

## Electronic Supplementary Information (ESI) for: Complexation of Al(III) with gluconate in alkaline to hyperalkaline solutions: formation, stability and structure

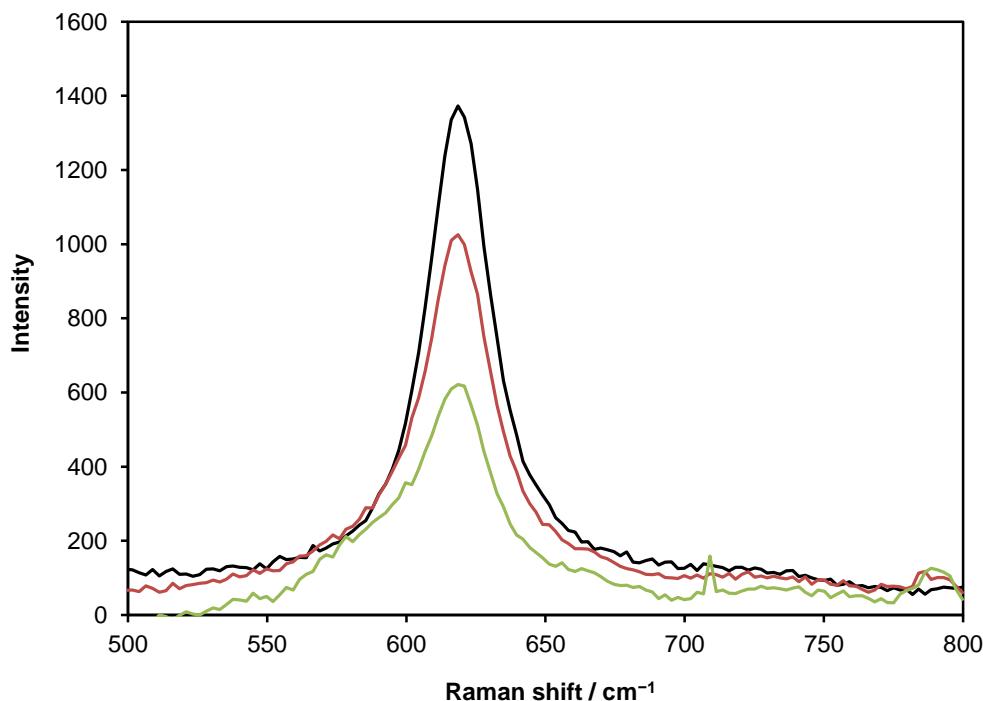
Attila Pallagi,<sup>a</sup> Ágost Gyula Tasi<sup>a</sup>, Gábor Peintler,<sup>b</sup> Péter Forgo,<sup>c</sup> István Pálinkó<sup>d</sup> and Pál Sipos\*<sup>a</sup>

<sup>5</sup> <sup>a</sup> Department of Inorganic and Analytical Chemistry, University of Szeged, Dóm tér 7, Szeged, H-6720 Hungary. Fax: +36 62 544 340; Tel: +36 62 544 054; E-mail: sipos@chem.u-szeged.hu

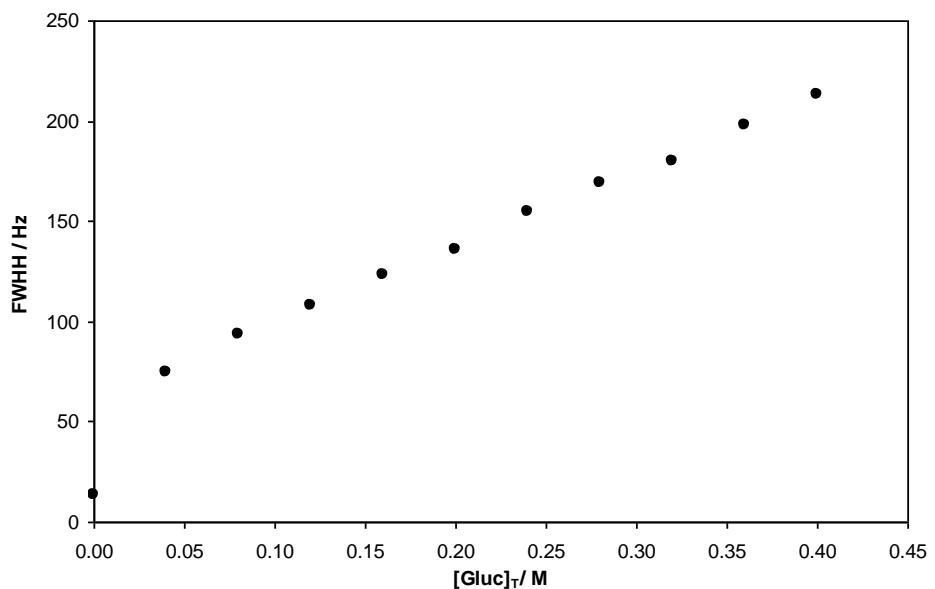
<sup>b</sup> Department of Physical Chemistry and Material Science, University of Szeged, Aradi Vértanúk tere 1, Szeged, H-6720 Hungary.

<sup>10</sup> <sup>c</sup> Eszterházy Károly College, Institute of Food Science, Leányka út 6/D, Eger, Hungary.

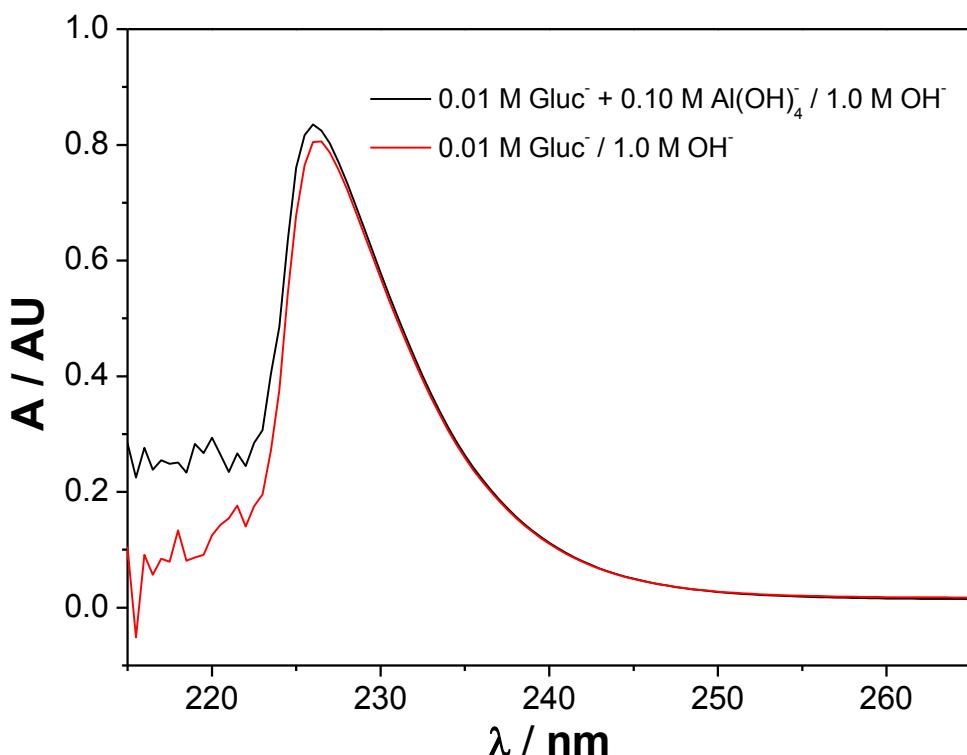
<sup>d</sup> Department of Organic Chemistry, University of Szeged, Dóm tér 8, Szeged, H-6720 Hungary



**Fig. S1** Background corrected Raman spectra of aluminate solutions  
( $[Al(OH)_4^-]_T = 0.4\text{ M}$ ;  $[NaOH]_T = 0.8\text{ M}$ ) with various added  $Gluc^-$  concentrations  
(—: 0 M; —: 0.2 M; —: 0.4 M)



**Fig. S2** The variation of the FWHH of Al on the  $^{27}\text{Al}$  NMR spectra with rising Gluc $^-$  concentration in solutions of  $[\text{Al}(\text{OH})_4^-]_T = 0.200\text{ M}$  and  $[\text{NaOH}]_T = 0.400\text{ M}$



**Fig. S3** The UV-Vis spectra of the  $[\text{Gluc}^-]_T$  (0.01 M) in the absence (red) and the presence of the  $[\text{Al}(\text{OH})_4^-]_T$  (black; 0.10 M) in 1.0 M  $[\text{OH}^-]_T$  medium at 25 °C