highly-pure α -Dawson type POM.

entry	date	author	counter cation	total experimental steps	total experimental time / days	yield / %	α -isomer purity / %
1	1997	Droege/Randall/frinke et al. ^{S1}	K ⁺	5	10	82	≥97
2	2004	Nadjo et al.	K ⁺	4-5	8-12	85-93	97-99
3	2012	This report	Li ⁺	3	1-2	87	≥97

Reference

S1. C. R. Graham, R. G. Finke, *Inorg. Chem.*, 2008, **47**, 3679.3679.