

Supporting Information

A manganese-doped polymeric framework of polyoxotitanate nanoclusters with a narrow band gap

Yang Chen,* Jesse D. Sokolow, Elzbieta Trzop and Philip Coppens*

Chemistry Department, University at Buffalo, SUNY, Buffalo, New York 14260-3000, United States.

E-mail: coppens@buffalo.edu; ychen37@buffalo.edu

Experimental Section

All of the reagents and solvent were purchased from commercial sources: Titanium(IV) ethoxide (99+%) was purchased from Alfa Aesar; Manganese(II) bromide tetrahydrate, 1,1,1-Tris(hydroxymethyl)ethane (99%), and ethanol (200 proof, anhydrous, $\geq 99.5\%$) were purchased from ALDRICH. The compounds containing titanium were stored and handled in a glove-box under a nitrogen atmosphere.

Synthesis of $\{\text{Ti}_{13}\text{Mn}_4\text{O}_{16}[\text{CH}_3\text{C}(\text{CH}_2\text{O})_3]_4(\text{OEt})_{12}\text{Br}_4\}_\infty$:

To a Teflon-lined Parr bomb with a 23 mL capacity was loaded Titanium(IV) ethoxide (684.3 mg, 3.0 mmol), 1,1,1-Tris(hydroxymethyl)ethane (120.2 mg, 1.0 mmol), Manganese(II) bromide tetrahydrate (143.4 mg, 0.5 mmol), and ethanol (5.0 mL). The resulting mixture was stirred for 5 minutes and sealed. The reactor was heated in an oven to 150 °C for 72 hours and then cooled to 40 °C over 48 hours. After the oven was totally cooled to the room temperature, the yellow block crystals of $\{\text{Ti}_{13}\text{Mn}_4\text{O}_{16}[\text{MeC}(\text{CH}_2\text{O})_3]_4(\text{OEt})_{12}\text{Br}_4\}_\infty$ were taken out from the Parr bomb, and washed with ethanol and dried in the glove-box. Yield: 26.1 mg (8.6% based on Mn). Elemental analysis calcd (%) for $\text{C}_{44}\text{H}_{96}\text{O}_{40}\text{Br}_4\text{Mn}_4\text{Ti}_{13}$: C 21.78, H 3.99; found: C 21.68, H 3.92.

Table S1. Crystal data and details of data collection and refinement for $\{\text{Ti}_{13}\text{Mn}_4\text{O}_{16}[\text{MeC}(\text{CH}_2\text{O})_3]_4(\text{OEt})_{12}\text{Br}_4\}_\infty$

Compound	$\{\text{Ti}_{13}\text{Mn}_4\text{O}_{16}[\text{MeC}(\text{CH}_2\text{O})_3]_4(\text{OEt})_{12}\text{Br}_4\}_\infty$
Chemical formula	$\text{C}_{44}\text{H}_{96}\text{O}_{40}\text{Br}_4\text{Mn}_4\text{Ti}_{13}$
Formula weight	2426.88
Crystal system	Tetragonal
Space group	$I4_1md$
$a / \text{Å}$	34.8012(10)
$b / \text{Å}$	34.8012(10)
$c / \text{Å}$	14.7872(6)
$V / \text{Å}^3$	17909.1(10)
Z	8
T / K	90(2)
$F(000)$	9647
μ (Mo $K\alpha$, mm^{-1})	3.490
Total reflections	9472
Unique reflections	4957
No. of parameters	485
R^a	0.0545
ωR^b	0.1579
GOF^c	1.043

^a $R = \sum ||F_o| - |F_c|| / \sum |F_o|$. ^b $\omega R = \{\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2\}^{1/2}$. ^c $GOF = \{\sum [w((F_o^2 - F_c^2)^2) / (n-p)]\}^{1/2}$, where n = number of reflections and p = total numbers of parameters refined.

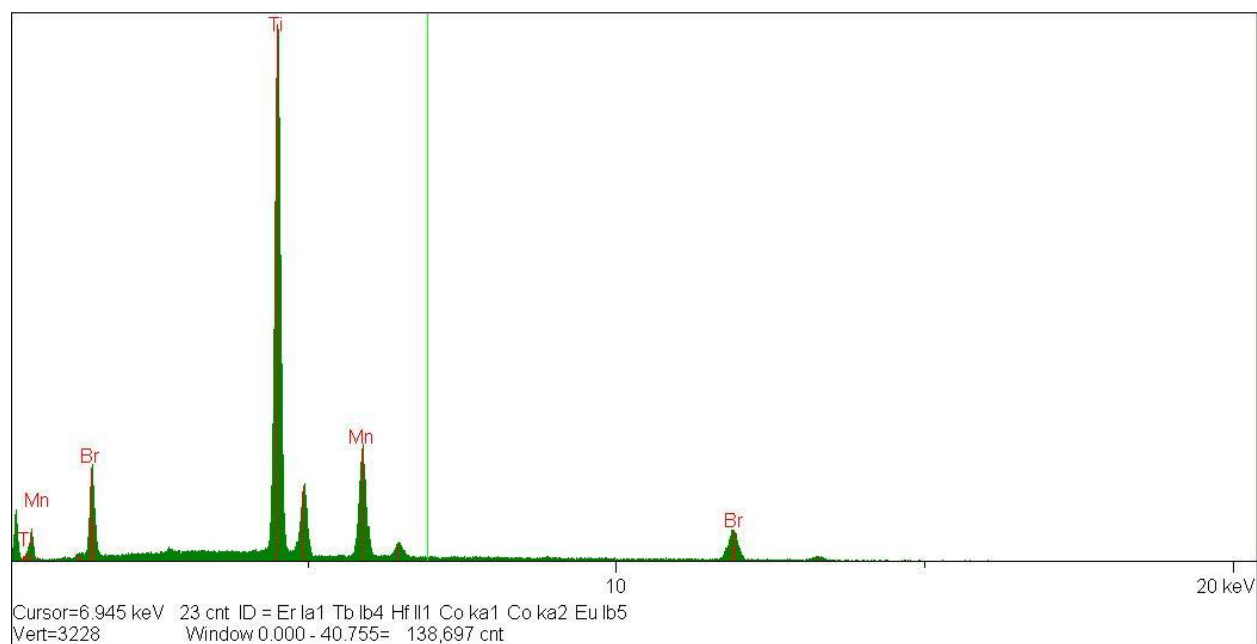


Fig. S1 EDS spectrum of $\{\text{Ti}_{13}\text{Mn}_4\text{O}_{16}[\text{MeC}(\text{CH}_2\text{O})_3]_4(\text{OEt})_{12}\text{Br}_4\}_\infty$.

El.	Line	Intensity (c/s)	Error 2-sig	Atomic %	Conc	Units
Ti	Ka	1,593.07	16.337	62.943	54.902	wt.%
Mn	Ka	370.71	8.168	19.470	19.491	wt.%
Br	Ka	131.11	5.020	17.587	25.607	wt.%
				100.000	100.000	wt.%
						Total

KV 30.0
Takeoff Angle 45.0°
Elapsed Livetime 25.0