

Electronic Supplementary Information

Comparison of ancillary ligand effects between 2,2'-bipyridine and 2-(2'-pyridyl)phenyl in the linkage and bridging isomerism of 5-methyltetrazolato iridium(III) and/or rhodium(III) complexes

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Fig. S1 An ORTEP (50% probability level, H-atoms are omitted for clarity) of $[\text{Cp}^*\text{Ir}(\text{ppy})(\text{MeCN}_4\text{-}\kappa\text{N}^1)]$ (**3**).

Fig. S2 Infrared spectra (Nujol mull) of bulk samples of $[\text{Cp}^*\text{Ir}(\text{bpy})(\text{MeCN}_4)]\text{PF}_6$ (**1**: black, lower) and $[\text{Cp}^*\text{Rh}(\text{bpy})(\text{MeCN}_4)]\text{PF}_6$ (**2**: red, upper).

Fig. S3 An ORTEP (50% probability level, H-atoms are omitted for clarity) of the cationic part in $[\{\text{Cp}^*\text{Rh}(\text{bpy})\}_2(\mu\text{-MeCN}_4)](\text{PF}_6)_3\bullet\text{CH}_3\text{CN}$ (**6**• CH_3CN).

Fig. S4 ^1H NMR (CDCl_3 , 300 MHz, 22 °C) spectrum of $[\{\text{Cp}^*\text{Rh}(\text{bpy})\}_2(\mu\text{-MeCN}_4)](\text{PF}_6)_3$ (**6**).

Fig. S5 ^1H NMR (CDCl_3 , 300 MHz, 22 °C) spectra of (a) $[\{\text{Cp}^*\text{Ir}(\text{ppy})\}_2(\mu\text{-MeCN}_4)]\text{PF}_6$ (**7**) and (b) $[\{\text{Cp}^*\text{Ir}(\text{ppy})\}_2(\mu\text{-MeCN}_4)]\text{BF}_4$ (**7B**).

Fig. S6 An ORTEP (50% probability level, H-atoms are omitted for clarity) of the cationic part in $[\{\text{Cp}^*\text{Rh}(\text{ppy})\}_2(\mu\text{-MeCN}_4\text{-}\kappa\text{N}^1\text{:}\kappa\text{N}^3)]\text{PF}_6\bullet\text{CH}_2\text{Cl}_2\bullet\text{Et}_2\text{O}$ (**8**• $\text{CH}_2\text{Cl}_2\bullet\text{Et}_2\text{O}$).

Fig. S7 ^1H NMR (300 MHz, 22 °C) spectra of $[\{\text{Cp}^*\text{Rh}(\text{ppy})\}_2(\mu\text{-MeCN}_4)]\text{PF}_6$ (**8**) in (a) CD_3CN , (b) CD_2Cl_2 , and (c) CDCl_3 .

Fig. S8 ^1H NMR (CDCl_3 , 300 MHz, 22 °C) spectrum of $[\{\text{Cp}^*\text{Rh}(\text{ppy})\}_2(\mu\text{-MeCN}_4)]\text{BF}_4$ (**8B**).

Fig. S9 ORTEP (50% probability level, H atoms omitted for clarity) of the cation in $[\text{Cp}^*\text{Ir}(\text{bpy})(\mu\text{-MeCN}_4)\text{Rh}(\text{bpy})\text{Cp}^*](\text{PF}_6)_3\bullet\text{CH}_3\text{CN}$ (**9**• CH_3CN : one of the possible orientations for the disordered Ir/Rh atoms).

Fig. S10 ^1H NMR spectra (at 22 °C, 300 MHz) of a CD_3CN solution of **11** (the isolated crystals): (a) immediately after dissolving and (b) after standing at room temperature for 24 h.

Fig. S11 ORTEPs (50% probability level, H atoms omitted for clarity) of the cationic parts in (a) $[\text{Cp}^*\text{Ir}(\text{ppy})\{\mu\text{-MeCN}_4\text{-}\kappa\text{N}^1(\text{Ir})\text{:}\kappa\text{N}^3(\text{Rh})\}\text{Rh}(\text{ppy})\text{Cp}^*]\text{BF}_4\bullet^{1/2}\text{CH}_3\text{CN}\bullet^{1/4}\text{H}_2\text{O}$ (**12B**• $^{1/2}\text{CH}_3\text{CN}\bullet^{1/4}\text{H}_2\text{O}$) and (b) $[\text{Cp}^*\text{Ir}(\text{ppy})\{\mu\text{-MeCN}_4\text{-}\kappa\text{N}^1(\text{Ir})\text{:}\kappa\text{N}^3(\text{Rh})\}\text{Rh}(\text{ppy})\text{Cp}^*]\text{BPh}_4\bullet\text{CH}_2\text{Cl}_2\bullet2\text{Et}_2\text{O}$ (**12BP**• $\text{CH}_2\text{Cl}_2\bullet2\text{Et}_2\text{O}$).

Fig. S12 ^1H NMR spectra (at 22 °C, 300 MHz) of CDCl_3 solutions of (a) the recrystallised sample of $[\text{Cp}^*\text{Ir}(\text{ppy})(\mu\text{-MeCN}_4)\text{Rh}(\text{ppy})\text{Cp}^*]\text{BF}_4$ (**12B**), (b) the crude reaction products of $[\text{Cp}^*\text{Ir}(\text{ppy})(\mu\text{-MeCN}_4)\text{Rh}(\text{ppy})\text{Cp}^*]\text{PF}_6$ (**12**) resulting from the addition of NH_4PF_6 , and (c) the recrystallised sample of $[\text{Cp}^*\text{Ir}(\text{ppy})(\mu\text{-MeCN}_4)\text{Rh}(\text{ppy})\text{Cp}^*]\text{BPh}_4\bullet\text{CH}_2\text{Cl}_2\bullet2\text{Et}_2\text{O}$ (**12BP**• $\text{CH}_2\text{Cl}_2\bullet2\text{Et}_2\text{O}$).

Table S1 Crystallographic data for complexes.

Table S2–S6 Selected bond lengths and angles.

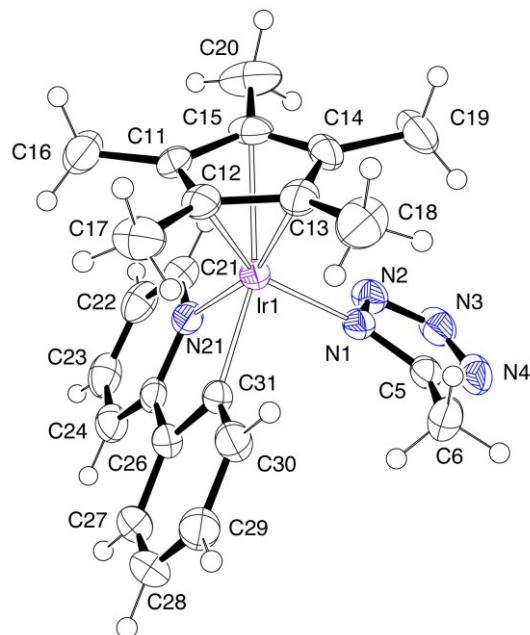


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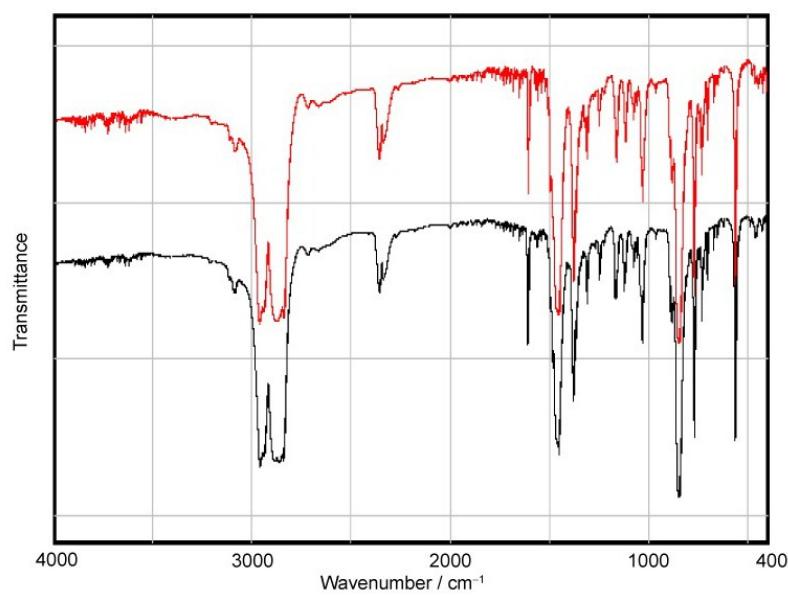


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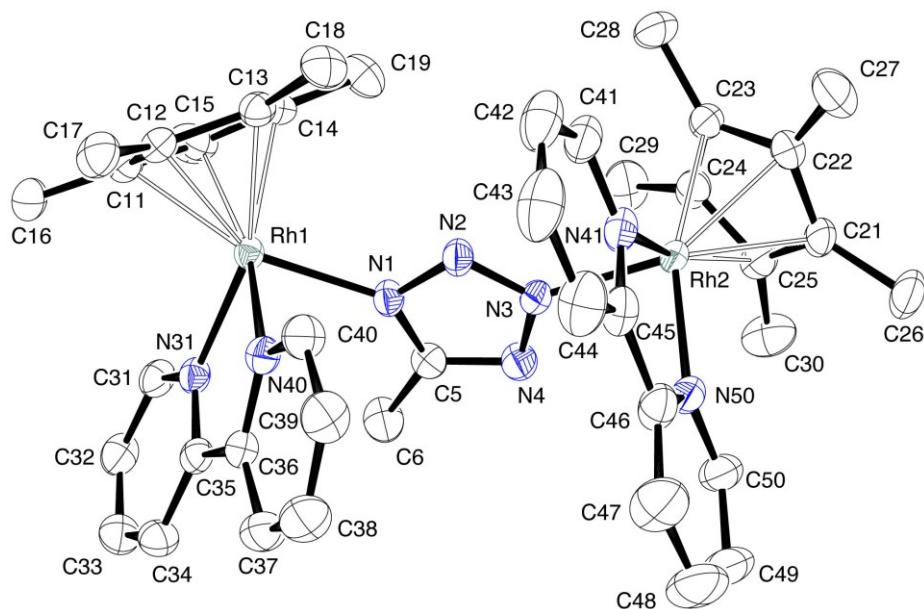


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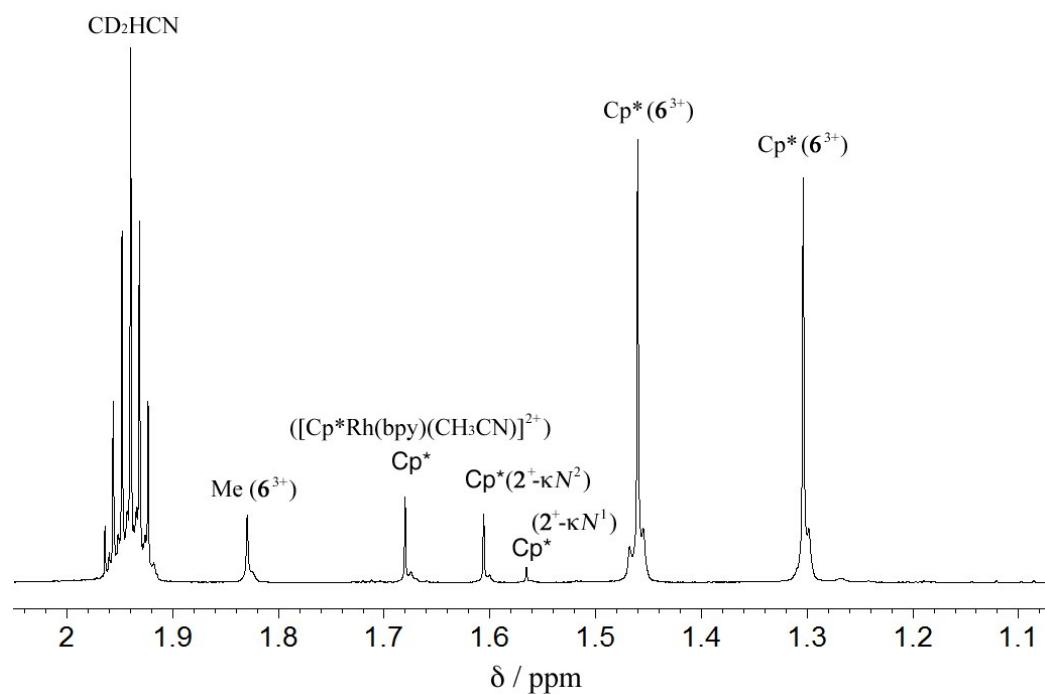


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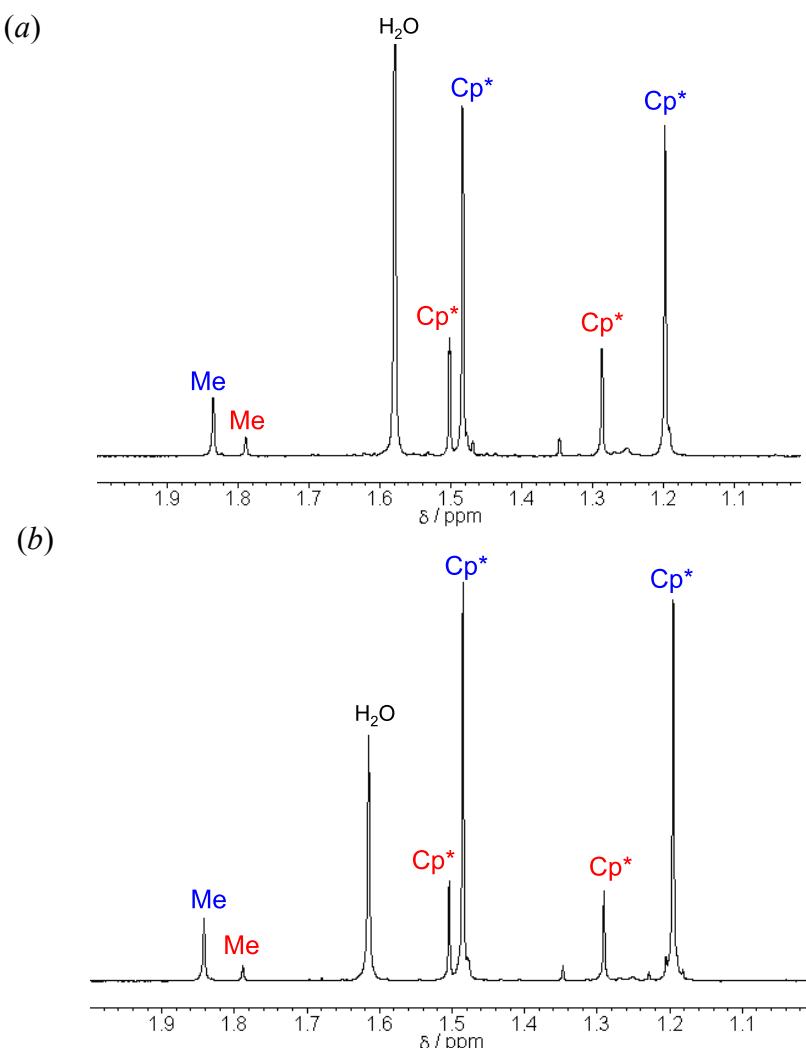


Fig. S5 ¹H NMR (CDCl₃, 300 MHz, 22 °C) spectra of (a) [{Cp*Ir(ppy)}₂(μ-MeCN₄)]PF₆ (**7**) and (b) [{Cp*Ir(ppy)}₂(μ-MeCN₄)]BF₄ (**7B**).

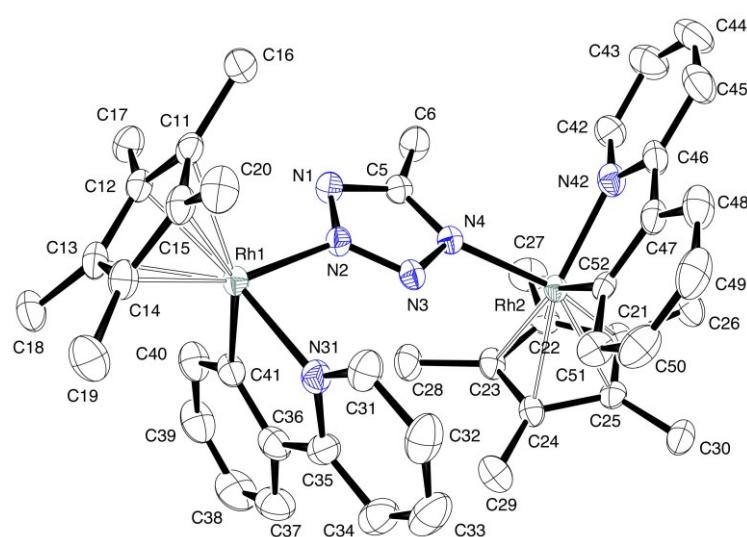


Fig. S6 An ORTEP (50% probability level, H-atoms are omitted for clarity) of the cationic part in [{Cp*Rh(ppy)}₂(μ-MeCN₄-κN¹:κN³)]PF₆•CH₂Cl₂•Et₂O (**8**•CH₂Cl₂•Et₂O).

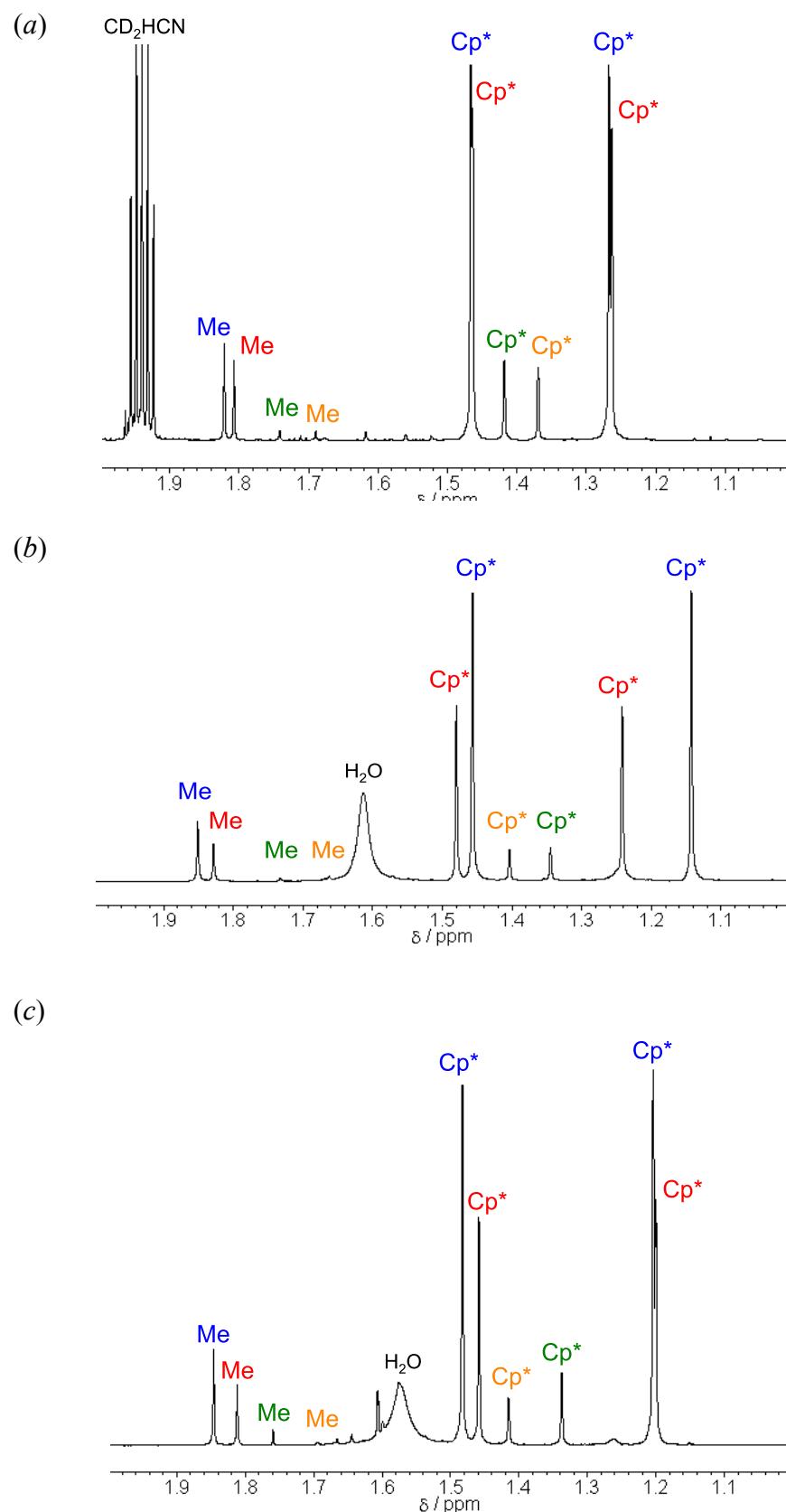


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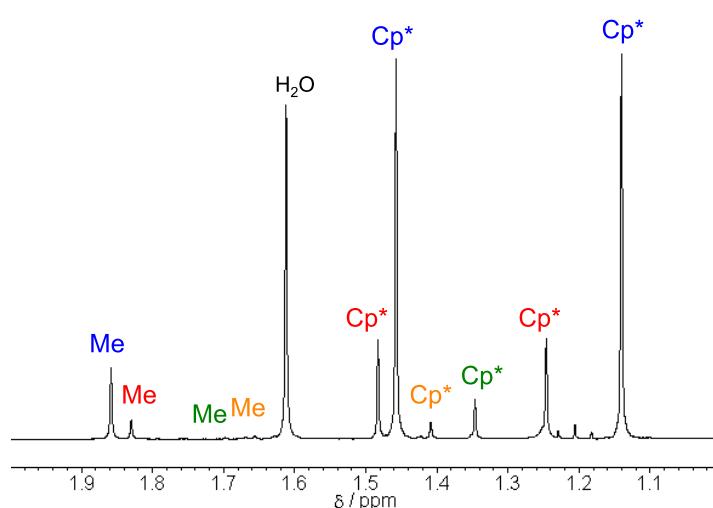


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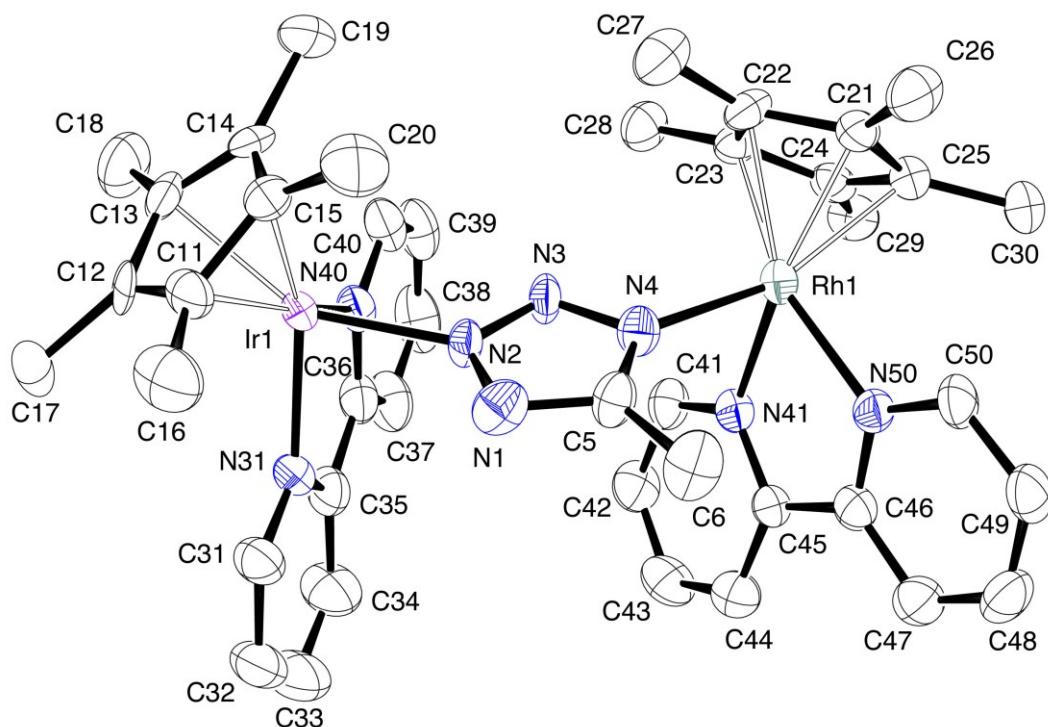


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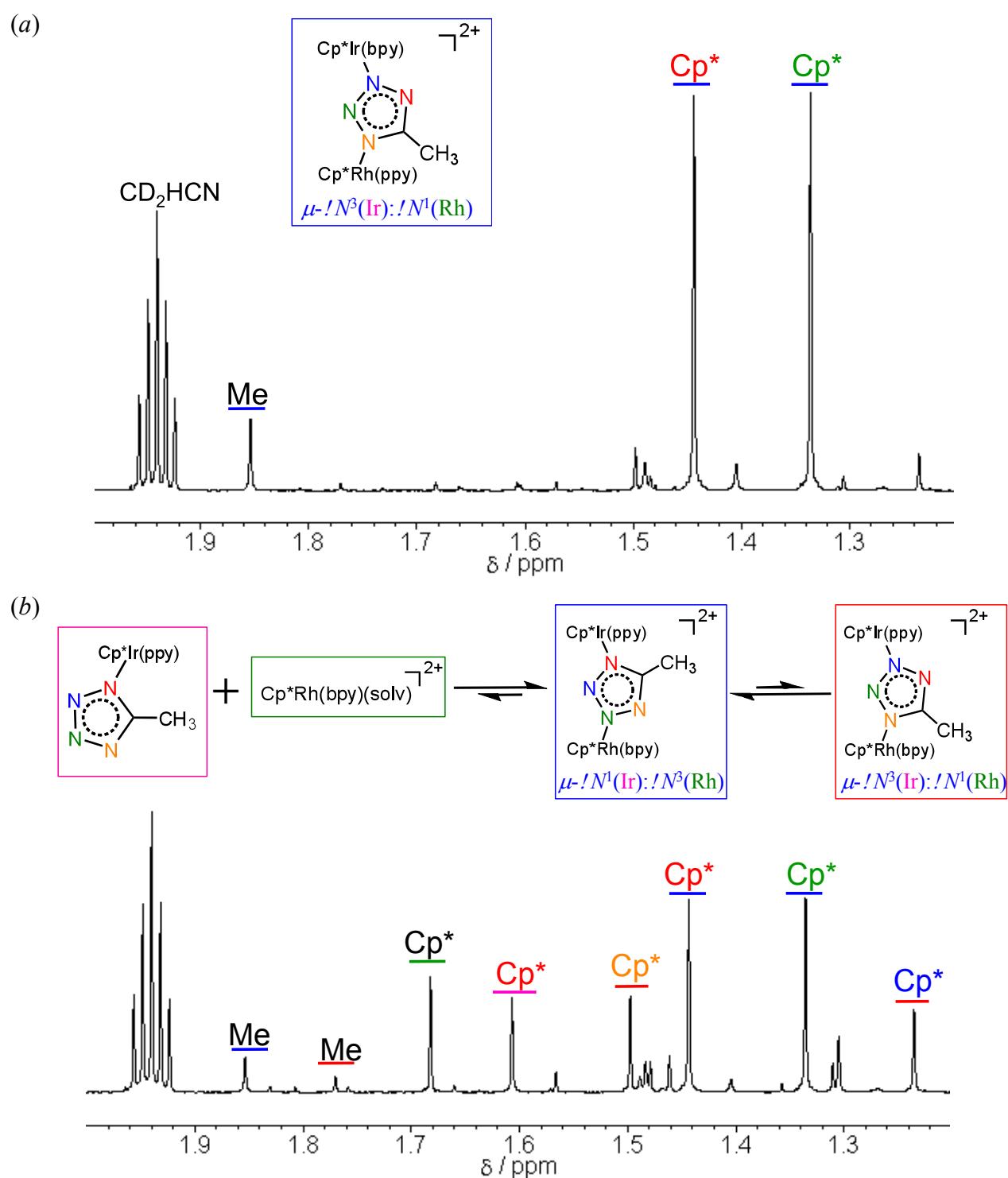
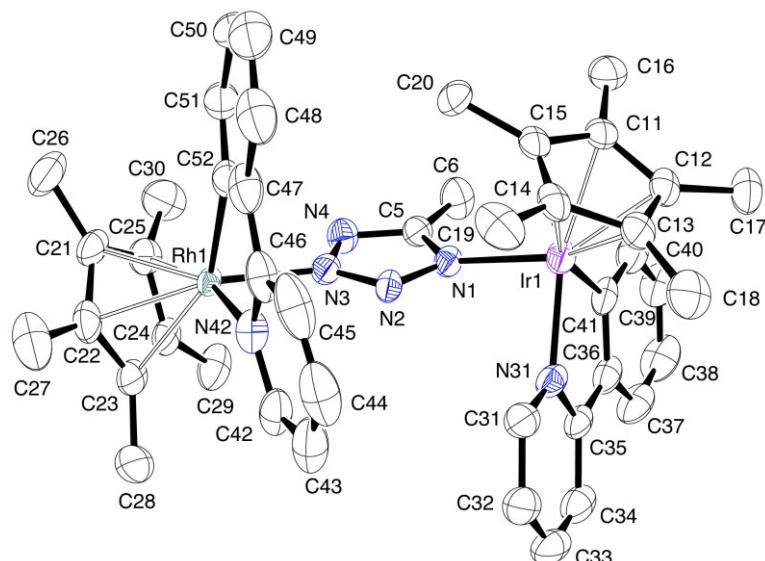


Fig. S10 ^1H NMR spectra (at 22 °C, 300 MHz) of a CD₃CN solution of **11** (the isolated crystals): (a) immediately after dissolving and (b) after standing at room temperature for 24 h.

(a)



(b)

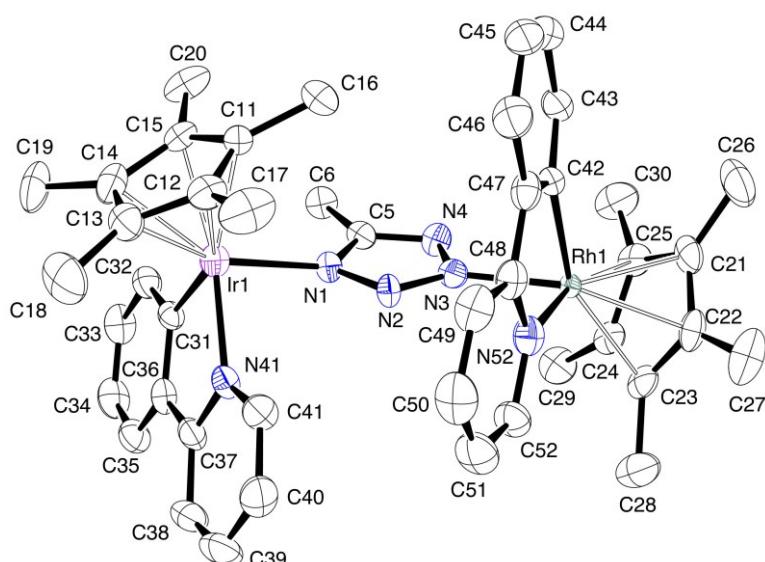


Fig. S11 ORTEPs (50% probability level, H atoms omitted for clarity) of the cationic parts in (a) $[\text{Cp}^*\text{Ir}(\text{ppy})\{\mu\text{-MeCN}_4\text{-}\kappa\text{N}^1(\text{Ir})\text{:}\kappa\text{N}^3(\text{Rh})\}\text{Rh}(\text{ppy})\text{Cp}^*]\text{BF}_4\bullet^{1/2}\text{CH}_3\text{CN}\bullet^{1/4}\text{H}_2\text{O}$ (**12B** $\bullet^{1/2}\text{CH}_3\text{CN}$ $\bullet^{1/4}\text{H}_2\text{O}$) and (b) $[\text{Cp}^*\text{Ir}(\text{ppy})\{\mu\text{-MeCN}_4\text{-}\kappa\text{N}^1(\text{Ir})\text{:}\kappa\text{N}^3(\text{Rh})\}\text{Rh}(\text{ppy})\text{Cp}^*]\text{BPh}_4\bullet\text{CH}_2\text{Cl}_2\bullet 2\text{Et}_2\text{O}$ (**12BP** $\bullet\text{CH}_2\text{Cl}_2\bullet 2\text{Et}_2\text{O}$).

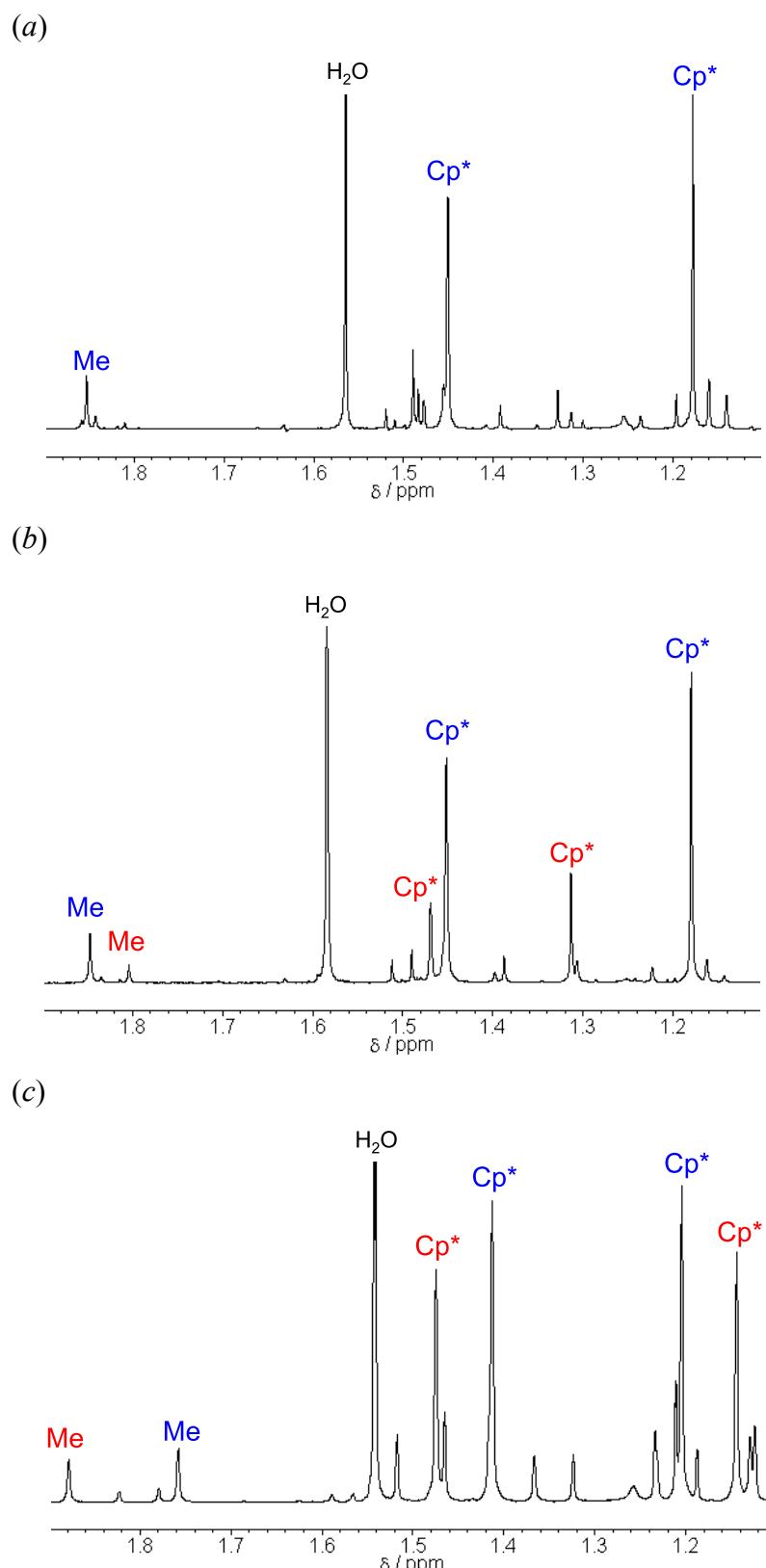


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Table S1 Crystallographic data for complexes.

	2	3	4•CH₂Cl₂	6•CH₃CN	8	8•CH₂Cl₂•Et₂O
Chemical formula	C ₂₂ H ₂₆ F ₆ N ₆ PRh	C ₂₃ H ₂₆ IrN ₅	C ₂₄ H ₂₈ Cl ₂ N ₅ Rh	C ₄₄ H ₅₂ F ₁₈ N ₉ P ₃ Rh ₂	C ₄₄ H ₄₉ F ₆ N ₆ PRh ₂	C ₄₉ H ₆₁ Cl ₂ F ₆ N ₆ OPRh ₂
Formula weight	622.37	564.69	560.32	1347.68	1012.68	1171.73
T / K	193(2)	192(2)	191(2)	193(2)	193(2)	193(2)
Crystal color and shape	yellow, block	yellow, columnar	orange, block	yellow, plate	yellow, plate	yellow, block
Size of specimen / mm	0.15 × 0.15 × 0.12	0.25 × 0.06 × 0.06	0.4 × 0.4 × 0.2	0.5 × 0.3 × 0.1	0.2 × 0.1 × 0.1	0.3 × 0.2 × 0.2
Crystal system	monoclinic	orthorhombic	monoclinic	monoclinic	triclinic	triclinic
Space group, Z	P2 ₁ /n, 4	Pbca, 8	C2/c, 8	P2 ₁ /c, 4	P $\bar{1}$, 2	P $\bar{1}$, 2
a / Å	8.8647(6)	8.7519(5)	34.5018(19)	12.9484(6)	13.0022(7)	13.8507(12)
b / Å	28.4028(15)	14.9783(7)	7.6323(5)	21.5240(9)	13.3883(8)	14.1135(12)
c / Å	9.9019(6)	31.9899(14)	22.3074(15)	20.3894(9)	14.0886(8)	14.1254(12)
α / °	90	90	90	90	71.928(3)	114.573(2)
β / °	93.770(4)	90	125.421(2)	108.669(1)	80.972(2)	96.515(3)
γ / °	90	90	90	90	63.631(2)	95.075(3)
U / Å ³	2487.7(3)	4193.5(3)	4787.0(5)	5383.6(4)	2088.4(2)	2466.7(4)
D _{calc} / Mg m ⁻³	1.662	1.789	1.555	1.663	1.610	1.578
μ (Mo K α) / mm ⁻¹	0.819	6.387	0.959	0.806	0.896	0.876
R _{int}	0.0436	0.0455	0.0222	0.0309	0.0653	0.0256
No. reflns / params.	5662/328	4801/268	5456/296	12324/682	9454/522	11230/579
R1 (F^2 : $F_o^2 > 2\sigma(F_o^2)$)	0.0497	0.0208	0.0246	0.0319	0.0613	0.0454
wR2 (F^2 : all data)	0.1219	0.0500	0.0666	0.0883	0.1833	0.1291
S	1.058	1.023	1.064	1.085	1.147	1.074

Table S1 (Continued)

	8B•$\frac{1}{2}$CH₃CN•$\frac{1}{4}$H₂O	9•CH₃CN	10•H₂O	11	12B•$\frac{1}{2}$CH₃CN•$\frac{1}{4}$H₂O	12BP•CH₂Cl₂•2Et₂O
Chemical formula	C ₄₅ H ₅₁ BF ₄ N _{6.5} O _{0.25} Rh ₂	C ₄₄ H ₅₂ F ₁₈ IrN ₉ P ₃ Rh	C ₄₃ H ₅₁ F ₁₂ IrN ₇ OP ₂ Rh	C ₄₃ H ₄₉ F ₁₂ IrN ₇ P ₂ Rh	C ₄₅ H ₅₁ BF ₄ IrN _{6.5} O _{0.25} Rh	C ₇₇ H ₉₁ BCl ₂ IrN ₆ O ₂ Rh
Formula weight	979.55	1436.97	1266.96	1248.94	1068.84	1509.38
T / K	185(2)	183(2)	193(2)	191(2)	192(2)	192(2)
Crystal color and shape	orange, block	yellow, needle	yellow, needle	yellow, needle	yellow, plate	orange, plate
Size of specimen / mm	0.4 × 0.4 × 0.2	0.2 × 0.1 × 0.1	0.3 × 0.15 × 0.15	0.35 × 0.15 × 0.1	0.4 × 0.2 × 0.1	0.35 × 0.3 × 0.05
Crystal system	triclinic	monoclinic	tetragonal	tetragonal	triclinic	triclinic
Space group, Z	P $\bar{1}$, 2	P2 ₁ /c, 4	P42 ₁ c, 8	P42 ₁ c, 8	P $\bar{1}$, 2	P $\bar{1}$, 2
a / Å	11.6756(8)	12.9890(4)	26.125(1)	26.0416(5)	11.6443(9)	13.7305(4)
b / Å	14.2627(9)	21.3738(5)	—	—	14.1518(8)	15.6553(9)
c / Å	14.4068(8)	20.4689(4)	14.4637(5)	14.3561(3)	14.4628(8)	18.2761(8)
α / °	99.704(2)	90	90	90	99.082(2)	67.466(3)
β / °	111.325(2)	109.311(1)	90	90	111.484(3)	76.436(2)
γ / °	96.083(2)	90	90	90	96.501(4)	82.757(2)
U / Å ³	2166.0(2)	5363.0(2)	9871.8(6)	9735.8(3)	2151.6(2)	3524.5(3)
D _{calc} / Mg m ⁻³	1.502	1.780	1.705	1.704	1.650	1.422
μ(Mo Kα) / mm ⁻¹	0.820	2.981	3.179	3.220	3.531	2.246
R _{int}	0.0217	0.0842	0.0737	0.0405	0.0349	0.0383
No. reflns / params.	9821/515	12261/670	11301/576	11104/567	9779/515	16053/811
R1 (F^2 : $F_o^2 > 2\sigma(F_o^2)$)	0.0404	0.0477	0.0508	0.0494	0.0463	0.0560
wR2 (F^2 : all data)	0.1079	0.1354	0.1389	0.1507	0.1241	0.1664
S	1.055	1.126	1.141	1.038	1.123	1.077

Table S2 Selected bond lengths (Å) and angles (°) of complexes **1–4**.

M = Ir or Rh	1^a	2	3	4•CH₂Cl₂
M1–N1 (MeCN ₄)	—	—	2.091(2)	2.0840(17)
M1–N2 (MeCN ₄)	2.075(4)	2.079(3)	—	—
M1–N31 (bpy)	2.096(4)	2.106(3)	—	—
M1–N40 (bpy)	2.086(4)	2.094(3)	—	—
M1–N21 (ppy)	—	—	2.093(2)	2.0932(16)
M1–C31 (ppy)	—	—	2.053(3)	2.0442(19)
M1–C11 (Cp*)	2.184(5)	2.183(4)	2.176(3)	2.1883(19)
M1–C12 (Cp*)	2.156(5)	2.148(4)	2.170(3)	2.2464(19)
M1–C13 (Cp*)	2.184(5)	2.177(4)	2.180(3)	2.2335(19)
M1–C14 (Cp*)	2.176(5)	2.173(4)	2.233(3)	2.1744(19)
M1–C15 (Cp*)	2.166(5)	2.158(4)	2.252(3)	2.1722(19)
N31–M1–N40	76.57(15)	77.06(12)	—	—
N21–M1–C31	—	—	77.97(11)	78.69(7)
N31–M1–N2	86.57(14)	87.82(12)	—	—
N40–M1–N2	85.27(15)	86.63(12)	—	—
N21–M1–N1	—	—	84.29(10)	86.40(6)
C31–M1–N1	—	—	91.42(10)	91.04(7)
M1–N2–N1	125.3(3)	125.4(2)	—	—
M1–N2–N3	123.0(3)	123.3(3)	—	—
M1–N1–N2	—	—	117.94(18)	116.62(12)
M1–N1–C5	—	—	135.4(2)	134.28(13)
N1–N2–N3	111.5(4)	111.1(3)	107.8(3)	108.06(16)
C5–N1–N2	103.1(4)	103.3(3)	106.2(2)	105.92(16)

^a M. Kotera, Y. Sekioka and T. Suzuki, *Inorg. Chem.*, 2008, **47**, 3498–3508.

Table S3 Selected bond lengths (Å) and angles (°) of complexes **5**• $2\text{H}_2\text{O}$, **6**• CH_3CN , and **9**• CH_3CN .

M = Ir or Rh	5 • $2\text{H}_2\text{O}^a$	6 • CH_3CN	M1 = Ir1 / Rh2 M2 = Rh1 / Ir2	9 • CH_3CN
M1–N1 (MeCN_4)	2.108(4)	2.138(2)	M2–N4 (MeCN_4)	2.133(4)
M1–N31 (bpy)	2.093(4)	2.104(2)	M2–N41 (bpy)	2.092(4)
M1–N40 (bpy)	2.094(4)	2.110(2)	M2–N50 (bpy)	2.108(4)
M2–N3 (MeCN_4)	2.092(3)	2.092(2)	M1–N2 (MeCN_4)	2.094(4)
M2–N41 (bpy)	2.093(4)	2.096(2)	M1–N31 (bpy)	2.099(4)
M2–N50 (bpy)	2.094(4)	2.104(2)	M1–N40 (bpy)	2.113(5)
M1–C11 (Cp^*)	2.164(4)	2.160(2)	M1–C11 (Cp^*)	2.151(6)
M1–C12 (Cp^*)	2.161(5)	2.165(2)	M1–C12 (Cp^*)	2.166(5)
M1–C13 (Cp^*)	2.171(5)	2.159(2)	M1–C13 (Cp^*)	2.172(5)
M1–C14 (Cp^*)	2.181(4)	2.184(2)	M1–C14 (Cp^*)	2.160(5)
M1–C15 (Cp^*)	2.169(4)	2.171(2)	M1–C15 (Cp^*)	2.168(5)
M2–C21 (Cp^*)	2.165(4)	2.162(2)	M2–C21 (Cp^*)	2.162(5)
M2–C22 (Cp^*)	2.161(4)	2.166(3)	M2–C22 (Cp^*)	2.171(5)
M2–C23 (Cp^*)	2.169(4)	2.161(3)	M2–C23 (Cp^*)	2.163(5)
M2–C24 (Cp^*)	2.168(4)	2.174(3)	M2–C24 (Cp^*)	2.175(5)
M2–C25 (Cp^*)	2.195(5)	2.152(3)	M2–C25 (Cp^*)	2.152(5)
N31–M1–N40	76.77(14)	76.89(8)	N50–M2–N41	77.06(17)
N31–M1–N1	89.68(14)	91.33(8)	N50–M2–N4	81.91(17)
N40–M1–N1	80.83(14)	81.90(8)	N41–M2–N4	90.36(17)
N41–M2–N50	76.62(15)	76.95(9)	N31–M1–N40	77.12(18)
N41–M2–N3	85.85(14)	86.28(8)	N31–M1–N2	85.62(17)
N50–M2–N3	85.24(14)	85.98(8)	N40–M1–N2	85.56(17)
M1–N1–N2	115.3(3)	114.30(15)	M2–N4–N3	114.2(3)
M1–N1–C5	135.9(3)	138.69(17)	M2–N4–C5	136.8(4)
C5–N1–N2	108.7(4)	106.7(2)	C5–N4–N3	108.7(4)
M2–N3–N2	123.1(3)	124.52(16)	M1–N2–N1	123.2(3)
M2–N3–N4	123.0(3)	123.34(16)	M1–N2–N3	123.3(3)
N2–N3–N4	111.8(4)	112.0(2)	N1–N2–N3	113.2(4)

^a M. Kotera, Y. Sekioka and T. Suzuki, *Inorg. Chem.*, 2008, **47**, 3498–3508.

Table S4 Selected bond lengths (Å) and angles (°) of complexes **8**, **8**• CH_2Cl_2 • Et_2O and **8B**• $\frac{1}{2}\text{CH}_3\text{CN}$ • $\frac{1}{4}\text{H}_2\text{O}$.

	8	8 • CH_2Cl_2 • Et_2O	12B • $\frac{1}{2}\text{CH}_3\text{CN}$ • $\frac{1}{4}\text{H}_2\text{O}$
Rh1–N1	2.110(5)	—	—
Rh1–N2	—	2.090(3)	2.092(3)
Rh1–N31	2.071(6)	2.086(3)	2.077(3)
Rh1–C41	2.061(7)	2.035(4)	2.042(3)
Rh2–N4	2.101(6)	2.092(3)	2.098(2)
Rh2–N42	2.092(5)	2.079(3)	2.093(3)
Rh2–C52	2.064(7)	2.055(3)	2.054(3)
Rh1–C11	2.210(7)	2.249(4)	2.245(3)
Rh1–C12	2.165(7)	2.160(4)	2.167(3)
Rh1–C13	2.204(7)	2.189(4)	2.186(3)
Rh1–C14	2.213(8)	2.166(4)	2.158(3)
Rh1–C15	2.190(7)	2.270(4)	2.253(3)
Rh2–C21	2.183(6)	2.173(4)	2.179(3)
Rh2–C22	2.207(7)	2.227(4)	2.249(3)
Rh2–C23	2.235(7)	2.221(4)	2.230(3)
Rh2–C24	2.187(6)	2.204(4)	2.188(3)
Rh2–C25	2.141(7)	2.180(4)	2.174(3)
N31–Rh1–C41	79.2(3)	78.99(15)	78.83(15)
N31–Rh1–N1	85.3(2)	—	—
N31–Rh1–N2	—	88.25(12)	87.92(11)
C41–Rh1–N1	91.8(2)	—	—
C41–Rh1–N2	—	83.29(14)	86.23(12)
N42–Rh2–C52	78.4(3)	78.77(14)	78.73(14)
N42–Rh2–N4	86.4(2)	92.12(12)	87.95(10)
C52–Rh2–N4	94.4(3)	88.13(12)	92.32(11)
Rh1–N1–N2	118.4(4)	—	—
Rh1–N1–C5	135.4(5)	—	—
C5–N1–N2	106.2(5)	103.7(3)	103.9(3)
Rh1–N2–N3	—	125.1(2)	125.5(2)
Rh1–N2–N1	—	122.2(2)	122.58(19)
N1–N2–N3	108.9(5)	112.2(3)	111.7(3)
Rh2–N4–N3	119.2(4)	118.7(2)	116.17(19)
Rh2–N4–C5	133.5(5)	133.7(3)	135.3(2)
N3–N4–C5	106.9(6)	106.4(3)	106.9(2)

Table S5 Selected bond lengths (Å) and angles (°) of complexes **10**• H_2O and **11**.

10 • H_2O		11	
Ir1–N2	2.078(7)	Rh1–N3	2.104(6)
Ir1–N31	2.082(8)	Rh1–N42	2.098(7)
Ir1–N40	2.108(8)	Rh1–N51	2.082(8)
Rh1–N4	2.122(7)	Ir1–N1	2.094(7)
Rh1–N41	2.102(7)	Ir1–N31	2.074(6)
Rh1–C51	2.054(8)	Ir1–C41	2.054(7)
Ir1–C11	2.197(8)	Ir1–C11	2.146(9)
Ir1–C12	2.191(9)	Ir1–C12	2.230(9)
Ir1–C13	2.183(9)	Ir1–C13	2.237(9)
Ir1–C14	2.185(9)	Ir1–C14	2.162(8)
Ir1–C15	2.158(9)	Ir1–C15	2.174(8)
Rh1–C21	2.176(9)	Rh1–C21	2.176(8)
Rh1–C22	2.170(9)	Rh1–C22	2.159(8)
Rh1–C23	2.230(8)	Rh1–C23	2.149(8)
Rh1–C24	2.227(8)	Rh1–C24	2.165(8)
Rh1–C25	2.156(9)	Rh1–C25	2.205(8)
N31–Ir1–N40	76.8(3)	N42–Rh1–N51	77.9(3)
N31–Ir1–N2	82.8(3)	N42–Rh1–N3	87.3(3)
N40–Ir1–N2	86.7(3)	N51–Rh1–N3	83.5(3)
N41–Rh1–C51	77.9(3)	N31–Ir1–C41	78.4(3)
N41–Rh1–N4	87.9(3)	N31–Ir1–N1	87.3(2)
C51–Rh1–N4	92.9(3)	C41–Ir1–N1	93.4(3)
Ir1–N2–N1	120.3(5)	Rh1–N3–N4	122.0(5)
Ir1–N2–N3	126.9(5)	Rh1–N3–N2	125.7(5)
N1–N2–N3	111.9(6)	N4–N3–N2	111.3(6)
Rh1–N4–N3	116.8(5)	Ir1–N1–N2	118.5(5)
Rh1–N4–C5	134.2(5)	Ir1–N1–C5	133.1(5)
C5–N4–N3	108.0(7)	C5–N1–N2	107.5(7)

Table S6 Selected bond lengths (Å) and angles ($^\circ$) of complexes **12B[•]** $\cdot\frac{1}{2}\text{CH}_3\text{CN}^{\bullet}\cdot\frac{1}{4}\text{H}_2\text{O}$ and **12BP[•]** $\cdot\text{CH}_2\text{Cl}_2\cdot2\text{Et}_2\text{O}$.

	12B[•] $\cdot\frac{1}{2}\text{CH}_3\text{CN}^{\bullet}\cdot\frac{1}{4}\text{H}_2\text{O}$		12BP[•] $\cdot\text{CH}_2\text{Cl}_2\cdot2\text{Et}_2\text{O}$
Ir1–N1	2.086(4)	Ir1–N1	2.099(5)
Ir1–N31	2.094(5)	Ir1–N41	2.090(5)
Ir1–C41	2.061(6)	Ir1–C31	2.043(6)
Rh1–N3	2.098(5)	Rh1–N3	2.082(5)
Rh1–N42	2.082(6)	Rh1–N52	2.093(5)
Rh1–C52	2.046(6)	Rh1–C42	2.055(6)
Ir1–C11	2.183(6)	Ir1–C11	2.233(6)
Ir1–C12	2.157(6)	Ir1–C12	2.251(6)
Ir1–C13	2.184(6)	Ir1–C13	2.164(7)
Ir1–C14	2.245(6)	Ir1–C14	2.148(7)
Ir1–C15	2.220(6)	Ir1–C15	2.161(6)
Rh1–C21	2.176(6)	Rh1–C21	2.195(6)
Rh1–C22	2.152(6)	Rh1–C22	2.168(6)
Rh1–C23	2.254(6)	Rh1–C23	2.228(6)
Rh1–C24	2.243(6)	Rh1–C24	2.222(6)
Rh1–C25	2.166(6)	Rh1–C25	2.200(6)
N31–Ir1–C41	77.8(2)	N41–Ir1–C31	78.6(2)
N31–Ir1–N1	86.28(19)	N41–Ir1–N1	88.15(19)
C41–Ir1–N1	91.4(2)	C31–Ir1–N1	92.3(2)
N42–Rh1–C52	78.5(3)	N52–Rh1–C42	78.5(2)
N42–Rh1–N3	87.7(2)	N52–Rh1–N3	87.4(2)
C52–Rh1–N3	86.4(2)	C42–Rh1–N3	84.9(2)
Ir1–N1–N2	117.3(4)	Ir1–N1–N2	117.4(4)
Ir1–N1–C5	133.8(4)	Ir1–N1–C5	133.5(4)
C5–N1–N2	107.3(5)	C5–N1–N2	106.7(5)
Rh1–N3–N4	122.4(4)	Rh1–N3–N4	121.1(4)
Rh1–N3–N2	124.9(4)	Rh1–N3–N2	126.5(4)
N4–N3–N2	112.3(5)	N4–N3–N2	112.1(5)