

SUPPORTING INFORMATION

In Search of Redox Noninnocence between a Tetrazine Pincer Ligand and Monovalent Copper

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MSC#12122 Single Crystal X-ray structure determination of Cu(btzp)₂PF₆ Hui

A dark red crystal (approximate dimensions 0.28 x 0.07 x 0.05 mm³) was placed onto the tip of a glass capillary and mounted on an Apex Kappa Duo diffractometer and measured at 150(2) K.

A preliminary set of cell constants was calculated from reflections harvested from three sets of 12 frames. These initial sets of frames were oriented such that orthogonal wedges of reciprocal space were surveyed. The data collection was carried out using Mo K α radiation (graphite monochromator) with a frame time of 90 seconds and a detector distance of 5.0 cm. A randomly oriented region of reciprocal space was surveyed to achieve complete data with a redundancy of 4. Sections of frames were collected with 0.50° steps in ω and ϕ . Data to a resolution of 0.84 Å were considered in the reduction. Final cell constants were calculated from the xyz centroids of 6125 strong reflections from the actual data collection after integration (SAINT).¹ The intensity data were corrected for absorption (SADABS).² The space group $P2_1/n$ was determined based on intensity statistics and systematic absences. The structure was solved using SHELXS-97³ and refined with SHELXL-97.³ A direct-methods solution was calculated, which provided most non-hydrogen atoms from the E-map. Full-matrix least squares / difference Fourier cycles were performed, which located the remaining non-hydrogen atoms. All non-hydrogen atoms were refined with anisotropic displacement parameters. The hydrogen atoms were placed in ideal positions and refined as riding atoms with individual relative isotropic displacement parameters. The final full matrix least squares refinement converged to $R1 = 0.0433$ and $wR2 = 0.1110$ (F^2 , all data). The remaining electron density is located along the bonds.

Two solvent molecules, acetonitrile, were found to be present in the crystal lattice, however they are all outside interaction range of copper, as judged by van der Waals radii sums. Hexafluorophosphate, the non-coordinating counterion was found to be disordered (79:21). The unit cell shows no unusual contacts

to PF_6^- (shortest Cu/P distance is 6.27 Å), and there is one contact of 3.06 Å from a guest MeCN nitrogen to a tetrazine ring carbon (C2).

¹ SAINT, Bruker Analytical X-Ray Systems, Madison, WI, current version.

² An empirical correction for absorption anisotropy, R. Blessing, Acta Cryst. A51, 33 - 38 (1995).

³ SHELXTL-Plus, Bruker Analytical X-Ray Systems, Madison, WI, current version.

Electronic spectrum.

Solid $\text{Cu}(\text{btzp})_2[\text{PF}_6]$ is almost black, and with 5mg of this in ~0.5 mL of CD_3CN , the solution color is black/red. With 1:10 dilution, this solution becomes increasingly red. In addition to showing the three peaks seen in btzp itself (Figure S3), hence assigned as ligand localized, the complex (Figure S1 and S2) has an additional very weak peak in the 400-450 nm region.

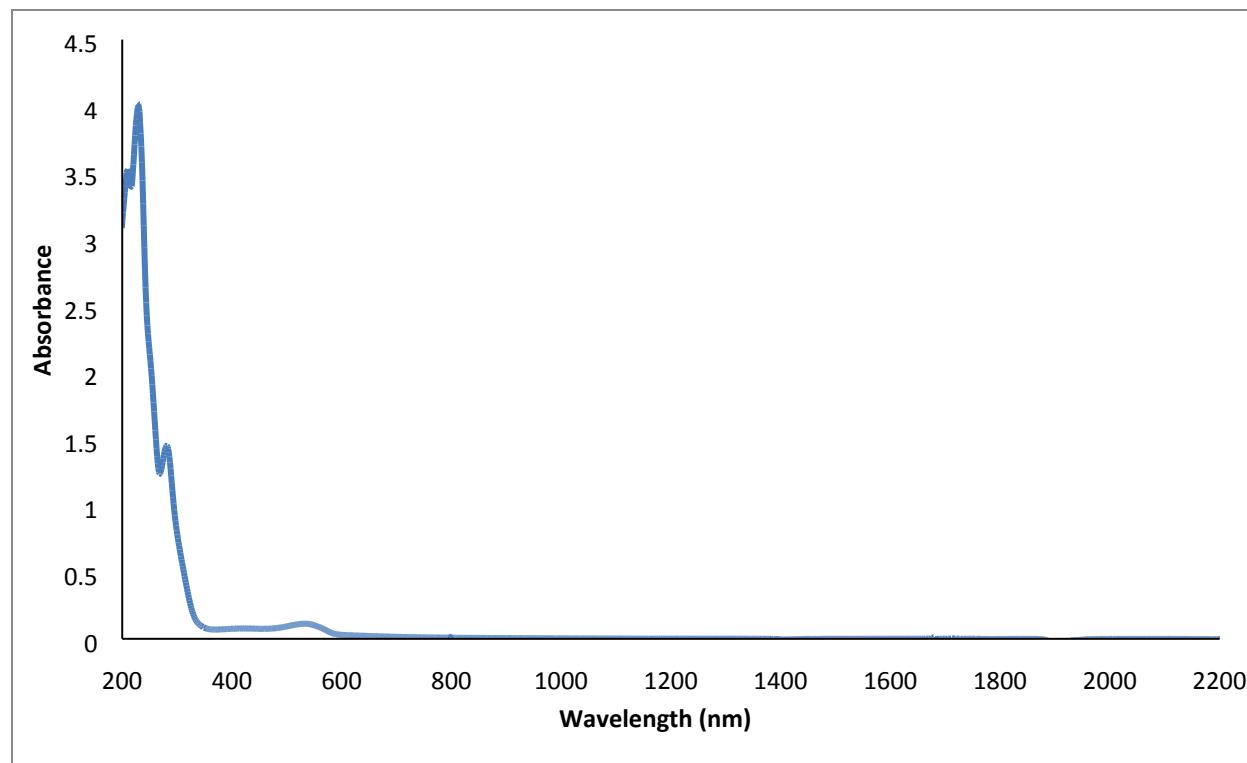


Figure S1. UV-Vis spectrum of 1mM $\text{Cu}(\text{btzp})_2\text{PF}_6$ in acetonitrile from 200 to 2200 nm.

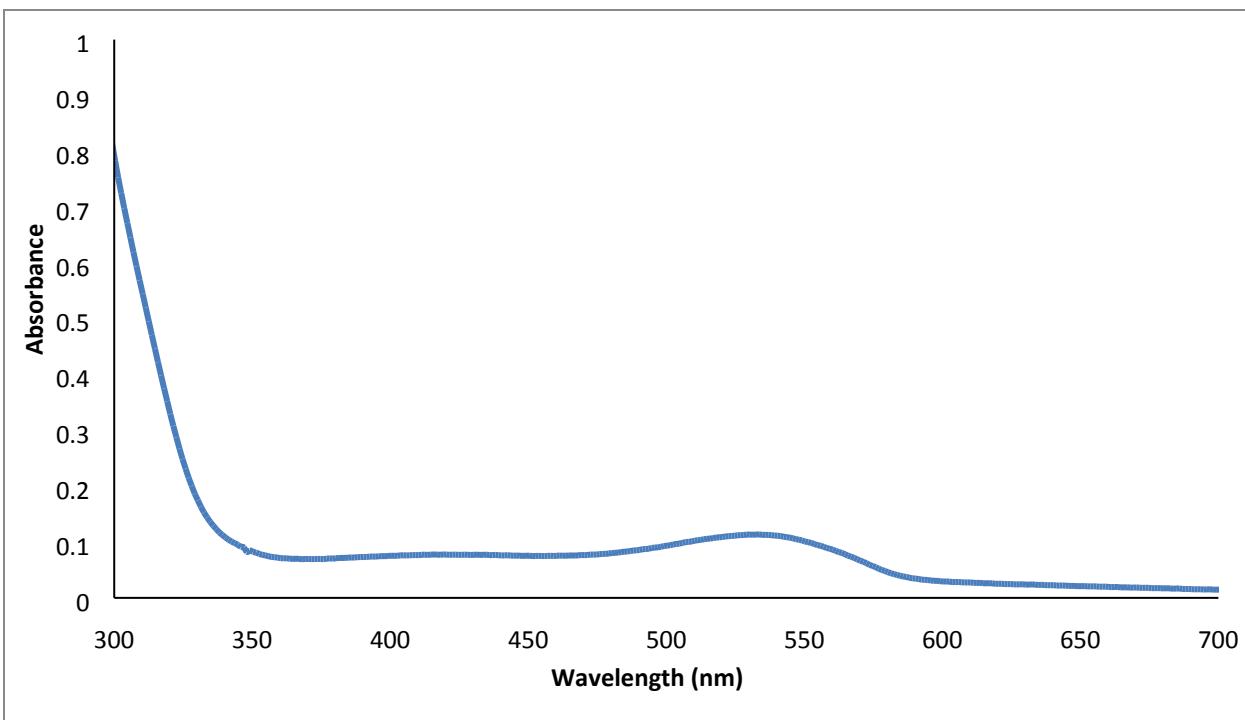


Figure S2. UV-Vis spectrum of 1mM CuL₂PF₆ in acetonitrile from 300 to 700 nm.

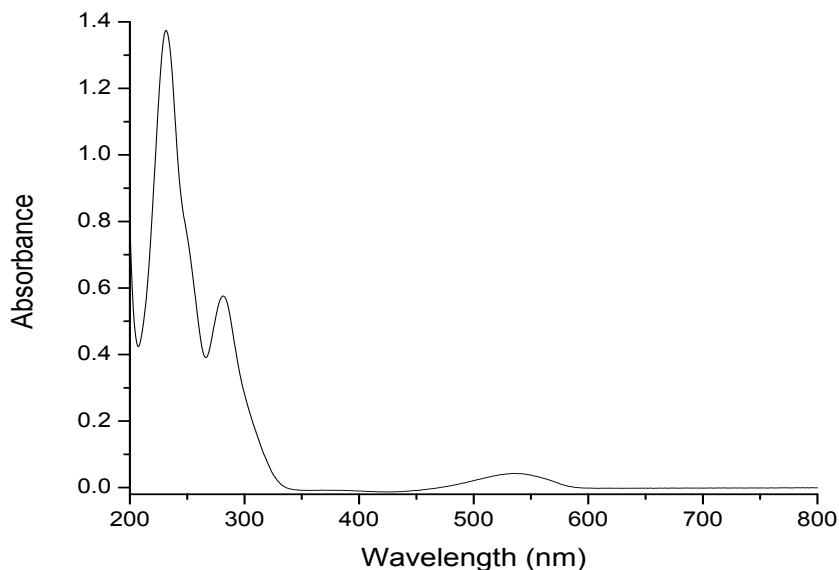


Figure S3. UV-Vis spectrum of 0.05 mM btzp in acetonitrile from 200 to 800 nm.

Reactivity

There has been a good deal of productive activity recently in using pendant Bronsted basicity to promote heterolytic splitting of H₂: the Bronsted base accepts H⁺ and the metal accepts H⁻. We therefore probed

whether such reactivity could exist for Cu(btzp)₂⁺. Exposure of this cation, dissolved in MeCN, to 1 atm of H₂ showed no color change nor any change in the ¹H NMR spectrum over a period of 24 h at 25 °C, indicating that the envisioned heterolysis does not occur. There was not even any reduction to bulk copper metal.

Electronic structure calculations were carried out using DFT¹ as implemented in Gaussian09.² Geometry optimizations were performed at the B3LYP/LANL2DZ/6-31G(d,p)³⁻⁷ level of theory with no symmetry constraints. All optimized structures were confirmed to have stable wavefunctions,⁸⁻⁹ and to be local minima by analyzing the harmonic frequencies.¹⁰⁻¹¹ Cartesian coordinates and frequencies for all optimized species may be found in Tables S1 and S2, respectively. We begin calculations assuming the wavefunction is unrestricted. Calculations that began with an unpaired electron on Cu and another of opposite spin on the ligand(s) collapsed back to a restricted wavefunction. Wavefunction stability analysis also suggested that, for the geometries we tested, the restricted wavefunction is stable with respect to a breaking of the alpha and beta spin densities.

1. Parr, R.G.; Yang, W. *Density-functional theory of atoms and molecules*; Oxford University Press: New York, 1989.
2. Gaussian 09, Revision C.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2010.
3. Vosko, S.H.; Wilk, L.; Nusair, M. *Can. J. Phys.* **1980**, 58, 1200.
4. Lee, C.; Yang, W.; Parr, R.G. *Phys. Rev. B* **1988**, 37, 785.
5. Becke, A.D. *J. Chem. Phys.* **1993**, 98, 5648.
6. Stephens, P.J.; Devlin, F.J.; Chabalowski, C.F.; Frisch, M.J. *J. Phys. Chem.* **1994**, 98, 11623.
7. a) Hay, P.J.; Wadt, W.R. *J. Chem. Phys.* **1985**, 82, 270. b) Wadt, W.R.; Hay, P.J. *J. Chem. Phys.* **1985**, 82, 284. c) Hay, P.J.; Wadt, W.R. *J. Chem. Phys.* **1985**, 82, 299.
8. Schlegel, H.B.; McDouall, J.J. In *Computational Advances in Organic Chemistry*; Oegretir, C.; Csizmadia, I.G., Eds.; Kluwer Academic: Amsterdam, The Netherlands, 1991.
9. Baurenschmitt, R.; Ahlrichs, R. *J. Chem. Phys.* **1996**, 104, 9047.
10. Schlegel, H.B. *J. Comput. Chem.* **1982**, 3, 214.
11. Schlegel, H.B. *WIREs Comput. Mol. Sci.* **2011**, 1, 790.

Table S1. Cartesian Coordinates (in Å) for Optimized Structures.[Cu(btzp)₂]⁺ S=0

Cu	-0.005052	0.511502	-0.023446
N	0.790088	-0.227229	-1.921829
C	0.255584	-1.172343	-2.710081
C	1.953134	0.335237	-2.288823
C	0.866451	-1.589992	-3.900103
C	2.632763	-0.020702	-3.458822
C	2.072233	-1.001506	-4.273548
H	0.392122	-2.355679	-4.501483
H	3.567730	0.467719	-3.705011
H	2.568697	-1.303499	-5.189827
C	2.481748	1.374634	-1.376959
C	3.370464	3.208092	0.211394
C	-1.031120	-1.767010	-2.268536
C	-3.287383	-2.781467	-1.536450
N	1.763331	1.671975	-0.281234
N	2.212208	2.609912	0.535861
N	4.105114	2.895571	-0.882512
N	3.648302	1.965112	-1.687158
N	-1.555358	-1.359064	-1.106404
N	-2.713429	-1.871630	-0.729421
N	-2.738766	-3.214688	-2.695492
N	-1.594355	-2.694531	-3.067731
C	-4.606529	-3.360439	-1.138456
H	-4.894657	-3.002655	-0.149834
H	-5.375177	-3.076459	-1.864847
H	-4.554066	-4.452976	-1.138618
C	3.889886	4.285639	1.107166
H	3.890074	5.243812	0.576905
H	3.269585	4.368557	1.999711
H	4.925348	4.074725	1.388878
N	-0.784543	-0.067203	1.936870
C	-1.949670	0.517151	2.260684
C	-0.238822	-0.936494	2.801135
C	-2.619699	0.260441	3.461729
C	-0.839913	-1.252820	4.026912
C	-2.047503	-0.641767	4.355312
H	-3.556639	0.762364	3.670309
H	-0.356663	-1.960039	4.689625
H	-2.536499	-0.865765	5.297627
C	1.049395	-1.559525	2.405455
N	1.567237	-1.248227	1.211057
N	2.725110	-1.787449	0.872483
C	3.305498	-2.625444	1.749588
C	-2.492066	1.467942	1.264125
N	-1.783249	1.672183	0.141186
N	-2.244796	2.531200	-0.752262
C	-3.405617	3.147140	-0.473650
N	-4.130487	2.927505	0.649079
N	-3.661131	2.075189	1.529397

N	1.620769	-2.413370	3.277792
N	2.765080	-2.959465	2.944840
C	4.623087	-3.234521	1.393655
H	5.391956	-2.904488	2.099901
H	4.566758	-4.324441	1.468255
H	4.912910	-2.946934	0.382867
C	-3.940516	4.136101	-1.458096
H	-3.311533	4.163528	-2.347974
H	-4.966315	3.875300	-1.734111
H	-3.973429	5.132748	-1.005875

[Cu(btzp)₂]⁺ S=1

Cu	-0.000210	-0.002729	-0.182409
N	-0.001542	-0.018285	-2.168494
C	0.182810	-1.184474	-2.814194
C	-0.187479	1.138011	-2.831187
C	0.187505	-1.233225	-4.208166
C	-0.195127	1.165889	-4.225802
C	-0.004522	-0.038844	-4.911574
H	0.337721	-2.181955	-4.710144
H	-0.346513	2.107073	-4.741437
H	-0.005645	-0.047055	-5.996368
C	-0.367983	2.279875	-1.935527
C	-0.674194	4.214589	-0.248285
C	0.364645	-2.312666	-1.901723
C	0.672830	-4.222160	-0.186414
N	-0.318034	1.996770	-0.604879
N	-0.477850	3.002516	0.265775
N	-0.724661	4.503125	-1.583331
N	-0.566003	3.499384	-2.429062
N	0.317019	-2.009808	-0.575531
N	0.477785	-3.002323	0.309552
N	0.721210	-4.530465	-1.517177
N	0.561564	-3.539767	-2.377421
C	0.856122	-5.359032	0.767987
H	0.796453	-5.007805	1.798355
H	0.090076	-6.120963	0.593325
H	1.825476	-5.837879	0.597582
C	-0.856510	5.365525	0.689356
H	-1.827163	5.840179	0.514613
H	-0.793292	5.030083	1.724766
H	-0.092289	6.125969	0.500842
N	0.002227	0.014144	1.928534
C	-1.144634	-0.140724	2.616204
C	1.149946	0.180630	2.611985
C	-1.185178	-0.134584	4.014667
C	1.192348	0.197581	4.010330
C	0.004069	0.037283	4.716817
H	-2.136159	-0.263674	4.516359
H	2.144008	0.334933	4.508526
H	0.004797	0.046180	5.801775
C	2.394219	0.350442	1.822460

N	2.300719	0.339931	0.490037
N	3.399183	0.489683	-0.225868
C	4.555084	0.646722	0.443436
C	-2.390015	-0.322489	1.831168
N	-2.297808	-0.337165	0.498762
N	-3.397773	-0.494426	-0.213378
C	-4.553802	-0.632547	0.459740
N	-4.645531	-0.624202	1.812317
N	-3.546130	-0.465860	2.506957
N	3.551043	0.505485	2.494308
N	4.649122	0.656205	1.795767
C	5.815198	0.825140	-0.337629
H	6.243537	1.812328	-0.135032
H	6.557606	0.084040	-0.027208
H	5.616787	0.728001	-1.404972
C	-5.817286	-0.807002	-0.316763
H	-5.604382	-0.838569	-1.385246
H	-6.321203	-1.728883	-0.011070
H	-6.503781	0.018910	-0.104212

Table S2. Frequencies (in cm⁻¹) for Optimized Structures.

[Cu(btzp)₂]⁺ S=0

9.7776	12.6871	13.7933
23.8243	24.0989	33.0893
37.6872	39.5819	42.5769
44.3351	46.2441	51.0939
53.5812	55.3059	58.0909
63.6922	67.6537	76.9314
104.8305	105.4081	108.5190
117.7926	155.8359	159.7597
161.2555	164.1570	170.1808
186.8874	203.1123	257.8860
264.9846	322.7447	322.9590
330.2466	330.3778	332.6312
336.1191	345.7590	346.0653
357.7656	357.9560	363.3884
368.6917	370.0150	375.6387
446.3320	447.3204	454.2999
458.2784	464.8146	470.6208
539.7633	541.2536	542.2704
543.4921	625.9592	626.1606
630.2989	632.1207	677.6515
677.9104	684.0636	684.4951
684.9660	685.6472	699.9920
700.1039	763.0977	763.9530
811.6930	811.9128	813.9604
813.9760	834.7677	836.1348
865.3456	865.6251	868.4859
870.0110	885.0505	885.5030
976.8914	977.6182	983.5236

984.1100	989.6766	990.3871
1024.6366	1026.1790	1036.8338
1037.0403	1059.2913	1059.4964
1060.1266	1060.2314	1060.3738
1060.5572	1065.6027	1066.0993
1080.8395	1080.9604	1104.1593
1104.3011	1115.4888	1116.6326
1140.0844	1140.7477	1164.4956
1165.0340	1195.8157	1197.3798
1299.8576	1300.4038	1324.9769
1325.2434	1329.4868	1329.5588
1332.3612	1333.7687	1364.3347
1364.7272	1397.6426	1398.0410
1409.1893	1409.5078	1436.2558
1436.3447	1439.3200	1439.4492
1459.5770	1460.5577	1473.3324
1473.8606	1482.3014	1482.3656
1482.7290	1482.9875	1492.0975
1492.4225	1501.2410	1501.8064
1511.5396	1513.0482	1518.6847
1519.0999	1542.0910	1542.8588
1548.3624	1549.8905	1629.1073
1629.3152	1638.5631	1638.8618
3066.6014	3066.7541	3066.9874
3067.1287	3129.6102	3129.8190
3130.3219	3130.6759	3176.9972
3177.0481	3178.2851	3178.6805
3219.2773	3219.4630	3245.8238
3246.1942	3248.2429	3248.4076

[Cu(btzp)₂]⁺ S=1

6.8035	13.7833	16.3241
21.5758	33.2996	33.4207
36.7146	44.0440	54.1718
61.1346	63.9488	79.1806
80.4371	84.2390	85.3106
88.2740	94.6730	97.3952
102.7720	135.4706	158.0567
159.1948	169.3575	179.1248
182.7502	192.0939	198.6109
234.4187	244.4021	266.4980
290.5676	316.4640	322.6071
328.5672	333.2201	338.1146
348.3906	350.0167	353.0445
366.3154	367.2983	368.5944
369.7216	373.0886	379.0998
450.0558	455.7078	455.8467
466.0911	473.0972	497.8737
534.3453	542.8742	557.1521
569.7021	617.5958	626.2136
634.2170	640.3432	667.3340
677.5850	679.6695	682.2281

683.0560	683.5918	703.6398
707.7344	745.7138	770.1507
777.3590	786.6469	811.1568
811.5547	817.1148	836.1574
846.7929	864.1342	866.5690
869.6656	870.4385	884.9230
965.1667	982.5993	985.1978
988.9900	990.8075	992.4775
1021.6939	1030.2974	1036.5743
1043.7182	1058.6256	1058.8036
1059.0731	1059.3048	1059.5165
1065.1206	1065.5658	1079.7278
1081.4731	1081.5867	1092.6146
1095.4430	1104.3572	1122.0786
1137.5512	1143.4391	1145.1647
1168.6280	1178.0620	1202.2598
1235.7767	1257.7346	1291.4839
1300.2667	1303.4115	1319.8789
1326.0487	1327.8356	1338.5891
1352.5769	1358.0020	1366.9518
1396.4994	1407.3228	1408.6011
1413.9050	1435.3748	1436.0376
1438.3724	1438.8984	1448.5663
1463.1019	1465.1376	1473.1908
1480.9657	1481.0788	1481.5537
1481.5761	1493.7621	1496.8488
1497.2196	1504.2605	1521.9198
1523.4559	1523.6632	1549.6911
1551.7230	1553.4100	1590.1116
1630.5150	1640.9897	1643.2650
3065.6083	3065.6481	3066.8981
3067.1590	3127.1710	3127.1945
3129.7377	3129.8986	3174.4308
3174.5435	3180.0678	3180.1247
3220.4716	3221.8735	3240.0190
3242.7713	3248.6711	3250.3522