Discriminate sensing of pyrophosphate using a new tripodal tetramine-based dinuclear Zn(II) complex under indicator displacement assay approach

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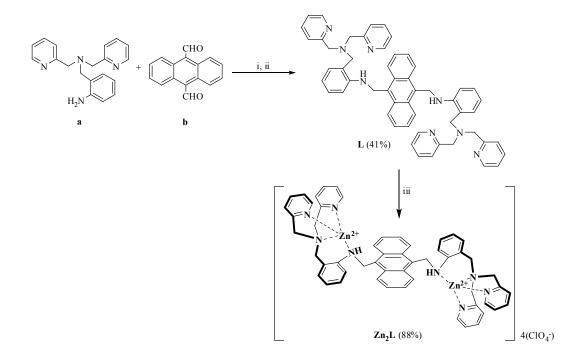
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Content

Scheme S1	Synthetic procedure of ligand L and Zn_2L .	S3
Figure S1	¹ H-NMR spectrum of ligand L in $CDCl_3$.	S3
Figure S2	¹³ C-NMR Spectrum of ligand L in CDCl ₃ .	S4
Figure S3	¹ H-NMR spectrum of $\mathbf{Zn}_{2}\mathbf{L}$ in 20% (v/v) $D_{2}O/CD_{3}CN$.	S4
Figure S4	¹³ C-NMR spectrum of $\mathbf{Zn}_2\mathbf{L}$ in 20% (v/v) D_2O/CD_3CN .	S5
Figure S5	HMQC-NMR spectrum of $\mathbf{Zn}_{2}\mathbf{L}$ in CD ₃ CN.	S5
Table S1	Total and relative energies of all the B3LYP/LANL2DZ–optimized structures of six conformers of the dimeric 2:2 species.	S6
Figure S6	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C1 conformer.	S7
Figure S7	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C2 conformer.	S8
Figure S8	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C3 conformer.	S9
Figure S 9	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P1 conformer.	S10
Figure S10	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P2 conformer.	S11
Figure S11	DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P3 conformer.	S12
Figure S12	¹ H NMR spectra of 1:1 ratio of ensemble formation between Zn₂L:MTB (5 mM) upon addition of various concentration of PPi	S13

(0.05 M) in 20% (v/v) D₂O:CD₃CN.

- Figure S13 1 H NMR spectra of 1:2 ratio of ensemble formation betweenS14Zn₂L:2MTB (5 mM) upon addition of various concentration of PPi(0.05 M) in 20% (v/v) D₂O:CD₃CN.
- **Figure S14** ³¹P NMR spectra of $[Zn_2L \cdot MTB]$ (5 mM) upon addition of various S15 concentration of **PPi** (0.05 M) in 20% (v/v) D₂O:CD₃CN.
- Figure S15 ³¹P NMR spectra of $[Zn_2L \cdot 2MTB]$ (5 mM) upon addition of various S16 concentration of PPi (0.05 M) in 20% (v/v) D₂O:CD₃CN.
- Figure S16 (a) UV/vis spectra obtained by addition of Zn_2L (400 μ M) to a S17 solution of indicator PV (20 μ M) in HEPES buffered pH 7.4 in 20% (v/v) H₂O/CH₃CN solution, (b) Job's plot analysis of PV-Zn₂L ensemble. (c) A plot of absorption against concentration of Zn_2L titrated in PV Zn₂L ensemble
- Figure S17 (a) UV/vis spectra obtained by addition of Zn₂L(400 μM) to a S18 solution of indicator BPG (20 μM) in HEPES buffered pH 7.4 in 20% (v/v) H₂O/CH₃CN solution, (b) Job's plot analysis of BPG-Zn₂L ensemble, C) A plot of absorption against concentration of Zn₂L titrated in BPG.
- Figure S18 (a) UV/vis spectra obtained by addition of $Zn_2L(400 \ \mu\text{M})$ to a S19 solution of indicator XO (20 μ M) in HEPES buffered pH 7.4 in 20% (v/v) H₂O/CH₃CN solution, (b) Job's plot analysis of XO-Zn₂Lensemble, C) A plot of absorption against concentration of Zn₂L titrated in XO.
- Figure S19 A plot of absorption against concentration of Zn_2L titrated in MTB. S19
- Figure S20 Calibration curve for detection of PPi using MTB-Zn₂L ensemble. S20



Scheme S1. Synthetic procedure of L and Zn_2L . (i) acetronitrile, reflux 12 h, (ii) NaBH₄, MeOH, reflux 12 h, (iii) $Zn(ClO_4)_2$, EtOH, reflux 12 h.

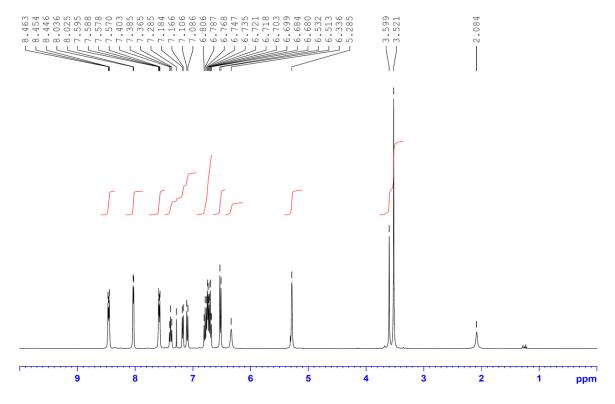


Figure S1.¹H-NMR spectrum of ligand L in CDCl₃.

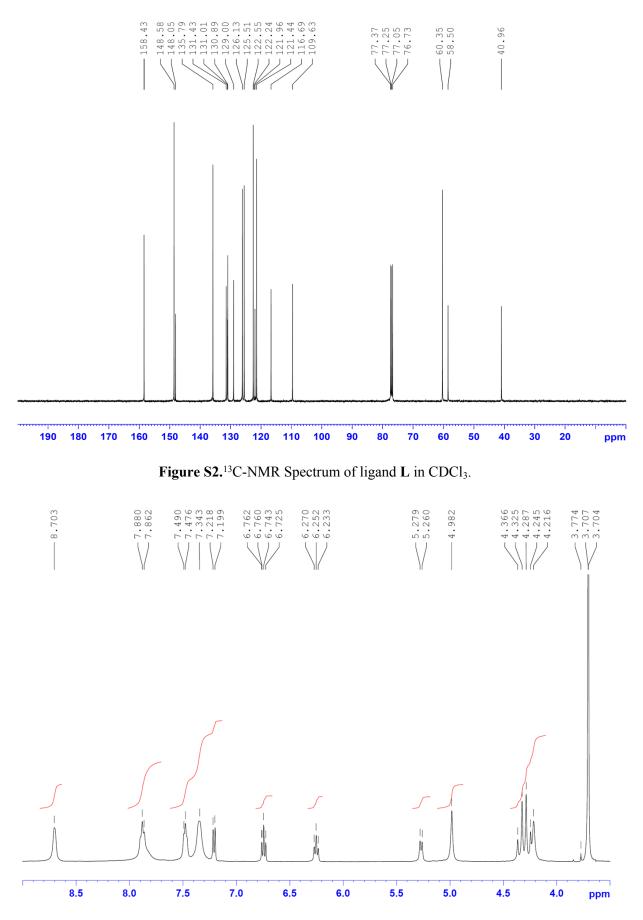


Figure S3.¹H-NMR spectrum of Zn_2L in 20% (v/v) D_2O/CD_3CN .

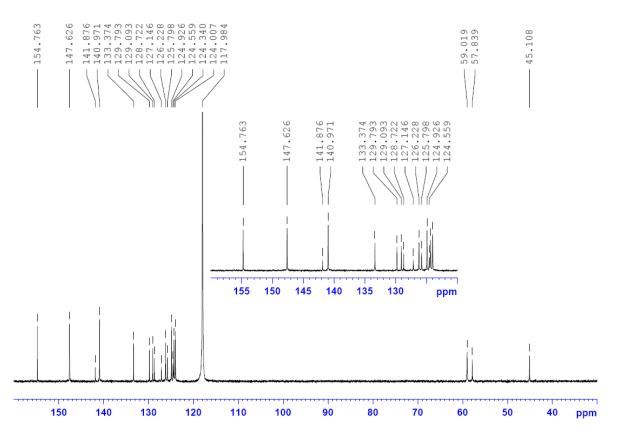


Figure S4.¹³C-NMR spectrum of Zn_2L in 20% (v/v) D_2O/CD_3CN .

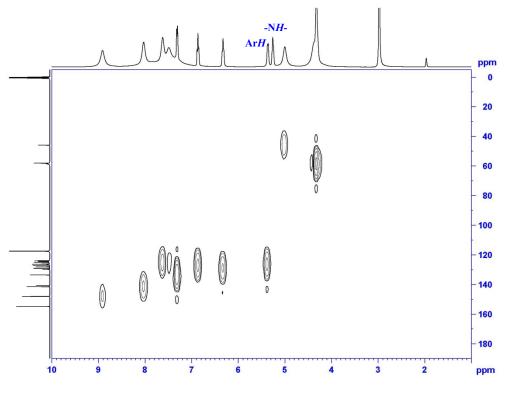


Figure S5. HMQC-NMR spectrum of Zn₂L in CD₃CN.

Conformers ^a	$E_{ m total}$ b	$\Delta E_{ m rel}$ °	
C1	-6392.7827862	75.37	
C2 ^d	-6392.9028947	0.00	
C3	-6392.8176487	53.49	
P1	-6392.7687205	84.20	
P2	-6392.7596734	89.87	
Р3	-6392.7567471	91.71	

Table S1 Total and relative energies of all the B3LYP/LANL2DZ–optimized structures of six conformers of the dimeric 2:2 species.

^a Conformers are named according to cross (C) or parallel (P) alignment of two anthracene units of the dimeric species. ^b Total energies are in au. ^c Relative energies compared with the most stable conformer (C2), in kcal/mol. ^d The most stable conformer.

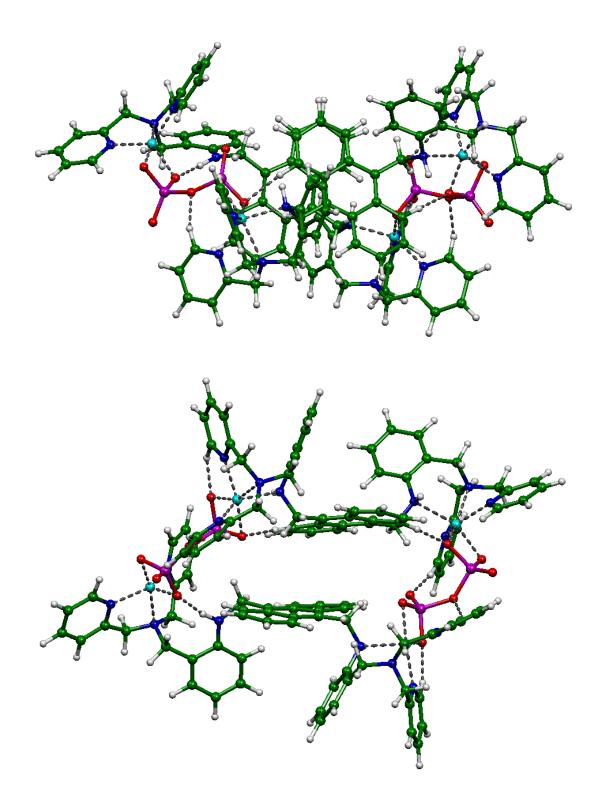


Figure S6. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C1 conformer. Top and bottom images are top and front views, respectively.

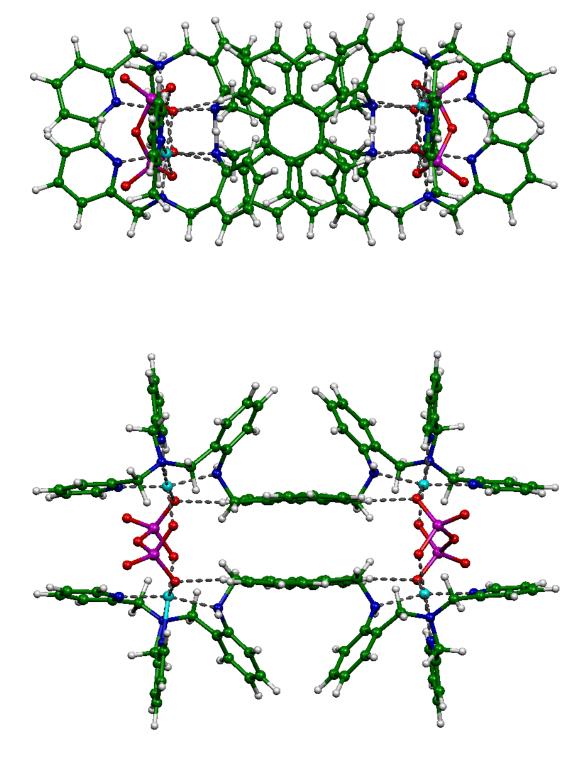


Figure S7. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C2 conformer as the most stable one. Top and bottom images are top and front views, respectively.

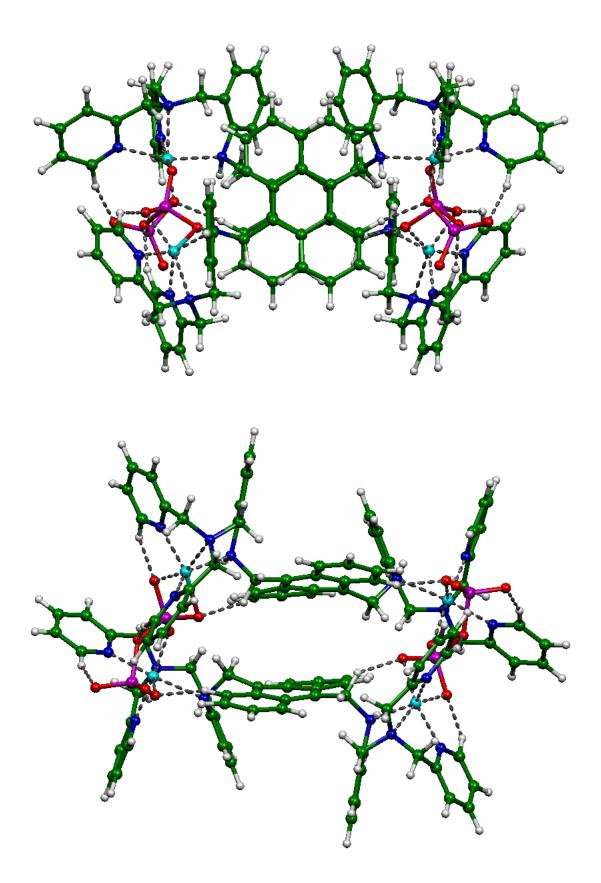


Figure S8. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of C3 conformer. Top and bottom images are top and front views, respectively.

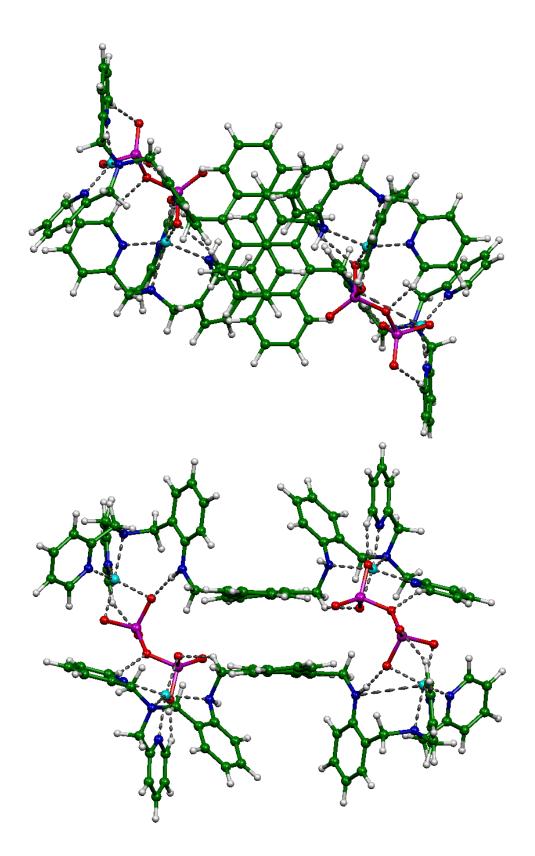


Figure S9. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P1 conformer. Top and bottom images are top and front views, respectively.

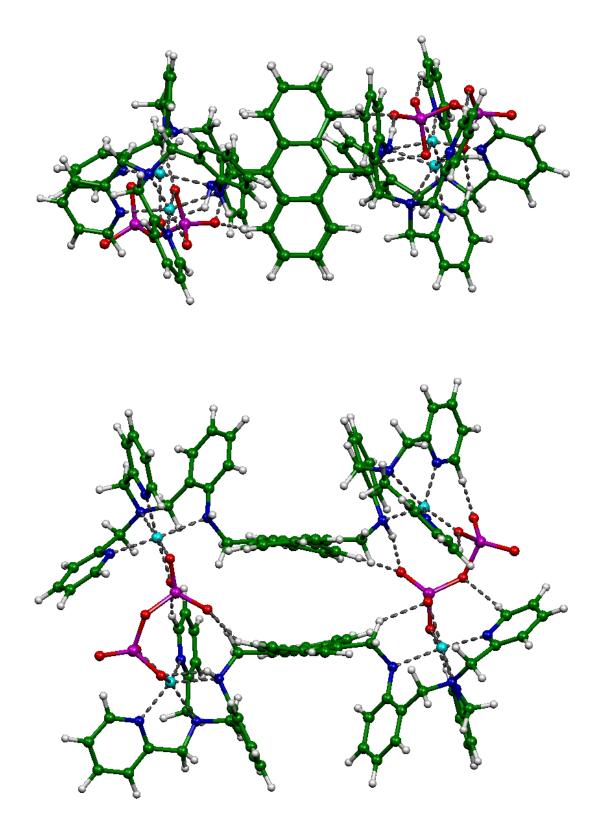


Figure S10. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P2 conformer. Top and bottom images are top and front views, respectively.

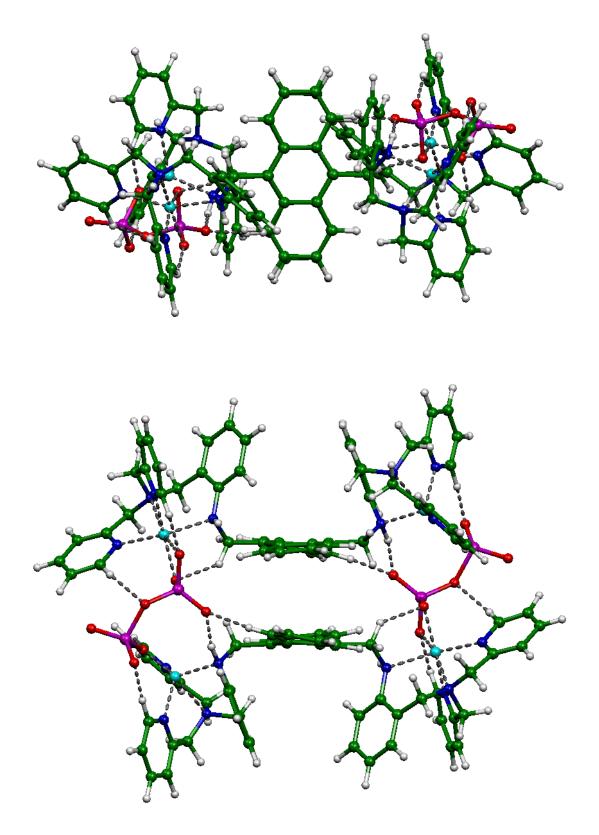


Figure S11. DFT/B3LYP/LANL2DZ-optimized structure of the dimeric 2:2 species of P3 conformer. Top and bottom images are top and front views, respectively.

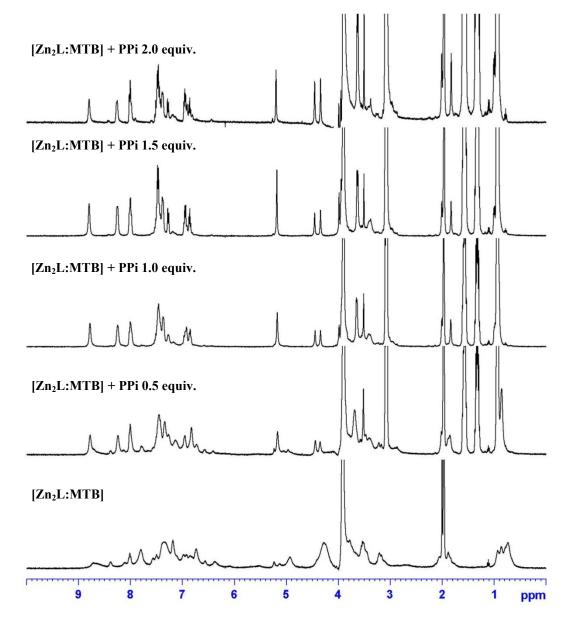


Figure S12. ¹H NMR spectra of 1:1 ratio of ensemble formation between $Zn_2L:MTB$ (5 mM) upon addition of various concentrations of **PPi** (0.05 M) in 20% (v/v) $D_2O:CD_3CN$.

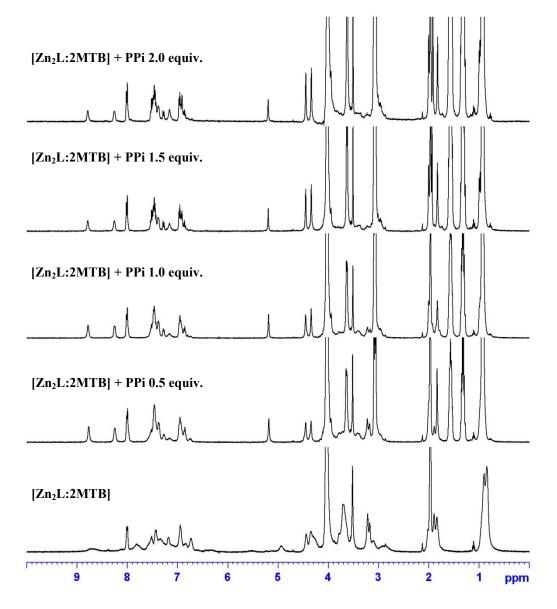


Figure S13.¹H NMR spectra of 1:2 ratio of ensemble formation between $Zn_2L:MTB$ (5 mM) upon addition of various concentration of **PPi** (0.05 M) in 20% (v/v) $D_2O:CD_3CN$.

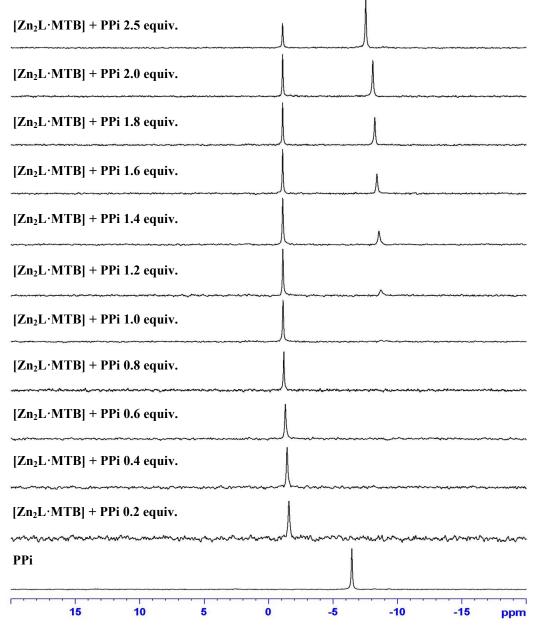


Fig S14. ³¹P NMR spectra of $[Zn_2L \cdot MTB]$ (5 mM) upon addition of various concentrations of PPi (0.05 M) in 20% (v/v) D₂O:CD₃CN.

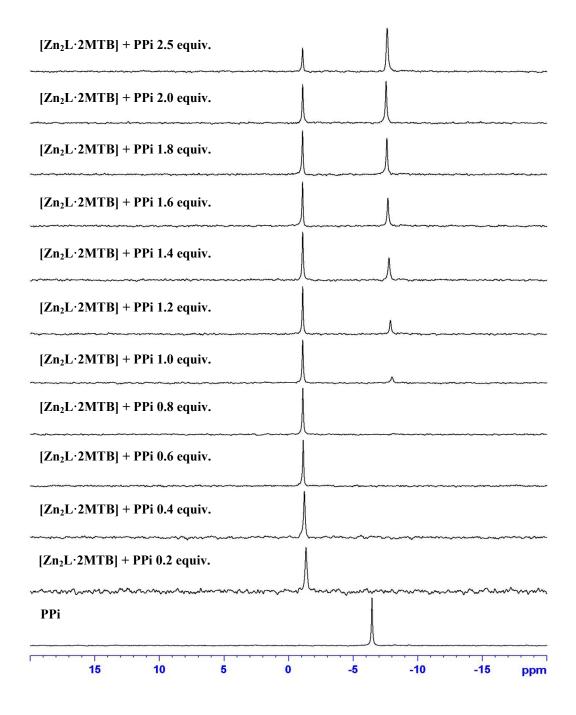


Fig S15. ³¹P NMR spectra of [$Zn_2L \cdot 2MTB$] (5 mM) upon addition of various concentrations of PPi (0.05 M) in 20% (v/v) D₂O:CD₃CN.

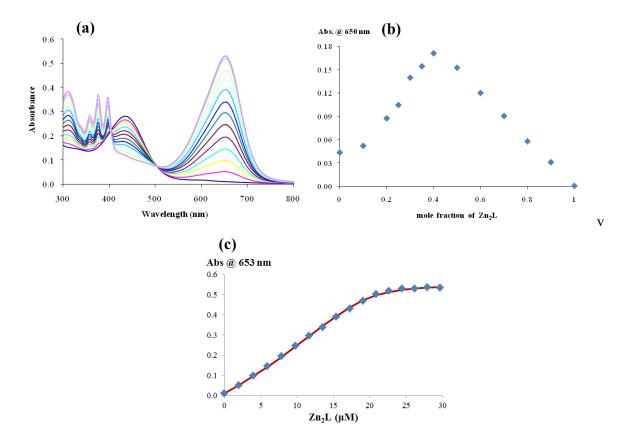


Figure S16. (a) UV/vis spectra obtained by addition of Zn_2L (400 µM) to a solution of indicator PV (20 µM) in HEPES buffered pH 7.4 in 20% (v/v) H₂O/CH₃CN solution , (b) Job's plot analysis of PV-Zn₂L ensemble, (c) A plot of absorption against concentration of Zn₂L titrated in PV. The red solid line is nonlinear least-squares fittings of the titration profiles using SPECFIT32 program.

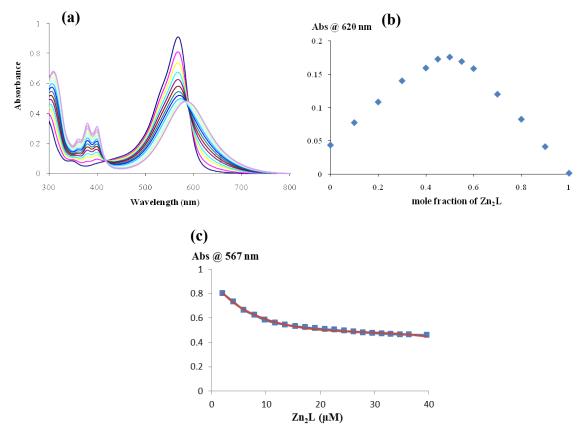


Figure S17. (a) UV/vis spectra obtained by addition of $Zn_2L(400 \ \mu M)$ to a solution of indicator BPG (20 μ M) in HEPES buffered pH 7.4 in 20% (v/v) H₂O/CH₃CN solution, (b) Job's plot analysis of BPG-Zn₂L ensemble, (c) A plot of absorption against concentration of Zn₂L titrated in BPG. The red solid line is nonlinear least-squares fittings of the titration profiles using SPECFIT32 program.

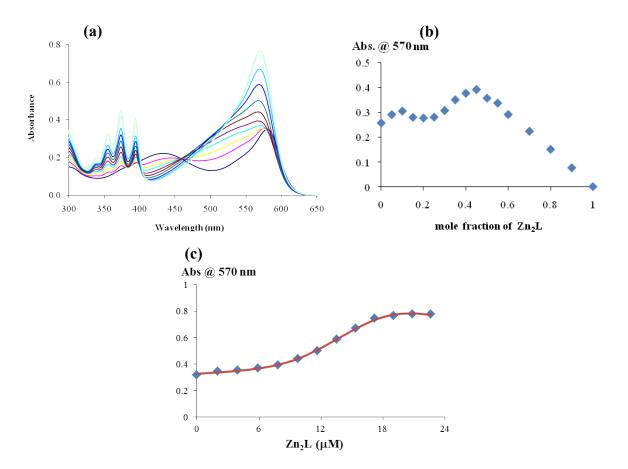


Figure S18. (a) UV/vis spectra obtained by addition of $Zn_2L(400 \ \mu\text{M})$ to a solution of indicator XO (20 μ M) in HEPES buffered pH 7.4 in 80/20 (% v/v) CH₃CN/H₂O solution, (b) Job's plot analysis of XO-Zn₂Lensemble, C) A plot of absorption against concentration of Zn₂L titrated in XO. The red solid line is nonlinear least-squares fittings of the titration profiles using SPECFIT32 program.

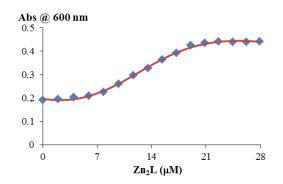


Figure S19. A plot of absorption against concentration of Zn_2L titrated in MTB. The red solid line is nonlinear least-squares fittings of the titration profiles using SPECFIT32 program.

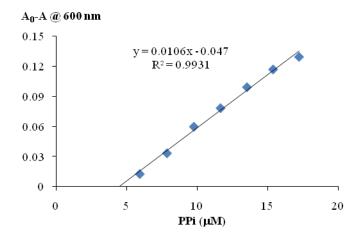


Figure S20. Calibration curve for detection of PPi using MTB-Zn₂L ensemble.