

Supporting information: Experimental conditions for potentiometry and microcalorimetry

Table S1. Potentiometric titrations of Fe(III)/ligand A

No.	Cup Solution					Titrant
	t, °C	V ⁰ , ml	C _A , mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _{Fe} , mmol•dm ⁻³	C _{NaOH} , mol•dm ⁻³
1	25	20.0	2.00	4.80	1.00	0.1013
2	25	20.0	9.50	23.6	4.05	0.1013
3	25	20.0	7.10	17.7	3.10	0.1013

Table S2. Potentiometric titrations of Cu(II)/ligand A

No.	Cup Solution					Titrant
	t, °C	V ⁰ , ml	C _A , mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _{Cu} , mmol•dm ⁻³	C _{NaOH} , mol•dm ⁻³
1	25	20.0	3.00	20.0	1.50	0.1066
2	25	20.0	2.25	17.5	1.50	0.1066
3	25	20.0	1.50	17.5	1.50	0.1066

Table S3. Potentiometric titrations of Pb(II)/ligand A

No.	Cup Solution					Titrant
	t, °C	V ⁰ , ml	C _A , mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _{Pb} , mmol•dm ⁻³	C _{NaOH} , mol•dm ⁻³
1	25	20.0	3.00	16.0	1.00	0.1066
2	25	20.0	2.00	13.0	1.00	0.1066
3	25	20.0	2.00	10.0	1.00	0.1066

Table S4. Potentiometric titrations of Ni(II)/ligand A

No.	Cup Solution					Titrant
	t, °C	V ⁰ , ml	C _A ,mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _{Ni} , mmol•dm ⁻³	C _{NaOH} , mol•dm ⁻³
1	25	20.0	3.00	18.0	1.50	0.1066
2	25	20.0	4.00	22.5	2.00	0.1066
3	25	20.0	4.00	20.0	2.00	0.1066

Table S5. Calorimetric titrations of Fe(III)/ligand A

No.	Cup Solution						Titrant	
	t, °C	V ⁰ , µl	C _A , mmol•dm ⁻³	C _{Fe} , mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _{NaOH} , mmol•dm ⁻³	C _H , mmol•dm ⁻³	C _A , mmol•dm ⁻³
1	25	900	1.889	0.911	4.079	21.32		
2	25	900	1.422	0.637	2.844	21.32		
3	25	900	1.422	0.711	1.777	10.66		
4	25	900	1.889	0.911	-1.132		20.0	
5	25	900		1.00	10.0		15.0	7.50
6	25	900		1.00	10.0		20.0	10.0