

## Electronic Supplementary Information

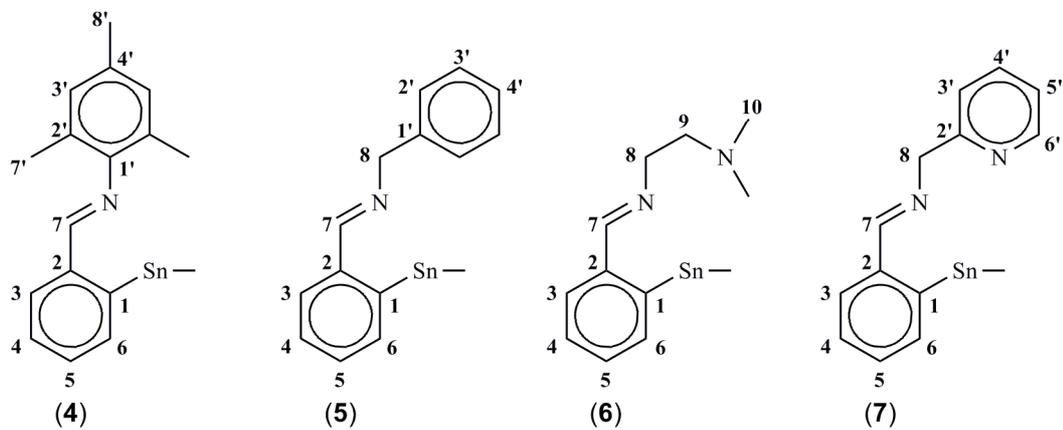
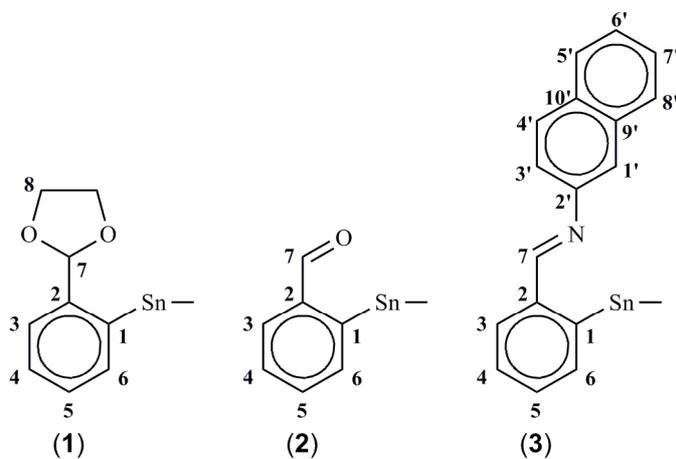
### Di(imino)aryltin(IV) dichlorides as tectons for heterometallic coordination compounds

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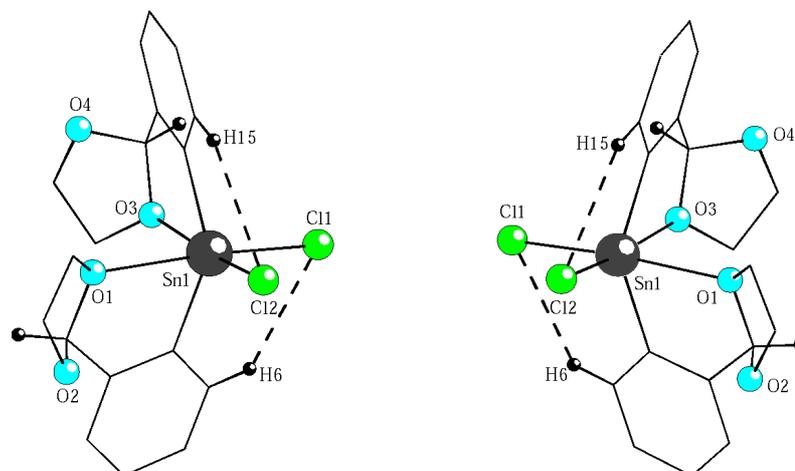
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### Numbering schemes for NMR resonance assignments



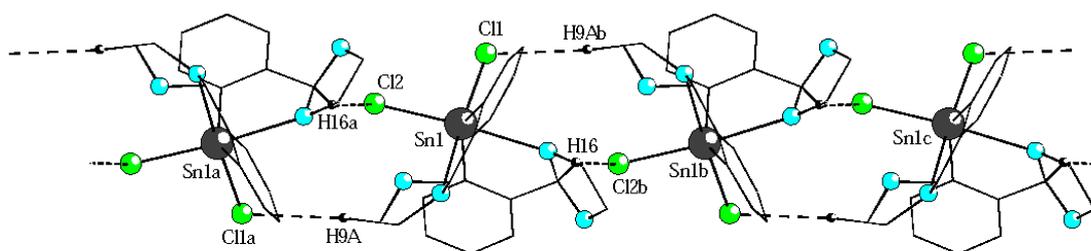
[2-((CH<sub>2</sub>O)<sub>2</sub>CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (**1**)

- the crystal contains a 1:1 mixture of  $\Delta$ -[*pS*<sub>O(1)</sub>*pS*<sub>O(3)</sub>]-[*S*<sub>C(7)</sub>*R*<sub>C(16)</sub>]-**1** and  $\Lambda$ -[*pR*<sub>O(1)</sub>*pR*<sub>O(3)</sub>]-[*R*<sub>C(7)</sub>*S*<sub>C(16)</sub>]-**1** isomers



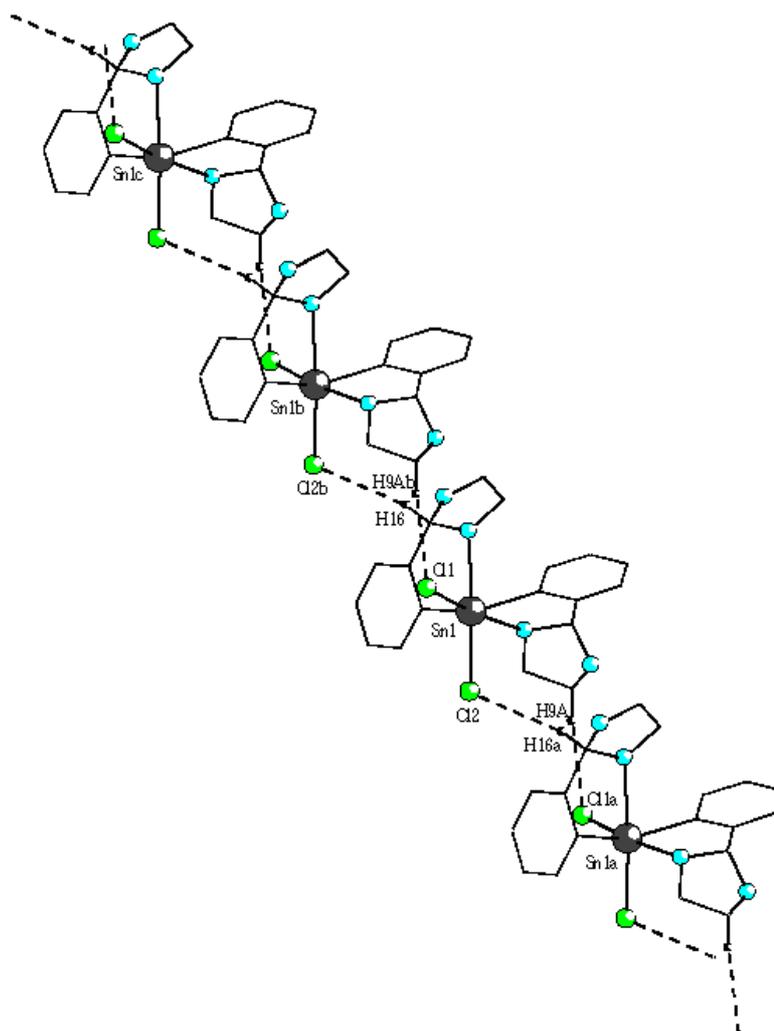
**Figure S1.** Molecular structure of  $\Delta$ -[*pS*<sub>O(1)</sub>*pS*<sub>O(3)</sub>]-[*S*<sub>C(7)</sub>*R*<sub>C(16)</sub>]-**1** isomer (*left*) and  $\Lambda$ -[*pR*<sub>O(1)</sub>*pR*<sub>O(3)</sub>]-[*R*<sub>C(7)</sub>*S*<sub>C(16)</sub>]-**1** isomer (*right*) in the crystal of **1**, showing the intramolecular chlorine-hydrogen contacts (only methine hydrogens and hydrogen atoms involved in intramolecular contacts are shown).

- intramolecular distance  $\text{Cl}(1)\cdots\text{H}(6)_{\text{aryl}}$  2.85 Å  $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å  
 $\text{Cl}(2)\cdots\text{H}(15)_{\text{aryl}}$  2.88 Å

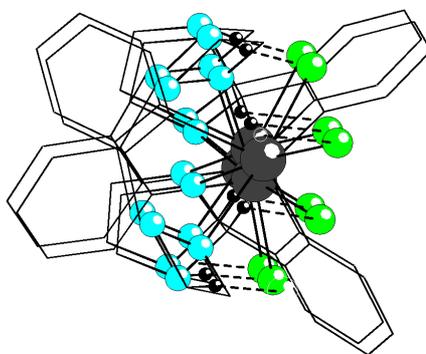


**Figure S2.** View along *a* axis of a chain polymer based on intermolecular chlorine-hydrogen contacts between alternating  $\Delta$ -[*pS*<sub>O(1)</sub>*pS*<sub>O(3)</sub>]-[*S*<sub>C(7)</sub>*R*<sub>C(16)</sub>]-**1** and  $\Lambda$ -[*pR*<sub>O(1)</sub>*pR*<sub>O(3)</sub>]-[*R*<sub>C(7)</sub>*S*<sub>C(16)</sub>]-**1** isomers in the crystal of **1** (only hydrogen atoms involved in C–H⋯Cl contacts are shown) [symmetry equivalent atoms ( $0.5+x, 1.5-y, 0.5+z$ ), ( $-0.5+x, 1.5-y, -0.5+z$ ) and ( $-1+x, y, -1+z$ ) are given by “a”, “b” and “c”, respectively].

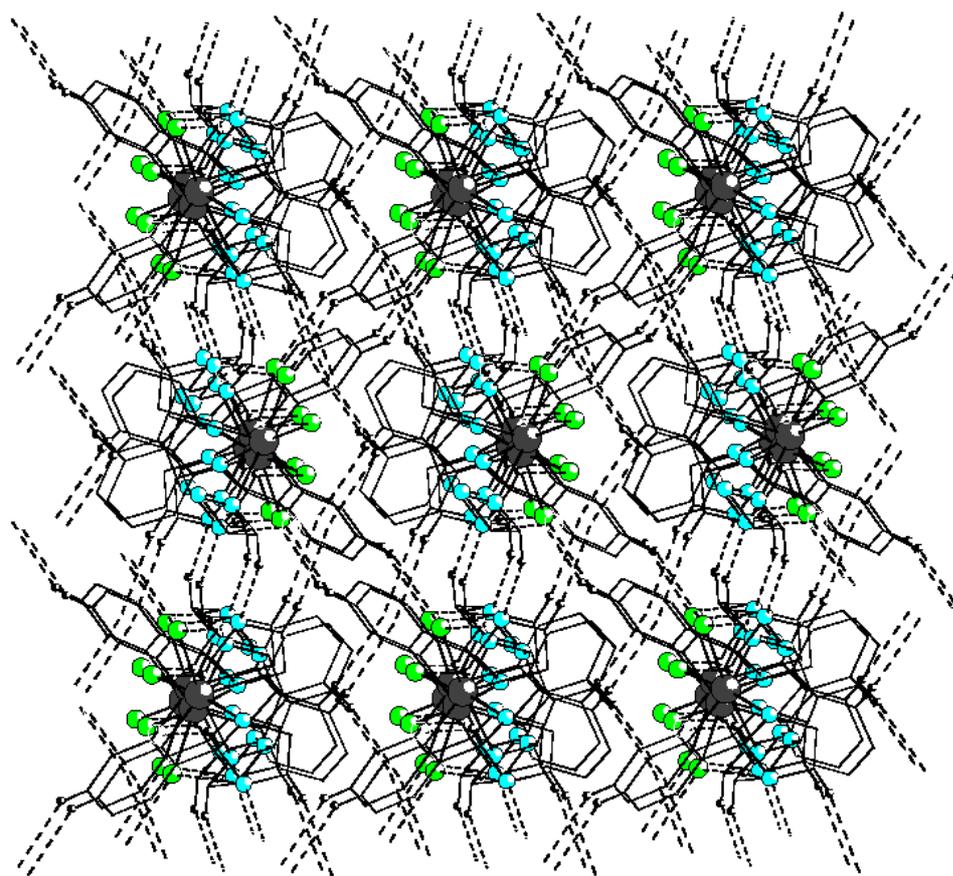
- intermolecular distance  $\text{Cl}(1)\cdots\text{H}(9\text{Ab})_{\text{methylene}}$  2.77 Å  $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å  
 $\text{Cl}(2)\cdots\text{H}(16\text{a})_{\text{methine}}$  2.90 Å



**Figure S3.** View along *b* axis of a chain polymer based on intermolecular chlorine-hydrogen contacts between alternating  $\Delta$ -[*pS*<sub>O(1)</sub>*pS*<sub>O(3)</sub>]-[*S*<sub>C(7)</sub>*R*<sub>C(16)</sub>]-**1** and  $\Lambda$ -[*pR*<sub>O(1)</sub>*pR*<sub>O(3)</sub>]-[*R*<sub>C(7)</sub>*S*<sub>C(16)</sub>]-**1** isomers in the crystal of **1** (only hydrogen atoms involved in C–H···Cl contacts are shown) [symmetry equivalent atoms ( $0.5+x$ ,  $1.5-y$ ,  $0.5+z$ ), ( $-0.5+x$ ,  $1.5-y$ ,  $-0.5+z$ ) and ( $-1+x$ ,  $y$ ,  $-1+z$ ) are given by “a”, “b” and “c”, respectively].



**Figure S4.** View along a chain polymer in the crystal of **1** (only hydrogen atoms involved in C–H···Cl contacts are shown).

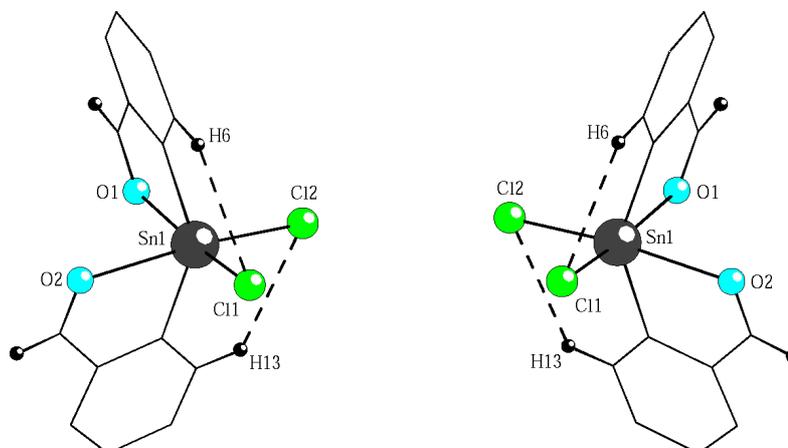


**Figure S5.** View of the 3D architecture in the crystal of **1**, based on chlorine-hydrogen, oxygen-hydrogen and C-H... $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts between parallel chain polymers (only hydrogen atoms involved in such contacts are shown) [symmetry equivalent atoms  $(-0.5-x, 0.5+y, 0.5-z)$ ,  $(-x, 1-y, -z)$ ,  $(-1+x, y, z)$  and  $(-x, 1-y, -z)$  are given by “d”, “e”, “f” and “g”, respectively].

- inter-chain distance	$\text{Cl}(1)\cdots\text{H}(14\text{d})_{\text{aryl}}$ 2.94 Å	$\sum r_{\text{vdW}}(\text{Cl},\text{H})$ 3.01 Å
	$\text{O}(4)\cdots\text{H}(8\text{Ae})_{\text{methine}}$ 2.53 Å	$\sum r_{\text{vdW}}(\text{O},\text{H})$ 2.60 Å
	$\text{C}(3\text{f})\text{--}\text{H}(3\text{f})_{\text{aryl}}\cdots\text{Ph}_{\text{centroid}}\{\text{C}(10)\text{--}\text{C}(15)\}$ 2.98 Å	$\gamma = 4.7^\circ$
	$\text{C}(18\text{g})\text{--}\text{H}(18\text{Bg})_{\text{methine}}\cdots\text{Ph}_{\text{centroid}}\{\text{C}(10)\text{--}\text{C}(15)\}$ 2.83 Å	$\gamma = 16.8^\circ$

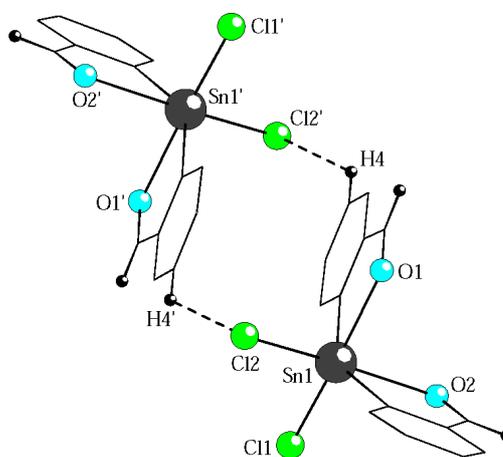
[2-(O=CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (**2**)

- the crystal contains a 1:1 mixture of  $\Delta$ -**2** and  $\Lambda$ -**2** isomers



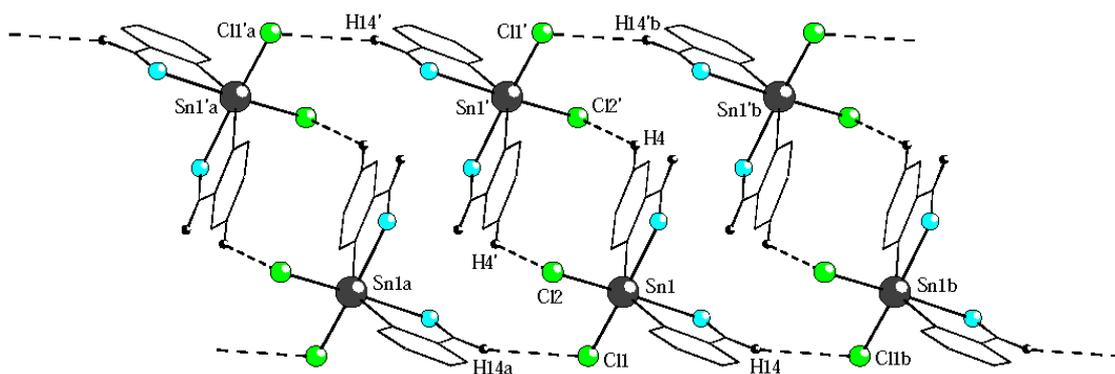
**Figure S6.** Molecular structure of  $\Delta$ -**2** isomer (*left*) and  $\Lambda$ -**2** isomer (*right*) in the crystal of **2**, showing the intramolecular chlorine-hydrogen contacts (only carbonyl hydrogens and hydrogen atoms involved in intramolecular contacts are shown).

- intramolecular distance  $\text{Cl}(1)\cdots\text{H}(6)_{\text{aryl}}$  2.87 Å  $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å  
 $\text{Cl}(2)\cdots\text{H}(13)_{\text{aryl}}$  2.84 Å



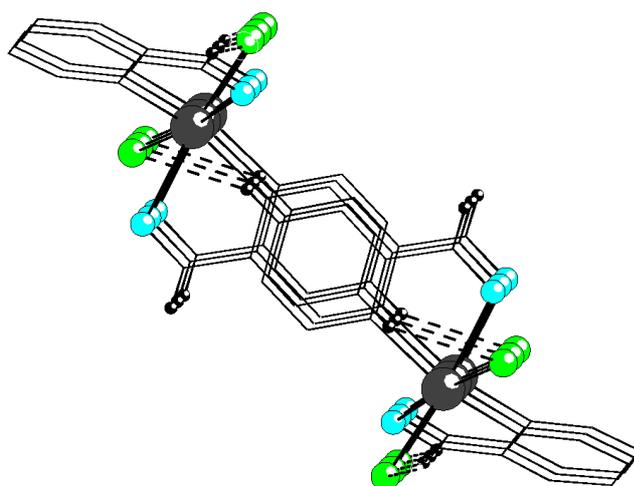
**Figure S7.** View along *a* axis of a dimer association of  $\Delta$ -**2** and  $\Lambda$ -**2** isomers based on intermolecular chlorine-hydrogen contacts in the crystal of **2** (only carbonyl hydrogens and hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $-x, 1-y, 2-z$ ) are given by “prime”].

- intermolecular distance  $\text{Cl}(2)\cdots\text{H}(4')_{\text{aryl}}$  2.93 Å  $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å

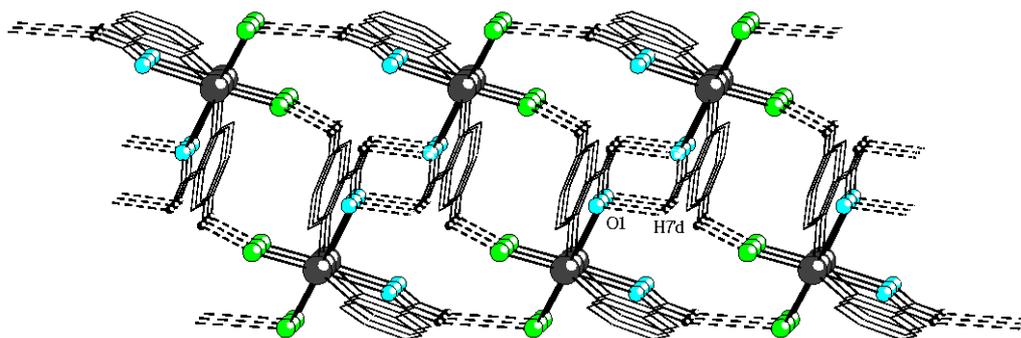


**Figure S8.** View along *a* axis of a ribbon-like polymer of dimers based on inter-dimer chlorine-hydrogen contacts in the crystal of **2** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms  $(-x, 1-y, 2-z)$ ,  $(x, y, 1+z)$ ,  $(-x, 1-y, 3-z)$ ,  $(x, y, -1+z)$  and  $(-x, 1-y, 1-z)$  are given by “prime”, “a”, “prime a”, “b” and “prime b”, respectively].

- inter-dimer distance  $\text{Cl}(1)\cdots\text{H}(14a)_{\text{carbonyl}}$  2.95 Å  $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å

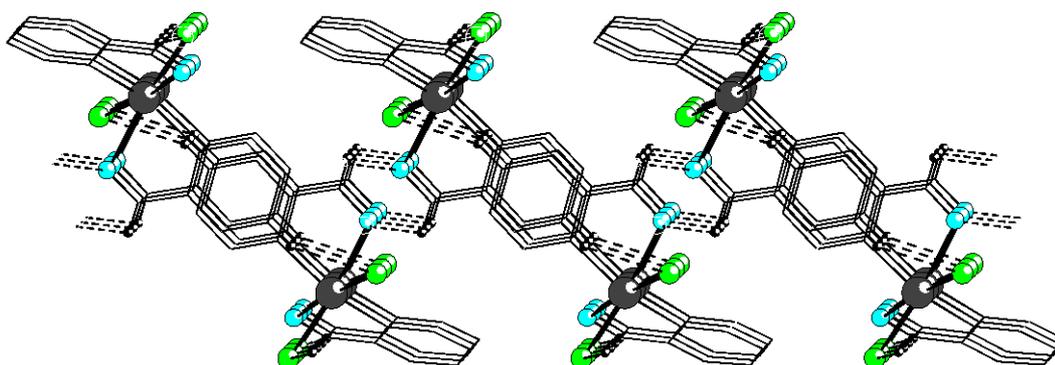


**Figure S9.** View along *c* axis of a ribbon-like polymer of dimers based on inter-dimer chlorine-hydrogen contacts in the crystal of **2** (only hydrogen atoms involved in intermolecular contacts are shown).



**Figure S10.** View along *a* axis of a layer based on inter-chain oxygen-hydrogen contacts in the crystal of **2** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $1-x$ ,  $1-y$ ,  $1-z$ ) are given by “prime d”].

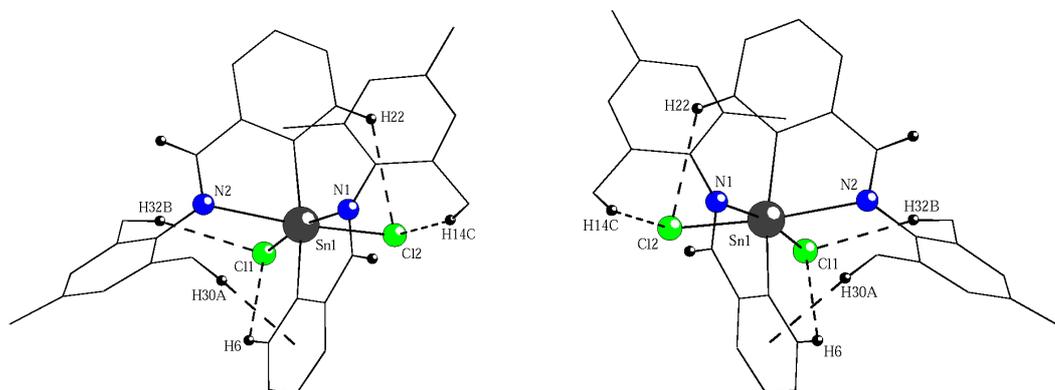
- inter-chain distance  $O(1)\cdots H(7'd)_{\text{carbonyl}}$  2.53 Å  $\sum r_{\text{vdW}}(O,H)$  2.60 Å
- no further contacts between parallel layers.



**Figure S11.** View along *c* axis of a layer based on inter-chain oxygen-hydrogen contacts in the crystal of **2** (only hydrogen atoms involved in intermolecular contacts are shown).

[2-(2',4',6'-Me<sub>3</sub>C<sub>6</sub>H<sub>2</sub>N=CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (**4**)

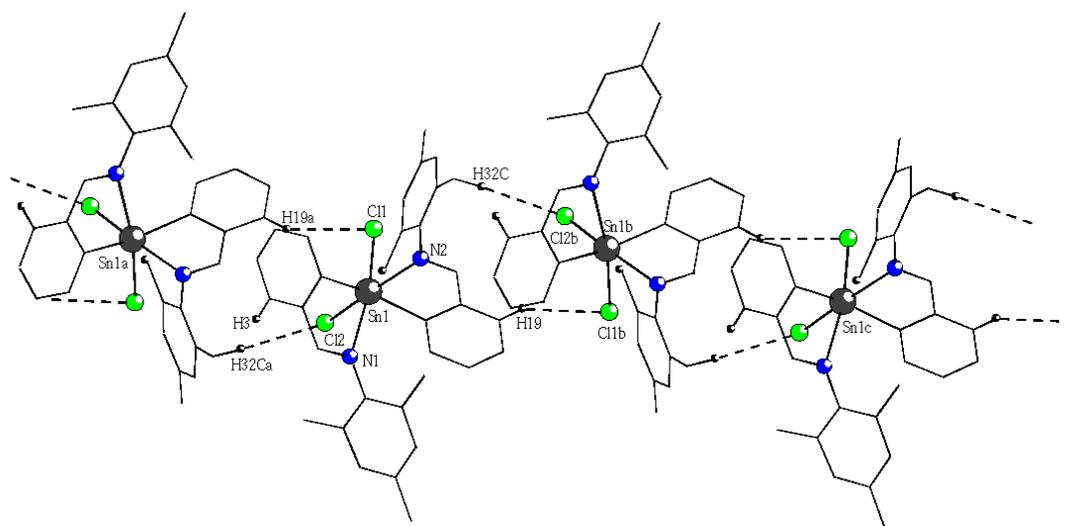
- the crystal contains a 1:1 mixture of  $\Delta$ -**4** and  $\Lambda$ -**4** isomers



**Figure S12.** Molecular structure of  $\Delta$ -**4** isomer (*left*) and  $\Lambda$ -**4** isomer (*right*) in the crystal of **4**, showing the intramolecular chlorine-hydrogen and C–H $\cdots\pi$  (Ph<sub>centroid</sub>) contacts (only imine hydrogens and hydrogen atoms involved in intramolecular contacts are shown).

- intramolecular distance
 

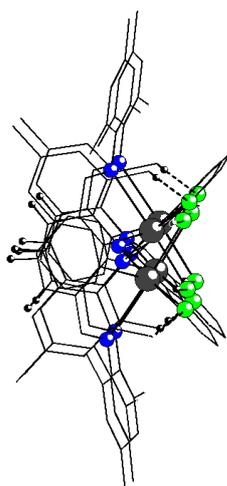
Cl(1) $\cdots$ H(6) <sub>aryl</sub>	2.93 Å	$\Sigma r_{\text{vdW}}(\text{Cl},\text{H})$ 3.01 Å
Cl(1) $\cdots$ H(32B) <sub>methyl</sub>	2.70 Å	
Cl(2) $\cdots$ H(22) <sub>aryl</sub>	2.96 Å	
Cl(2) $\cdots$ H(14C) <sub>methyl</sub>	2.72 Å	
C(30)–H(30A) <sub>methyl</sub> $\cdots$ Ph <sub>centroid</sub> {C(1)–C(6)}	2.71 Å	$\gamma = 19.1^\circ$



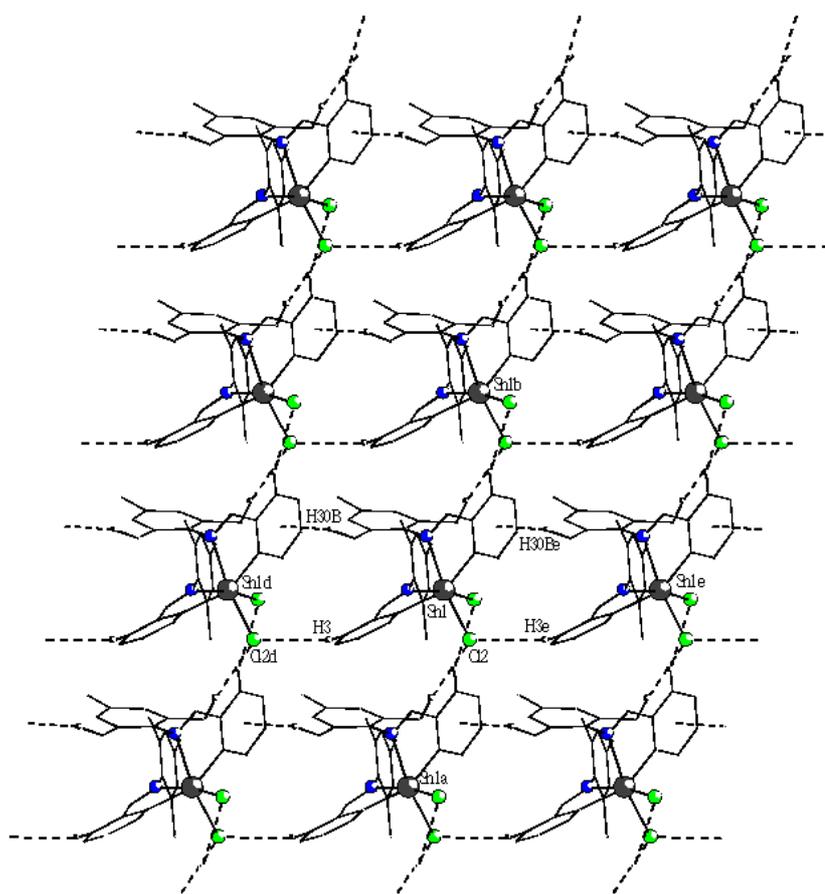
**Figure S13.** View of a chain polymer of alternating  $\Delta$ -**4** and  $\Lambda$ -**4** isomers based on intermolecular chlorine-hydrogen contacts in the crystal of **4** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $x, 0.5-y, 0.5+z$ ), ( $x, 0.5-y, -0.5+z$ ) and ( $x, y, -1+z$ ) are given by “a”, “b” and “c”, respectively].

- intermolecular distance
 

Cl(1) $\cdots$ H(19a) <sub>aryl</sub>	2.89 Å	$\Sigma r_{\text{vdW}}(\text{Cl},\text{H})$ 3.01 Å
Cl(2) $\cdots$ H(32Ca) <sub>methyl</sub>	2.93 Å	

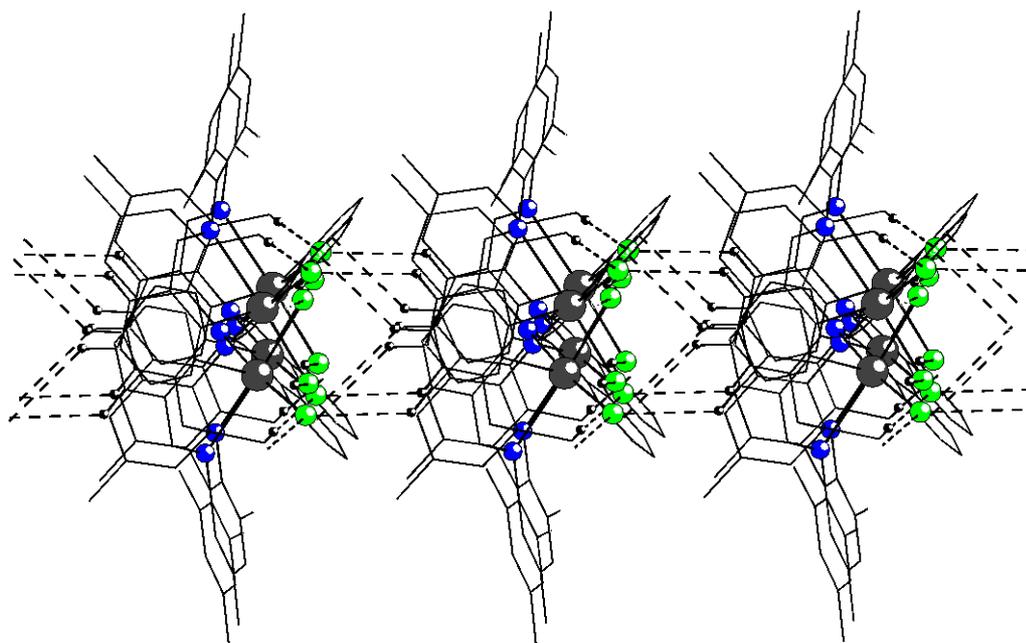


**Figure S14.** View along *c* axis of a chain polymer of alternating  $\Delta$ -4 and  $\Lambda$ -4 isomers in the crystal of **4**.



**Figure S15.** View along *b* axis of a layer based on inter-chain chlorine-hydrogen and C–H... $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts in the crystal of **4** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $-I+x, y, z$ ) and ( $I+x, y, z$ ) are given by “d” and “e”, respectively].

- inter-chain distance	$\text{Cl}(2) \cdots \text{H}(3\text{e})_{\text{aryl}}$ 2.91 Å	$\sum r_{\text{vdW}}(\text{Cl}, \text{H})$ 3.01 Å
	$\text{C}(30) - \text{H}(30\text{B})_{\text{methyl}} \cdots \text{Ph}_{\text{centroid}}\{\text{C}(17\text{d}) - \text{C}(22\text{d})\}$ 2.92 Å	$\gamma = 10.0^\circ$

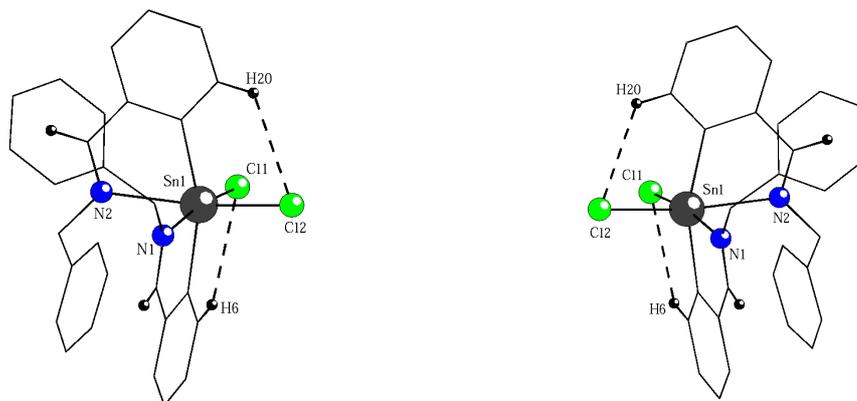


**Figure S16.** View along *c* axis of a layer based on inter-chain chlorine-hydrogen and C–H... $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts in the crystal of **4** (only hydrogen atoms involved in intermolecular contacts are shown).

- no further contacts between parallel layers.

**[2-(PhCH<sub>2</sub>N=CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (5)**

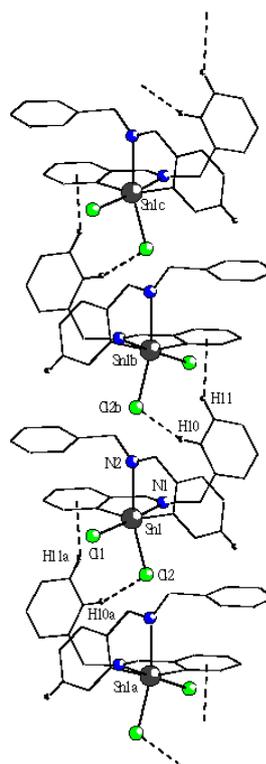
- the crystal contains a 1:1 mixture of  $\Delta$ -5 and  $\Lambda$ -5 isomers



**Figure S17.** Molecular structure of  $\Delta$ -5 isomer (*left*) and  $\Lambda$ -5 isomer (*right*) in the crystal of **5**, showing the intramolecular chlorine-hydrogen contacts (only imine hydrogens and hydrogen atoms involved in intramolecular contacts are shown).

- intramolecular distance

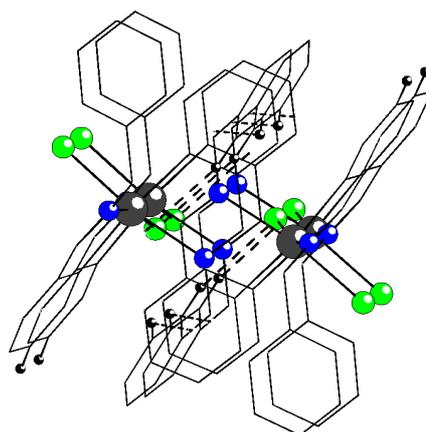
	Cl(1)···H(6) <sub>aryl</sub> 2.90 Å	$\sum r_{\text{vdW}}(\text{Cl}, \text{H})$ 3.01 Å
	Cl(2)···H(20) <sub>aryl</sub> 2.86 Å	



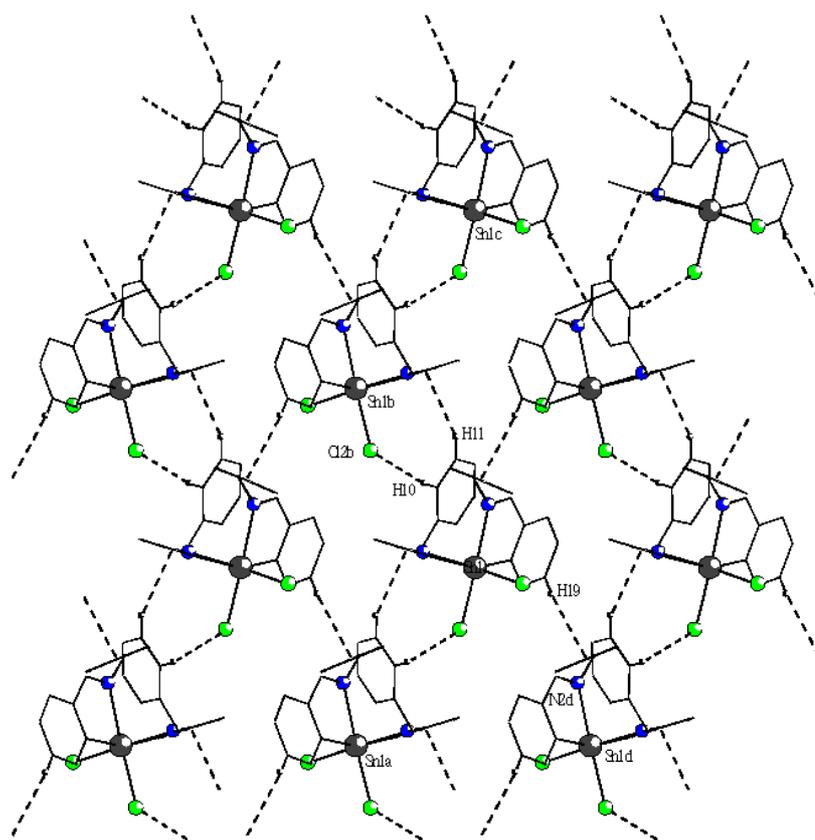
**Figure S18.** View along *a* axis of a zig-zag chain polymer of  $\Lambda$ -5 isomers based on intermolecular chlorine-hydrogen and C–H··· $\pi$  (Ph<sub>centroid</sub>) contacts in the crystal of **5** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $0.5-x$ ,  $-0.5+y$ ,  $1.5-z$ ), ( $0.5-x$ ,  $0.5+y$ ,  $1.5-z$ ) and ( $x$ ,  $1+y$ ,  $z$ ) are given by “a”, “b” and “c”, respectively].

- intermolecular distance

	Cl(2)···H(10a) <sub>aryl</sub> 2.95 Å	$\sum r_{\text{vdW}}(\text{Cl}, \text{H})$ 3.01 Å
	C(11)–H(11) <sub>aryl</sub> ···Ph <sub>centroid</sub> {C(1b)–C(6b)} 2.75 Å	$\gamma = 7.5^\circ$

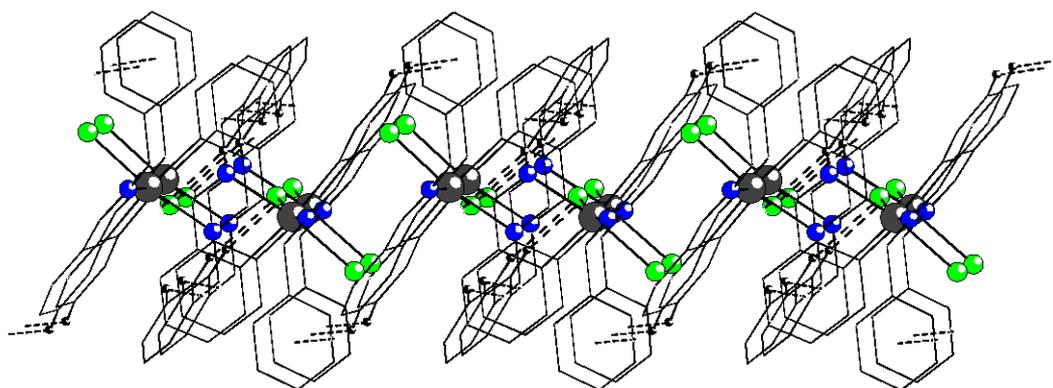


**Figure S19.** View along *b* axis of a *zig-zag* chain polymer of  $\Delta$ -**5** isomers based on intermolecular chlorine-hydrogen contacts in the crystal of **5** (only hydrogen atoms involved in intermolecular contacts are shown).



**Figure S20.** View along *c* axis of a layer based on C–H $\cdots$  $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts between alternating chains of  $\Delta$ -**5** and  $\Lambda$ -**5** isomers, respectively, in the crystal of **5** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms  $(0.5-x, -0.5+y, 1.5-z)$ ,  $(0.5-x, 0.5+y, 1.5-z)$ ,  $(x, 1+y, z)$  and  $(1.5-x, -0.5+y, 1.5-z)$  are given by “a”, “b”, “c” and “d”, respectively].

- inter-chain distance      C(19)–H(19)<sub>aryl</sub> $\cdots$ Ph<sub>centroid</sub>{C(23d)–C(28d)} 2.76 Å  
 $\gamma = 6.6^\circ$

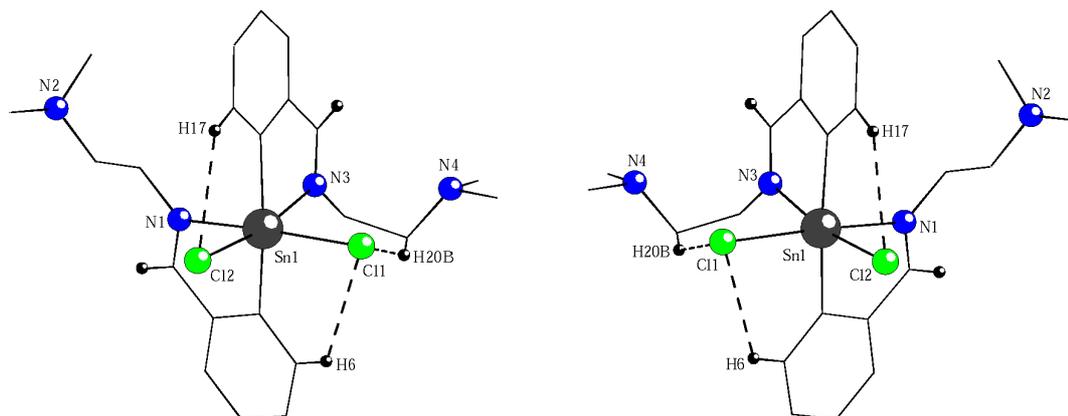


**Figure S21.** View along *b* axis of a layer based on C–H··· $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts between alternating chains of  $\Delta$ -**5** and  $\Lambda$ -**5** isomers, respectively, in the crystal of **5** (only hydrogen atoms involved in intermolecular contacts are shown).

- no further contacts between parallel layers.

[2-(Me<sub>2</sub>NCH<sub>2</sub>CH<sub>2</sub>N=CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (**6**)

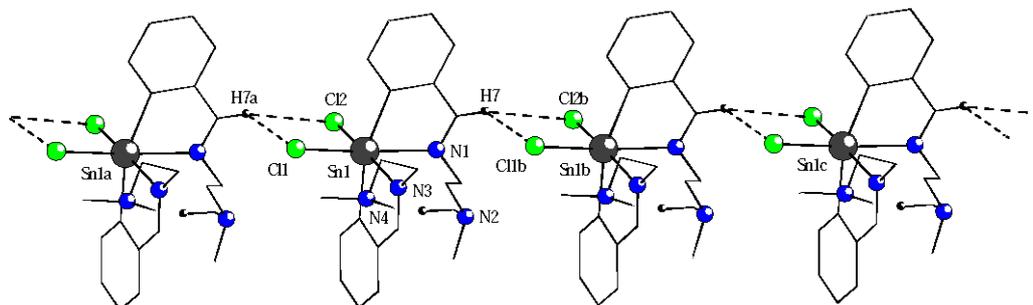
- the crystal contains a 1:1 mixture of  $\Delta$ -**6** and  $\Lambda$ -**6** isomers



**Figure S22.** Molecular structure of  $\Delta$ -**6** isomer (*left*) and  $\Lambda$ -**6** isomer (*right*) in the crystal of **6**, showing the intramolecular chlorine-hydrogen contacts (only imine hydrogens and hydrogen atoms involved in intramolecular contacts are shown).

- intramolecular distance

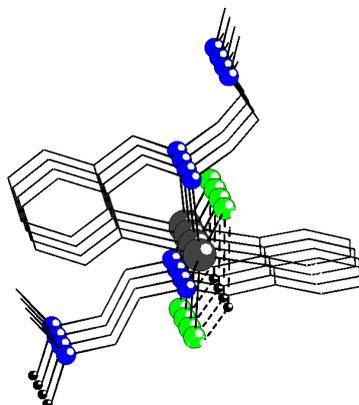
Cl(1)···H(6) <sub>aryl</sub>	2.97 Å	$\sum r_{vdW}(Cl,H)$ 3.01 Å
Cl(1)···H(20B) <sub>methylene</sub>	2.90 Å	
Cl(2)···H(17) <sub>aryl</sub>	2.93 Å	



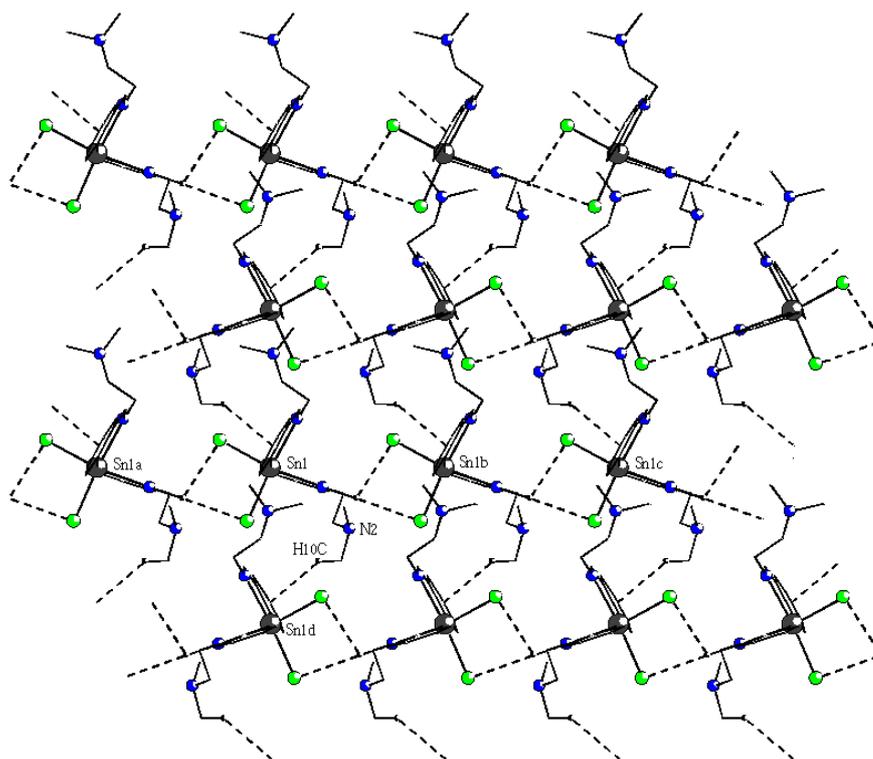
**Figure S23.** View along *b* axis of a chain polymer of  $\Lambda$ -**6** isomers based on intermolecular chlorine-hydrogen contacts in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms  $(-I+x, y, z)$ ,  $(I+x, y, z)$  and  $(2+x, y, z)$  are given by “a”, “b” and “c”, respectively].

- intermolecular distance

Cl(1)···H(7a) <sub>imine</sub>	2.92 Å	$\sum r_{vdW}(Cl,H)$ 3.01 Å
Cl(2)···H(7a) <sub>imine</sub>	2.95 Å	

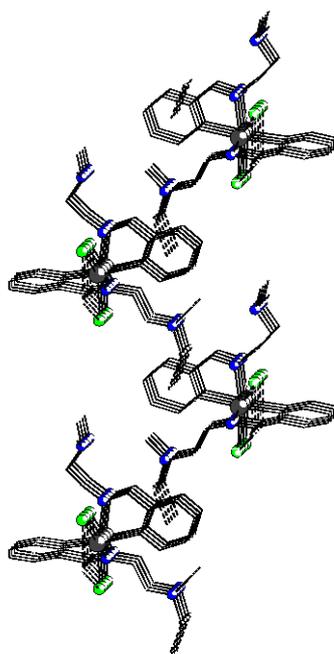


**Figure S24.** View along *a* axis of a chain polymer of  $\Lambda$ -**6** isomers based on intermolecular chlorine-hydrogen contacts in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown).



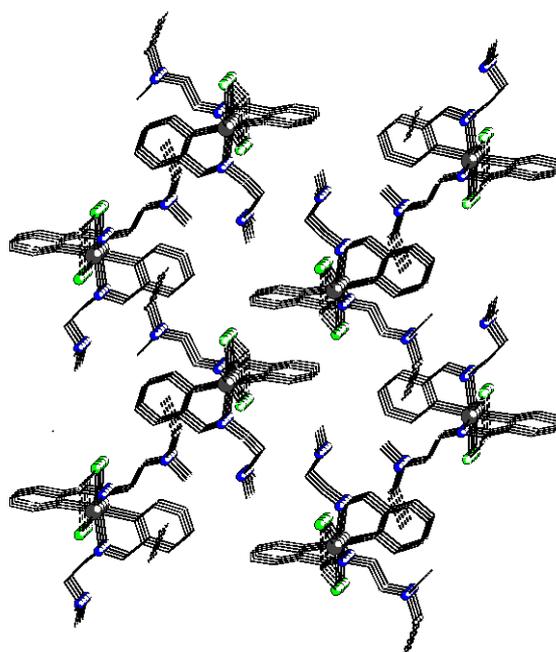
**Figure S25.** View along *c* axis of a wave-like layer based on C–H··· $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts between parallel chains of  $\Lambda$ -6 isomers in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms  $(-I+x, y, z)$ ,  $(I+x, y, z)$ ,  $(2+x, y, z)$  and  $(0.5+x, -0.5+y, 0.5-z)$  are given by “a”, “b”, “c” and “d”, respectively].

- inter-chain distance      C(10)–H(10C)<sub>methyl</sub>··· $\text{Ph}_{\text{centroid}}\{\text{C}(12\text{d})\text{--}\text{C}(17\text{d})\}$  2.82 Å  
 $\gamma = 12.3^\circ$



**Figure S26.** View along *b* axis of a wave-like layer based on C–H··· $\pi$  ( $\text{Ph}_{\text{centroid}}$ ) contacts between parallel chains of  $\Lambda$ -6 isomers in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown).

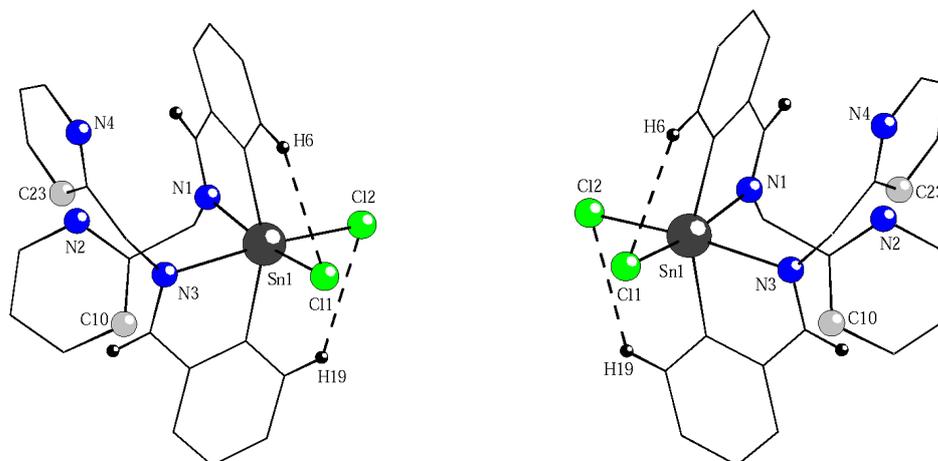
- no further contacts between parallel layers of  $\Delta$ -**6** and  $\Lambda$ -**6** isomers, respectively.



**Figure S27.** View along *b* axis of parallel wave-like layers of  $\Delta$ -**6** and  $\Lambda$ -**6** isomers, respectively, in the crystal of **6** (only hydrogen atoms involved in intermolecular contacts are shown).

[2-(2'-PyCH<sub>2</sub>N=CH)C<sub>6</sub>H<sub>4</sub>]<sub>2</sub>SnCl<sub>2</sub> (7)

- the crystal contains a 1:1 mixture of  $\Delta$ -7 and  $\Lambda$ -7 isomers



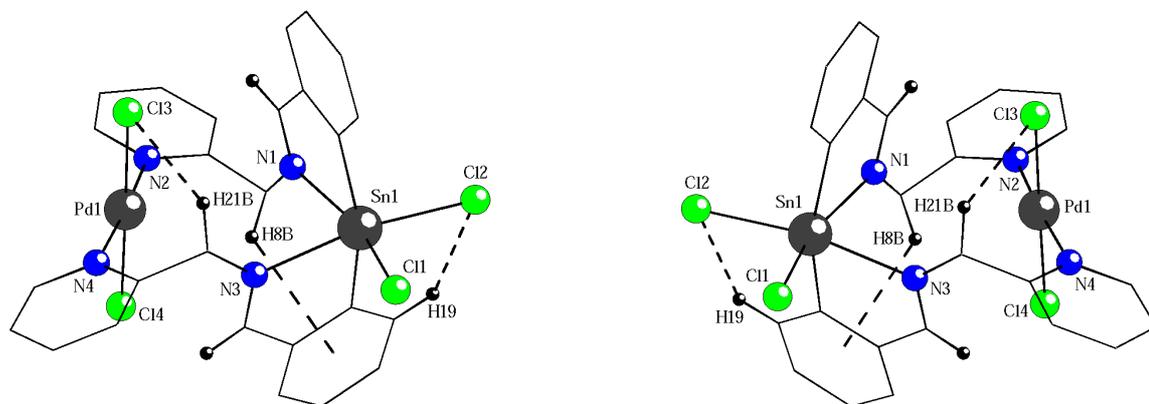
**Figure S28.** Molecular structure of  $\Delta$ -7 isomer (*left*) and  $\Delta$ -7 isomer (*right*) in the crystal of **6**, showing the intramolecular chlorine-hydrogen contacts (only imine hydrogens and hydrogen atoms involved in intramolecular contacts are shown). The occupancy degree is: 0.93 for N2 (0.07 for C10B) / 0.93 for C10 (0.07 for N2B) and 0.74 for N4 (0.26 for C23B) / 0.74 for C23 (0.26 for N4B).

- intramolecular distance

Cl(1)···H(6) <sub>aryl</sub>	2.85 Å	$\sum r_{\text{vdw}}(\text{Cl},\text{H})$ 3.01 Å
Cl(2)···H(19) <sub>aryl</sub>	2.92 Å	

[Cl<sub>2</sub>Pd{2-(2'-PyCH<sub>2</sub>N=CH)C<sub>6</sub>H<sub>4</sub>}<sub>2</sub>SnCl<sub>2</sub>]·CH<sub>3</sub>CN (8·CH<sub>3</sub>CN)

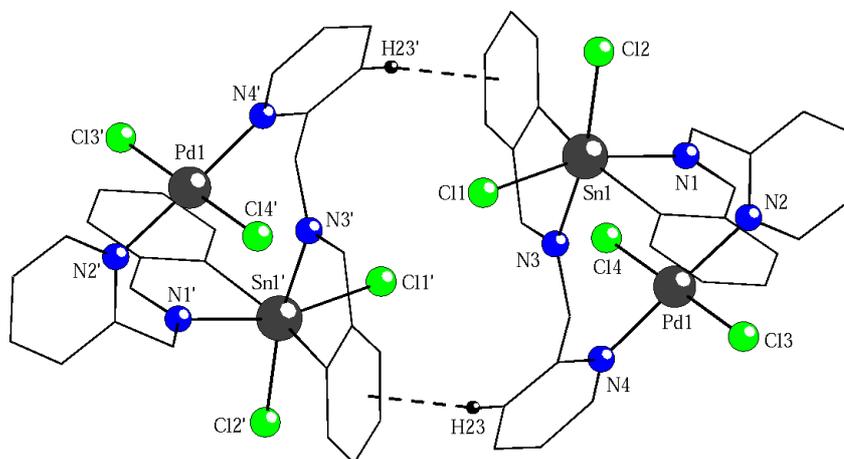
- the crystal contains a 1:1 mixture of Δ-8 and Λ-8 isomers



**Figure S29.** Molecular structure of Δ-8 isomer (*left*) and Λ-8 isomer (*right*) in the crystal of 8·CH<sub>3</sub>CN, showing the intramolecular chlorine-hydrogen and C–H···π (Ph<sub>centroid</sub>) contacts (only imine hydrogens and hydrogen atoms involved in intramolecular contacts are shown; the solvent molecule is omitted).

- intramolecular distance

Cl(2)···H(19) <sub>aryl</sub>	2.81 Å	$\sum r_{vdw}(Cl,H)$	3.01 Å
Cl(3)···H(21B) <sub>methylene</sub>	2.74 Å		
C(8)–H(8B) <sub>methylene</sub> ···Ph <sub>centroid</sub> {C(14)–C(19)}	2.97 Å		
		$\gamma$	25.5°

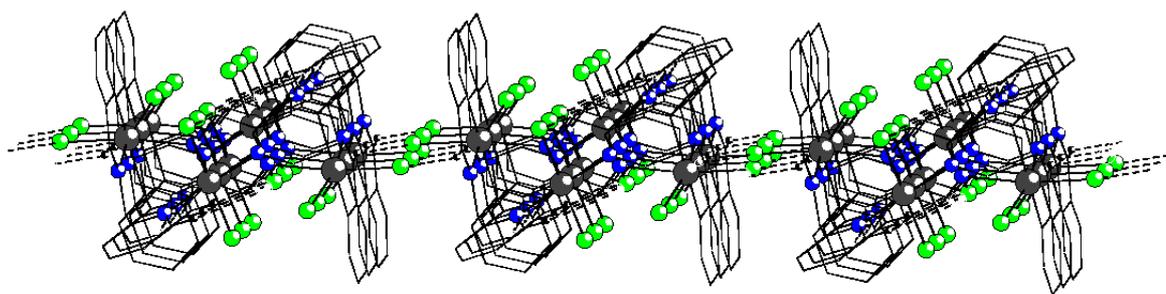


**Figure S30.** View along *a* axis of a dimer association of Δ-8 and Λ-8 isomers based on intermolecular C–H···π (Ph<sub>centroid</sub>) contacts in the crystal of 8·CH<sub>3</sub>CN (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms (*I*–*x*, *I*–*y*, *I*–*z*) are given by “prime”].

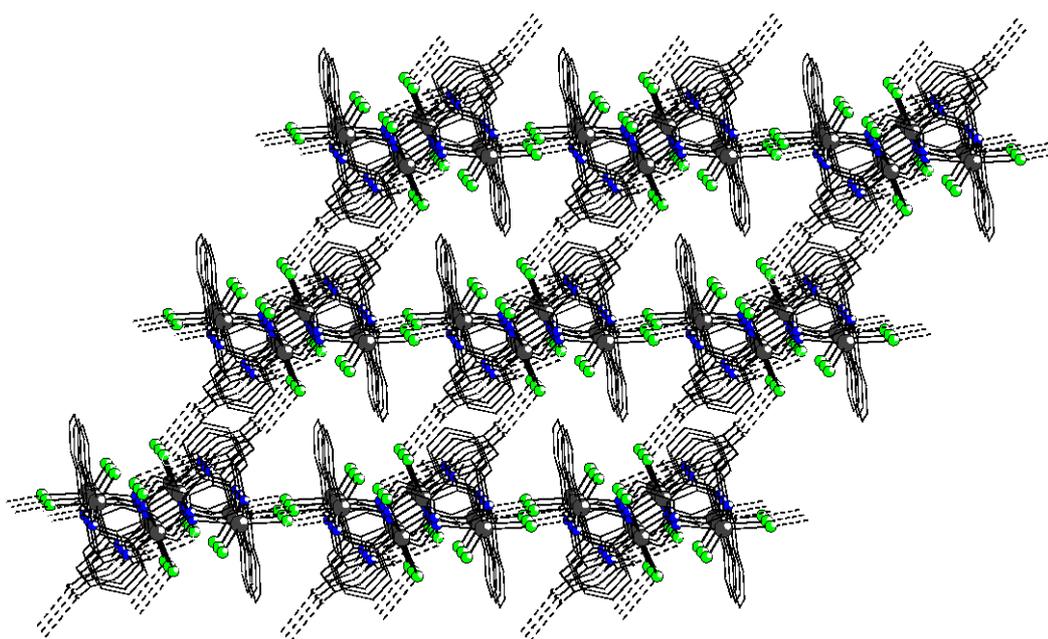
- intermolecular distance

C(23)–H(23) <sub>aryl</sub> ···Ph <sub>centroid</sub> {C(14')–C(19')}	2.61 Å		
		$\gamma$	2.8°





**Figure S33.** View along chain polymers of a layer based on inter-chain chlorine-hydrogen interactions in the crystal of **8**·CH<sub>3</sub>CN (only hydrogen atoms involved in intermolecular contacts are shown).



**Figure S34.** View of the 3D architecture in the crystal of **8**·CH<sub>3</sub>CN, based on chlorine-hydrogen contacts between parallel layers (only hydrogen atoms involved in intermolecular contacts are shown) [symmetry equivalent atoms ( $-x, 1-y, -z$ ) are given by “prime e”].

- inter-layer distance                      Cl(4)···H(11'e)<sub>aryl</sub> 2.83 Å                       $\sum r_{\text{vdW}}(\text{Cl},\text{H})$  3.01 Å