

Supporting Information

Copper (II) complex of methionine conjugated bis-pyrazole based ligand promotes dual pathway for DNA cleavage

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Fig. S6 Mechanistic study of oxidative DNA cleavage of pUC 19 DNA (300ng) in presence of 5 eq. ascorbic acid (H_2A) by complex **2**(100 μ M) in 50mM Tris-HCl/NaCl buffer (pH=7.4) after 1.5h incubation at 37 °C using DMSO (1eq.), NaN_3 (1eq.), histidine(1 eq.), catalase (10U) and 4-carboxy TEMPO(1eq.)

Fig. S7 DNA cleavage of pUC 19 DNA(300ng)after irradiation with 365nm UV light (6W,15min) followed by 1h dark incubation at 37 °C in presence of **1-3** in 50mM Tris-HCl/NaCl buffer(pH=7.4).

Fig. S8 Effects of treatment of the complexes **1-3** with MCF-7 tumor cell line for 48 h with rising concentrations: (A) **1**, (B) **2**, (C) **3**.The IC_{50} obtained from the curves are calculated using GraphPad Prism. Data are means $\pm$ SD of three independent experiments.

Table S1. Literature reports on DNA cleavage activity of methionine based metal complexes in presence of light and in dark.

complexes	Dark Cytotoxicity		Photo-cytotoxicity		ROS species involved	Ref
	Conc (μM)	%NC	Conc (μM)	%NC		
[Cu ₂ (L ¹) ₂ (H ₂ O)]	80 ^a		-	-	¹ O ₂	1
[Cu ₂ (L ²) ₂ (H ₂ O)].H ₂ O	88 ^a		-	-		
[(η ⁵ -C ₅ Me ₅) Ir (H ₂ metOMe) (dppz)](CF ₃ SO ₃) ₃	70 ^b		60 ^c			2
[(η ⁵ -Cp*) Ir (AcmetOMe)(dppz)] ²⁺			60 ^c	-		
[Cu(phen)(met)(OMe)](ClO ₄)	200	14	200	89 ^d	¹ O ₂	3
				40 ^e		
				43 ^f		
[Cu(L-met)(phen)(MeOH)](ClO ₄)	100	7	50 ⁱ	40	•OH ^t , ¹ O ₂ ^u	4
	100 ^h	40				
[Cu(L-met)(bpy)(H ₂ O)] (ClO ₄)	100 ^h	10	50 ⁱ	10		
[Cu(L-met)(dpq)(H ₂ O)](ClO ₄)	100 ^h	83	50 ⁱ	77		
[Cu(L-met)(bppz)(H ₂ O)](ClO ₄)	100 ^h	87	50 ⁱ	60		
[VO(salmet)(phen)]	67	15 ^h /11 ^p	33 ^j /67 ^k	15 ^j /4 ^k	•OH ^t / ¹ O ₂ , •OH ^{u,n}	5
[VO(salmet)(dpq)]		13 ^h /12 ^o /14 ^p		82 ^j /20 ^k		
[VO(salmet)(dppz)]		9 ^g /10 ^p		91 ^j /28 ^k		
[Cu(Fc-met)(phen)](NO ₃)	30	6	30	50 ^j /60 ^m	•OH ^u	6
[Cu(Fc-met)(dppz)](NO ₃)		5		91 ^j /97 ^m		
[Cu(Fc-met)(phen)](NO ₃)	20	5	20	62 ^q	•OH ^{t,u}	7
	10	53 ^o /68 ^p	25	55 ^r /58 ^s		
[Cu(Fc-met)(dpq)](NO ₃)	20	5	20	85 ^q		
	10	78 ^o /82 ^p	25	78 ^r /82 ^s		
[Cu(Fc-met)(dppz)](NO ₃)	20	6	20	95 ^q		
	10	87 ^o /94 ^p	25	92 ^r /97 ^s		
[Cu(Fc-met)(nip)](NO ₃)	20	5	20	82 ^q		
	10	72 ^o /79 ^p	25	77 ^r /79 ^s		
[Cu(Ph-met)(phen)](NO ₃)	20	12	20	33 ^q		
	10	49 ^o /32 ^p	25	39 ^r /38 ^s		
[Cu(Ph-met)(dppz)](NO ₃)	20	18	20	63 ^q		
	10	85 ^o /61 ^p	25	75 ^r /73 ^s		

L¹ = (E)-2-(2-hydroxybenzylideneamino)-4-(methylthio)butanoic acid, L² = 2-(2-hydroxybenzylamino)-4-(methylthio)butanoic acid, bpy=2,2'-bipyridine, dpq=dipyrido[3,2-d':2',3'-f]quinoxaline, dppz=dipyrido[3,2-a:2',3'-c]phenazine, phen=1,10-phenanthroline, Fc-met=ferrocene conjugate reduced schiff base of methionine, salmet=N-salicylidene-L-methionate, Ph-met= reduced Schiff base from benzaldehyde and L-methionine, L-met=L-methionine ^a 25μM sodium ascorbate was used as reducing agent, ^b no external reducing agent used, ^c 500W Hg Lamp(3min), ^d 312nm (96W, 20 min), ^e 436 nm (125W,10 min), ^f 532 nm (125W, 30 min), ^g 312nm (5 min), ^h mercaptopropionic acid as reducing agent, ⁱ 365nm (12W, 5min), ^j 365nm (12W, 2h), ^k >750nm(150mW, 2h irradiation), ^l 454nm(30mW,2h), ^m 633nm(12mW,2h), ⁿ only singlet oxygen involves and for >750nm both the ROS involved, ^o in presence of glutathione, ^p in presence of H₂O₂, ^q 454 nm (50mW, 2h), ^r 568 nm (50mW, 2h), ^s 647 nm (50mW, 2h), ^t in dark, ^u in presence of photo irradiation.

Table S2 Crystallographic data and structure refinement parameters for **1-3**.

	1	2.4CH₃CN.2H₂O	3.CH₂Cl₂
Empirical formula	[C ₃₈ H ₅₆ Cu N ₁₂ O ₆](ClO ₄) ₂	[C ₃₈ H ₆₀ CuN ₁₂ O ₆](Cl) ₂ .4(CH ₃ CN).2H ₂ O	[C ₃₇ H ₅₄ CuN ₁₀ O ₆ S ₂](ClO ₄) ₂ .(CH ₂ Cl ₂)
<i>F_w</i> (g mol ⁻¹)	1037.8	1115.65	1132.39
Temperature(K)	100(2)	100(2)	100(2)
Crystal system	Monoclinic	Triclinic	Triclinic
space group	<i>C2/c</i>	<i>P</i> $\bar{1}$	<i>P1</i>
<i>a</i> (Å)	39.869(4)	9.9692(10)	10.9216(11)
<i>b</i> (Å)	13.5647(12)	10.5011(11)	11.5633(11)
<i>c</i> (Å)	20.4661(18)	14.4625(16)	13.5707(13)
α (°)	90	69.076(2)	71.247(2)
β (°)	90.536(2)	70.144(2)	69.310(2)
γ (°)	90	82.168(2)	85.255(2)
<i>V</i> (Å ³)	11067.9(17)	1329.9(2)	1517.3(3)
<i>Z</i>	8	1	1
<i>D_c</i> (g cm ⁻³)	1.246	1.393	1.239
μ (mm ⁻¹)	0.56	0.578	0.684
<i>F</i> (000)	4337.6	591	673
Limiting indices	-50< <i>h</i> <50, -17< <i>k</i> <17, -20< <i>l</i> <26	-13< <i>h</i> <13, -13< <i>k</i> <13, -19< <i>l</i> <19	-13< <i>h</i> <12, -14< <i>k</i> <14, -17< <i>l</i> <17
Reflections collected	91296	23922	18417
Unique reflections/ <i>R_{int}</i>	12090/0.0388	6563/0.0262	9450/ 0.0219
Data / restraints / parameters	12090/0/639	6563/0/340	9450/3/625
Goodness-of-fit ^a on <i>F</i> ²	1.058	1.063	1.070
<i>R</i> ₁ ^b , <i>wR</i> ₂ ^c [<i>I</i> >2σ(<i>I</i>)]	0.0423, 0.1156	0.0467, 0.1419	0.0524, 0.1468
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0493, 0.1190	0.0497, 0.1447	0.0570, 0.1500
Δρ _{max/min} (e Å ⁻³)	0.896/-0.595	0.584/-1.328	1.704/-0.559

^aGoodness-of-fit=[Σw(*F_o*² - *F_c*²)²]/(N_{refln} - N_{params})^{1/2}, based on all data. ^b *R*₁ = Σ (|*F_o*| - |*F_c*|)/|*F_o*|. ^c *wR*₂ = [R [Σ (*F_o*² - *F_c*²)²]/R[Σ(*F_o*²)]]^{1/2}.

Table S3 Bond Valence Sum calculation for **1-3** with two possible oxidation states

Complex	Cu ^I	Cu ^{II}
1	1.584621	1.657387
2	1.568981	1.649399
3	1.558736	1.628816

Table S4 Lipophilicity of metal complexes and the ligands in octanol/water system.

	LogP _{o/w}
L1	2.08(3)
L2	1.67(6)
1	-0.08(2)
2	-0.49(3)
3	-0.35(2)

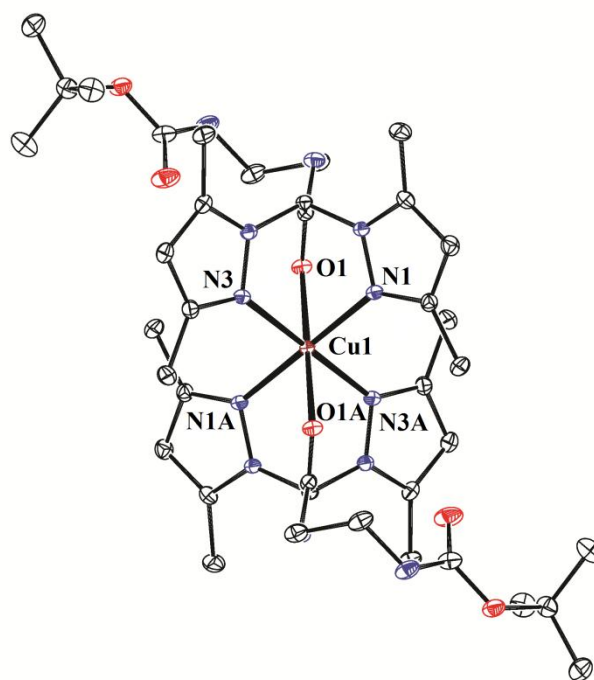


Fig. S1 ORTEP diagram of **2**. Thermal ellipsoids are drawn at 50% probability level. Hydrogen atoms, counter anions and solvents have been omitted for clarity. Symmetry transformations used to generate equivalent atoms A: $-x, -y+1, -z+1$.

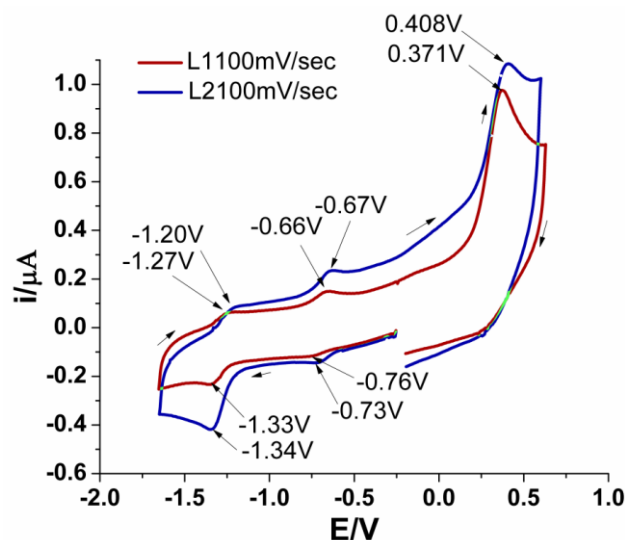


Fig. S2 Cyclic Voltammogram of a 1.0 mM solution of L1 & L2 in DMF and TBAP (0.1 M) as supporting electrolyte. Indicated peak potentials are in V vs. 0.01 M non aqueous Ag/Ag⁺ reference electrode (0.55 V vs. NHE, evaluated using Fc/Fc⁺ couple)

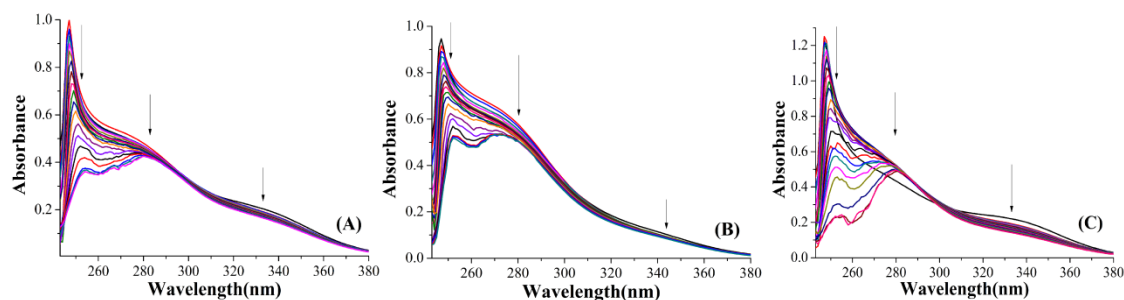


Fig. S3 Absorption spectral change after addition of increased amount of CT DNA (4×10^{-3} M) to complex (4×10^{-4} M) in Tris-NaCl/DMF (9:1) media (pH 7.4) at room 25°C: (A) **1**, (B) **2**, (C) **3**. Inset: Plot of $[\text{DNA}]/\Delta\epsilon$ vs $[\text{DNA}]$ for calculation of apparent binding constant (K_b).

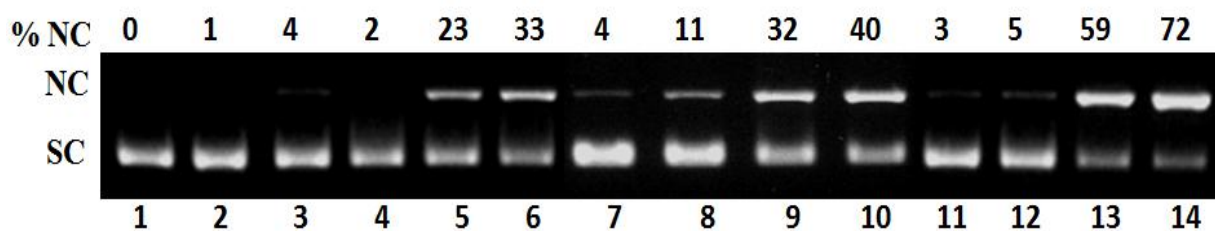


Fig. S4 DNA cleavage of pUC 19 DNA(300ng)in presence of hydrogen peroxide(H_2O_2) by **1-3** in 50mM Tris-HCl/NaCl buffer(pH=7.4) after 1.5h incubation at 37 °C .Lane 1,DNA control; lane 2, DNA+ H_2O_2 (700 μ M); lane 3, DNA+**1**(170 μ M); lane 4, DNA+**1**(100 μ M)+ H_2O_2 (700 μ M); lane 5, DNA+**1** (140 μ M)+ H_2O_2 (700 μ M); lane 6, DNA+**1**(170 μ M)+ H_2O_2 (700 μ M); lane 7, DNA+**2**(170 μ M); lane 8, DNA+**2**(100 μ M)+ H_2O_2 (700 μ M); lane 9, DNA+**2** (140 μ M)+ H_2O_2 (700 μ M); lane 10, DNA+**2**(170 μ M)+ H_2O_2 (700 μ M); lane 11, DNA+**3**(170 μ M); lane 12, DNA+**3**(100 μ M)+ H_2O_2 (700 μ M); lane 13, DNA+**3** (140 μ M)+ H_2O_2 (700 μ M); lane 14, DNA+**3**(170 μ M)+ H_2O_2 (700 μ M).

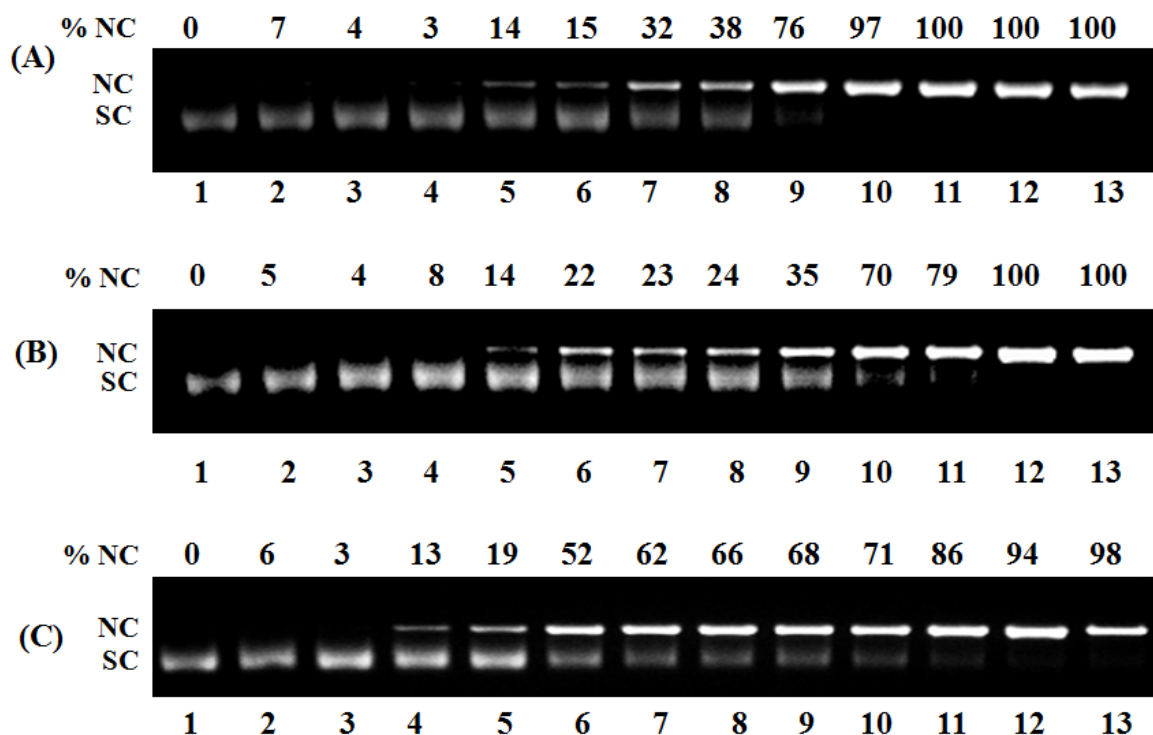


Fig. S5 pUC19 SC DNA cleavage activity of complexes **1-3** monitored by 1% agarose gel electrophoresis, where [DNA] =300ng, [Complex] =100 μ M and [H_2A] =500 μ M in the same dilution and running buffer used for the usual gel electrophoresis. The gel images for (A) – (C) are for complexes **1-3** respectively. The representations are Lane 1, DNA control; lane 2, DNA + H_2A ; lane 3, DNA+ complex; lane 4-13, DNA+complex+ H_2A at 0, 5, 10 ,15, 20, 25, 30, 40, 50, 60 min respectively.

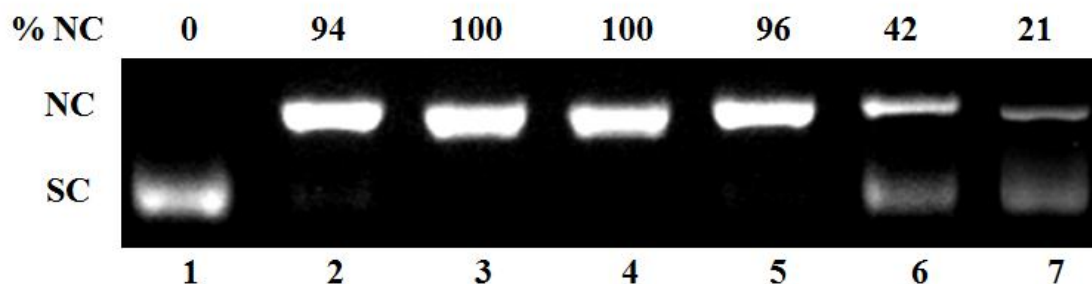


Fig. S6 Mechanistic study of oxidative DNA cleavage of pUC 19 DNA (300ng) in presence of 5 eq. ascorbic acid(H_2A) by complex 2 (100 μ M) in 50mM Tris-HCl/NaCl buffer(pH=7.4) after 1.5h incubation at 37 °C using DMSO(1eq.), NaN_3 (1eq.), histidine(1 eq.), catalase(10U) and 4-carboxy TEMPO(1eq.). Lane 1, DNA control; lane 2, DNA+2; lane 3, DNA+2+DMSO; lane 4, DNA+2+ NaN_3 ; lane 5, DNA+2+histidine; lane 6, DNA+2+catalase; lane 7, DNA+2+4-carboxy TEMPO.

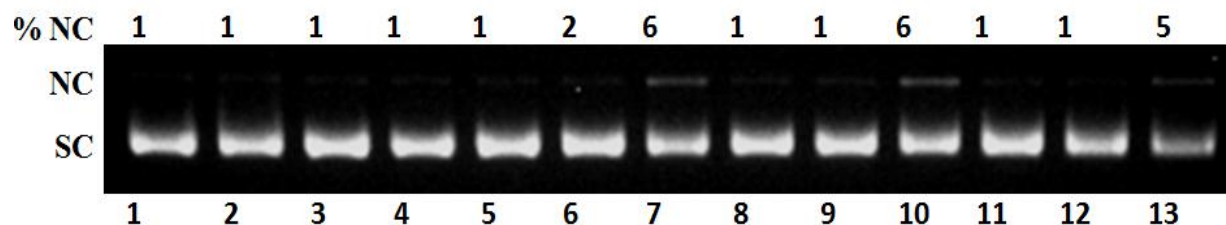


Fig. S7 DNA cleavage of pUC 19 DNA(300ng) after irradiation with 365nm UV light(4W,30min) followed by 1h dark incubation at 37 °C in presence of 1-3 in 50mM Tris-HCl/NaCl buffer(pH=7.4).Lane 1,DNA control; lane 2, DNA+1(100 μ M); lane 3, DNA+2(100 μ M); lane 4, DNA+3(100 μ M); lane 5, DNA+1(50 μ M); lane 6, DNA+1(100 μ M); lane 7, DNA+1(170 μ M); lane 8, DNA+2(50 μ M); lane 9, DNA+2 (100 μ M); lane 10, DNA+2(170 μ M); lane 11, DNA+3(50 μ M); lane 12, DNA+3(100 μ M); lane 13, DNA+3 (170 μ M).

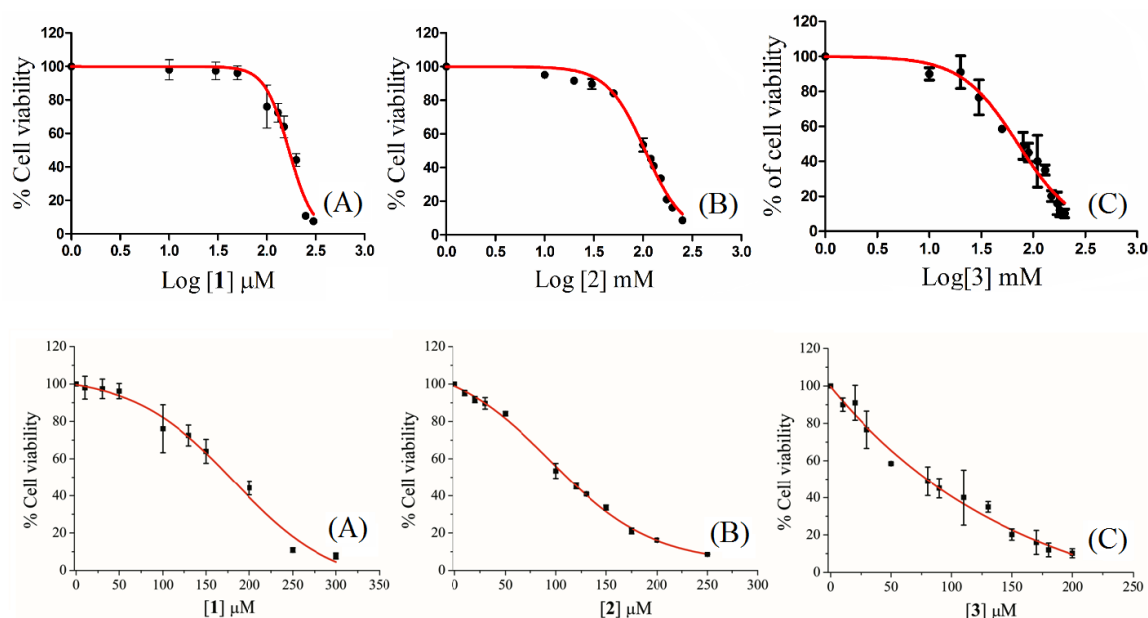


Fig. S8 Effects of treatment of the complexes **1-3** with MCF-7 tumor cell line for 48 h with rising concentrations: (A) **1**, (B) **2**, (C) **3**. Top panel of graphs showing % survival against log of concentration of complexes and bottom panel showing % survival against concentration of complexes). The IC_{50} obtained from the curves are calculated using GraphPad Prism. Data are means $\pm$ SD of three independent experiments.

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