

Supporting Information

Modulating the $M_2\delta$ -to-Ligand Charge Transfer Transition by the use of Diarylboron Substituents

Samantha E. Brown-Xu, Malcolm H. Chisholm*, Christopher B. Durr, Thomas F. Spilker, and Philip J. Young

Department of Chemistry and Biochemistry, The Ohio State University, 100 West 18th Avenue, Columbus, Ohio 43210-1185

Figure S1. The electronic absorption and emission spectra of **I**

Figure S2. The phosphorescence spectra of **I**

Figure S3. Cyclic voltammogram of compounds **I** and **II**

Figure S4. Kinetic trace from fs-TA spectra of **I** in THF

Figure S5. fs-TA spectrum of **I** in toluene

Figure S6. Kinetic trace from fs-TA spectra of **I** in toluene

Figure S7. fs-TA spectrum of **II**

Figure S8. Kinetic trace from fs-TA spectra of **II**

Figure S9. ns-TA spectrum of **I** in THF

Figure S10. Kinetic trace from ns-TA spectra of **I**

Figure S11. ns-TA spectrum of **II**

Figure S12. Kinetic trace from ns-TA spectra of **II**

Figure S13. Absorption spectrum of **II** in THF titrated with fluoride ions.

Table S1. Crystallographic supporting information

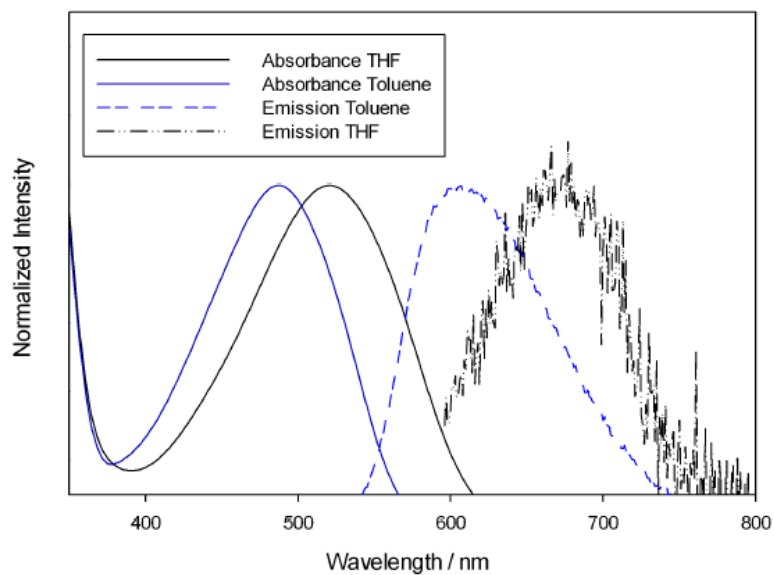


Fig S1. The electronic absorption and emission spectra of **I** in toluene and THF.

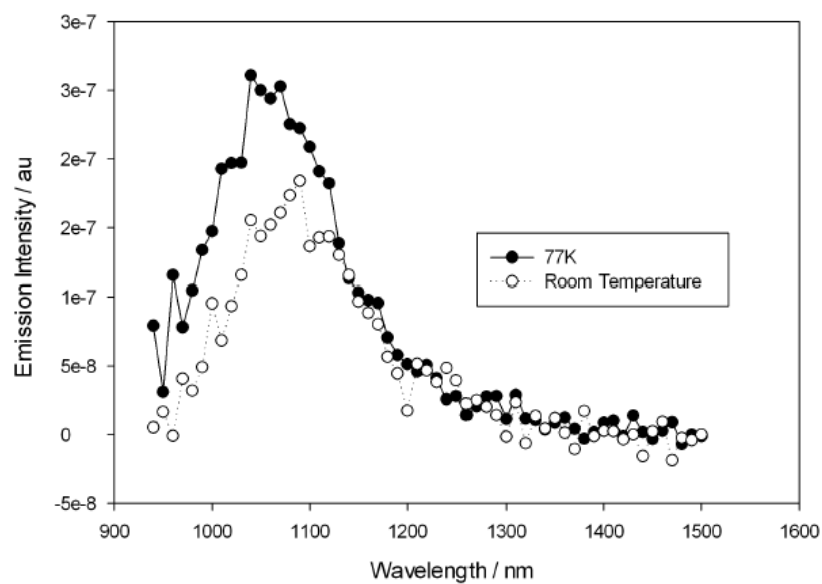


Fig S2. The phosphorescence spectra of **I** at RT and 77 K, $\lambda_{\text{ex}} = 405$.

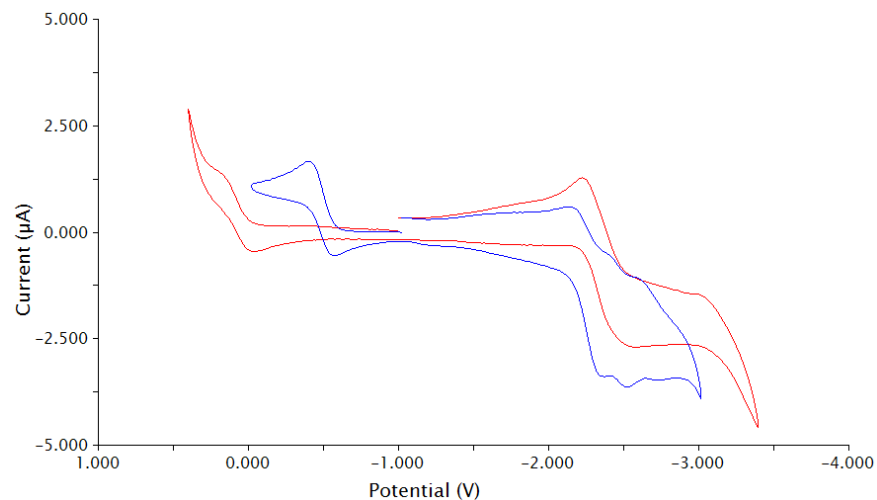


Fig S3. Cyclic voltammogram of compounds **I** and **II** in THF.

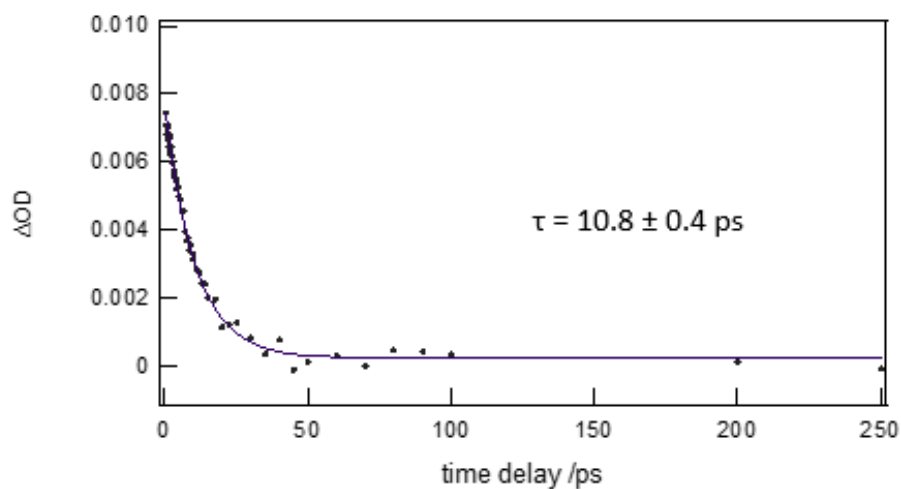


Fig S4. Kinetic trace from fsTA spectra of **I** in THF monitored at 390 nm.

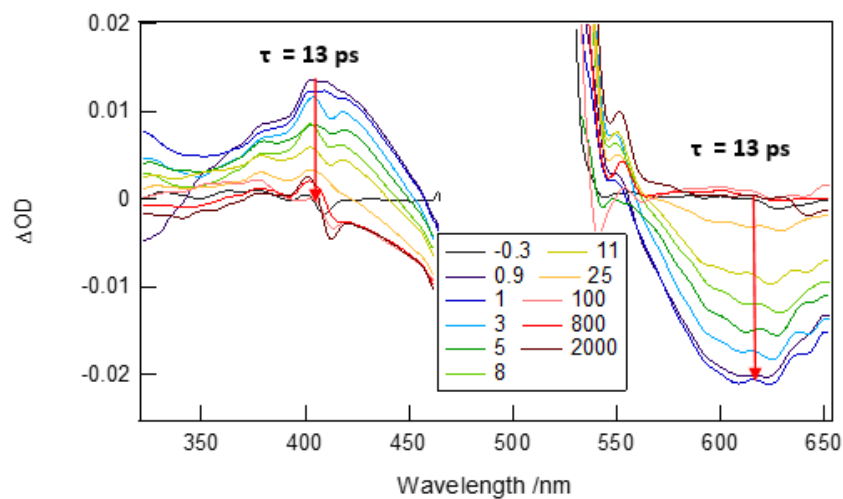


Fig S5. fs-TA spectrum of **I** in toluene at RT, $\lambda_{\text{ex}} = 568$ nm.

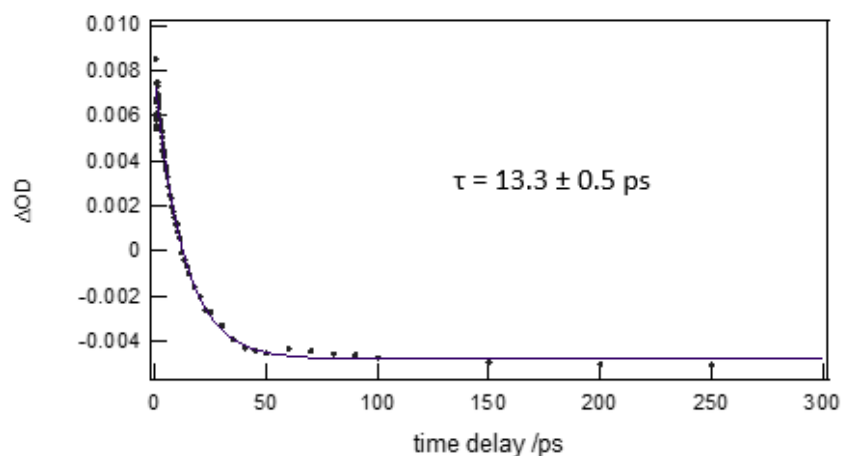


Fig S6. Kinetic trace from fsTA spectra of **I** in toluene monitored at 419 nm.

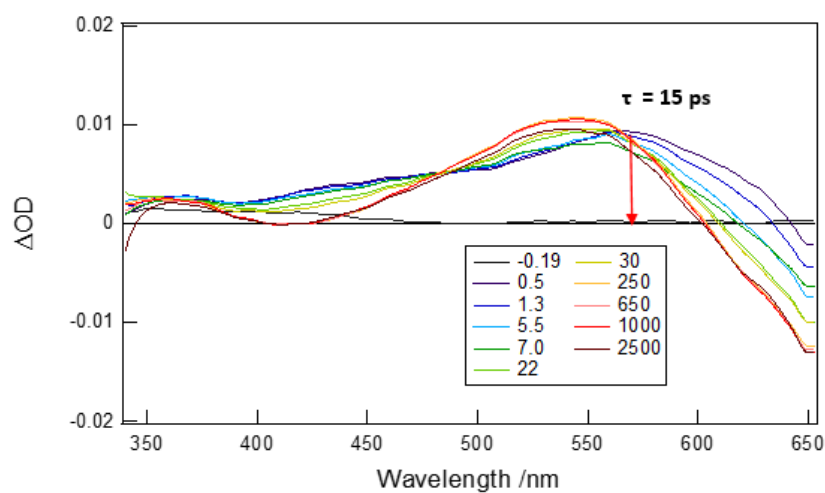


Fig S7. fs-TA spectrum of **II** in THF at RT, $\lambda_{\text{ex}} = 800$ nm.

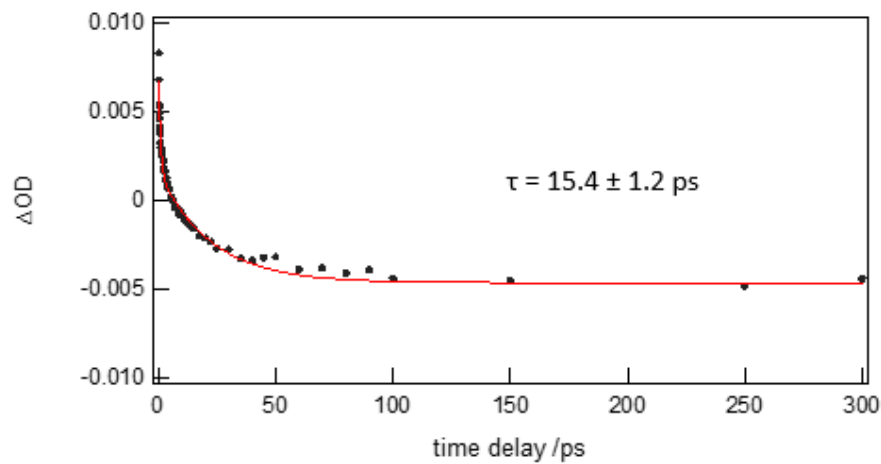


Fig S8. Kinetic trace from fsTA spectra of **II** monitored at 620 nm.

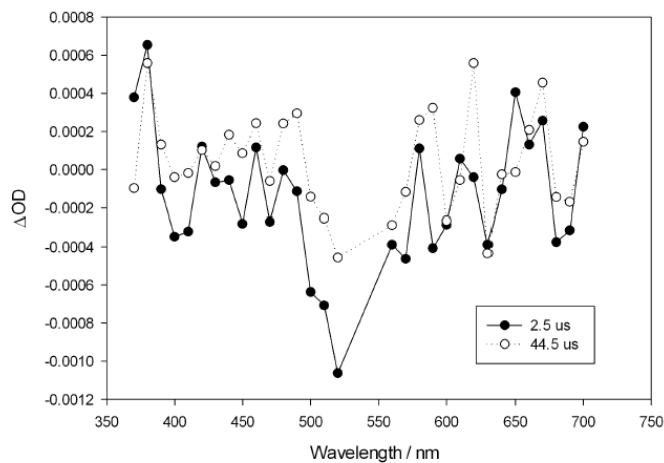


Fig S9. ns-TA spectrum of **I** in THF at RT, $\lambda_{\text{ex}} = 532$ nm.

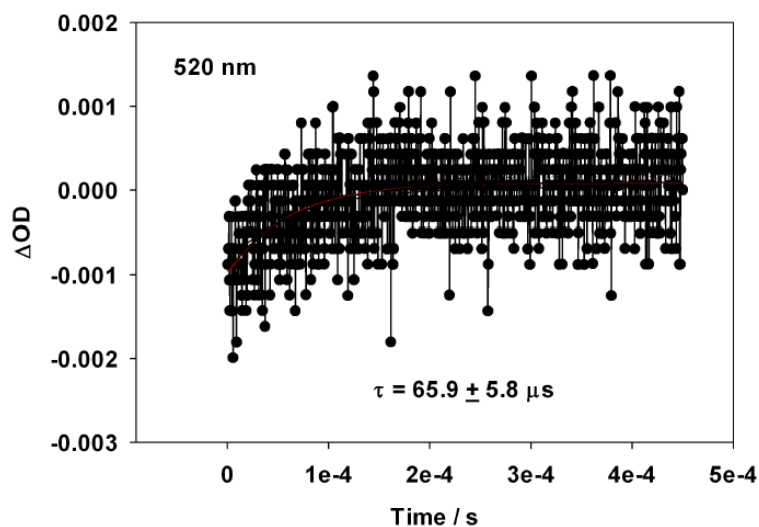


Fig S10. Kinetic trace from nsTA spectra of **I** monitored at 520 nm.

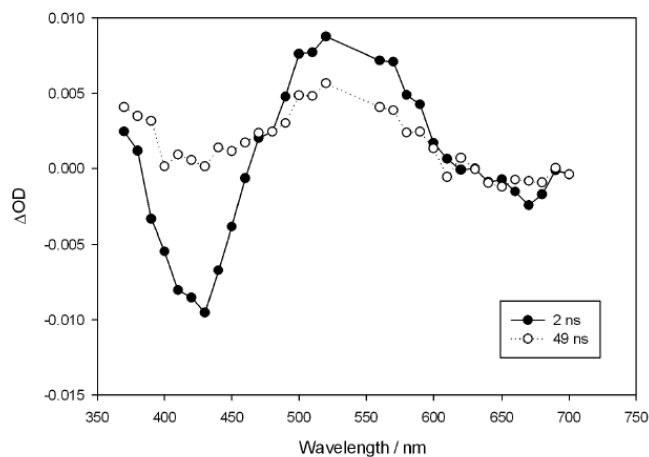


Fig S11. ns-TA spectrum of **II** in THF at RT, $\lambda_{\text{ex}} = 355$ nm.

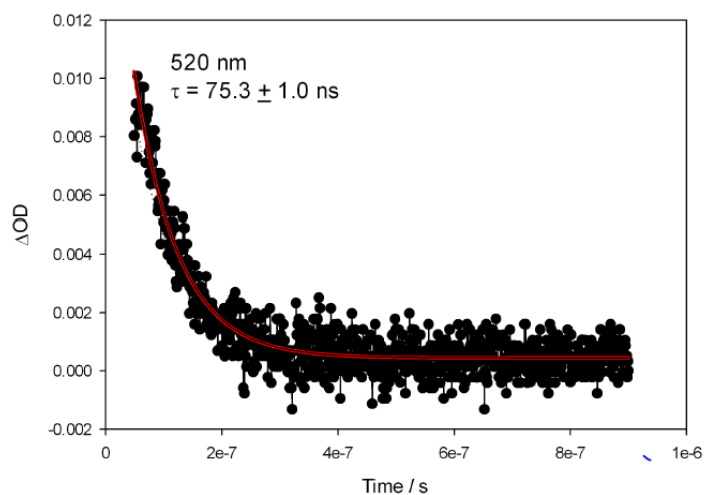


Fig S12. Kinetic trace from nsTA spectra of **II** monitored at 520 nm.

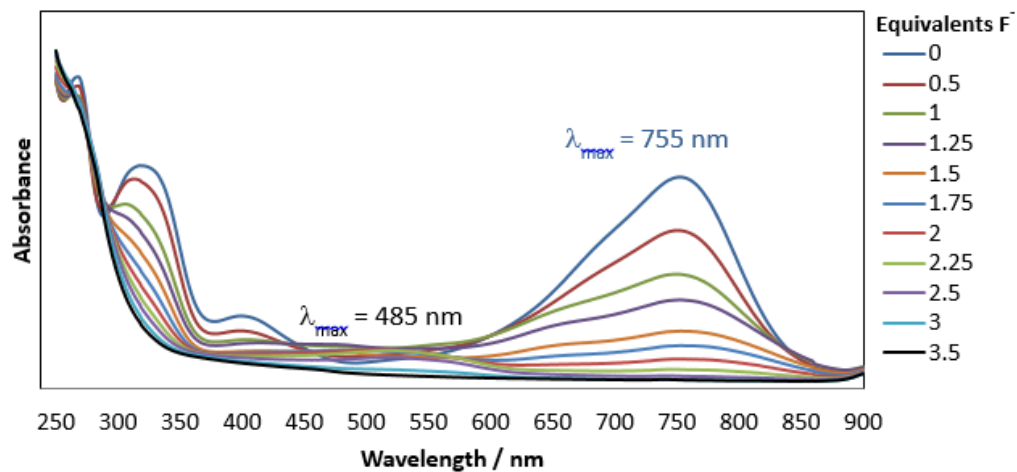


Fig S13. Absorption spectrum of **II** in THF titrated with fluoride ions. The compound decomposes upon the addition of water and could not be regenerated.

Table S1. Crystallographic supporting information.

Compound	Mo ₂ (TPB) ₂ (Bz) ₂
Chemical formula	C ₅₄ H ₇₂ Mo ₂ O ₁₀
Formula weight	1072.99
Temperature (K)	150(2)
Space group	Triclinic, P-1
<i>a</i> (Å)	10.2136(2)
<i>b</i> (Å)	10.3526(2)
<i>c</i> (Å)	12.6092(2)
α (°)	83.0670(10)
β (°)	80.8170(10)
γ (°)	84.8880(10)
<i>V</i> (Å ³)	1303.23(4)
<i>Z</i>	1
<i>D</i> _{calcd} (Mg/m ³)	1.367
Crystal Size (mm)	0.31 X 0.15 X 0.12
Theta range for data collection	0.74 to 27.49°
μ , (Mo, K α) (mm ⁻¹)	0.536
<i>F</i> (000)	560
Reflections collected	38070
Unique reflections	5984 [R(int)= 0.059]
Completeness to $\theta_{\max}$	99.7%
Data/restraints/parameters	5984 / 103 / 462
<i>R</i> ¹ (%) (all data)	4.26 (7.75)
<i>wR</i> ² (%) (all data)	10.05 (13.63)
Goodness-of-fit on <i>F</i> ²	1.119
Largest diff. peak and hole (e Å ⁻³)	0.647 and -1.192

$$aR1 = \Sigma ||F_o|-|F_c|| / \Sigma |F_o| \times 100$$

$$b_wR2 = \{ \Sigma [w(F_o^2 - F_c^2)^2] / \Sigma [w(F_o^2)^2] \}^{1/2} \times 100$$