

Supplementary material

Design and fabrication of a series of metal-mediated assemblies with tetrapyridylporphyrins for supramolecular solar cells

Wen-Ting Deng^a, Jia-Cheng Liu*^a, Jing Cao^a, Dong-Cheng Hu^a, Ren-Zhi Li*^b,
Neng-Zhi Jin^c

^a Key Laboratory of Eco-Environment-Related Polymer Materials of Ministry of Education, Key Laboratory of Polymer Materials of Gansu Province, Key Laboratory of Bioelectrochemistry & Environmental Analysis of Gansu Province, College of Chemistry and Chemical Engineering, Northwest Normal University, Lanzhou 730070, P. R. China.

^b State Key Laboratory of Polymer Physics and Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, P. R. China.

^c Gansu Computing Center, Lanzhou 730030, P. R. China.

*Corresponding authors.

Tel: +869317971039, fax: +869317971989.

E-mail addresses: jcliu8@nwnu.edu.cn (J. C. Liu), renzhi.li@ciac.jl.cn (R. Z. Li).

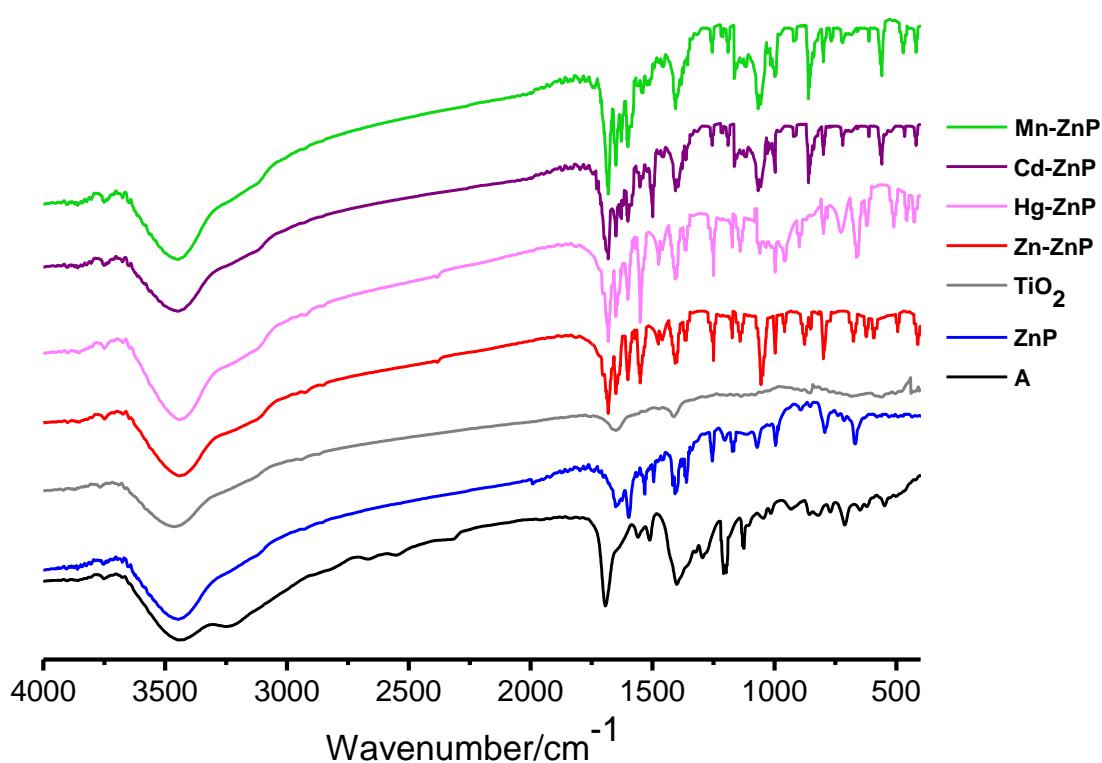


Fig. S1 The IR spectra of **A**, **TiO₂**, **ZnP** and **ZnP-M-A**.

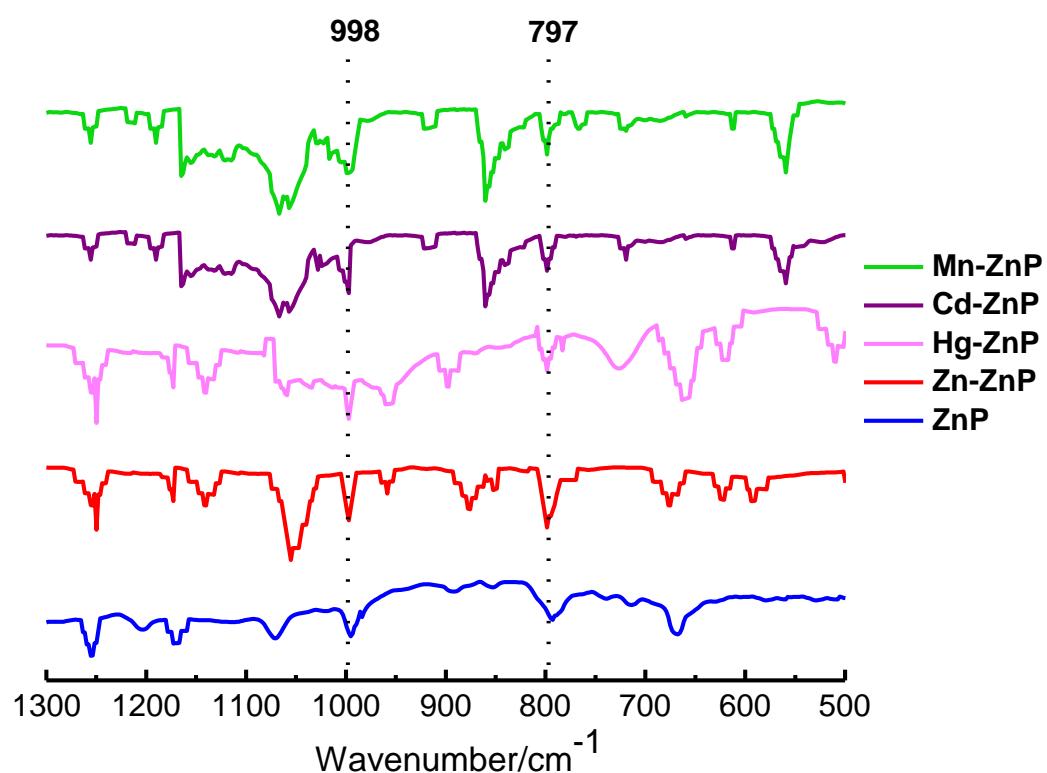


Fig. S2 The IR spectra of **ZnP** and **ZnP-M-A** in the region of 500–1300 cm^{−1}.

Table S1. The peak wavelength of the UV-vis absorption and fluorescence spectra of **ZnP** in DMF solution and these assemblies absorbed on TiO₂ thin films.

Dye	Absorption		Emission/nm
	$\lambda_{\text{max}}/\text{nm}$		
ZnP	424, 577, 598		608, 653
ZnP-Zn-A	445, 575, 618		601
ZnP-Hg-A	438, 569, 611		599
ZnP-Cd-A	437, 571, 612		603
ZnP-Mn-A	438, 572, 613		604

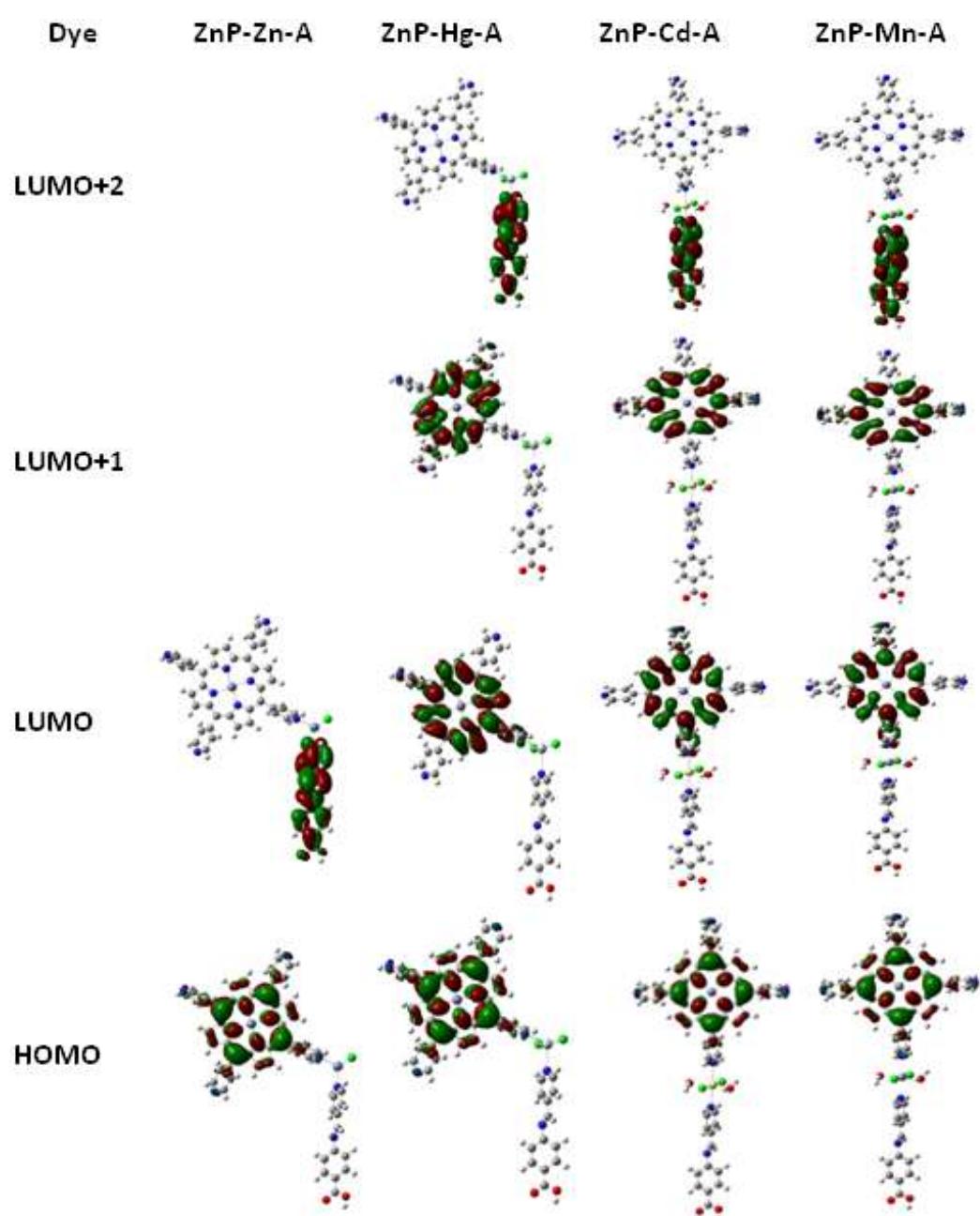


Fig. S3 Molecular orbital distributions of these assemblies.

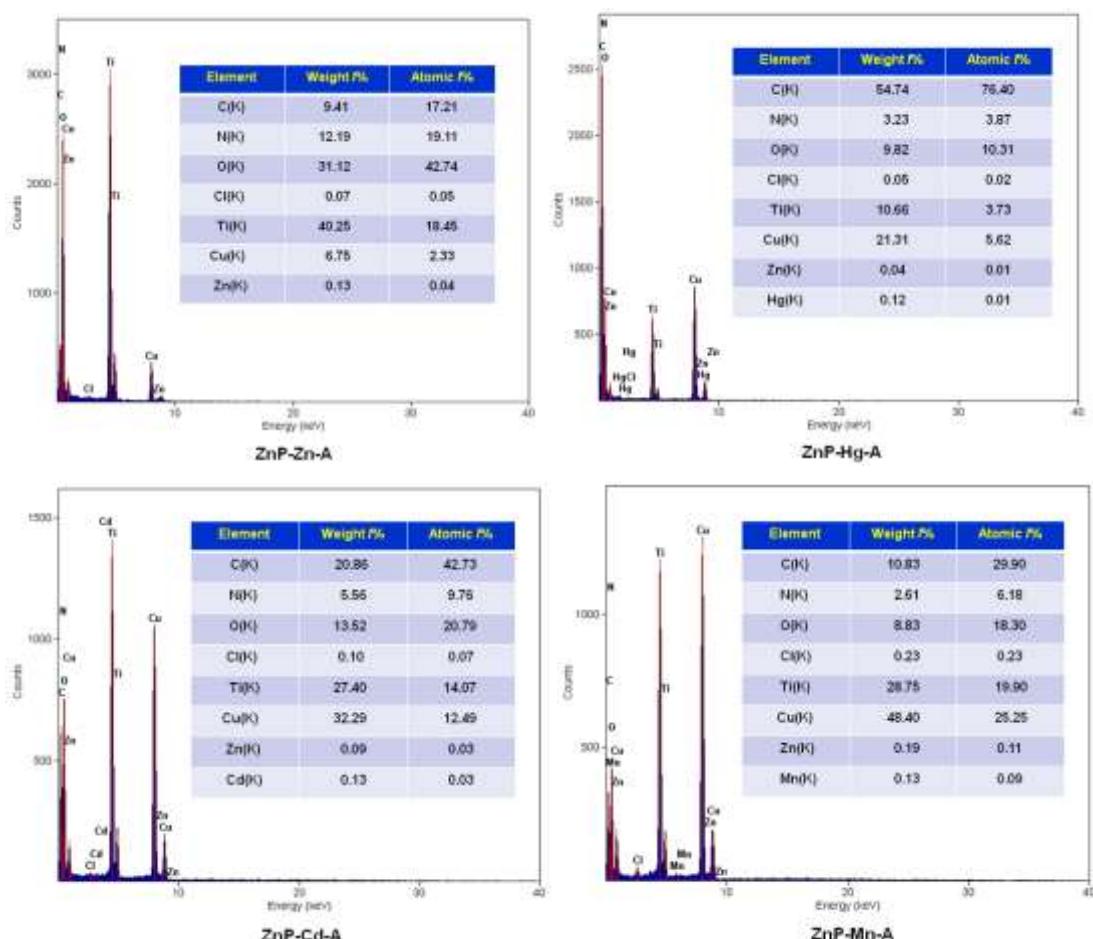


Fig. S4 EDX spectra of the assemblies immobilized on the TiO_2 nanoparticles. The inset tables give the data of elemental analyses.