

SUPPLEMENTARY INFORMATION

Title: Copper(II) and Triphenylphosphine Copper(I) Ethylene Glycol Carboxylates: Synthesis, Characterisation and Copper Nanoparticle Generation

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Content

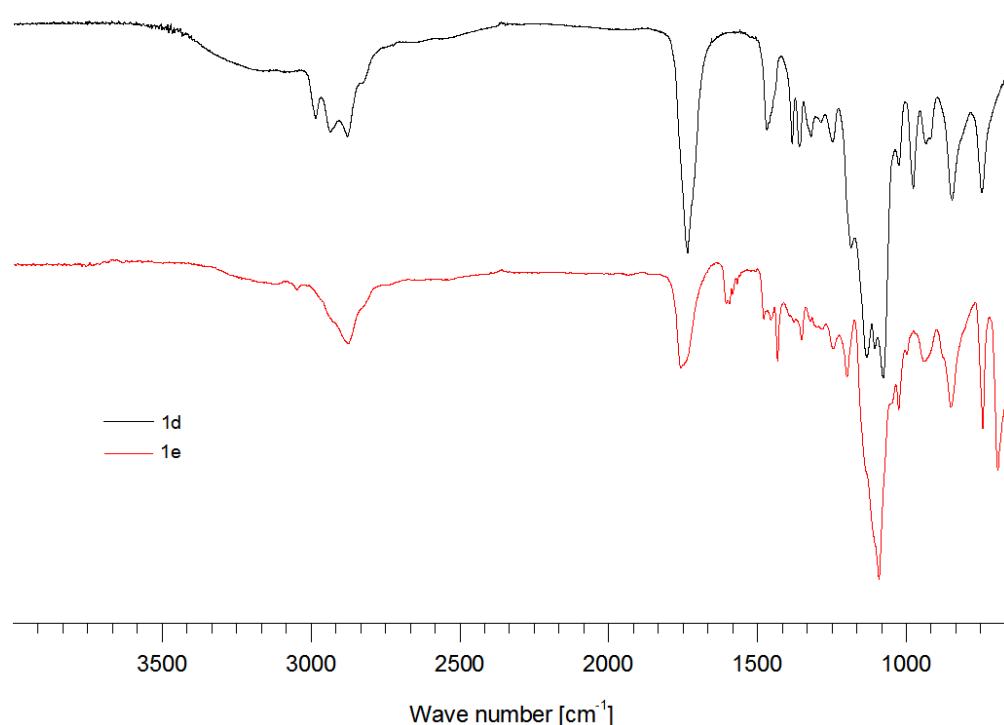
1	Precursor Characterisation	3
1.1	IR Spectra	3
1.2	NMR Spectra	5
1.3	X-Ray structure analysis	17
2	Thermal Characterisation	21
3	Additional TEM images	24

1 Precursor Characterisation

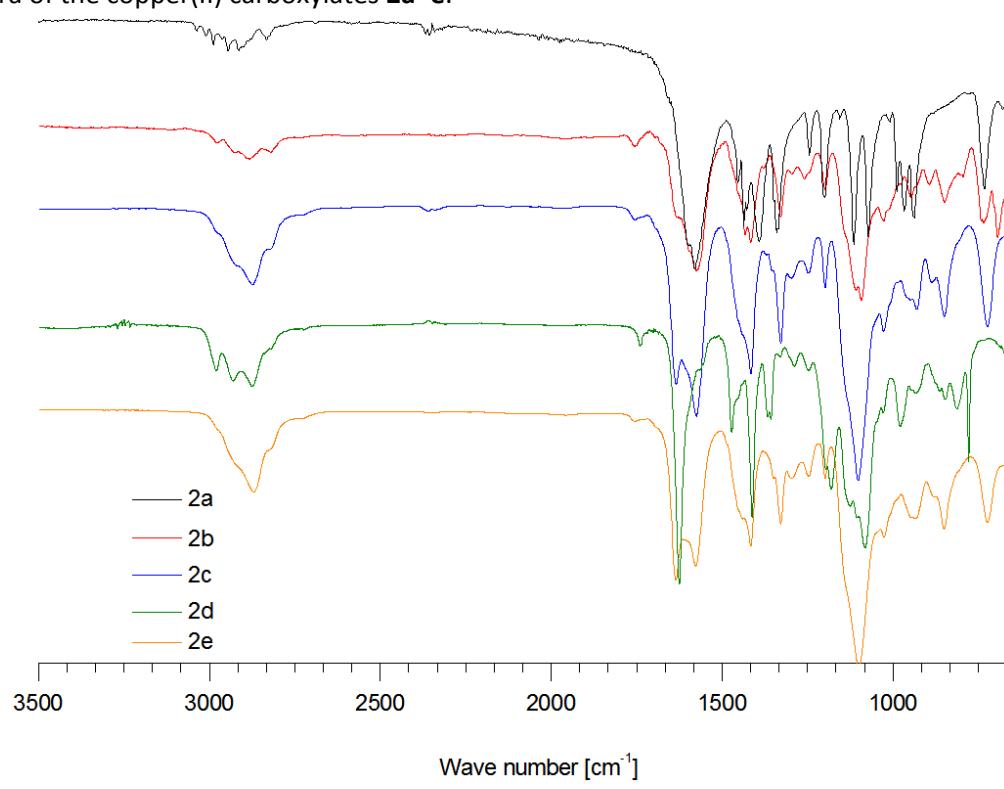
1.1 IR Spectra

All infrared spectra were recorded at ambient conditions as ATR-FTIR spectra by using a Biorad FTS-165 or a Nicolet iS 10 spectrometer from Thermo Scientific.

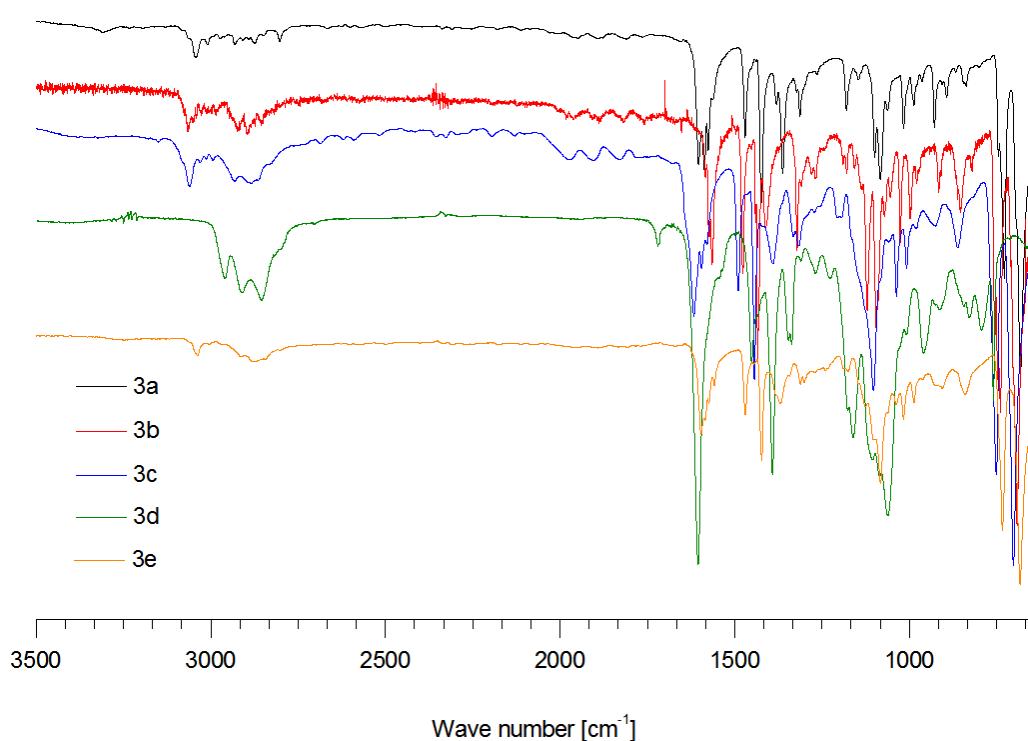
IR spectra of the carbonic acids **1d** and **1e**:



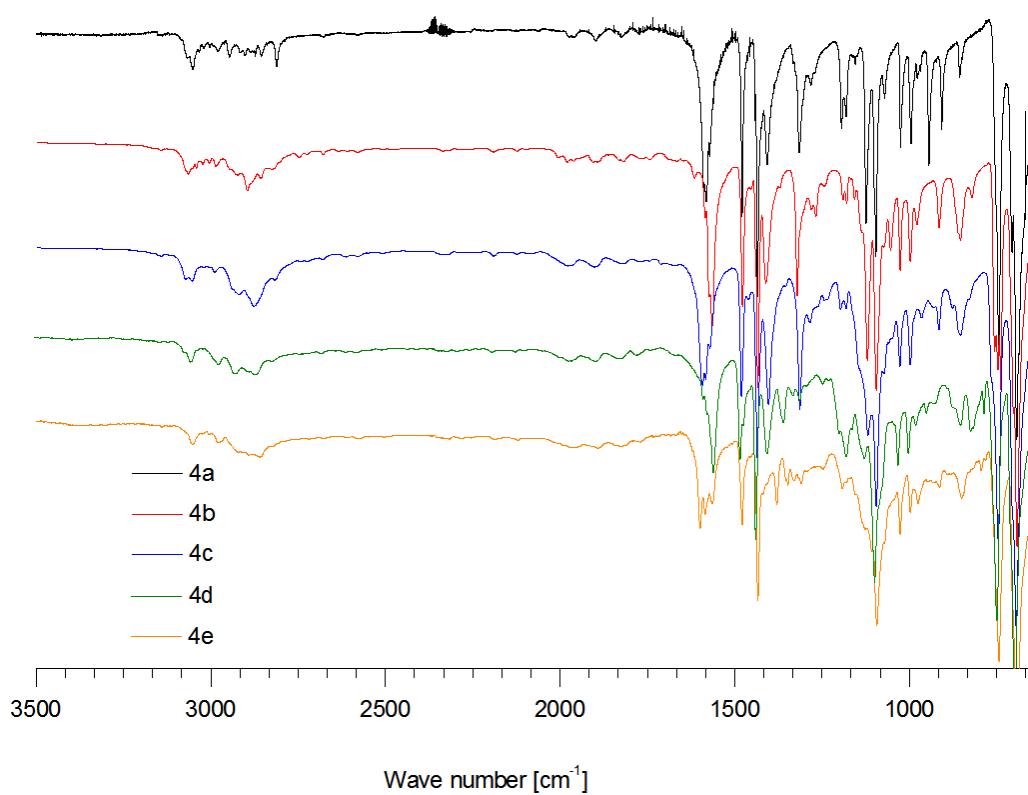
IR spectra of the copper(II) carboxylates **2a–e**:



IR spectra of the tris(triphenylphosphine)copper(I) carboxylates **3a–e**:



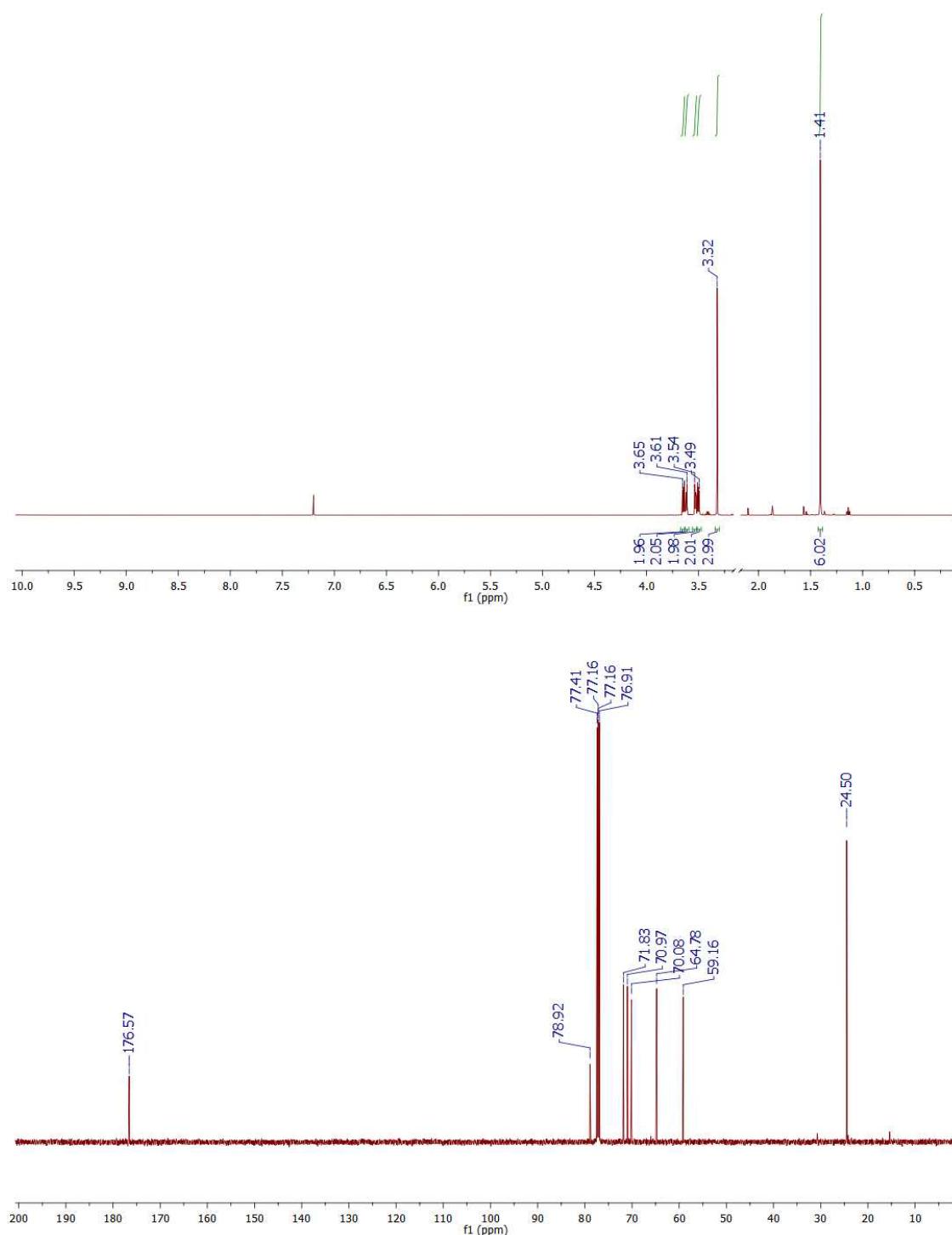
IR Spectra of the bis(triphenylphosphine)copper(I) carboxylates **4a–e**:



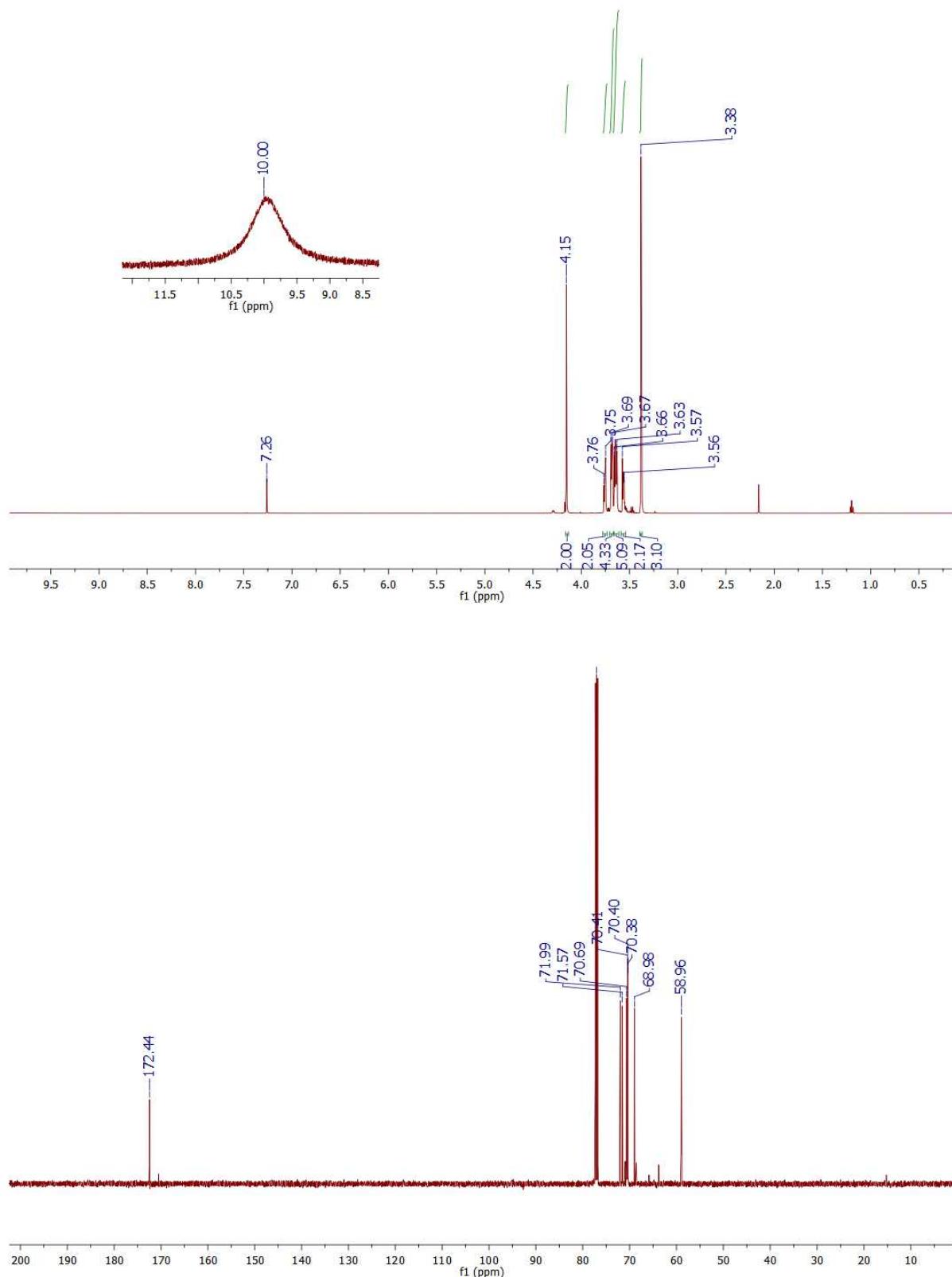
1.2 NMR Spectra

All ^1H NMR spectra were recorded in CDCl_3 at 25 °C with a Bruker Avance III 500 spectrometer operating at 500.30 MHz in the Fourier transform mode; $^{13}\text{C}\{^1\text{H}\}$ NMR spectra were recorded at 125.80 MHz. Chemical shifts are given relative to the internal standard tetramethylsilane ($\delta = 0.00$ ppm).

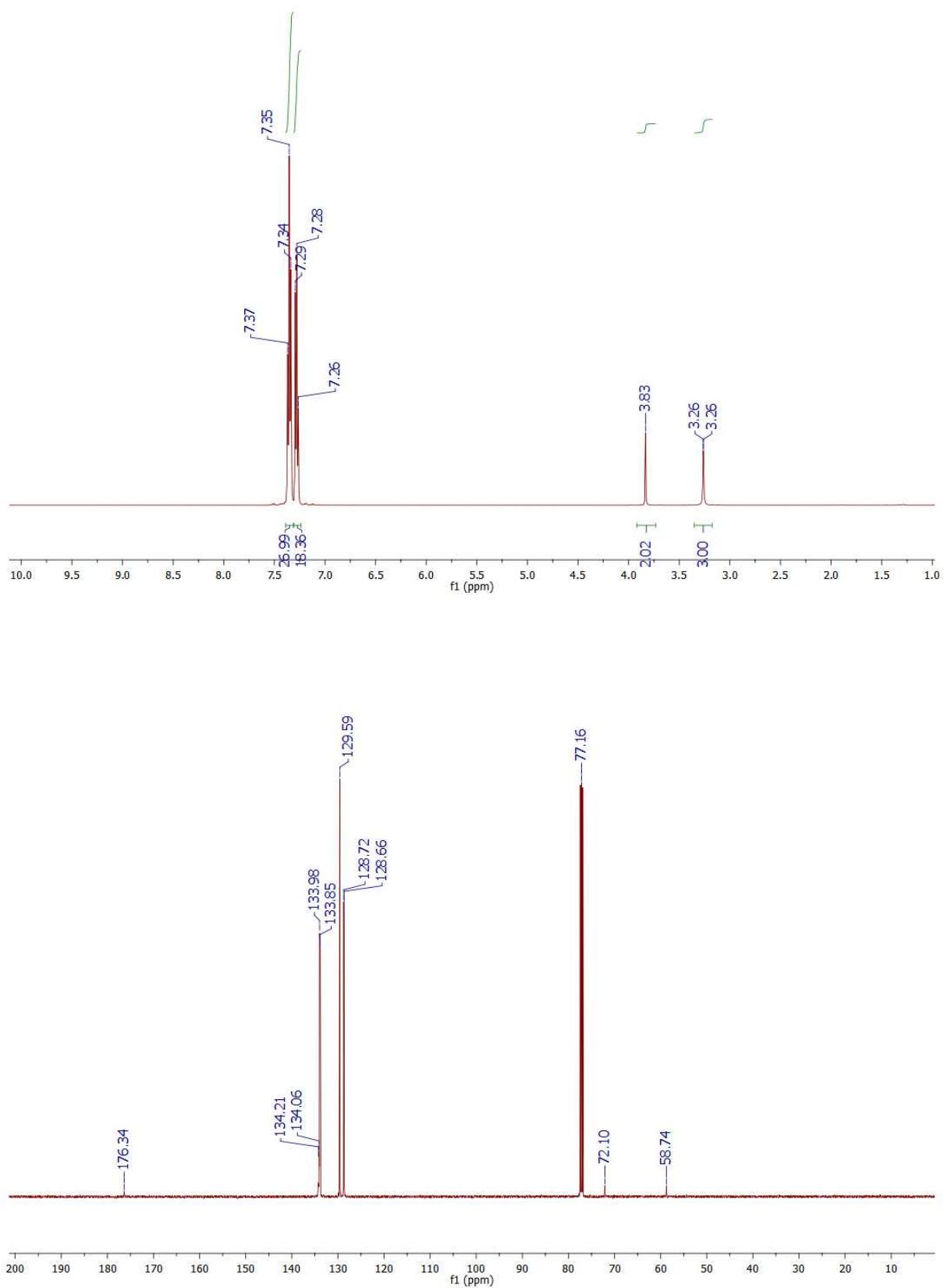
¹H (top) and ¹³C{¹H} (bottom) NMR spectra of 2-[2-(2-methoxyethoxy)ethoxy]-2-methylpropanoic acid (**1d**):



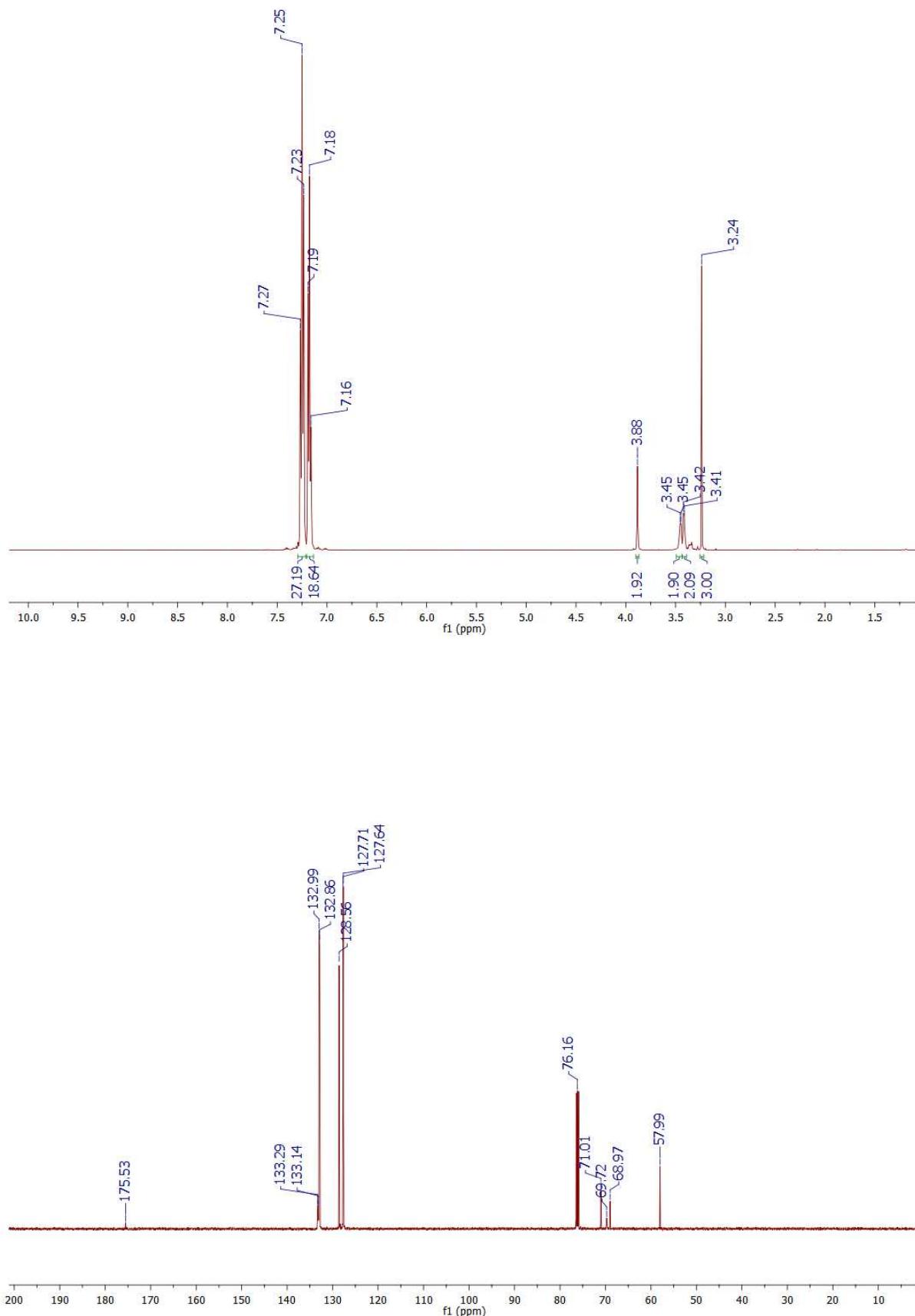
^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of 2-{2-[2-(2-methoxyethoxy)ethoxy]ethoxy}acetic acid (**1e**):



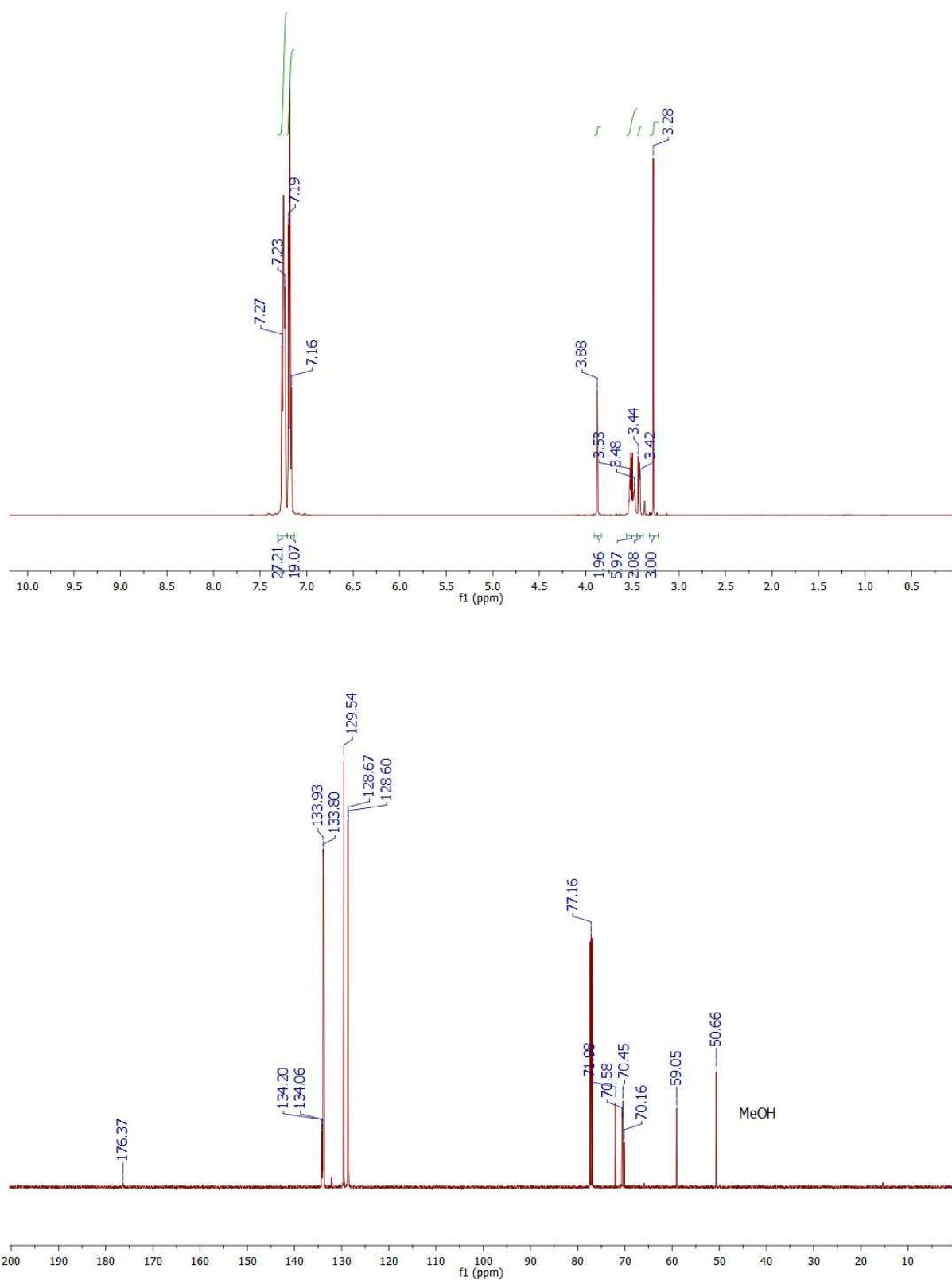
^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of tris(triphenylphosphine)copper(I)-2-methoxyacetate (**3a**):



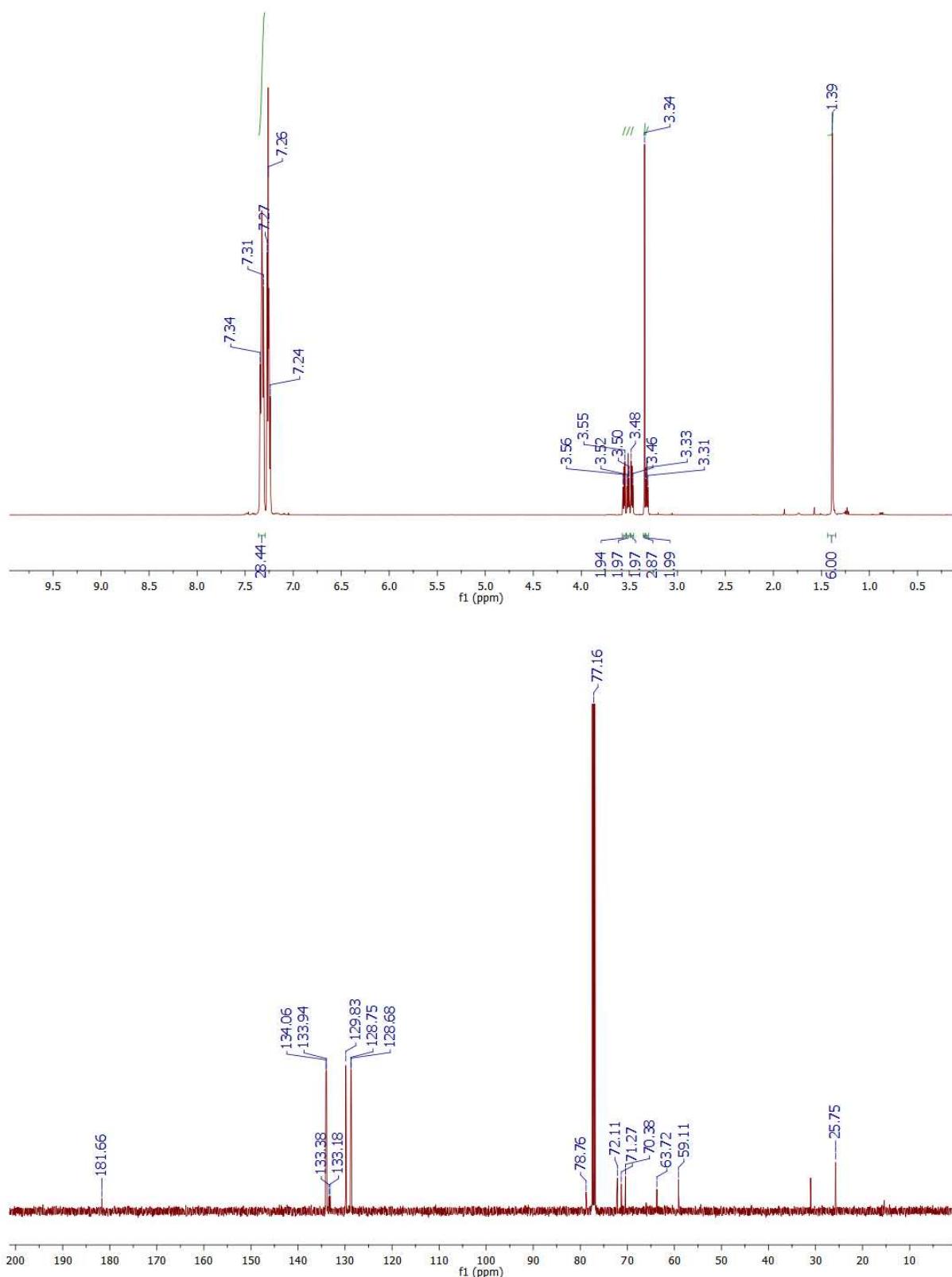
^1H (top) and $^{13}\text{C}\{\text{H}\}$ (bottom) NMR spectra of tris(triphenylphosphine)copper(I)-2-(2-methoxyethoxy)acetate (**3b**):



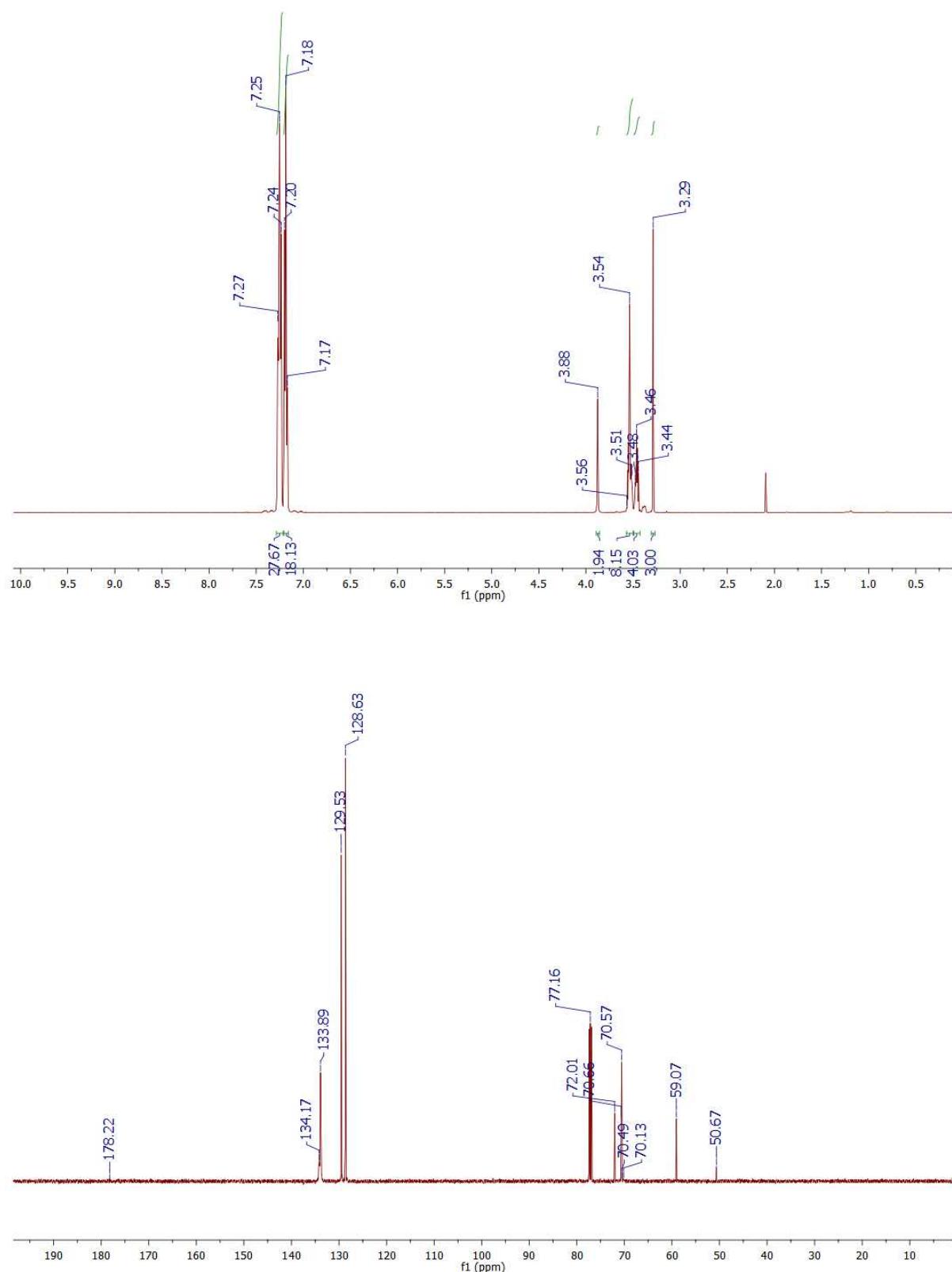
¹H (top) and ¹³C{¹H} (bottom) NMR spectra of tris(triphenylphosphine)copper(I)-2-[2-(2-methoxyethoxy)ethoxy]acetate (**3c**):



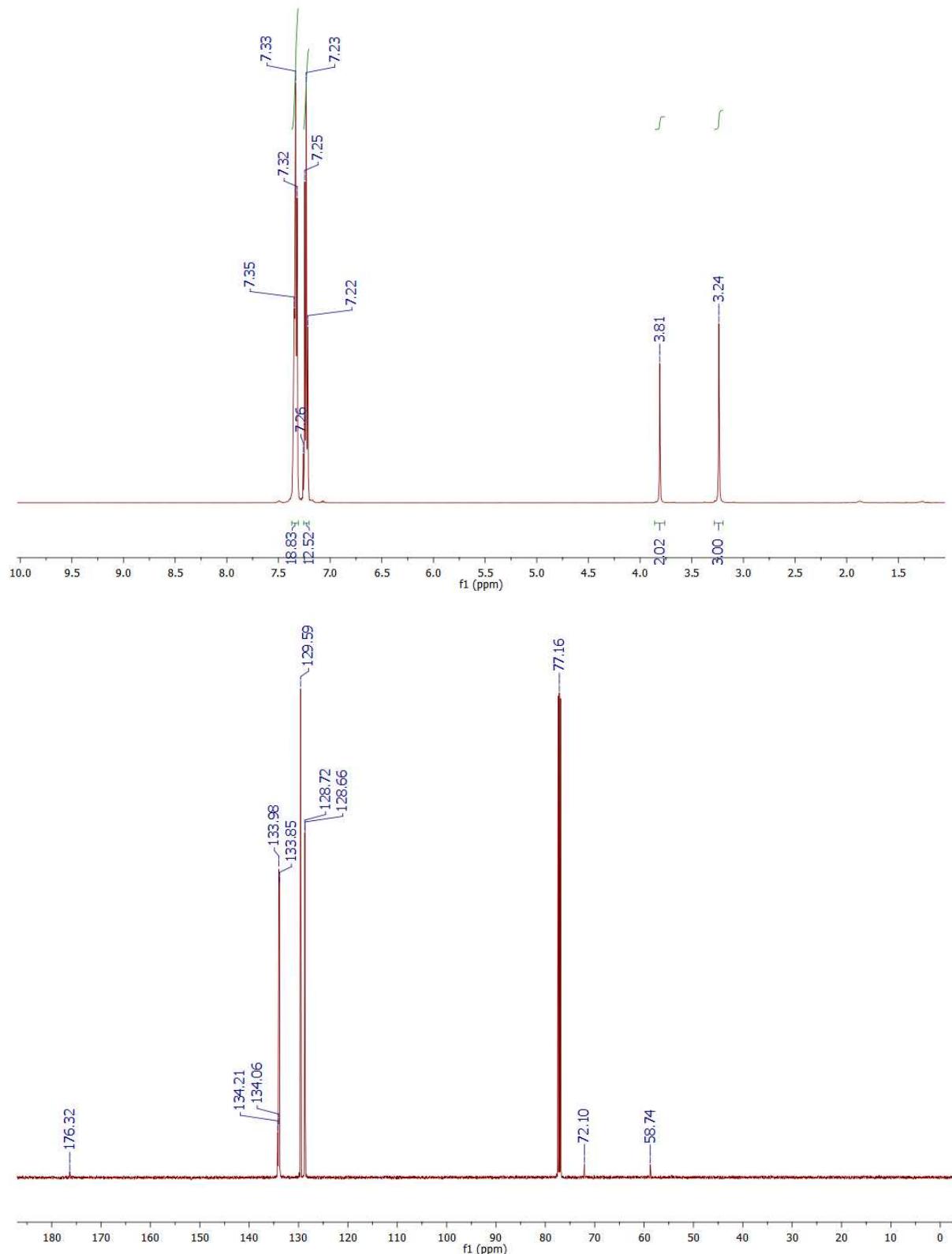
^1H (top) and $^{13}\text{C}\{\text{H}\}$ (bottom) NMR spectra of tris(triphenylphosphine)copper(I)-2-[2-(2-methoxyethoxy)ethoxy]-2-methylpropanoate (**3d**):



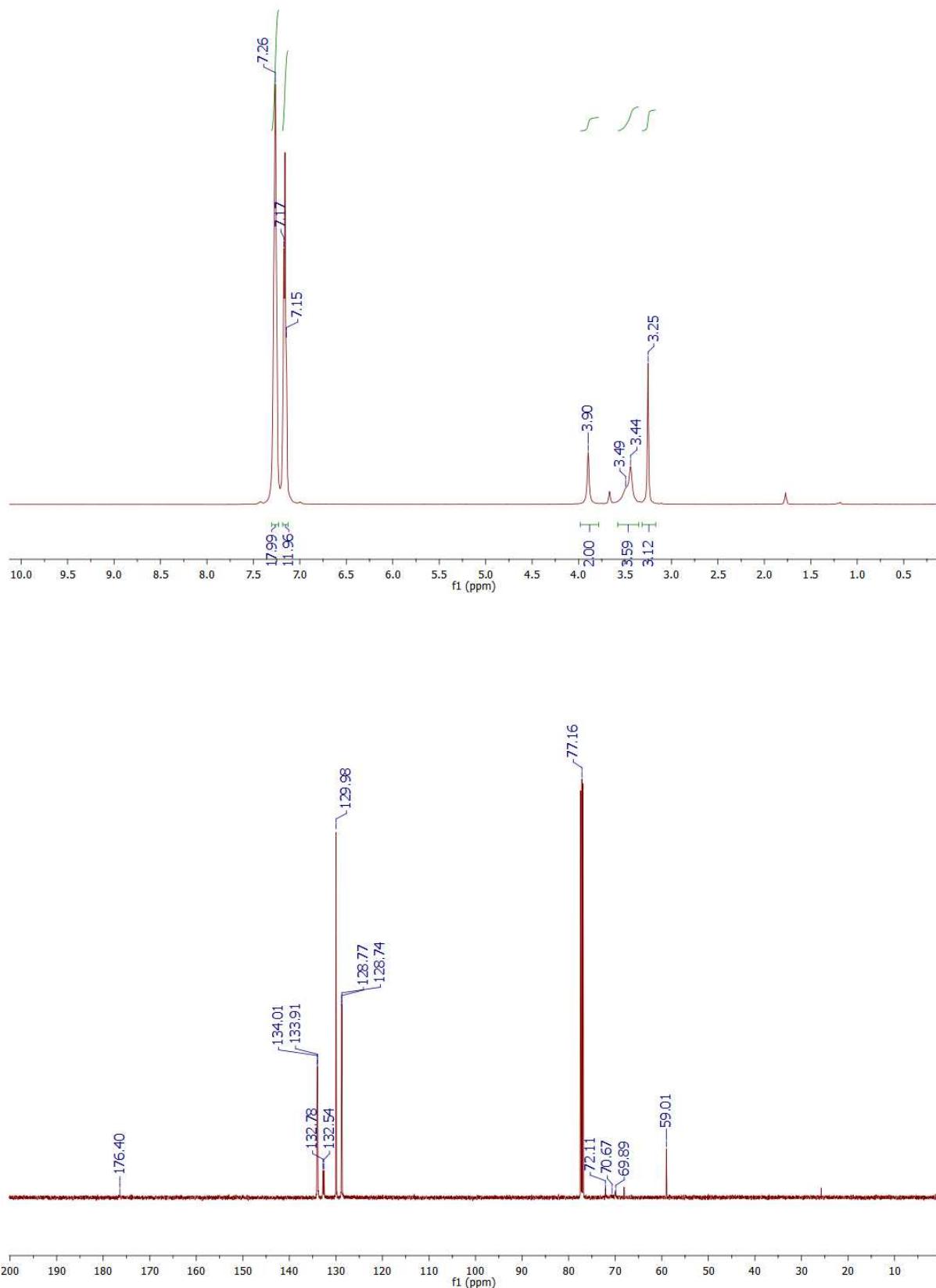
^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of tris(triphenylphosphine)copper(I)-2-{2-[2-(2-methoxyethoxy)ethoxy]ethoxy}acetate (**3e**):



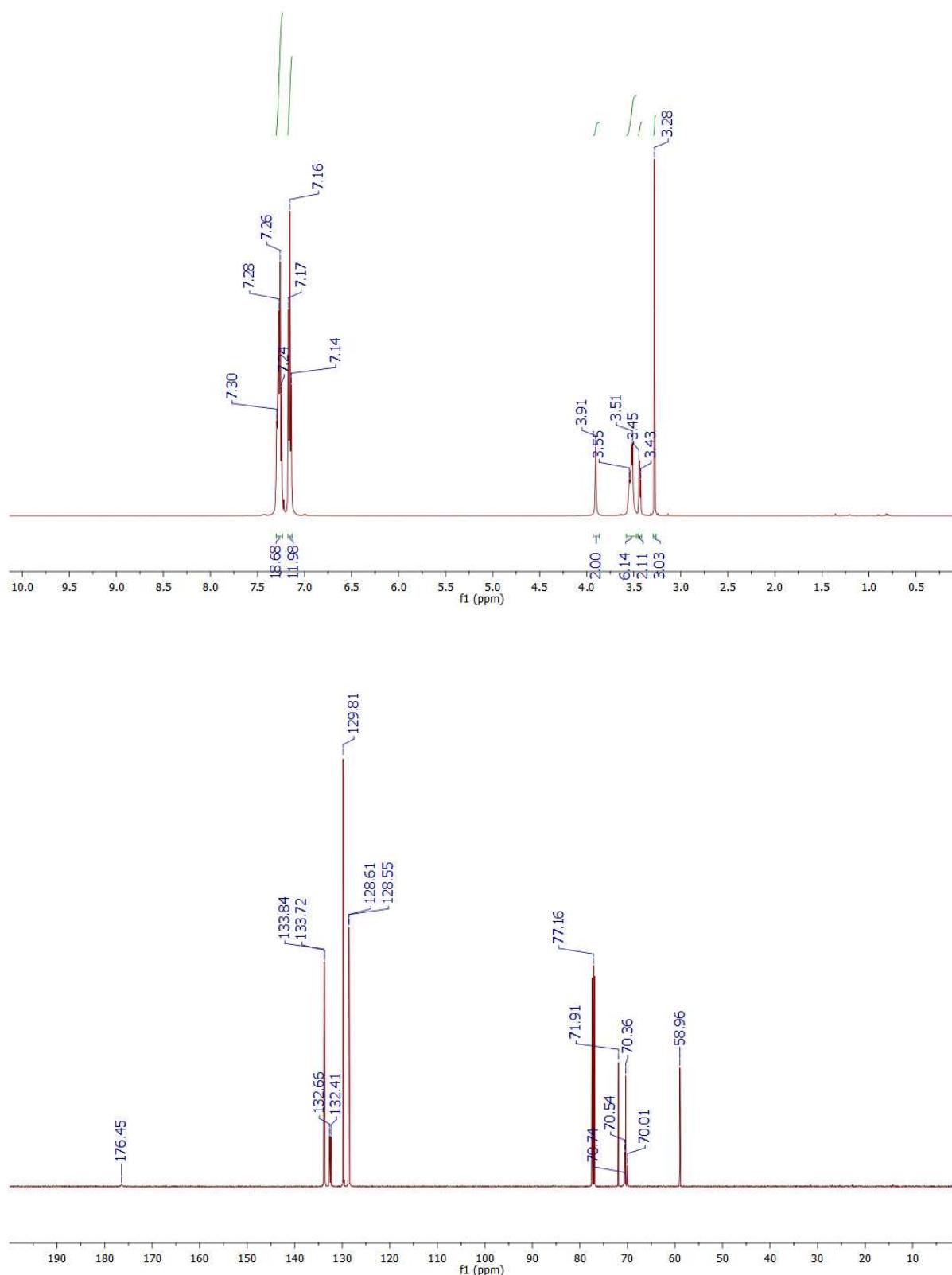
^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of bis(triphenylphosphine)copper(II)-methoxyacetate (**4a**):



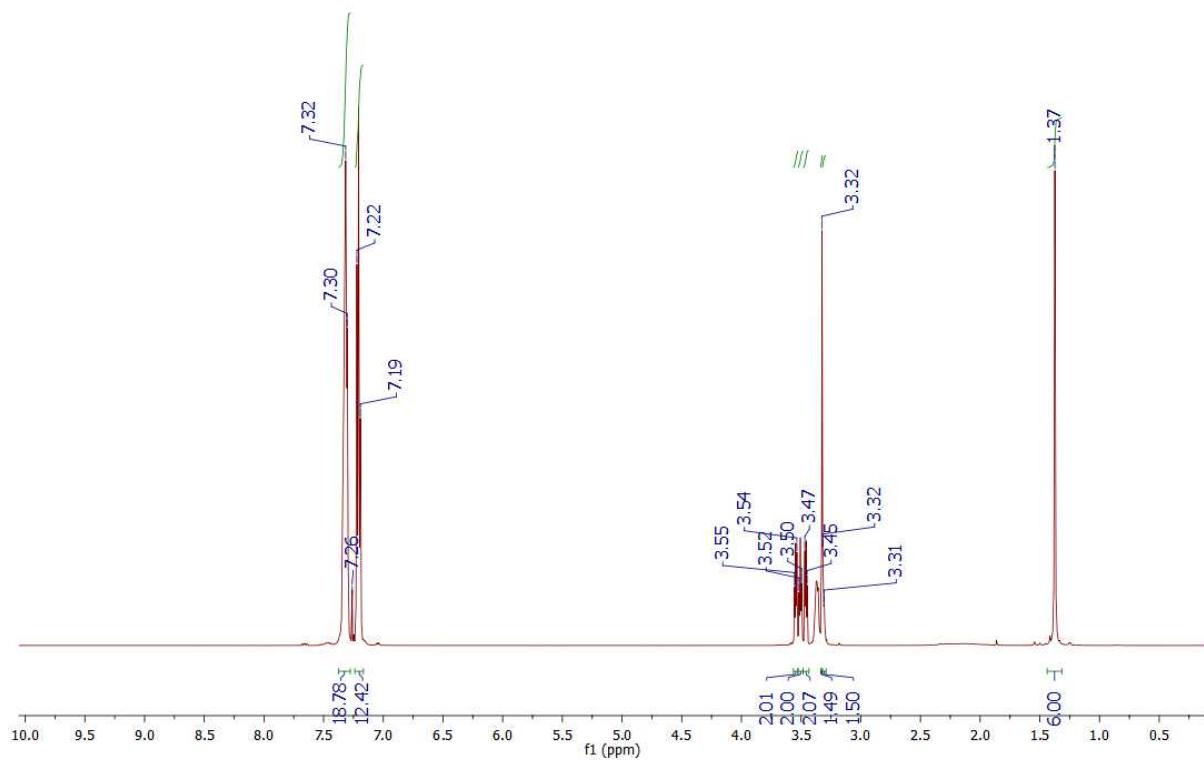
^1H (top) and $^{13}\text{C}\{\text{H}\}$ (bottom) NMR spectra of bis(triphenylphosphine)copper(I)-2-(2-methoxyethoxy)-acetate (**4b**):



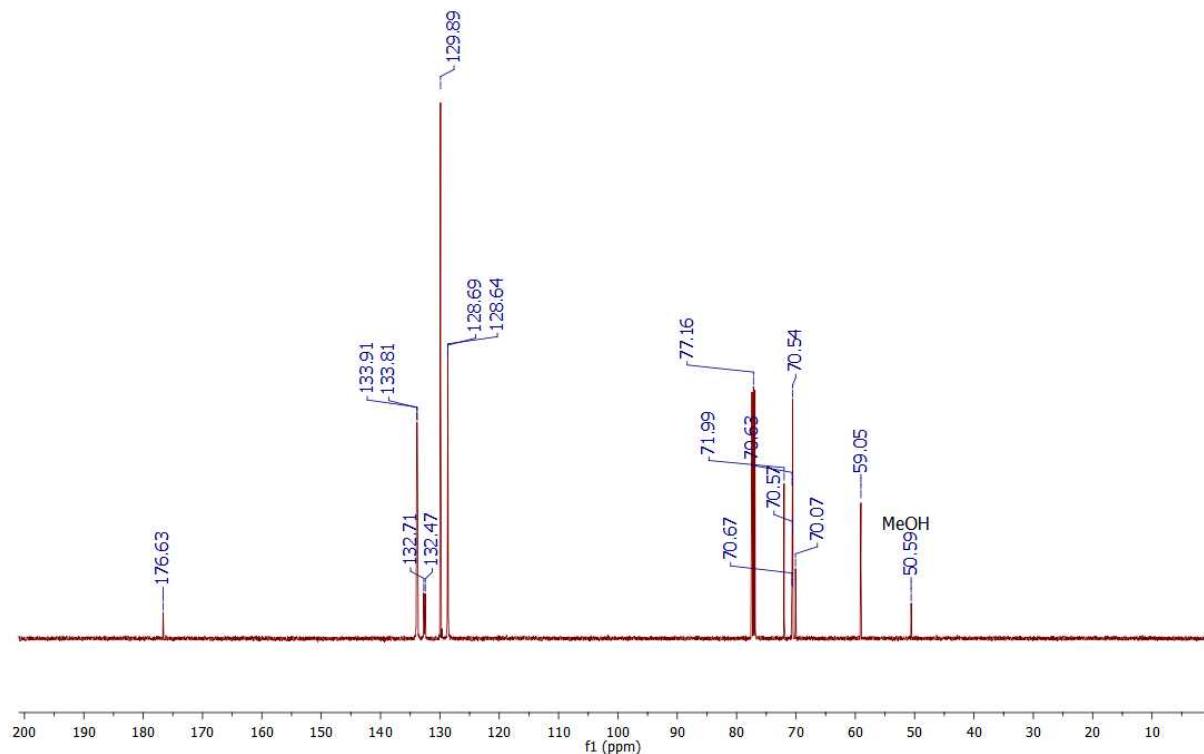
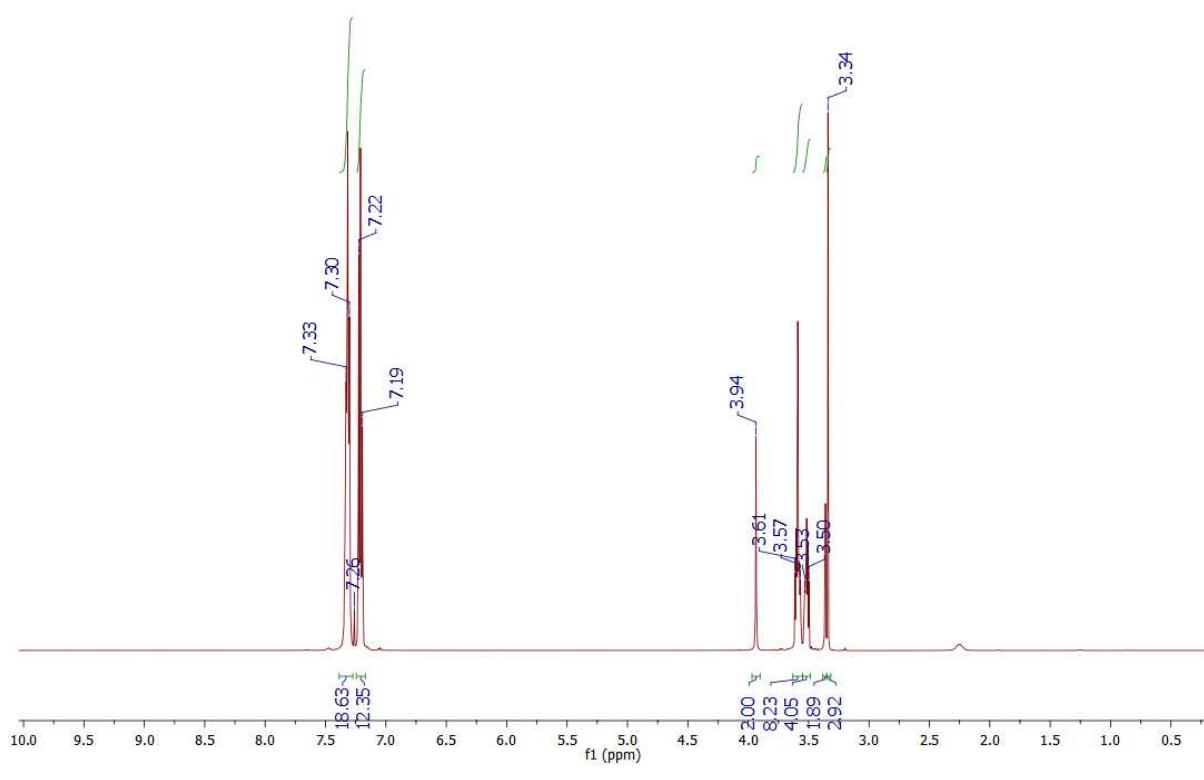
^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of bis(triphenylphosphine)copper(I)-2-[2-(2-methoxyethoxy)ethoxy]acetate (**4c**):



^1H (top) and $^{13}\text{C}\{^1\text{H}\}$ (bottom) NMR spectra of bis(triphenylphosphine)copper(I)-2-[2-(2-methoxyethoxy)ethoxy]-2-methylpropanoate (**4d**):



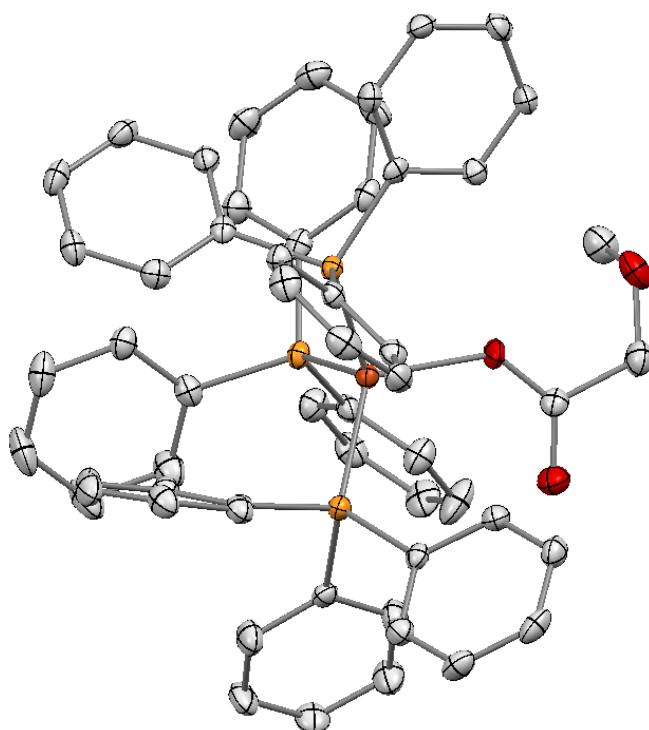
^1H (top) and $^{13}\text{C}\{\text{H}\}$ (bottom) NMR spectra of bis(triphenylphosphine)copper(I)-2-{2-[2-(2-methoxyethoxy)ethoxy]ethoxy}acetate (**4e**):



1.3 X-Ray structure analysis

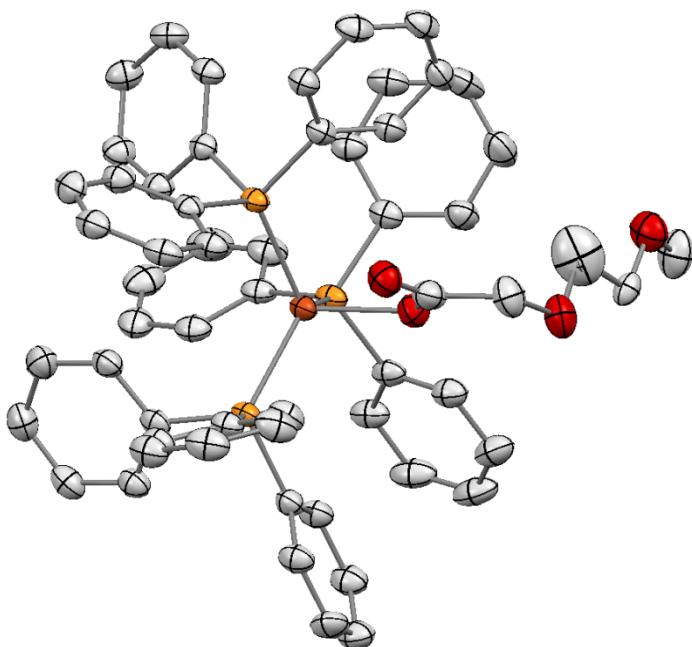
Data for single crystal X-ray diffraction analysis were collected with an Oxford Gemini S diffractometer at 110 K using Cu K α ($\lambda = 154.184$ pm) radiation. All figures show ORTEP diagrams with 50 % probability level. Hydrogen atoms and solvent molecules are omitted for clarity. In case of disordered atoms, only one conformation is shown. Colour code: gray – carbon, red – oxygen, yellow – phosphorus, orange – copper.

Crystal structure of tris(triphenylphosphine)copper(I)-methoxyacetate (**3a**, CCDC 932853):



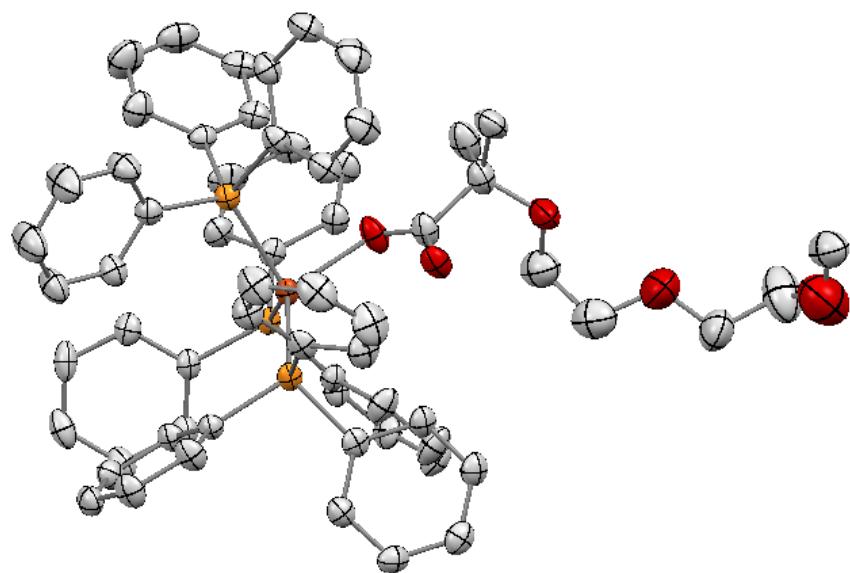
Crystal Data for 3a: $C_{57}H_{50}CuO_3P_3$, $M_r = 939.42$ g · mol $^{-1}$, crystal dimensions $0.40 \times 0.38 \times 0.30$ mm, $T = 115$ K, $\lambda = 154.184$ pm, orthorhombic, $Pna2_1$, $a = 18.31189(10)$ Å, $b = 13.00000(10)$ Å, $c = 19.6264(2)$ Å, $V = 4672.13(6)$ Å 3 , $Z = 4$, $\rho_{\text{calcd.}} = 1.336$ g · cm $^{-3}$, $\mu = 1.984$ mm $^{-1}$, ϑ range = $4.08 - 62.91^\circ$, reflections collected: 9408, independent: 5316 ($R_{\text{int}} = 0.0209$), $R_1 = 0.0340$, $wR_2 = 0.0887$ [$I > 2\sigma(I)$].

Crystal structure of tris(triphenylphosphine)copper(I)-2-(2-methoxyethoxy)acetate (**3b**, CDCC 932858):



Crystal Data for **3b:** $C_{255.60}H_{294.40}Cu_4O_{35.60}P_{12}$, $M_r = 4561.91 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.40 \cdot 0.40 \cdot 0.20 \text{ mm}$, $T = 110 \text{ K}$, $\lambda = 154.184 \text{ pm}$, monoclinic, C2/c, $a = 41.396(3) \text{ \AA}$, $b = 12.6912(4) \text{ \AA}$, $c = 26.4725(18) \text{ \AA}$, $\beta = 227.373(8)^\circ$, $V = 12350.6(12) \text{ \AA}^3$, $Z = 2$, $\rho_{\text{calcd.}} = 1.227 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 1.659 \text{ mm}^{-1}$, ϑ range = $3.41 - 64.10^\circ$, reflections collected: 32252, independent: 10102 ($R_{\text{int.}} = 0.0415$), $R_1 = 0.0653$, $wR_2 = 0.1801 [I > 2\sigma(I)]$.

Crystal structure of tris(triphenylphosphine)copper(I)-2-[2-(2-methoxyethoxy)ethoxy]2-methylpropanoate (**3d**, CDCC 932854):



Crystal Data for 3a: $C_{64}H_{63}CuO_5P_3$, $M_r = 1174.94 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.40 \cdot 0.38 \cdot 0.38 \text{ mm}$, $T = 100 \text{ K}$, $\lambda = 154.184 \text{ pm}$, monoclinic, $P2_1/n$, $a = 13.1082(6) \text{ \AA}$, $b = 22.7413(12) \text{ \AA}$, $c = 19.7501(11) \text{ \AA}$, $\beta = 104.039(5)^\circ$, $V = 5711.6(5) \text{ \AA}^3$, $Z = 4$, $\rho_{\text{calcd.}} = 1.366 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 3.022 \text{ mm}^{-1}$, ϑ range = $3.68 - 62.50^\circ$, reflections collected: 18682, independent: 9054 ($R_{\text{int}} = 0.0613$), $R_1 = 0.0816$, $wR_2 = 0.2237$ [$I > 2\sigma(I)$].

Crystal Data for 4a: $C_{39}H_{35}CuO_3P_2$, $M_r = 677.15 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.20 \cdot 0.10 \cdot 0.05 \text{ mm}$, $T = 110 \text{ K}$, $\lambda = 71.073 \text{ pm}$, triclinic, $P\bar{1}$, $a = 12.4723(6) \text{ \AA}$, $b = 12.4970(7) \text{ \AA}$, $c = 12.7037(6) \text{ \AA}$, $\alpha = 67.651(5)^\circ$, $\beta = 75.301(4)^\circ$, $\gamma = 62.402(5)^\circ$, $V = 1615.69(14) \text{ \AA}^3$, $Z = 2$, $\rho_{\text{calcd.}} = 1.392 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 0.813 \text{ mm}^{-1}$, ϑ range = $2.97\text{--}25.75^\circ$, reflections collected: 12298, independent: 6090 ($R_{\text{int}} = 0.0380$), $R_1 = 0.0390$, $wR_2 = 0.0805$ [$I > 2\sigma(I)$].

Crystal Data for 4b: $C_{41}H_{39}CuO_4P_2$, $M_r = 721.20 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.30 \cdot 0.20 \cdot 0.20 \text{ mm}$, $T = 110 \text{ K}$, $\lambda = 154.184 \text{ pm}$, orthorhombic, $Pbca$, $a = 21.2391(2) \text{ \AA}$, $b = 10.7028(1) \text{ \AA}$, $c = 30.5782(3) \text{ \AA}$, $V = 6950.97(11) \text{ \AA}^3$, $Z = 8$, $\rho_{\text{calcd.}} = 1.378 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 2.090 \text{ mm}^{-1}$, ϑ range = $3.56\text{--}62.94^\circ$, reflections collected: 11178, independent: 6420 ($R_{\text{int}} = 0.0223$), $R_1 = 0.0507$, $wR_2 = 0.1381$ [$I > 2\sigma(I)$].

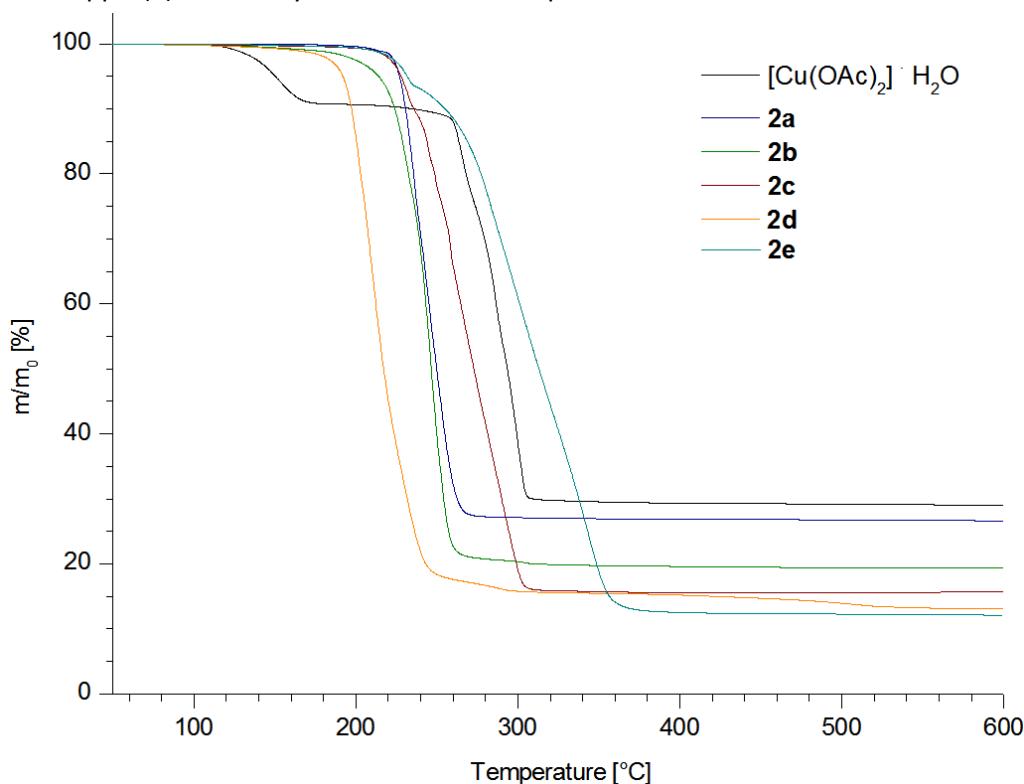
Crystal Data for 4d: $C_{45}H_{47}CuO_5P_2$, $M_r = 793.31 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.30 \cdot 0.25 \cdot 0.20 \text{ mm}$, $T = 105 \text{ K}$, $\lambda = 154.184 \text{ pm}$, triclinic, $P\bar{1}$, $a = 12.0902(10) \text{ \AA}$, $b = 13.2571(9) \text{ \AA}$, $c = 13.3691(10) \text{ \AA}$, $\alpha = 86.456(6)^\circ$, $\beta = 89.537(7)^\circ$, $\gamma = 64.306(7)^\circ$, $V = 1926.8(3) \text{ \AA}^3$, $Z = 2$, $\rho_{\text{calcd.}} = 1.367 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 1.954 \text{ mm}^{-1}$, ϑ range = $3.31\text{--}66.98^\circ$, reflections collected: 14265, independent: 6810 ($R_{\text{int}} = 0.0362$), $R_1 = 0.0549$, $wR_2 = 0.1424$ [$I > 2\sigma(I)$].

Crystal Data for 4e: $C_{45}H_{47}CuO_6P_2$, $M_r = 809.31 \text{ g} \cdot \text{mol}^{-1}$, crystal dimensions $0.20 \cdot 0.10 \cdot 0.10 \text{ mm}$, $T = 104 \text{ K}$, $\lambda = 71.073 \text{ pm}$, monoclinic, $P2_1/n$, $a = 15.9464(8) \text{ \AA}$, $b = 10.6978(3) \text{ \AA}$, $c = 23.2985(11) \text{ \AA}$, $V = 3972.8(3) \text{ \AA}^3$, $Z = 4$, $\rho_{\text{calcd.}} = 1.353 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 0.679 \text{ mm}^{-1}$, ϑ range = $2.88\text{--}25.50^\circ$, reflections collected: 19549, independent: 7368 ($R_{\text{int}} = 0.0399$), $R_1 = 0.0354$, $wR_2 = 0.0814$ [$I > 2\sigma(I)$].

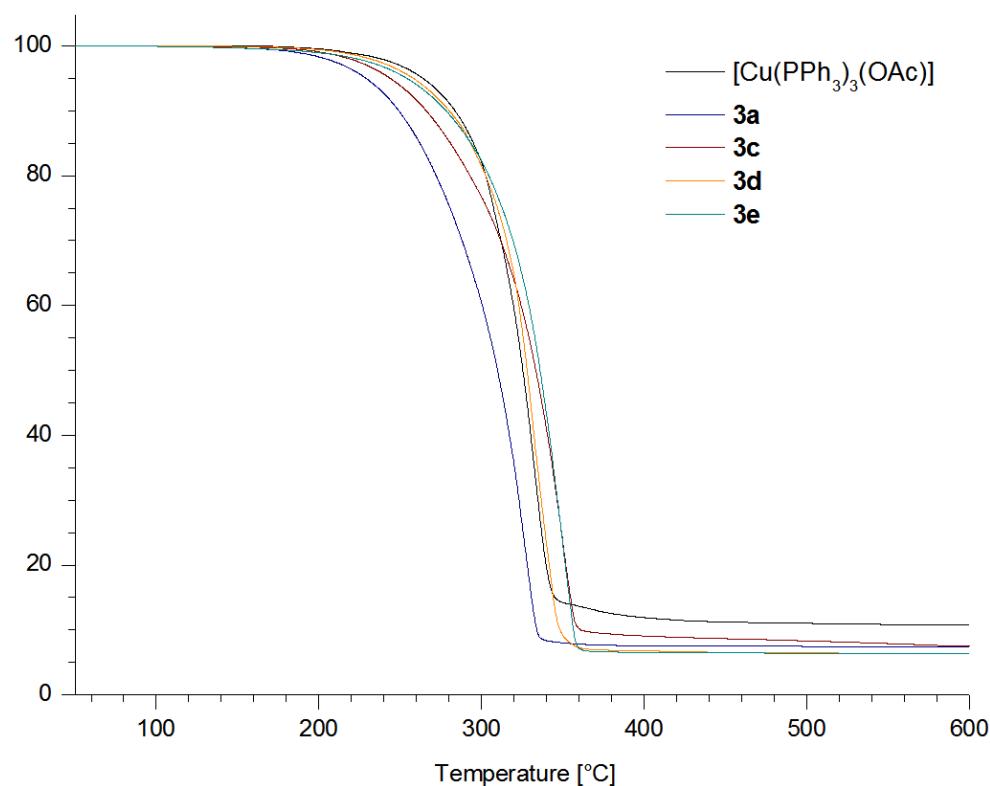
2 Thermal Decomposition

All thermogravimetric measurements were performed under N₂ flow (60 mL · min⁻¹) with a heating rate of 10 K · min⁻¹ with a Mettler Toledo TGA/DSC1 1600 system.

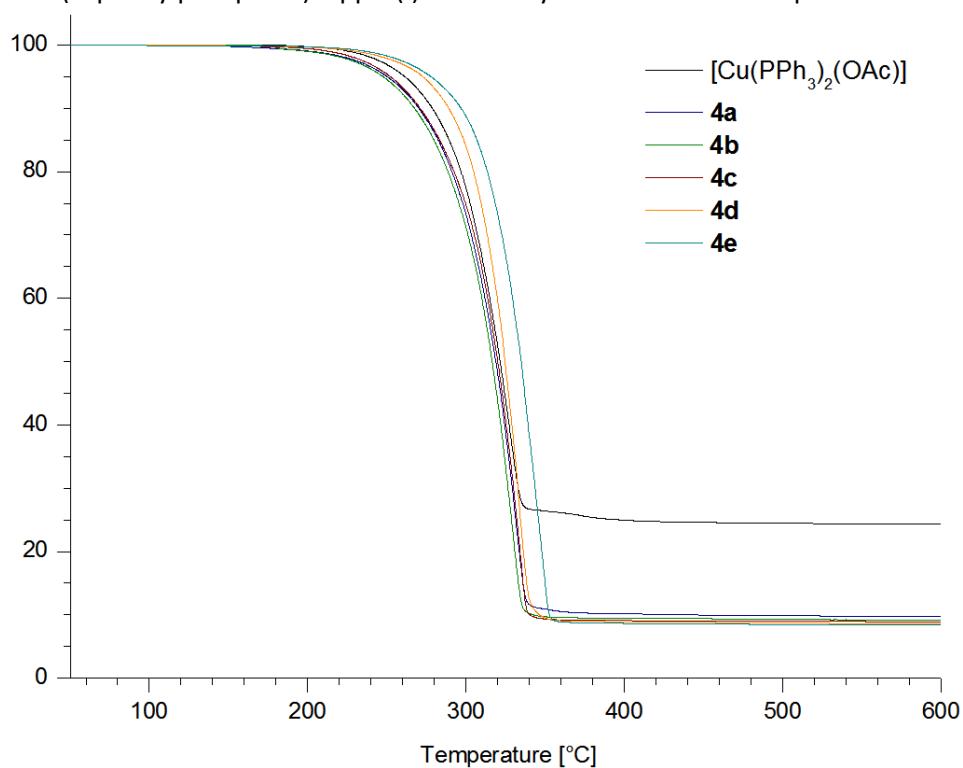
TG traces of copper(II) acetate hydrate as well as complexes **2a–e**:



TG traces of tris(triphenylphosphine)copper(I) acetate hydrate as well as complexes **3a,c–e**:

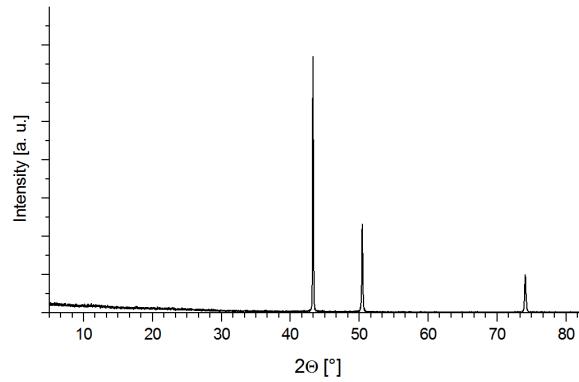


TG traces of bis(triphenylphosphine)copper(I) acetate hydrate as well as compounds **4a–e**:

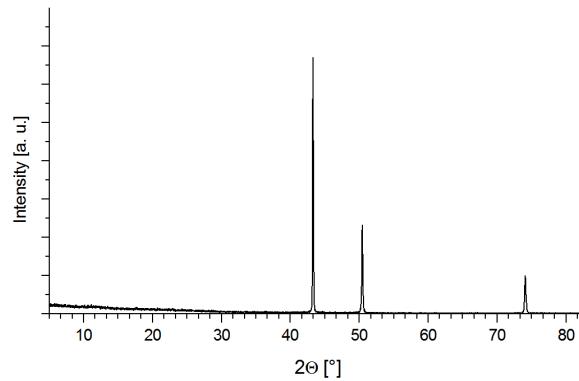


The non-volatile decomposition products of the thermal decompositions have been investigated by X-ray powder diffraction using a STOE-STAD IP device with Cu K_α ($\lambda = 154.184$ pm) radiation.

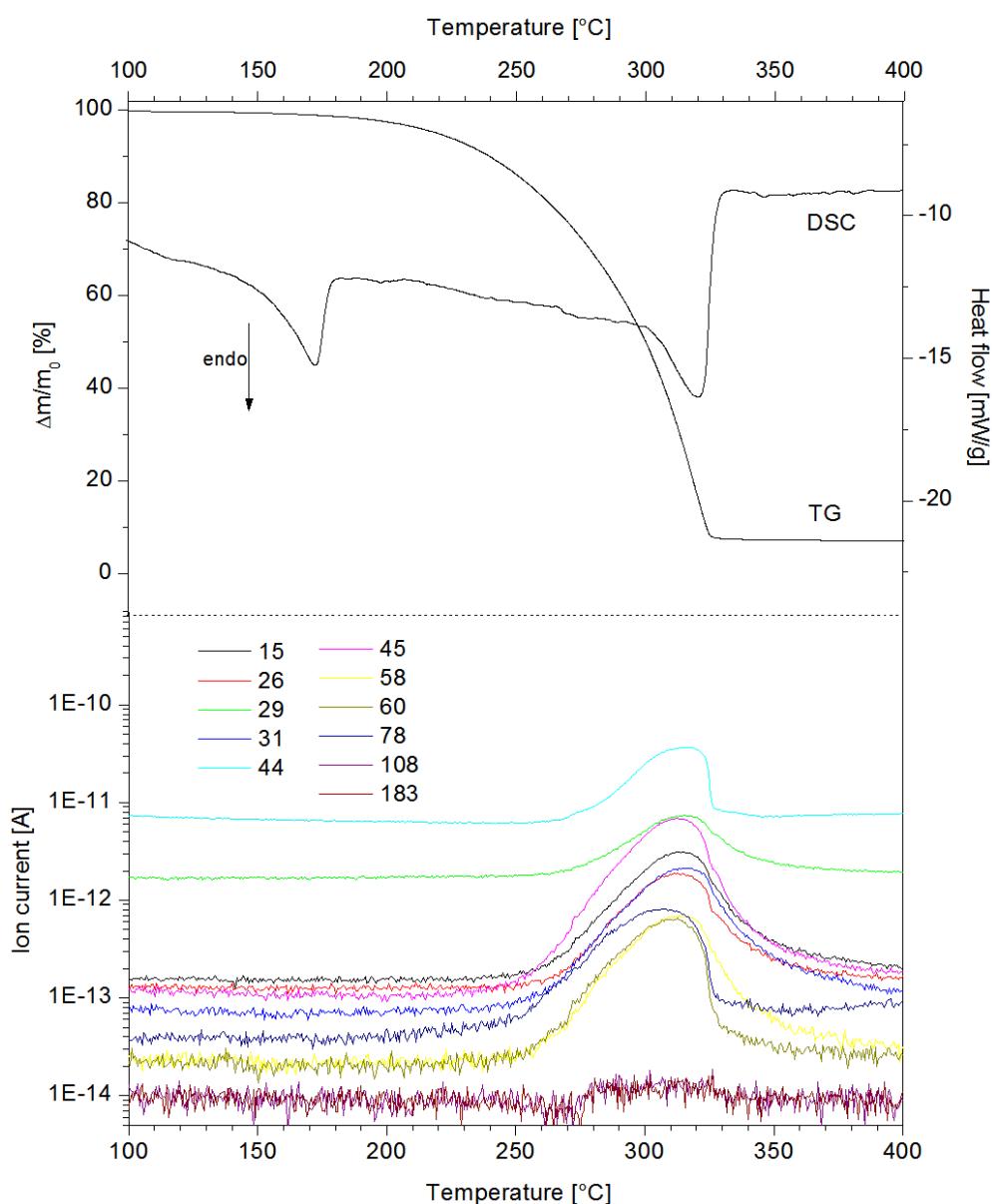
X-ray powder diffractogram of the non-volatile decomposition products of **2c**:



X-ray powder diffractogram of the non-volatile decomposition products of **4c**:



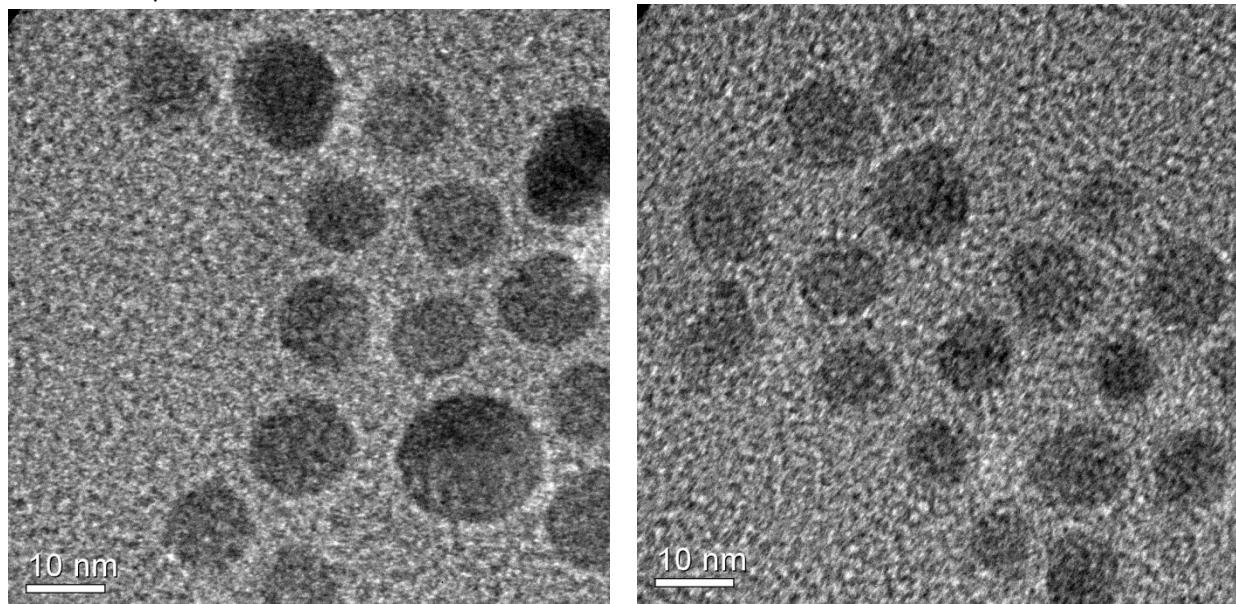
TG-MS characterisation of tris(triphenylphosphine)copper(I) 2-(2-methoxyethoxy)acetate (**3b**), measured by using a Pfeiffer Vacuum MS ThermoStar GSC 301 TS mass spectrometer coupled to the TG system:



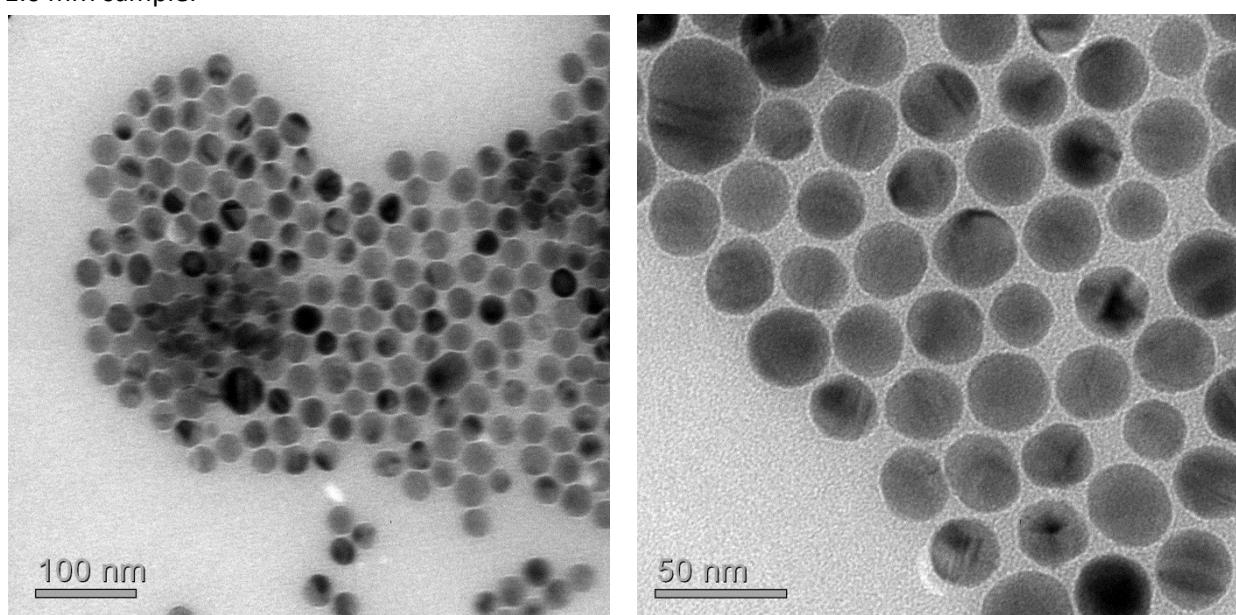
3 Additional TEM images

All TEM measurements were performed with a PHILIPS CM 20 FEG instrument operated at 200 kV.

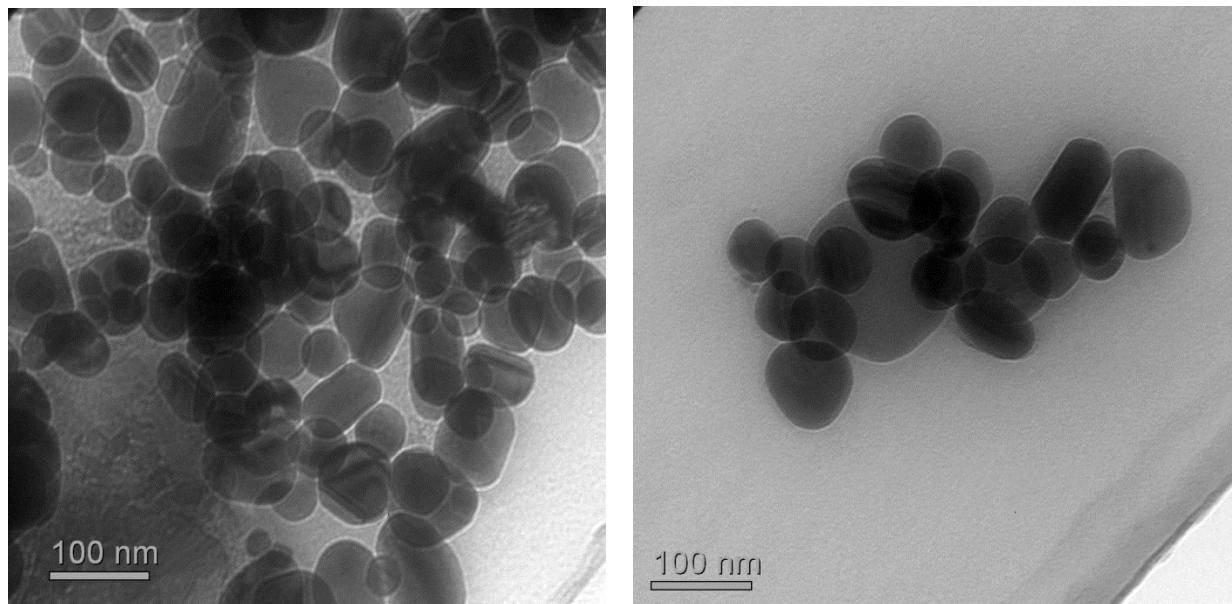
0.5 mM sample:



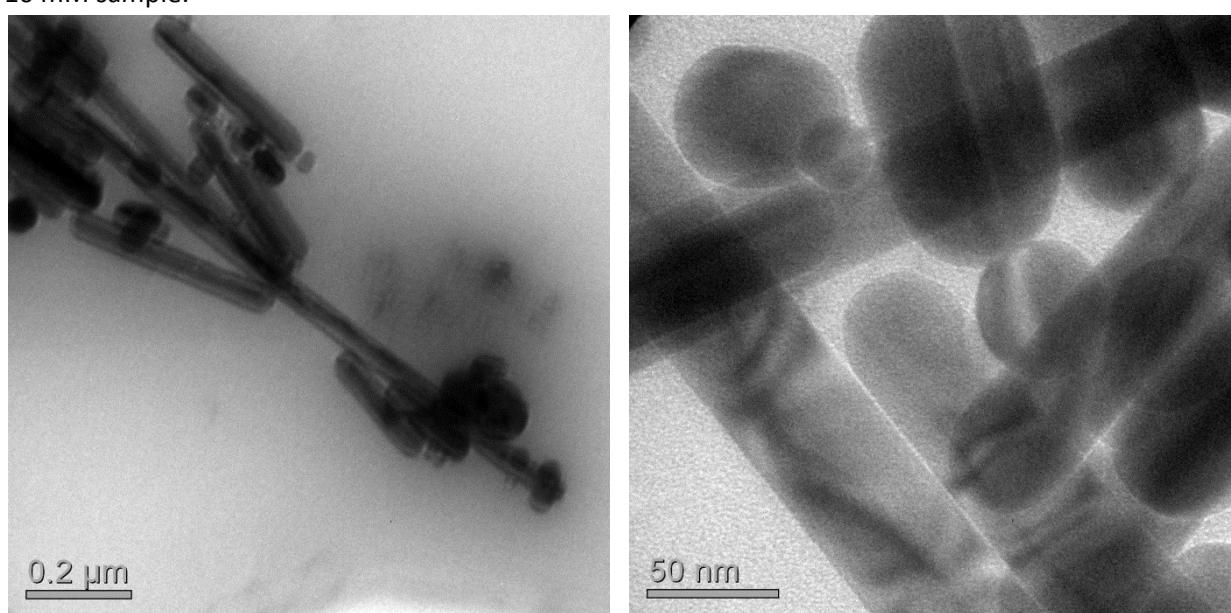
1.0 mM sample:



2.0 mM sample:



10 mM sample:



Electron diffractogram of 1.0 mM sample (red rings: copper, fcc, $a = 3.61 \text{ \AA}$):

