# Synthesis and Structural Characterization of a Single-Crystal to Single-Crystal Transformable Coordination Polymer 

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## Supporting Information

## Elemental Analysis of CHNS

The elemental analysis on $\mathrm{C}, \mathrm{H}, \mathrm{N}$ and S of the products was taken place on a Carlo Erba CHNS analyser. The weight contents of $\mathrm{C}, \mathrm{H}, \mathrm{N}$ and S are $35.94 \%, 2.51 \%, 3.65 \%$ and $7.05 \%$. The calculated values are $34.49 \%$, $2.43 \%, 3.10 \%$ and $7.08 \%$.

## Thermal Stability

The TGA curve of compound 1 was shown in Fig S1. An obvious weight loss of $7.42 \%$ ranging from 473 K to 513 K in air atmosphere can be referred to the removal of the coordinated molecules (calculated weight loss from the complete removal of water is $7.96 \%$ ).


Fig. S1, TGA curve for hydrated sample of compound $\mathbf{1}$ in air atmosphere at the heating rate of $10 \mathrm{Kmin}^{-1}$.

Compound 1 was put in a tube furnace, heated under air at a $10 \mathrm{Kmin}^{-1}$ rate. After calcined at $520 \mathrm{~K}, 540 \mathrm{~K}, 580 \mathrm{~K}$ and 600 K for 10 min respectively, the residuals were recovered to room temperature. Powder XRD was carried on these residuals. The patterns are shown in Fig S2.


## Coordination geometry of $\mathrm{Cu}-\mathrm{BPy}-\mathrm{Cu}$ chain

The coordination geometry of Cu 2 in compound $\mathbf{1}$ is shown in Fig. S3a. It is 6-coordinated by two STP molecules and two Bpy molecules in an inverse way. The changes of Cu 2 after dehydration are shown in fig. S 3 b . The coordination geometry of copper atoms remains 6 -coordinated but the $\mathrm{Cu}-\mathrm{Bpy}-\mathrm{Cu}$ chain has a torsion angle of $19.131(821)^{0}$.

b


Fig. S3. Coordination environment of Cu 2 atoms in hydrated structure (a) and the changes after dehydration (b). (Cu: turquoise, O: red, S: yellow, C: grey, N : blue. H is not shown for clarity.)

## Crystallographic data

The crystal data and structure refinements of compound $\mathbf{1}$ and compound $\mathbf{2}$ are shown in table S1.
Table S1. Crystal data and structure refinements of compounds $\mathbf{1}$ and compound 2

| Compound | $\mathbf{1}$ | $\mathbf{2}$ |
| :--- | :--- | :--- |
| Formula | $\mathrm{C}_{26} \mathrm{H}_{22} \mathrm{Cu}_{3} \mathrm{~N}_{2} \mathrm{O}_{18} \mathrm{~S}_{2}$ | $\mathrm{C}_{26} \mathrm{H}_{14.5} \mathrm{Cu}_{3} \mathrm{~N}_{2} \mathrm{O}_{14.25} \mathrm{~S}_{2}$ |
| Formula weight | 905.20 | 837.64 |
| Crystal system | Monoclinic | Monoclinic |
| Space group | $\mathrm{P} 21 / \mathrm{c}$ | $\mathrm{P} 21 / \mathrm{c}$ |
| $a(\AA)$ | $11.017(5)$ | $10.075(5)$ |
| $b(\AA \AA)$ | $11.814(5)$ | $12.806(5)$ |
| $c(\AA)$ | $12.135(5)$ | $19.502(5)$ |
| $\alpha(\mathrm{O})$ | $90.000(5)$ | $90.000(5)$ |
| $\beta(\mathrm{O})$ | $109.701(5)$ | $95.342(5)$ |
| $\gamma(\mathrm{O})$ | $90.000(5)$ | $90.000(5)$ |
| $V\left(\AA^{3}\right)$ | $1487.0(11)$ | $2505.2(17)$ |


| $Z$ | 2 | 4 |
| :--- | :--- | :--- |
| $D_{\mathrm{c}}\left(\mathrm{g} / \mathrm{cm}^{-3}\right)$ | 2.022 | 2.221 |
| $F(000)$ | 910 | 1670 |
| $\mu\left(\mathrm{~mm}^{-1}\right)$ | 2.890 | 3.411 |
| $\vartheta$ for data collection $\left({ }^{\circ}\right)$ | 2.85 to 34.65 | 2.96 to 22.79 |
| Reflections collected | 21375 | 23264 |
| Unique reflections | 4736 | 2610 |
| $R_{\text {int }}$ | 0.0578 | 0.0959 |
| Data/restraints $/$ parameters | $4736 / 6 / 244$ | $2610 / 0 / 412$ |
| Goodness-of-fit on $F^{2}$ | 1.023 | 1.045 |
| $R_{1}, \mathrm{w} R_{2}[I>2 \sigma(\mathrm{I})]$ | $0.0327,0.0904$ | $0.1038,0.2118$ |
| $R_{1}, \mathrm{w} R_{2}($ all data $)$ | $0.0378,0.0938$ | $0.0713,0.1854$ |
| $\mathrm{R}_{1}=\sum\left(\left\|\mathrm{F}_{\mathrm{o}}\right\|-\left\|\mathrm{F}_{\mathrm{c}}\right\|\right) / \sum\left\|\mathrm{F}_{\mathrm{o}}\right\|, \mathrm{w} R_{2}=\left[\sum \mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2} / \sum \mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}\right)^{2}\right]^{1 / 2}$. |  |  |

Table S2. Selected Bond lengths $[\AA]$ for mor20.

| $\mathrm{Cu}(1)-\mathrm{O}(3)$ | $1.9321(17)$ |
| :--- | :--- |
| $\mathrm{Cu}(1)-\mathrm{O}(1 \mathrm{~W})$ | $1.9757(16)$ |
| $\mathrm{Cu}(1)-\mathrm{O}(2 \mathrm{~W})$ | $1.9909(16)$ |
| $\mathrm{Cu}(1)-\mathrm{O}(5) \# 1$ | $1.9913(16)$ |
| $\mathrm{Cu}(1)-\mathrm{O}(9) \# 2$ | $2.2895(16)$ |
| $\mathrm{Cu}(2)-\mathrm{N}(1) \# 3$ | $1.9744(18)$ |
| $\mathrm{Cu}(2)-\mathrm{N}(1)$ | $1.9744(18)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(6)$ | $2.0435(16)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(6) \# 3$ | $2.0435(16)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(7)$ | $2.2887(17)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(7) \# 3$ | $2.2887(17)$ |

Symmetry transformations used to generate equivalent atoms:
\#1 $x-1,-y+3 / 2, z-1 / 2 \quad \# 2-x+2,-y+2,-z+1 \quad \# 3-x+3,-y+2,-z+2$
$\# 4 x+1,-y+3 / 2, z+1 / 2 \quad \# 5-x+2,-y+2,-z+2$

Table S3. Selected Bond lengths $[\AA]$ for mor21.

| $\mathrm{Cu}(1)-\mathrm{O}(13)$ | $1.825(8)$ |
| :--- | :--- |
| $\mathrm{Cu}(1)-\mathrm{O}(11) \# 1$ | $1.907(10)$ |
| $\mathrm{Cu}(1)-\mathrm{O}(12) \# 1$ | $1.948(9)$ |
| $\mathrm{Cu}(1)-\mathrm{O}(1)$ | $1.960(9)$ |
| $\mathrm{Cu}(1)-\mathrm{C}(15) \# 1$ | $2.253(15)$ |
| $\mathrm{Cu}(1)-\mathrm{S}(1)$ | $2.769(4)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(5) \# 2$ | $1.856(9)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(7) \# 3$ | $1.869(12)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(10)$ | $1.975(10)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(6) \# 3$ | $2.121(14)$ |
| $\mathrm{Cu}(2)-\mathrm{O}(9) \# 4$ | $2.193(10)$ |
| $\mathrm{Cu}(2)-\mathrm{C}(8) \# 3$ | $2.29(2)$ |
| $\mathrm{Cu}(3)-\mathrm{N}(2) \# 5$ | $1.889(12)$ |
| $\mathrm{Cu}(3)-\mathrm{N}(2)$ | $1.889(12)$ |
| $\mathrm{Cu}(3)-\mathrm{O}(14)$ | $2.027(9)$ |
| $\mathrm{Cu}(3)-\mathrm{O}(14) \# 5$ | $2.027(9)$ |
| $\mathrm{Cu}(3)-\mathrm{O}(8)$ | $2.334(10)$ |
| $\mathrm{Cu}(3)-\mathrm{O}(8) \# 5$ | $2.335(10)$ |
| $\mathrm{Cu}(4)-\mathrm{N}(1) \# 2$ | $1.895(13)$ |
| $\mathrm{Cu}(4)-\mathrm{N}(1)$ | $1.895(13)$ |
| $\mathrm{Cu}(4)-\mathrm{O}(4)$ | $1.923(9)$ |
| $\mathrm{Cu}(4)-\mathrm{O}(4) \# 2$ | $1.923(9)$ |

Symmetry transformations used to generate equivalent atoms:

```
#1 x,-y+1/2,z+1/2 #2 -x,-y+1,-z+1 #3 -x,y-1/2,-z+3/2
#4 -x,-y,-z+1 #5 -x-1,-y,-z+1 #6 -x,y+1/2,-z+3/2
#7 x,-y+1/2,z-1/2 #8 -x-1,y+1/2,-z+1/2 #9 -x-1,y-1/2,-z+1/2
```

