

New series of hybrid fluoroferrates synthesized with triazoles: various dimensionalities and magnetic properties

Mouna Smida,^{a,b} Jérôme Lhoste,^{a,*} Vanessa Pimenta,^a Annie Ribaud,^a Laurent Jouffret,^a Marc Leblanc,^a Jean-Marc Grenèche,^a Mohamed Dammak,^b and Vincent Maisonneuve.^{a,*}

^a*LUNAM Université du Maine. IMMM-UMR 6283 CNRS. Faculté des Sciences. 72085 Le Mans. Cedex 9. France*

^b*Laboratoire de Chimie Inorganique. Faculté des Sciences de Sfax. Université de Sfax. B.P 1171 Sfax 3000. Tunisia*

* To whom correspondence should be addressed. E-mail: jerome.lhoste@univ-lemans.fr Phone: (33).(0)2 43 83 35 61. Fax: (33).(0)2 43 83 35 095

SUPPLEMENTARY INFORMATION

Figure S1. Composition space of the $\text{FeF}_2\text{-FeF}_3\text{-Htaz-HF}_{\text{aq.}}\text{-DMF}$ system established under solvothermal conditions (120°C , 72 h) for a constant concentration $[\text{Fe}^{\text{II}}] + [\text{Fe}^{\text{III}}] = 0.15 \text{ mol.L}^{-1}$ and a ratio $[\text{Fe}^{\text{II}}]/[\text{Fe}^{\text{III}}] = 1$.

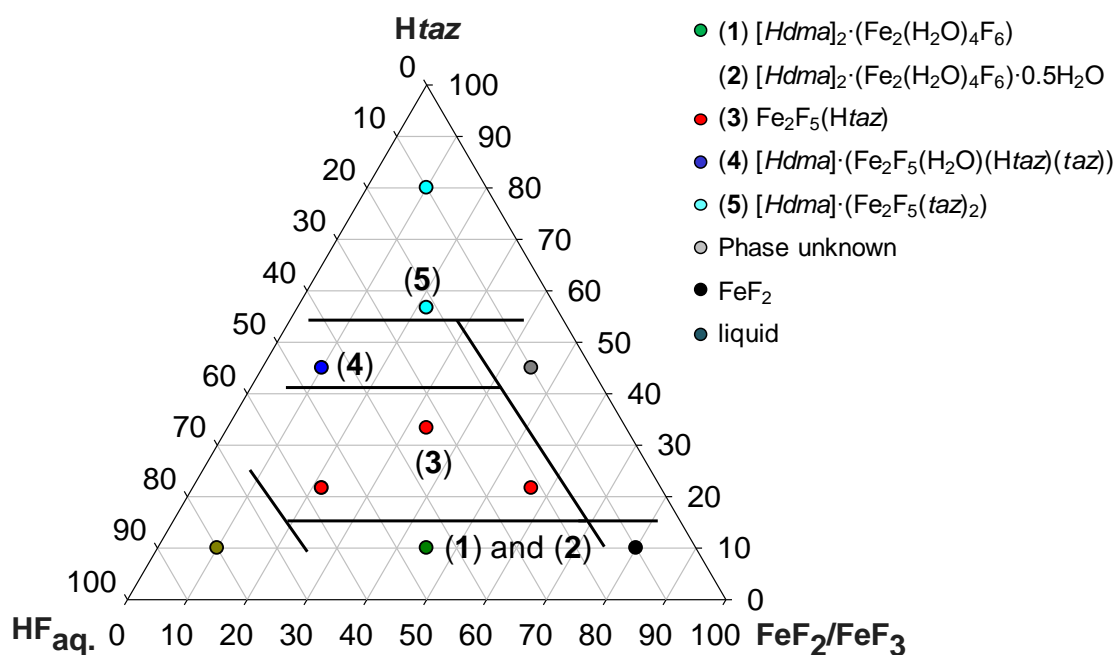


Figure S2. Diagram pattern obtained in the $\text{FeF}_2\text{-FeF}_3\text{-Htaz-HF}_{\text{aq.}}\text{-DMF}$ system with a 2/2/1/2/323 molar ratio : (1) and (2)

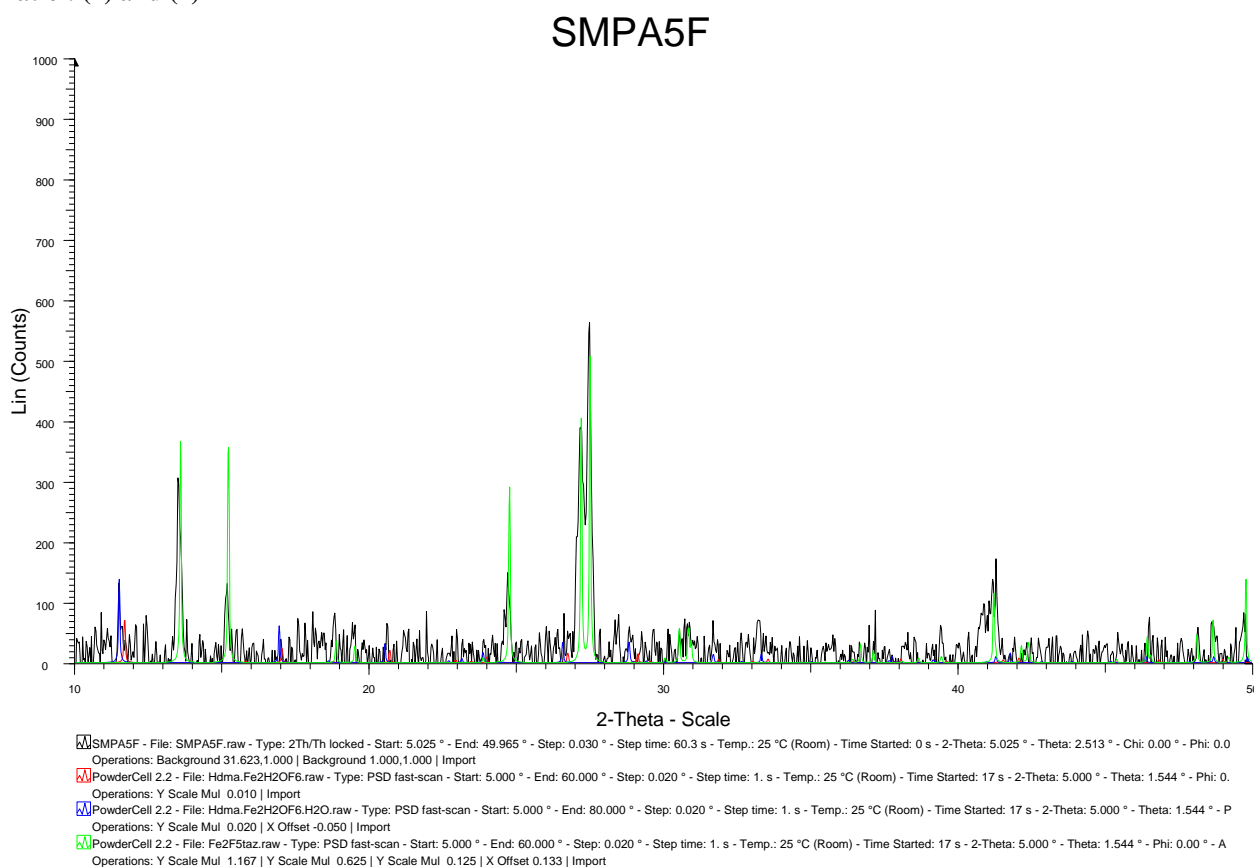


Figure S3. Diagram pattern obtained in the $\text{FeF}_2\text{-FeF}_3\text{-Htaz-HF}_{\text{aq}}\text{-DMF}$ system with a 2/2/17/17/323 molar ratio (4)

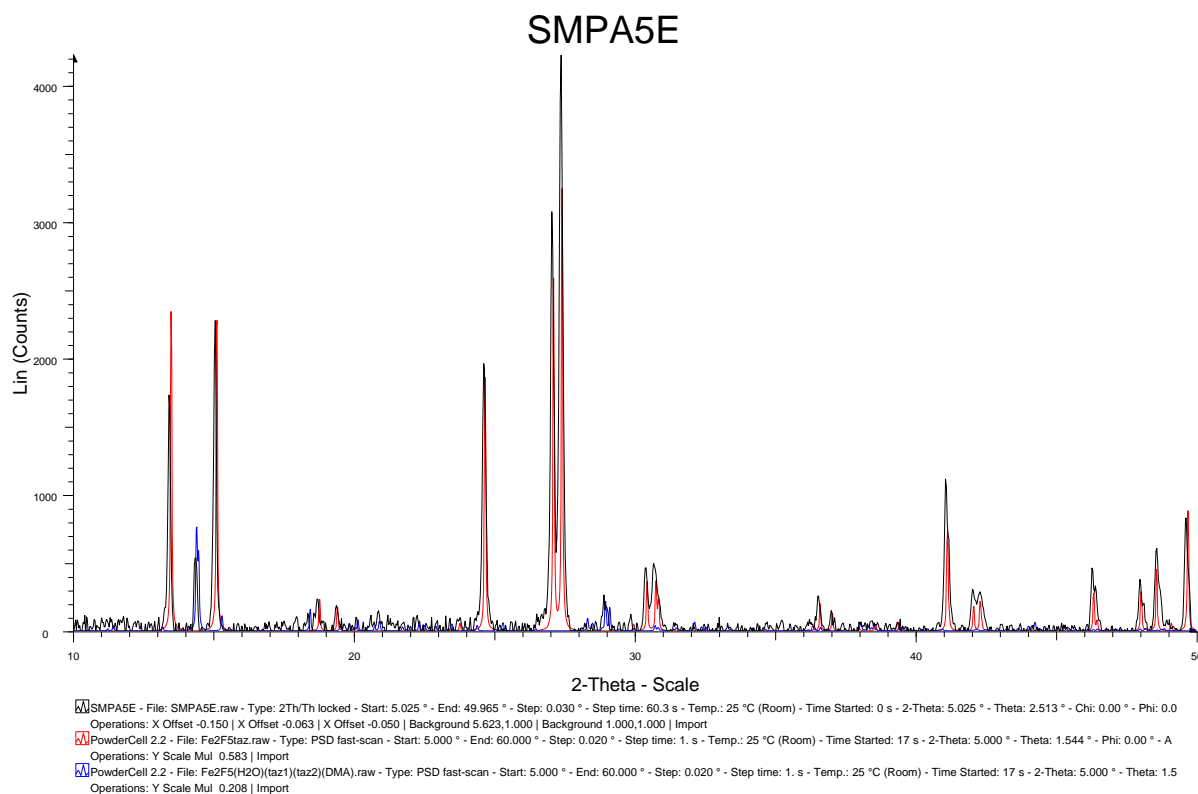


Figure S4. Diagram pattern obtained in the $\text{FeF}_2/\text{FeF}_3\text{-Htaz-HF}_{\text{aq}}\text{-DMF}$ system with a 2/2/10/4/323 molar ratio (5)

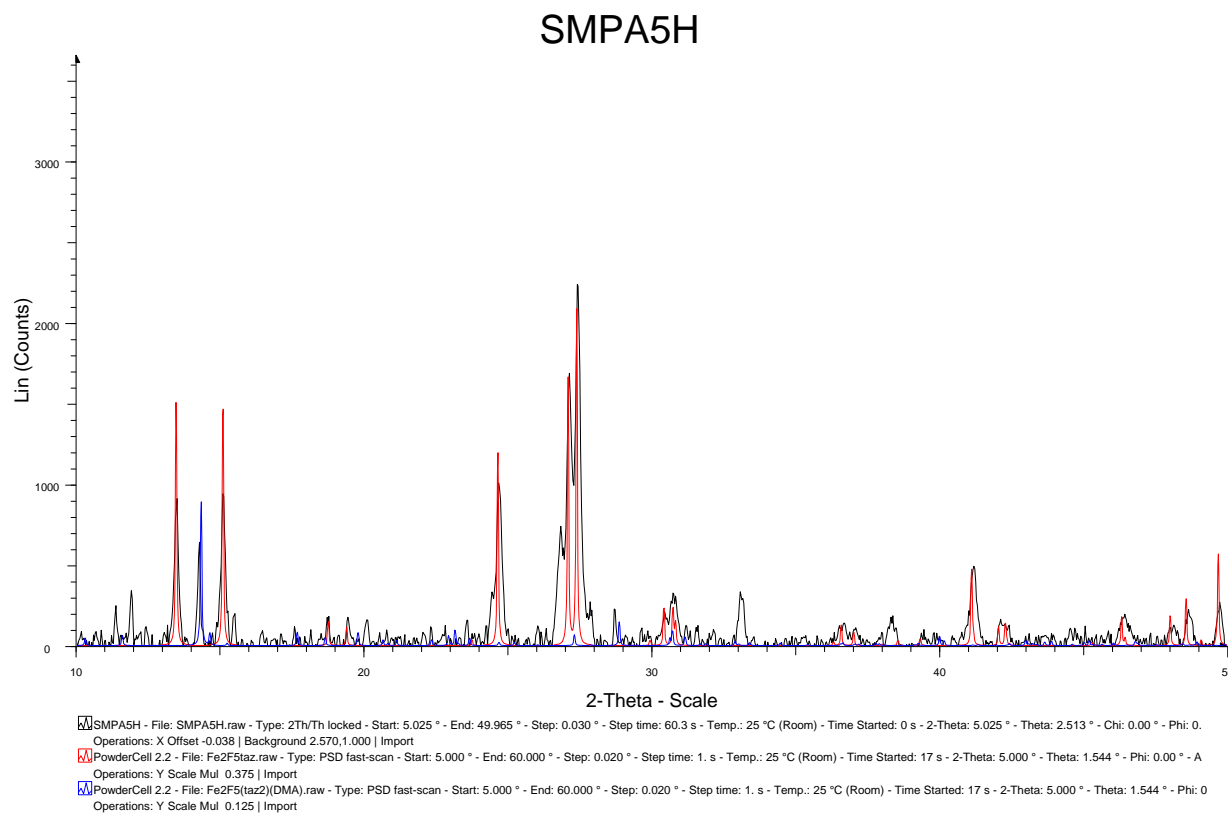


Table S1. Atomic coordinates and equivalent APD of [Hdma]·(Fe₂(H₂O)₄F₆) (1)

Atom	Wyckoff position	x	Y	z	B _{eq} (Å ²)
Fe(1)	4a	½	½	½	1,29(2)
Fe(2)	4b	½	1	½	1,29(2)
F(1)	8f	½	0.7570(2)	0.54882(17)	1,86(5)
F(2)	16h	0.63082(15)	0.9287(2)	0.41707(11)	2,15(3)
OW(1)	16h	0.36125(18)	0.4262(2)	0.59871(14)	1,95(4)
C(1)	8g	0.3813(7)	0.7462(6)	¾	3,79(11)
N(1)	4c	½	0.6359(7)	¾	2,88(9)

Table S2. Atomic coordinates and equivalent APD of [Hdma]·(Fe₂(H₂O)₄F₆)·0.5H₂O (2)

Atom	Wyckoff position	x	y	z	B _{eq} (Å ²)
Fe(1)	4a	½	½	½	1,34(2)
Fe(2)	4b	½	1	½	1,27(2)
F(1)	8f	½	0.7570(2)	0.54779(17)	1,82(4)
F(2)	16h	0.63035(16)	0.9294(2)	0.41655(12)	2,30(4)
OW(1)	16h	0.36143(18)	0.4269(3)	0.59472(14)	2,07(4)
C(1)	8g	0.3826(7)	0.7331(8)	¾	4,02(12)
N(1)	4c	½	0.61587(7)	¾	4,03(14)
OW(2)	4c	0	0.7109(8)	¾	5,52(40)

Table S3. Atomic coordinates and equivalent APD of Fe₂F₅(Htaz) (3)

Atom	Wyckoff position	x	y	z	B _{eq} (Å ²)
Fe(1)	4a	½	½	½	1,10(8)
Fe(2)	4c	¾	¾	¾	1,67(8)
F(1)	4e	½	¾	0.4318(5)	1,11(2)
F(2)	16j	0.6461(3)	0.5537(4)	0.6367(2)	1,56(1)
N(1)	8i	0.9258(5)	¾	0.5845(6)	1,74(3)
N(2)	4e	1	¾	0.3695(7)	1,74(3)
C(1)	8i	0.8845(7)	¾	0.4590(9)	3,47(4)

Table S4. Atomic coordinates and equivalent APD of [Hdma]·(Fe₂F₅(H₂O)(Htaz)(taz)) (4)

Atom	Wyckoff position	x	y	z	B _{eq} (Å ²)
Fe(1)	2i	0.32425(7)	0.27136(6)	0.38112(6)	1,24(15)
Fe(2)	1a	0.0000	0.0000	0.0000	1,12(2)
Fe(3)	1e	0.5000	0.5000	0.0000	1,22(2)
F(1)	2i	0.4751(3)	0.4337(2)	0.2221(2)	1,76(4)
F(2)	2i	0.1494(3)	0.4150(3)	0.4347(3)	2,05(5)
F(3)	2i	0.3709(3)	0.3357(3)	0.5416(3)	2,17(5)
F(4)	2i	0.1617(3)	0.1213(2)	0.5193(2)	1,95(4)
F(5)	2i	0.5015(3)	0.1403(3)	0.3306(3)	2,31(5)
O(1)	2i	0.2102(4)	0.0119(3)	-0.2356(3)	2,06(6)
N(6)	2i	0.0976(4)	0.8035(4)	0.0830(4)	1,52(6)
N(1)	2i	0.2656(4)	0.2266(4)	0.1986(4)	1,56(6)
N(2)	2i	0.1536(4)	0.1337(4)	0.0614(4)	1,67(6)
N(3)	2i	0.3285(4)	0.3105(4)	0.0520(4)	1,57(6)
N(4)	2i	0.2846(4)	0.6418(4)	0.0915(4)	1,88(6)
N(5)	2i	0.1320(4)	0.6079(4)	0.2177(4)	1,78(6)
N(7)	2i	0.3087(5)	0.6561(4)	0.5018(5)	3,45(9)
C(1)	2i	0.1631(5)	0.1239(4)	0.1982(4)	1,68(7)
C(2)	2i	0.2590(5)	0.2503(4)	-0.0243(5)	1,80(7)
C(3)	2i	0.0240(5)	0.7056(4)	0.2111(4)	1,65(7)
C(4)	2i	0.2583(4)	0.7608(3)	0.0132(3)	1,97(8)
C(5)	2i	0.2947(4)	0.7963(3)	0.4075(3)	5,09(14)
C(6)	2i	0.2019(4)	0.6460(3)	0.6726(3)	4,43(12)

Table S5. Atomic coordinates and equivalent APD of [Hdma]·(Fe₂F₅(taz)₂) (5)

Atom	Wyckoff position	x	y	z	B _{eq} (Å ²)
Fe(1)	4c	0.42845(7)	¼	0.31073(15)	1,64(3)
Fe(2)	4c	0.22819(7)	¼	0.24328(16)	1,25(2)
F(1)	4c	0.4961(3)	¼	0.4773(7)	3,15(13)
F(2)	4c	0.3368(3)	¼	0.1618(6)	1,85(9)
F(3)	8d	0.4739(2)	0.4041(4)	0.1949(5)	3,11(9)
F(4)	4c	0.1316(3)	¼	0.3356(6)	2,89(12)
N(1)	8d	0.2256(3)	-0.0909(6)	-0.0997(6)	1,73(9)
N(2)	8d	0.1464(3)	-0.0892(6)	-0.0683(6)	1,73(9)
N(3)	8d	0.2033(3)	0.0805(6)	0.0811(6)	1,66(9)
C(1)	8d	0.1362(4)	0.0140(7)	0.0400(7)	1,62(11)
C(2)	8d	0.2573(3)	0.0127(7)	-0.0078(6)	1,58(11)
C(4)	4c	0.4012(3)	¾	0.1994(6)	3,80(2)
C(3)	4c	0.5386(3)	¾	0.1466(6)	4,12(2)
N(4)	4c	0.4624(3)	¾	0.1006(6)	13,2(8)

Table S6. Selected H-bond, D-H... A (D = N, O ; A = F, O) distances (Å) in [Hdma]·(Fe₂(H₂O)₄F₆) (1), [Hdma]·(Fe₂(H₂O)₄F₆)·0.5H₂O (2), [Hdma]·(Fe₂F₅(Htaz)(taz)) (4) and [Hdma]·(Fe₂F₅(taz)₂) (5)

D-H...A	d(D...A)
(1)	
OW(1)-H(1)WA...F(2)	2.644(3)
OW(1)-H(1) WB...F(2)	2.639(3)
N(1)-H(1)A... OW(1) x2	3.123(4)
N(1)-H(1)B...OW(1) x2	3.123(4)
(2)	
OW(1)-H(1)WA...F(2)	2.642(3)
OW(1)-H(1) WB...F(2)	2.655(3)
N(1)-H(1)A... OW(1) x2	3.128(4)
N(1)-H(1)B...OW(1) x2	3.128(4)
(3)	
N(5)-H(5)...F(2)	2.599 (5)
N(7)-H(7)A...F(2)	2.932(6)
N(7)-H(7)A...F(3)	2.993(5)
N(7)-H(7)B...F(3)	2.674(6)
(4)	
N(4)-H(4)A...F(3) x 2	3.079(8)
N(4)-H(4)A...F(4)	2.779(8)

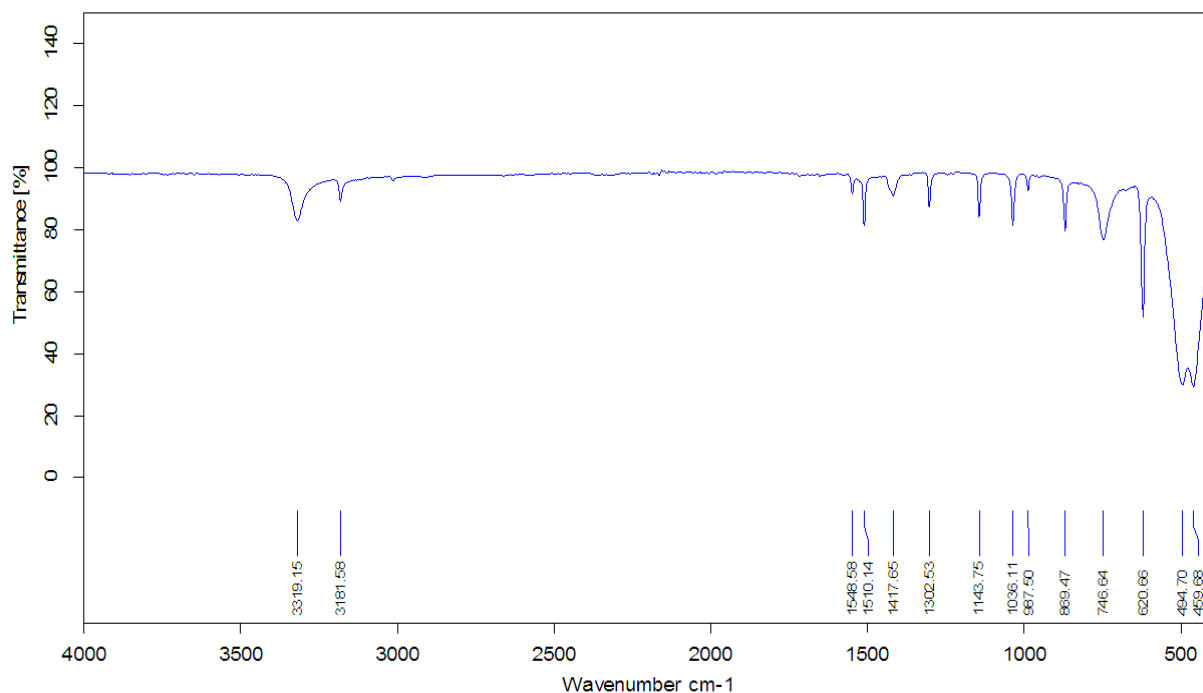


Figure S5. Infrared spectroscopy de Fe₂F₅(Htaz)