## **Supporting Information**

## A Cap-Type Schiff Base Acting as Fluorescence Sensor for Zinc(II) and Colorimetric Sensor for Iron(II), Copper(II), and Zinc(II) in Aqueous Media

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**Fig. S1.** Job plot of a 1:1 complex of L and  $Zn^{2+}$ , where the intensity at 462 nm was plotted against the mole fraction of zinc ions.



Fig. S2. Benesi-Hildebrand plot (absorption at 397 nm) of L, assuming a 1:1 stoichiometry for association between L and  $Zn^{2+}$ .



Fig. S3. Detection limit of L (20  $\mu$ M) with Zn<sup>2+</sup>.



Fig. S4. (a) Fluorescence spectra of L (10  $\mu$ M) after the addition of zinc ions (1 equiv) at different pH in 10 mM HEPES buffer-CH<sub>3</sub>OH (99:1, v/v). (b) Plot of the fluorescence intensity of L at 462 nm against pH.



**Fig. S5.** Job plot of a 1:1 complex of L and  $Fe^{2+}$ , where the absorption at 515 nm was plotted against the mole fraction of ferrous ions.



**Fig. S6.** Benesi-Hildebrand plot (absorption at 400 nm) of L, assuming a 1:1 stoichiometry for association between L and  $Fe^{2+}$ .



Fig. S7. Detection limit of L (20  $\mu$ M) with Fe<sup>2+</sup>.



**Fig. S8.** (a) Absorption spectra of L (20  $\mu$ M) after the addition of ferrous ions (2 equiv) in 10 mM HEPES buffer and CH<sub>3</sub>OH (99:1, v/v). (b) Plot of the absorbance of L at 509 nm against pH.



**Fig. S9.** Job plot of a 1:1 complex of L and  $Cu^{2+}$ , where the absorption at 396 nm was plotted against the mole fraction of copper ions.



Fig. S10. Positive-ion electrospray ionization mass spectrum of L upon addition of 1 equiv of  $Cu^{2+}$  in  $CH_3OH$ .



**Fig. S11.** Benesi-Hildebrand plot (absorption at 400 nm) of L, assuming a 1:1 stoichiometry for association between L and  $Cu^{2+}$ .



Fig. S12. Detection limit of L (20  $\mu$ M) with Cu<sup>2+</sup>.