

## Supplementary Information for

# Complexation of Cm(III) and Eu(III) with CyMe<sub>4</sub>-BTPhen and CyMe<sub>4</sub>-BTBP studied by Time Resolved Laser Fluorescence Spectroscopy

Antje Bremer,<sup>\*a,b</sup> Daniel M. Whittaker,<sup>c</sup> Clint A. Sharrad,<sup>c,d,e</sup> Andreas Geist<sup>a</sup> and Petra J. Panak<sup>a,b</sup>

<sup>a</sup> Karlsruher Institut für Technologie, Institut für Nukleare Entsorgung, P. O. Box 3640, 76021 Karlsruhe, Germany.

<sup>b</sup> Ruprecht-Karls-Universität Heidelberg, Institut für Physikalische Chemie, Im Neuenheimer Feld 234, 69120 Heidelberg, Germany.

<sup>c</sup> Centre of Radiochemistry Research, School of Chemistry, The University of Manchester, Oxford Road, Manchester M13 9PL, United Kingdom.

<sup>d</sup> School of Chemical Engineering and Analytical Science, The University of Manchester, Oxford Road, Manchester M13 9PL, United Kingdom.

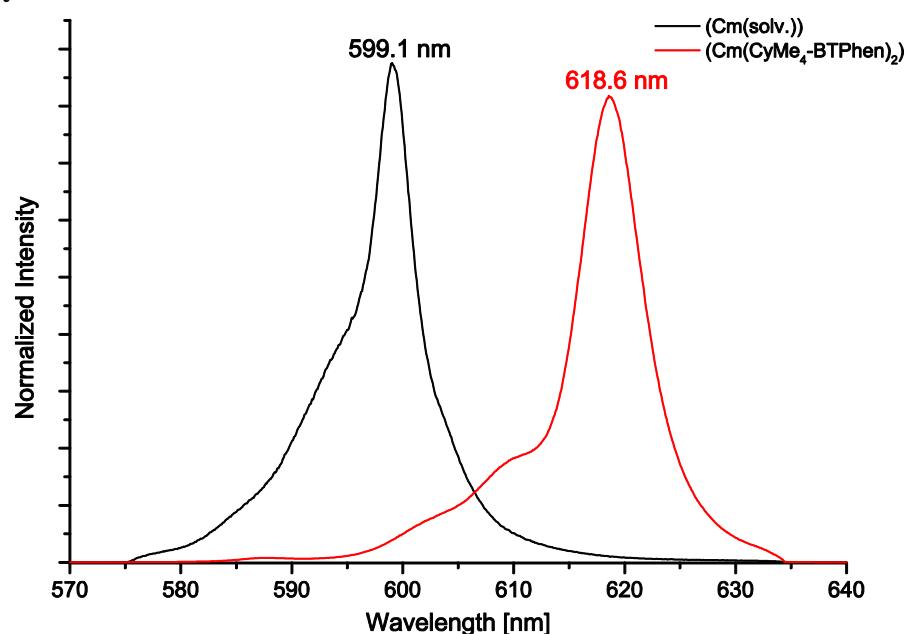
<sup>e</sup> Research Centre for Radwaste and Decommissioning, Dalton Nuclear Institute, The University of Manchester, Oxford Road, Manchester M13 9PL, United Kingdom.

## Table of Contents

Emission spectra of the (M(solv.)) and (ML <sub>2</sub> ) complexes .....	3
Cm(III) – CyMe <sub>4</sub> -BTPhen .....	3
Cm(III) – CyMe <sub>4</sub> -BTBP .....	3
Eu(III) – CyMe <sub>4</sub> -BTPhen .....	4
Eu(III) – CyMe <sub>4</sub> -BTBP .....	5
Overview Cm(III) batch experiments.....	6
Batch experiment Cm(III) – CyMe <sub>4</sub> -BTBP .....	7
Emission spectra.....	7
Speciation diagram.....	7
Overview Eu(III) batch experiments .....	8
Batch experiment Eu(III) – CyMe <sub>4</sub> -BTPhen – slope analysis.....	9
Batch experiments Eu(III) – CyMe <sub>4</sub> -BTBP .....	10
Emission spectra.....	10
Speciation diagram.....	11
Slope analysis .....	12
Comparison of mono- and biphasic experiments with CyMe <sub>4</sub> -BTBP .....	13
Cm(III) .....	13
Eu(III) .....	14

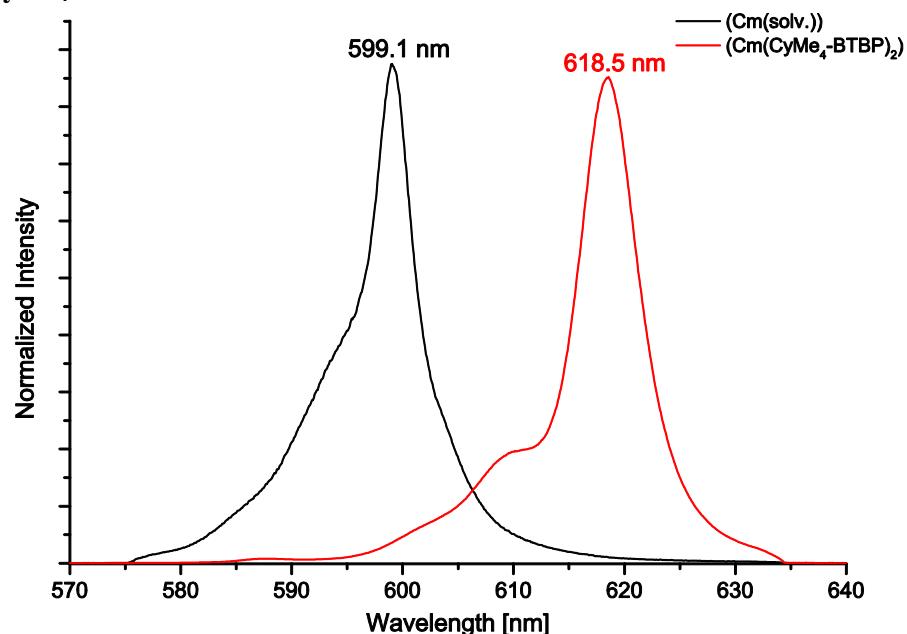
### Emission spectra of the (M(solv.)) and (ML<sub>2</sub>) complexes

#### Cm(III) – CyMe<sub>4</sub>-BTPhen



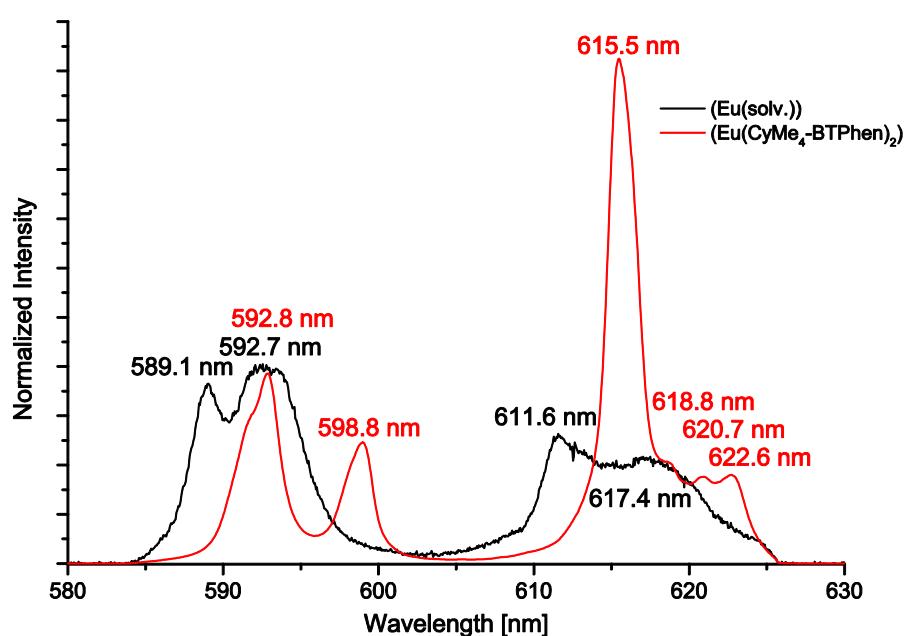
**Fig. S1** Normalized fluorescence spectra of the solvated Cm(III) ion and the Cm(III)-CyMe<sub>4</sub>-BTPhen 1:2 complex in methanol (3.3 mol% water, c(HClO<sub>4</sub>) = 91.2 mM).

#### Cm(III) – CyMe<sub>4</sub>-BTBP

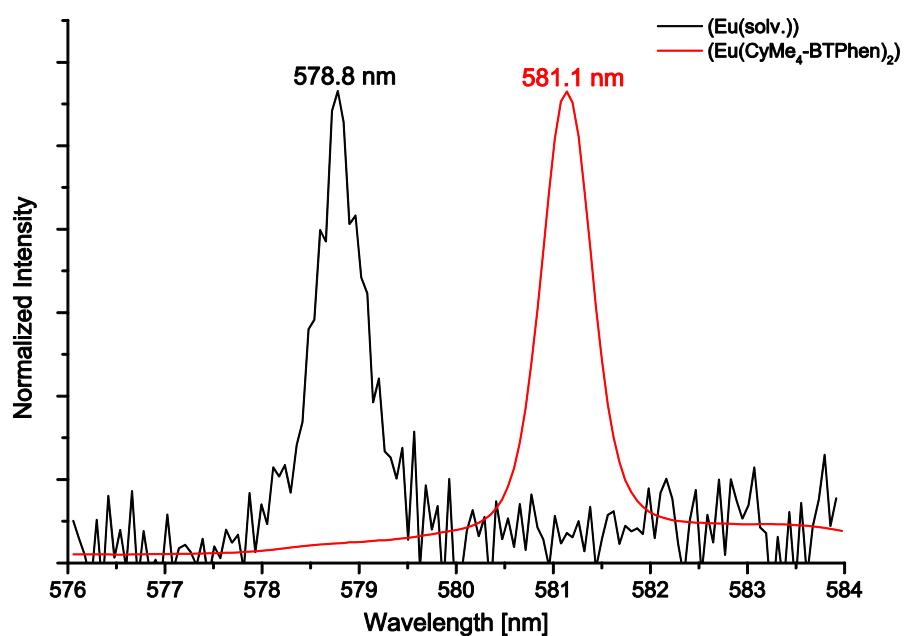


**Fig. S2** Normalized fluorescence spectra of the solvated Cm(III) ion and the Cm(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (3.3 mol% water, c(HClO<sub>4</sub>) = 91.2 mM).

**Eu(III) – CyMe<sub>4</sub>-BTPhen**

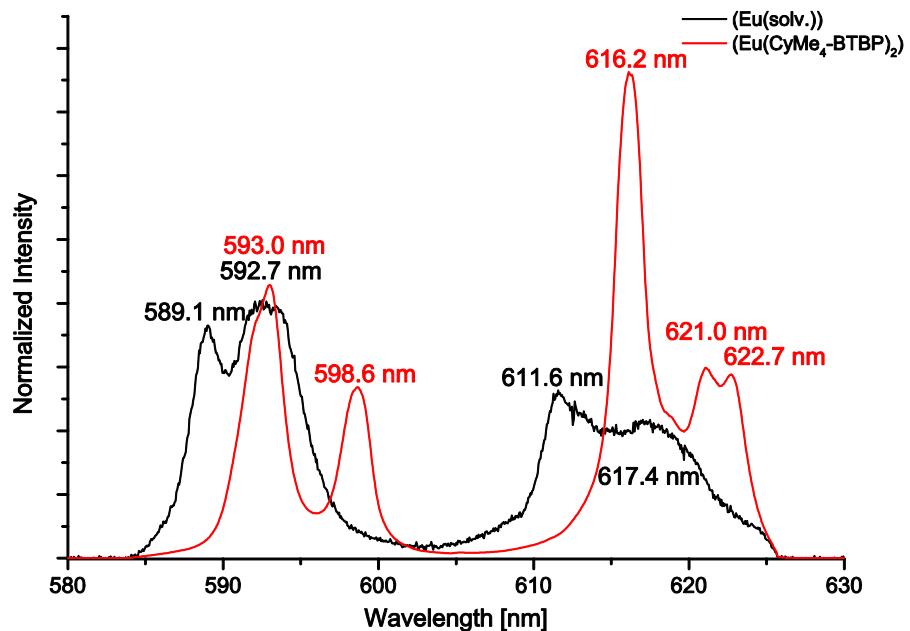


**Fig. S3** Normalized fluorescence spectra resulting from the  $^5D_0 \rightarrow ^7F_1$  and  $^5D_0 \rightarrow ^7F_2$  transitions of the solvated Eu(III) ion and the Eu(III)-CyMe<sub>4</sub>-BTPhen 1:2 complex in methanol (3.3 mol% water,  $c(\text{HClO}_4) = 91.2 \text{ mM}$ ).

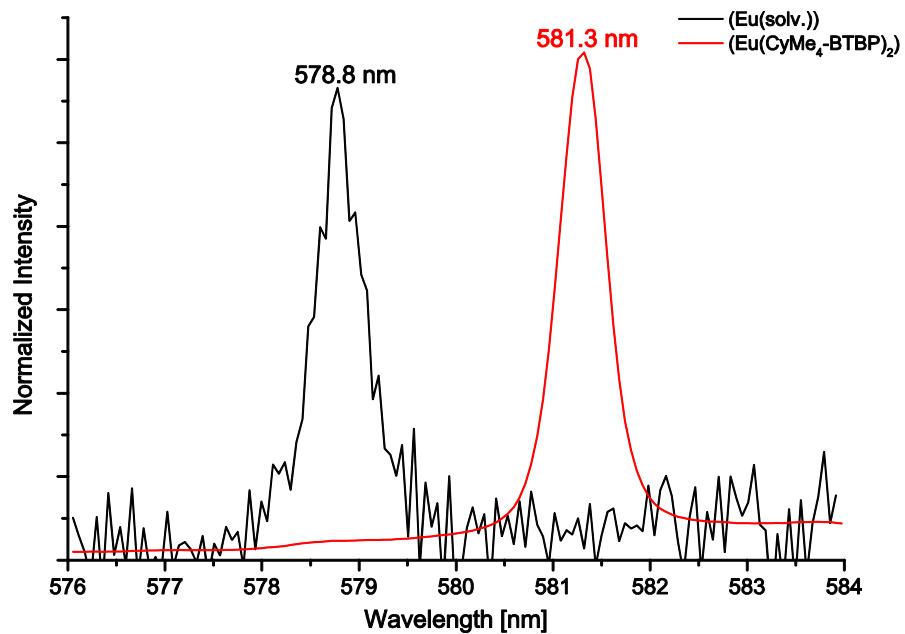


**Fig. S4** Normalized fluorescence spectra resulting from the  $^5D_0 \rightarrow ^7F_0$  transition of the solvated Eu(III) ion and the Eu(III)-CyMe<sub>4</sub>-BTPhen 1:2 complex in methanol (3.3 mol% water,  $c(\text{HClO}_4) = 91.2 \text{ mM}$ ).

**Eu(III) – CyMe<sub>4</sub>-BTBP**



**Fig. S5** Normalized fluorescence spectra resulting from the  ${}^5D_0 \rightarrow {}^7F_1$  and  ${}^5D_0 \rightarrow {}^7F_2$  transitions of the solvated Eu(III) ion and the Eu(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (3.3 mol% water,  $c(\text{HClO}_4) = 91.2 \text{ mM}$ ).



**Fig. S6** Normalized fluorescence spectra resulting from the  ${}^5D_0 \rightarrow {}^7F_0$  transition of the solvated Eu(III) ion and the Eu(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (3.3 mol% water,  $c(\text{HClO}_4) = 91.2 \text{ mM}$ ).

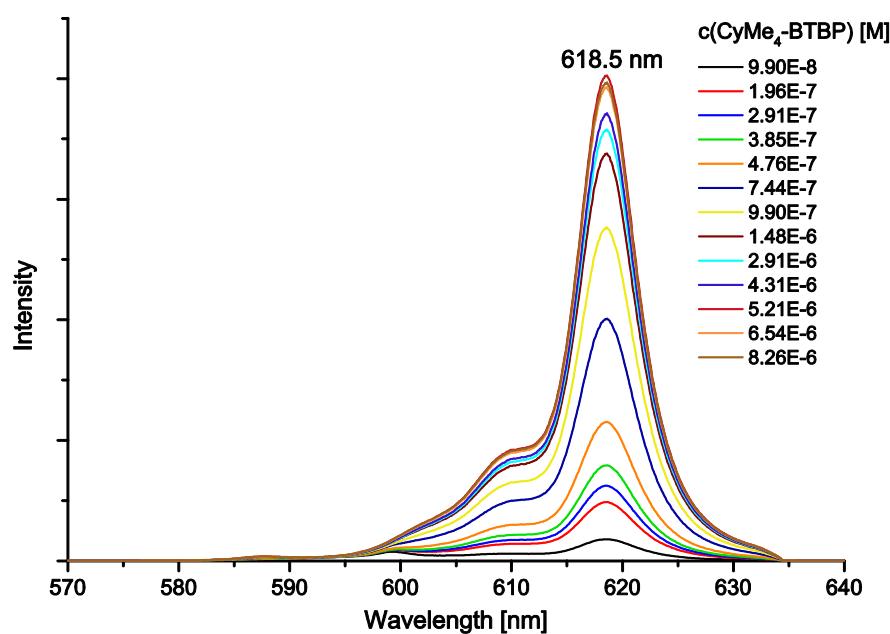
## Overview Cm(III) batch experiments

**Table S1** Overview of Cm(III) batch experiments performed with CyMe<sub>4</sub>-BTPhen and CyMe<sub>4</sub>-BTBP.

experiment name	log β <sub>2</sub>	average
Cm-BTPhen 1	13.6	
Cm-BTPhen 2	13.9	13.8
Cm-BTBP 1	12.4	
Cm-BTBP 2	12.4	12.4

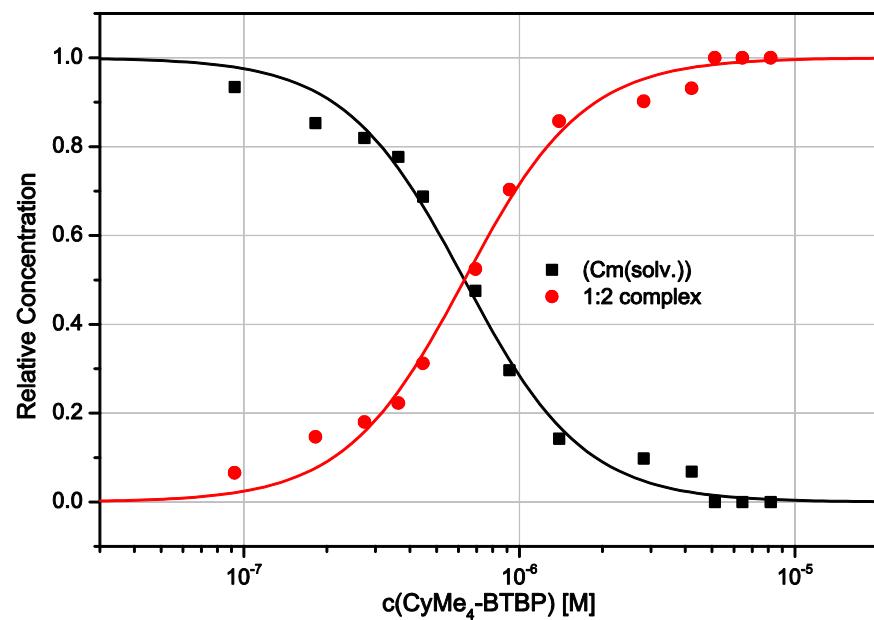
### Batch experiment Cm(III) – CyMe<sub>4</sub>-BTBP

#### Emission spectra



**Fig. S7** Fluorescence spectra of Cm(III) in methanol (with 3.3 mol% water) upon increasing CyMe<sub>4</sub>-BTBP concentration ( $c(\text{Cm(III)})_{\text{ini}} = 5 \cdot 10^{-8} \text{ M}$ ,  $c(\text{CyMe}_4\text{-BTBP}) = 9.90 \cdot 10^{-8} \text{ M} - 8.26 \cdot 10^{-6} \text{ M}$ ,  $c(\text{H}^+) = 91.2 \text{ mM}$ ).

#### Speciation diagram



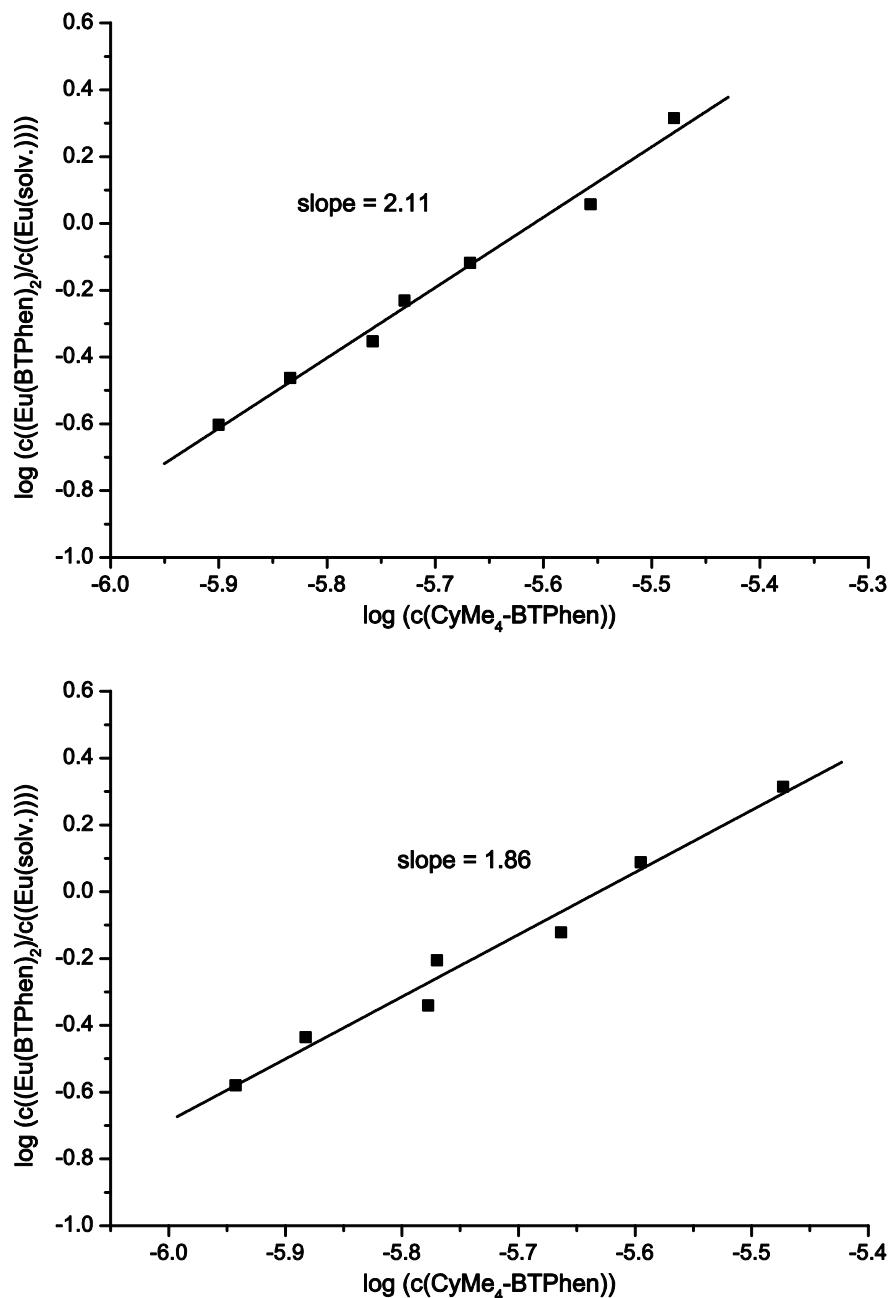
**Fig. S8** Relative concentrations of (Cm(solv.)) and the Cm(III)-CyMe<sub>4</sub>-BTBP 1:2 complex as a function of the free CyMe<sub>4</sub>-BTBP concentration in methanol (with 3.3 mol% water). Symbols represent experimental data, lines designate calculations.

## Overview Eu(III) batch experiments

**Table S2** Overview of Eu(III) batch experiments performed with CyMe<sub>4</sub>-BTPhen and CyMe<sub>4</sub>-BTBP.

experiment name	emission band(s)	log β <sub>2</sub>	average
Eu-BTPhen 1	<sup>7</sup> F <sub>1</sub> and <sup>7</sup> F <sub>2</sub>	11.9	11.6
	<sup>7</sup> F <sub>0</sub>	12.0	
Eu-BTPhen 2	<sup>7</sup> F <sub>1</sub> and <sup>7</sup> F <sub>2</sub>	11.2	11.3
	<sup>7</sup> F <sub>0</sub>	11.3	
Eu-BTBP 1	<sup>7</sup> F <sub>1</sub> and <sup>7</sup> F <sub>2</sub>	11.6	11.3
	<sup>7</sup> F <sub>0</sub>	11.6	
Eu-BTBP 2	<sup>7</sup> F <sub>1</sub> and <sup>7</sup> F <sub>2</sub>	11.0	11.3
	<sup>7</sup> F <sub>0</sub>	11.0	
Eu-BTBP 3	<sup>7</sup> F <sub>1</sub> and <sup>7</sup> F <sub>2</sub>	11.3	11.3
	<sup>7</sup> F <sub>0</sub>	11.3	

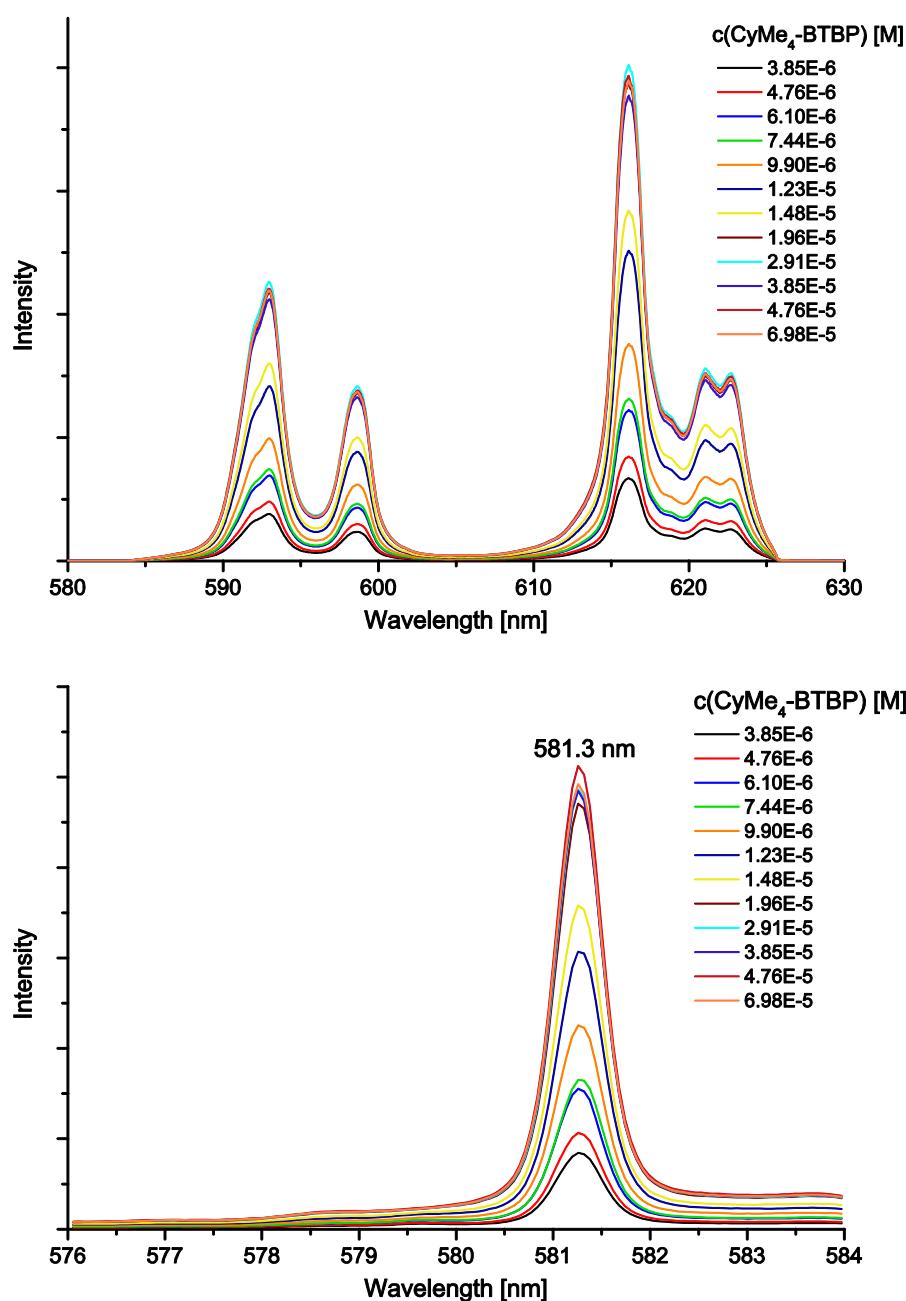
**Batch experiment Eu(III) – CyMe<sub>4</sub>-BTPhen – slope analysis**



**Fig. S9** Double logarithmic plot of  $c\left(\text{Eu}(\text{CyMe}_4\text{-BTPhen})_2\right)/c\left(\text{Eu}(\text{solv.})\right)$  concentration ratio vs free ligand concentration resulting from the analysis of the  $^5\text{D}_0 \rightarrow ^7\text{F}_1$  and  $^5\text{D}_0 \rightarrow ^7\text{F}_2$  emission bands (top) or  $^5\text{D}_0 \rightarrow ^7\text{F}_0$  emission band (bottom).

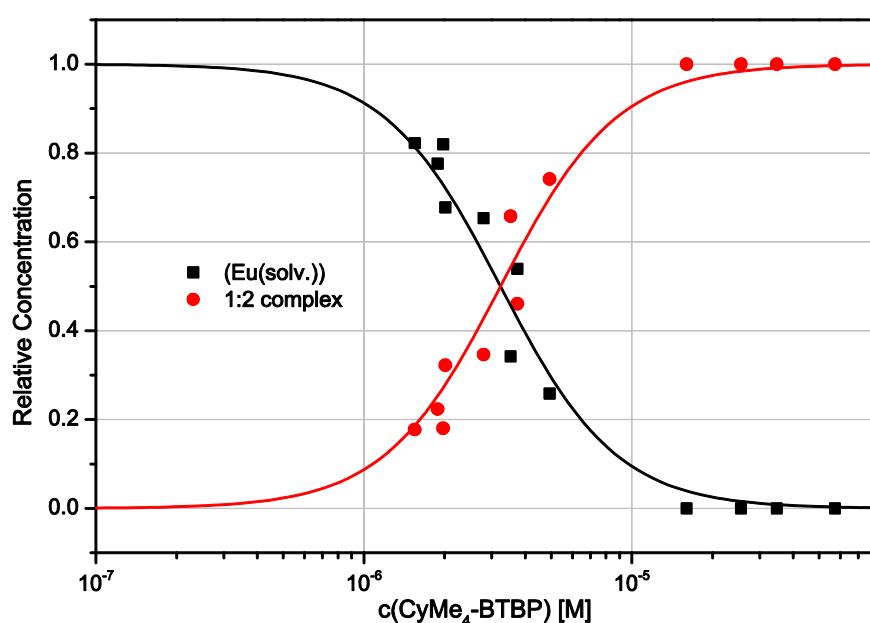
## Batch experiments Eu(III) – CyMe<sub>4</sub>-BTBP

### Emission spectra



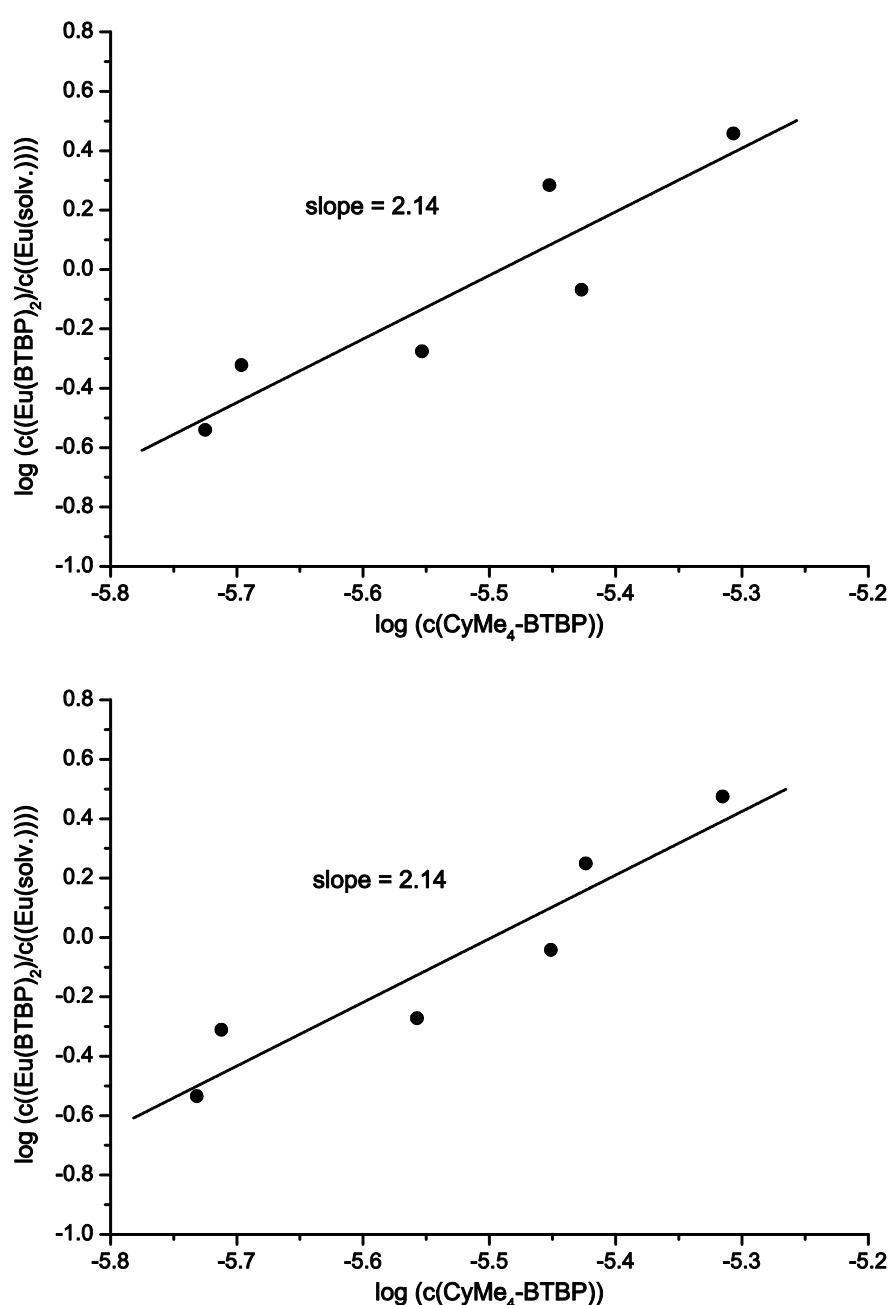
**Fig. S10** Fluorescence spectra resulting from the  $^5D_0 \rightarrow ^7F_1$  and  $^5D_0 \rightarrow ^7F_2$  transitions (top) or  $^5D_0 \rightarrow ^7F_0$  transition (bottom) of Eu(III) in methanol (with 3.3 mol% water) upon increasing CyMe<sub>4</sub>-BTBP concentration ( $c(\text{Eu(III)})_{\text{ini}} = 5 \cdot 10^{-8} \text{ M}$ ,  $c(\text{CyMe}_4\text{-BTBP}) = 3.85 \cdot 10^{-6} \text{ M} - 6.98 \cdot 10^{-5} \text{ M}$ ,  $c(\text{H}^+) = 91.2 \text{ mM}$ ).

### Speciation diagram



**Fig. S11** Relative concentrations of (Eu(solv.)) and the Eu(III)-CyMe<sub>4</sub>-BTBP 1:2 complex as a function of the free CyMe<sub>4</sub>-BTBP concentration in methanol (with 3.3 mol% water). Symbols represent experimental data resulting from the analysis of the <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>1</sub> and <sup>5</sup>D<sub>0</sub> → <sup>7</sup>F<sub>2</sub> emission bands, lines designate calculations.

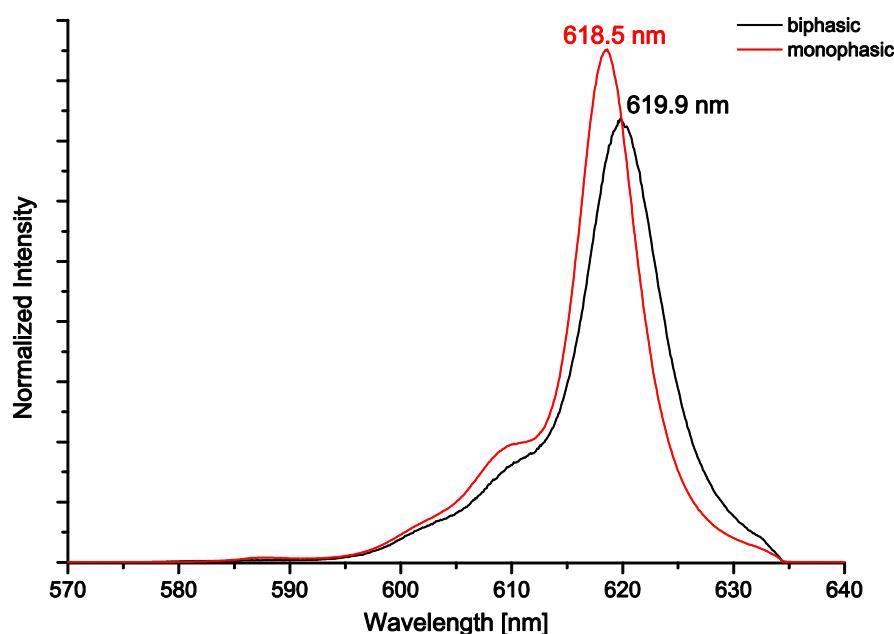
### Slope analysis



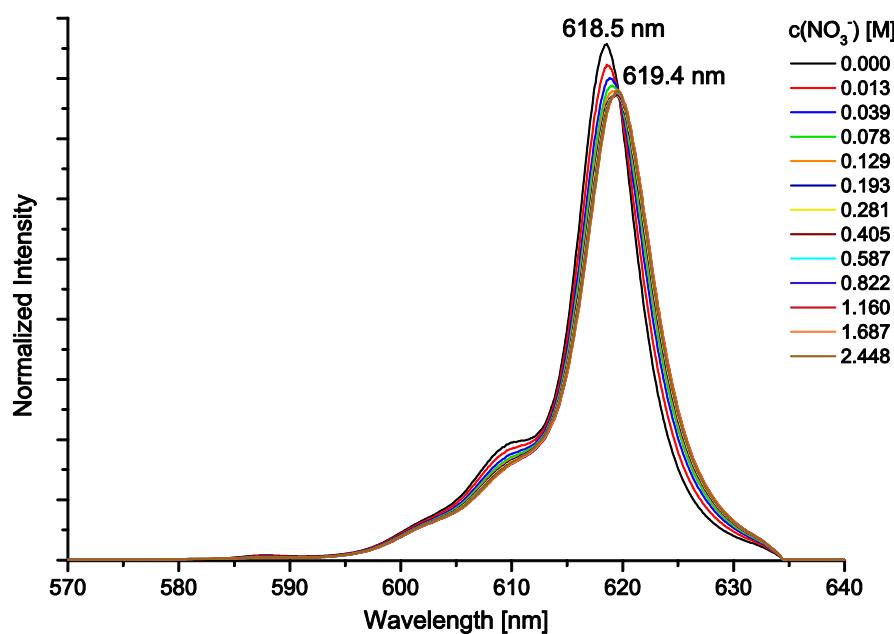
**Fig. S12** Double logarithmic plot of  $c\left(\text{Eu}(\text{CyMe}_4\text{-BTBP})_2\right)/c\left(\text{Eu}(\text{solv.})\right)$  concentration ratio vs free ligand concentration resulting from the analysis of the  $^5\text{D}_0 \rightarrow ^7\text{F}_1$  and  $^5\text{D}_0 \rightarrow ^7\text{F}_2$  emission bands (top) or  $^5\text{D}_0 \rightarrow ^7\text{F}_0$  emission band (bottom).

### Comparison of mono- and biphasic experiments with CyMe<sub>4</sub>-BTBP

Cm(III)

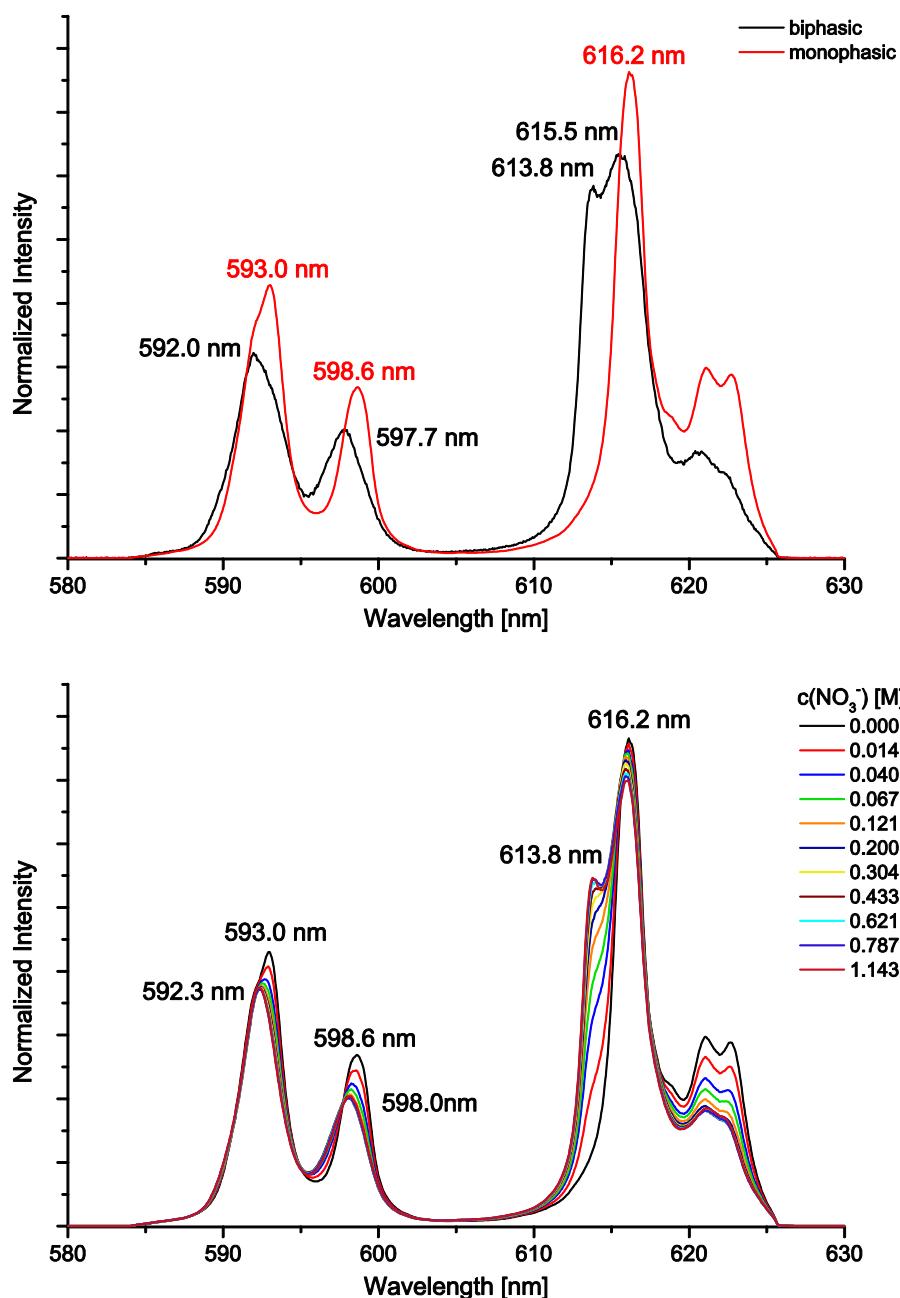


**Fig. S13** Normalized fluorescence spectra of Cm(III) in the organic phase of an extraction experiment with CyMe<sub>4</sub>-BTBP and of the Cm(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (with 3.3 mol% water).

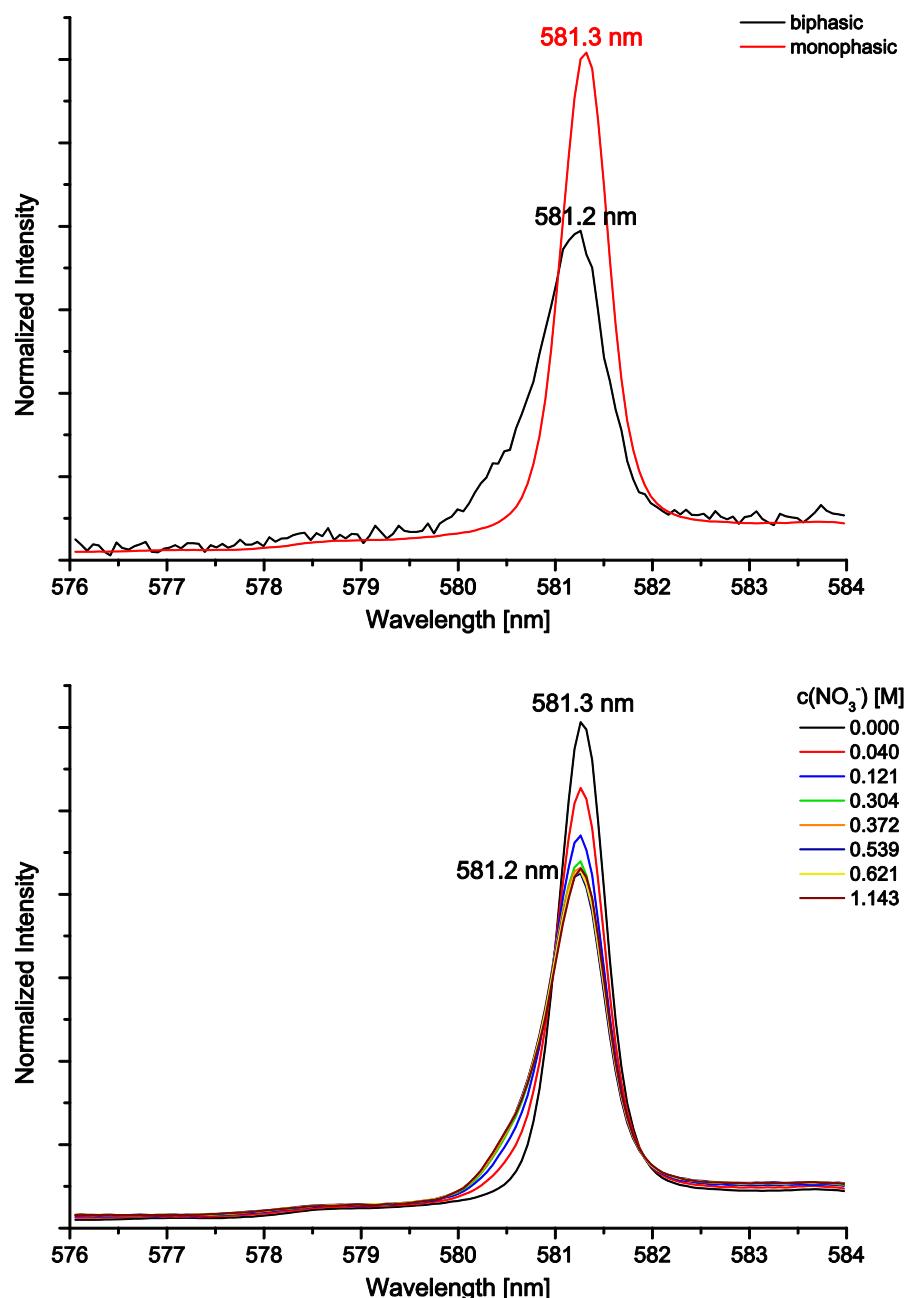


**Fig. S14** Normalized fluorescence spectra of Cm(III)-CyMe<sub>4</sub>-BTBP species upon increasing nitrate concentration in methanol (3.3 mol% water, c(Cm(III))<sub>ini</sub> = 4.59 · 10<sup>-8</sup> M, c(CyMe<sub>4</sub>-BTBP)<sub>ini</sub> = 8.26 · 10<sup>-6</sup> M).

**Eu(III)**



**Fig. S15** Normalized fluorescence spectra resulting from the  $^5D_0 \rightarrow ^7F_1$  and  $^5D_0 \rightarrow ^7F_2$  transitions of Eu(III) in the organic phase of an extraction experiment with CyMe<sub>4</sub>-BTBP and of the Eu(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (with 3.3 mol% water) (top). Normalized fluorescence spectra resulting from the  $^5D_0 \rightarrow ^7F_1$  and  $^5D_0 \rightarrow ^7F_2$  transitions of Eu(III)-CyMe<sub>4</sub>-BTBP species upon increasing nitrate concentration in methanol (with 3.3 mol% water,  $c(\text{Eu(III)})_{\text{ini}} = 6.49 \cdot 10^{-6} \text{ M}$ ,  $c(\text{CyMe}_4\text{-BTBP})_{\text{ini}} = 4.76 \cdot 10^{-5} \text{ M}$ ) (bottom).



**Fig. S16** Normalized fluorescence spectra resulting from the  $^5\text{D}_0 \rightarrow ^7\text{F}_0$  transition of Eu(III) in the organic phase of an extraction experiment with CyMe<sub>4</sub>-BTBP and of the Eu(III)-CyMe<sub>4</sub>-BTBP 1:2 complex in methanol (with 3.3 mol% water) (top). Normalized fluorescence spectra resulting from the  $^5\text{D}_0 \rightarrow ^7\text{F}_0$  transition of Eu(III)-CyMe<sub>4</sub>-BTBP species upon increasing nitrate concentration in methanol (with 3.3 mol% water,  $c(\text{Eu(III)})_{\text{ini}} = 6.49 \cdot 10^{-6}$  M,  $c(\text{CyMe}_4\text{-BTBP})_{\text{ini}} = 4.76 \cdot 10^{-5}$  M) (bottom).