

## Electronic Supplementary Information

### Supramolecular Interactions

The complex shows supramolecular layer structure with inter layer separation (shortest nickel-nickel distance) of 9.5781(7) Å. The H atom, H(10B), attached with O(100) of the lattice water forms bifurcated H-bonding with the methoxy oxygen atom O(9) and phenoxy oxygen atom, O(8) of Schiff base ligand. Another H atom, H(10A), attached with O(100) of the lattice water is H-bonded with N(24)<sup>#</sup> of a (PTZ)<sup>-</sup> ligand from a symmetry related (<sup>#</sup> = 2-x, 1/2+y, 5/2-z) molecule. On the other hand, the H atom, H(3), attached with amine nitrogen atom, N(3), is H-bonded with an oxygen atom, O(100)<sup>\*</sup> of a symmetry related (<sup>\*</sup> = 2-x, -y, 2-z) water of crystallization. The combination of these three H-bonds resulted a two dimensional H-bonded network as shown in Figure S1. The details of H bonding dimensions are given in Table S3.

The complex also shows significant intra and intermolecular C-H... $\pi$  interactions. One methylene C(sp<sup>3</sup>) hydrogen, H(2A) of N-ethyl group C(2) in Schiff base *HL* shows intramolecular interaction with chelate ring Cg(2) produced by *HPTZ* ligand. An aromatic C(sp<sup>2</sup>) hydrogen, H11 of phenyl ring is involved in intermolecular interactions with the pyrazine ring Cg(6) of *HPTZ* forming an 1D array in the crystal packing of **1** (Figure S2). Geometric features of the C-H... $\pi$  interactions are given in Table S4. Considering the centrosymmetric dimer as a node, each node is connected with two other node by C-H... $\pi$  interactions and four other node by hydrogen bonding interactions through water molecule. These combined supramolecular force (C-H... $\pi$  interactions and H-bond) creates a 3<sup>6</sup>-**hxl** topological supramolecular network as shown in Figure S3. It is noteworthy that acetonitrile molecule present in the crystal structure is actually

encapsulated as a hydrophobic guest within the 2D supramolecular network as shown in Figure S7.

**Table S1:** Selected bond angles (°) around nickel(II) in complex **1**.

|                       |           |                         |            |
|-----------------------|-----------|-------------------------|------------|
| O(8) – Ni(1) – N(3)   | 174.15(6) | N(6) – Ni(1) – N(14)    | 93.57(6)   |
| O(8) – Ni(1) – N(6)   | 91.53(6)  | N(6) – Ni(1) – N(21)    | 170.82(6)  |
| O(8) – Ni(1) – N(14)  | 86.76(6)  | N(6) – Ni(1) – N(22)*   | 93.60(6)   |
| O(8) – Ni(1) – N(21)  | 90.20(6)  | N(14) – Ni(1) – N(21)   | 77.52(6)   |
| O(8) – Ni(1) – N(22)* | 92.06(6)  | N(14) – Ni(1) – N(22)*  | 172.76(6)  |
| N(3) – Ni(1) – N(6)   | 82.88(6)  | N(21) – Ni(1) – N(22)*  | 95.35(6)   |
| N(3) – Ni(1) – N(14)  | 91.84(6)  | Ni(1) – N(21) – N(22)   | 139.85(12) |
| N(3) – Ni(1) – N(21)  | 95.04(6)  | Ni(1) – N(22)* – N(21)* | 124.60(12) |
| N(3) – Ni(1) – N(22)* | 90.03(6)  | -                       | -          |

Symmetry element: \* = 2-x,-y,2-z

**Table S2:** The details data of the photoluminescence and time-resolved photoluminescence decays of complex **1**.

| Complex  | $\lambda_{\text{ex}}$ (nm) | $\lambda_{\text{em}}$ (nm) | T <sub>1</sub> (ns) | T <sub>2</sub> (ns) | B <sub>1</sub> | B <sub>2</sub> | $\chi^2$ | $\langle \tau \rangle$ |
|----------|----------------------------|----------------------------|---------------------|---------------------|----------------|----------------|----------|------------------------|
| <b>1</b> | 385                        | 413                        | 0.542               | 2.168               | 0.519          | 0.481          | 1.054933 | 1.322 ns               |
|          |                            | 436                        | 0.230               | 0.990               | 0.101          | 0.899          | 1.069045 | 0.914 ns               |

**Table S3:** Hydrogen bond distances (Å) and angles (°) for complex **1**.

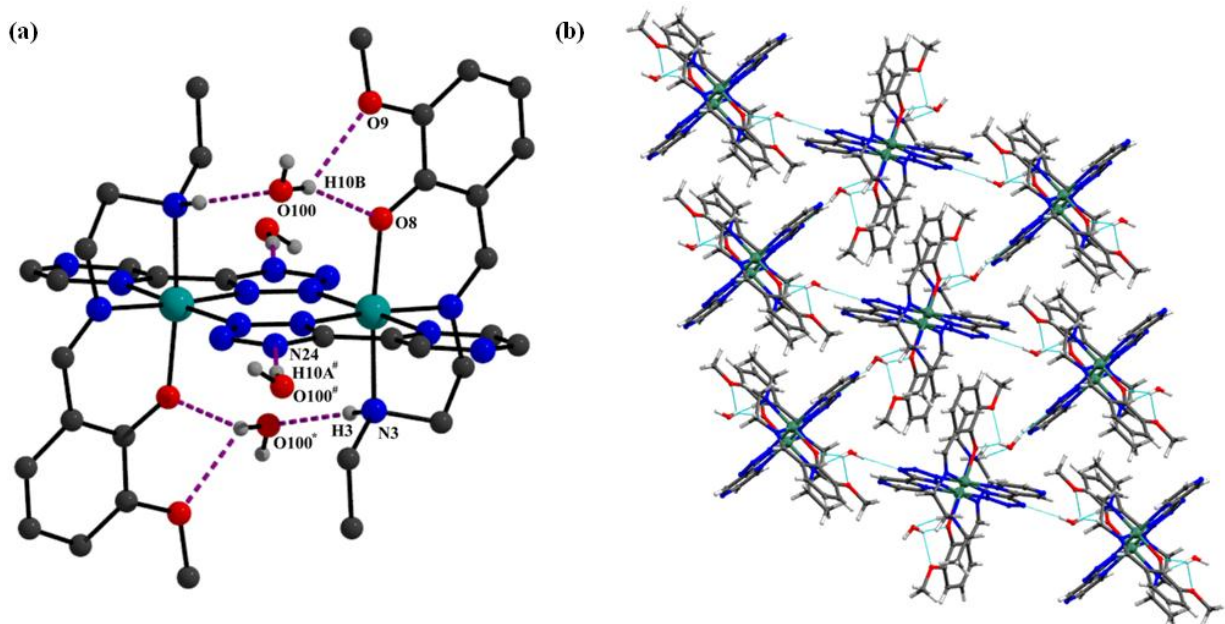
| D–H···A                            | D–H     | D···A    | H···A   | ∠D–H···A |
|------------------------------------|---------|----------|---------|----------|
| O(100)–H(10B)···O(8)               | 0.82(3) | 3.017(2) | 2.24(3) | 159(3)   |
| O(100)–H(10B)···O(9)               | 0.82(3) | 2.911(2) | 2.30(2) | 132(3)   |
| O(100)–H(10A)···N(24) <sup>#</sup> | 0.77(3) | 2.977(2) | 2.22(3) | 169(3)   |
| N(3)–H(3)···O(100) <sup>*</sup>    | 0.91(3) | 3.004(2) | 2.13(3) | 160(2)   |

Symmetry element <sup>\*</sup> = 2-x,-y,2-z; <sup>#</sup> = 2-x,1/2+y,5/2-z D, donor; H, hydrogen; A, acceptor

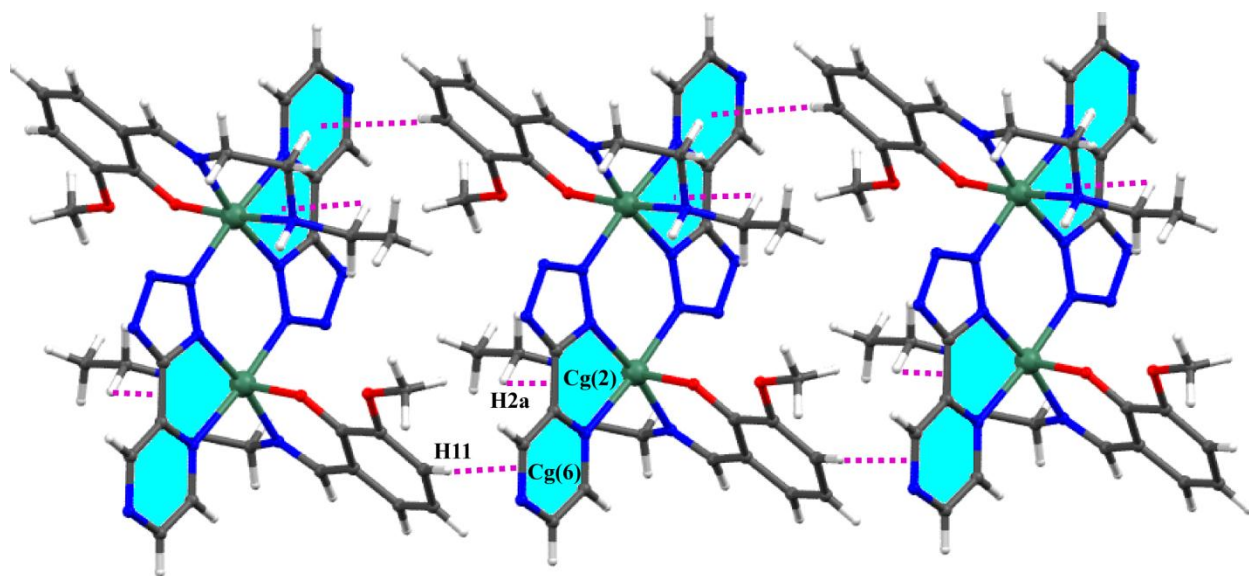
**Table S4:** Geometric features (distances in Å and angles in degrees) of the C–H··· $\pi$  interactions obtained for **1**.

| Complexes | C–H···Cg(Ring)      | H···Cg (Å) | C–H···Cg (°) | C···Cg (Å) |
|-----------|---------------------|------------|--------------|------------|
| <b>1</b>  | C(2)–H(2a)···Cg(2)  | 2.71       | 3.083(2)     | 103        |
|           | C(11)–H(11)···Cg(6) | 2.98       | 3.872(2)     | 157        |

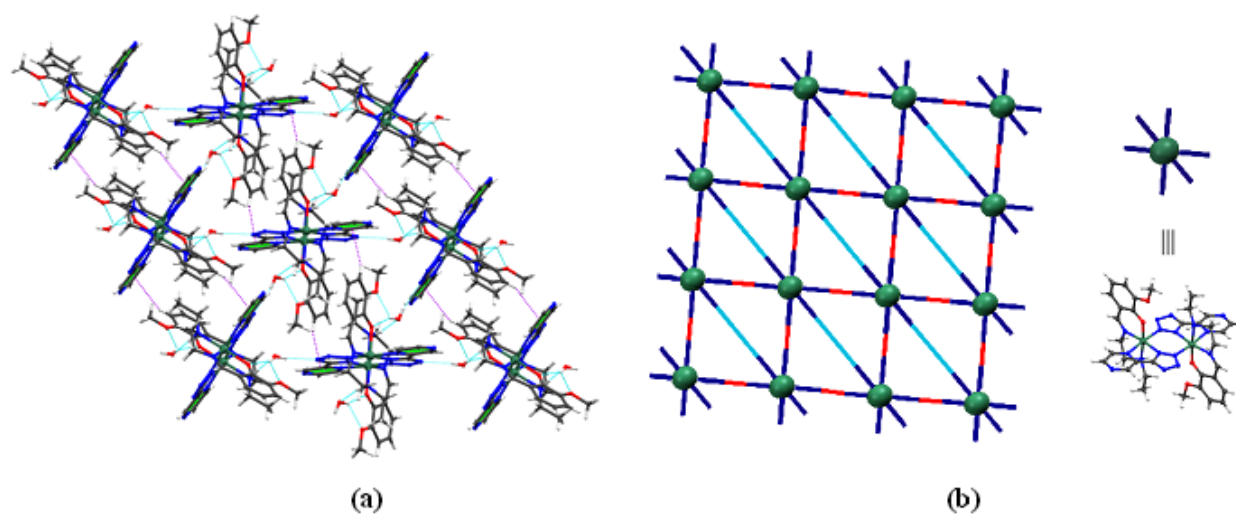
Cg(2) = Centre of gravity of the ring [Ni(1)–N(14)–C(19)–C(20)–N(21)] and Cg(6) = Centre of gravity of the ring [N(14)–C(15)–C(16)–N(17)–C(18)–C(18)]



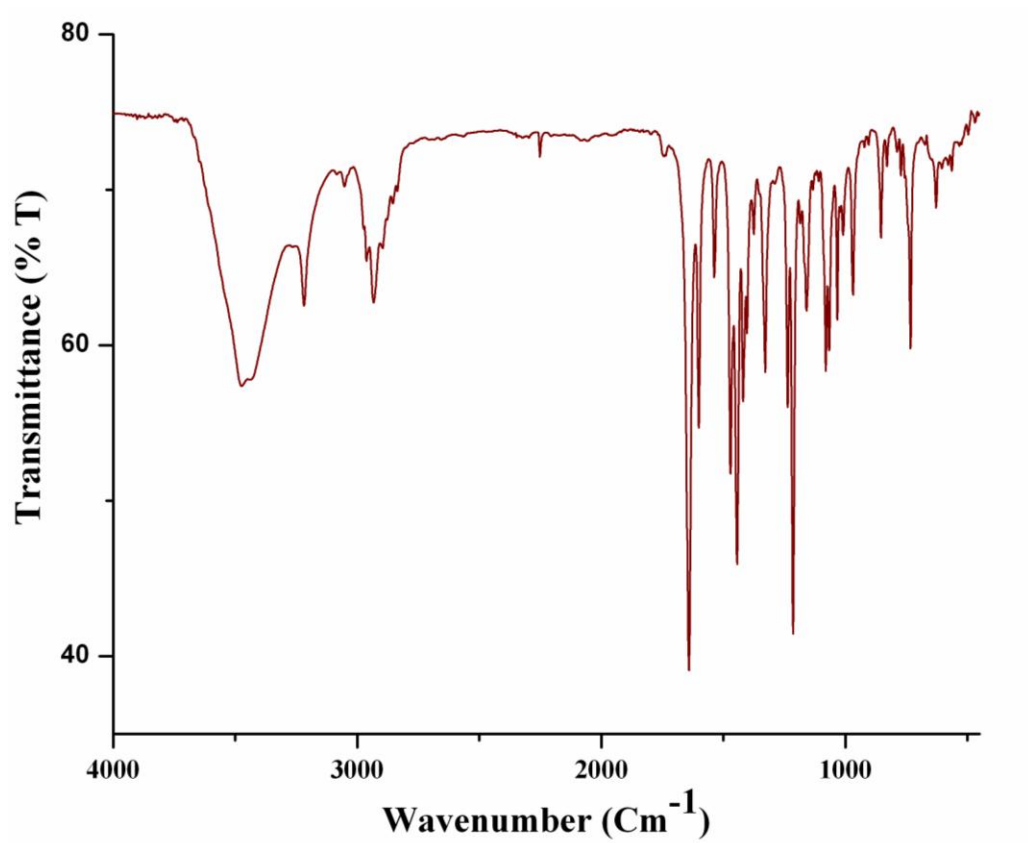
**Figure S1:** Perspective view of (a) H-bonding interaction and (b) two dimensional H-bonded layer structure of complex **1**. Only the relevant hydrogen atoms are shown for clarity. Symmetry element  $^* = 2-x, -y, 2-z$ ;  $^{\#} = 2-x, 1/2+y, 5/2-z$ .



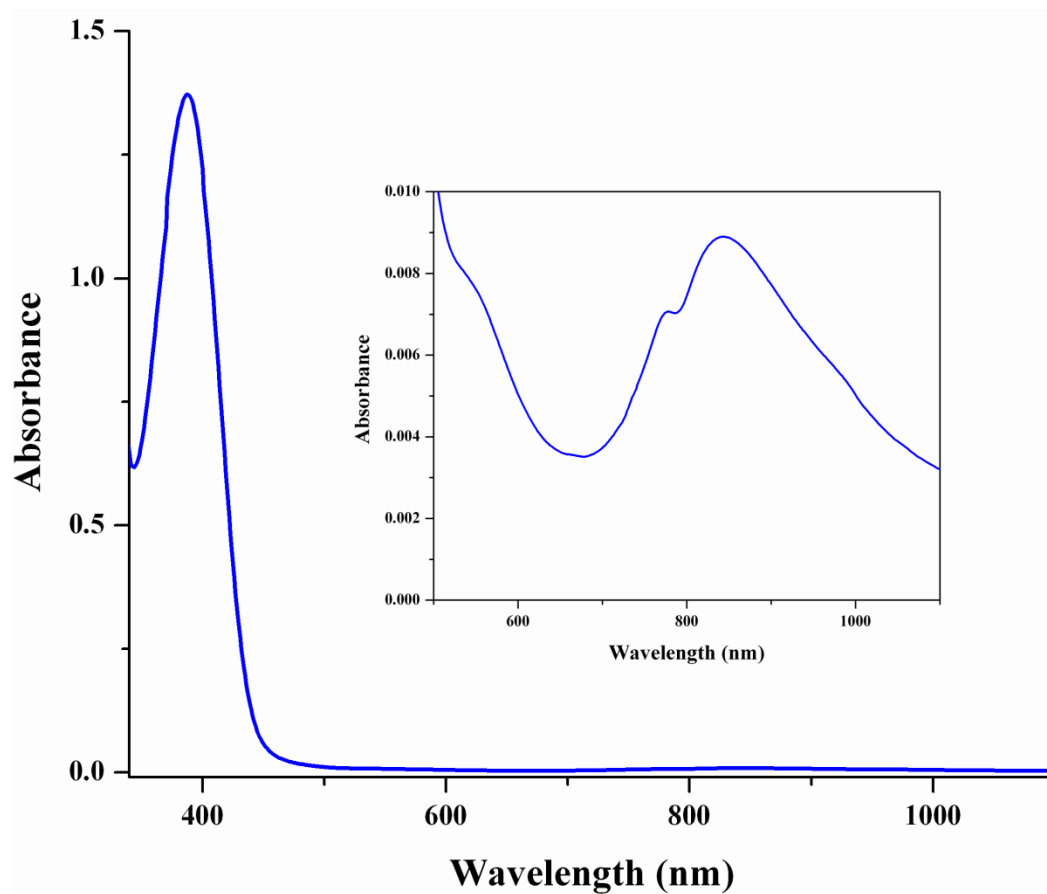
**Figure S2:** Perspective view of one dimensional array in crystal packing via C-H... $\pi$  interactions.



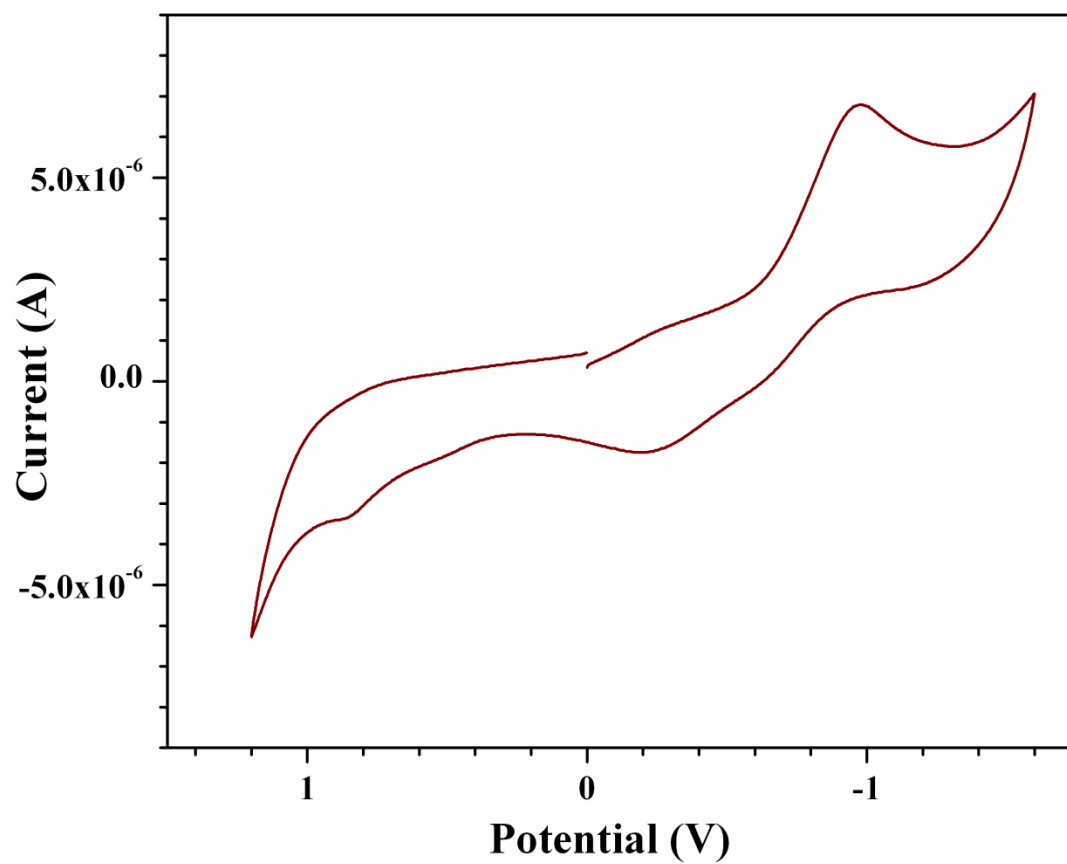
**Figure S3:** (a) Perspective view of two-dimensional supramolecular layer structure highlighting H-bonding and intermolecular C-H... $\pi$  interactions (b) A view of 3<sup>6</sup>-hxl topological supramolecular network. The acetonitrile molecules have been omitted for clarity. Red and aquamarine lines represent water mediated hydrogen bonding interactions and intermolecular C-H... $\pi$  interactions respectively.



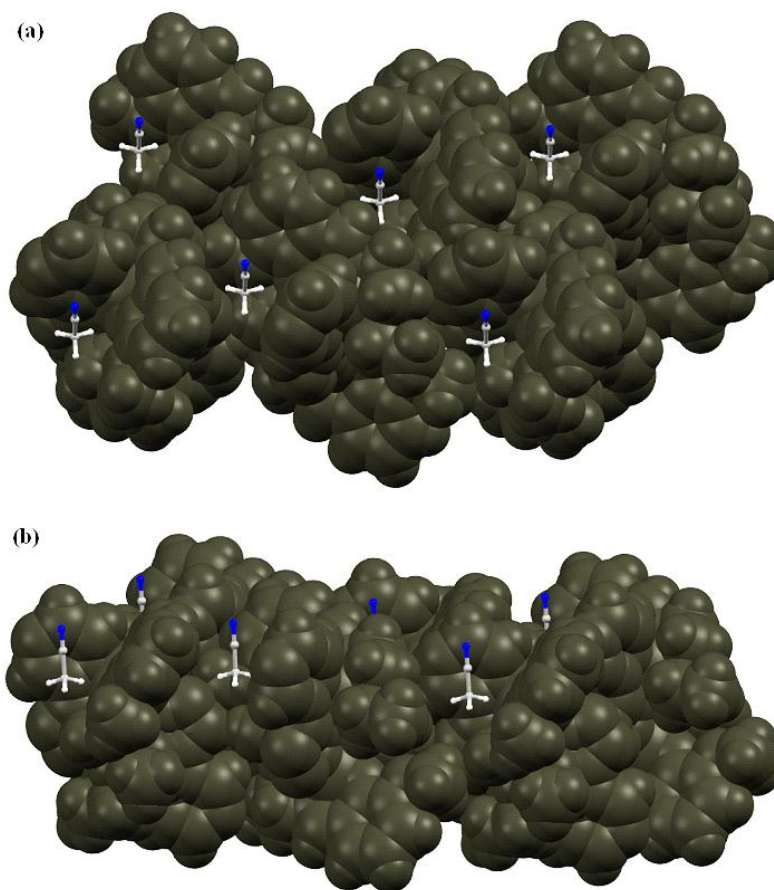
**Figure S4:** IR spectrum (450-4000 cm<sup>-1</sup>) of complex **1**.



**Figure S5:** The absorption spectrum of complex **1** in DMSO medium.



**Figure S6:** Cyclic voltammogram of complex **1** in DMSO at room temperature.



**Figure S7:** Top view **(a)** and side view **(b)** of encapsulated acetonitrile (hydrophobic) molecule in the 2D supramolecular self-assembly of complex **1**. Dinuclear Schiff base complex and water of crystallization forming supramolecular self-assembly has been shown in space-fill model.