

Supporting Information for

Synthesis and Characterization of 1D Iron(II) Spin Crossover Coordination Polymers with Hysteresis

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Table S1. Crystallographic data of **1**·3 MeOH and **4**·(0.5 bipy)(MeOH).

compound	1 ·3 MeOH	4 ·(0.5 bipy) (2 MeOH)
formula	C ₃₃ H ₄₂ FeN ₄ O ₁₁	C ₃₅ H ₃₈ FeN ₅ O ₈
M _r / g mol ⁻¹	726.56	712.55
crystal system	monoclinic	triclinic
space group	P2 ₁ /c	P ₁
a / Å	11.1275(18)	11.1228(9)
b / Å	16.408(2)	11.8129(9)
c / Å	21.317(3)	13.3170(9)
α / °	90	102.269(5)
β / °	115.206(11)	103.440(4)
γ / °	90	95.947(4)
V / Å ³	3521.5(9)	1641.3(2)
Z	4	2
ρ / g cm ⁻³	1.370	1.442
μ / mm ⁻¹	0.492	0.521
crystal size	0.23 × 0.07 × 0.03	0.19 × 0.07 × 0.04
T / K	173(3)	173(2)
diffractometer	Oxford XCalibur	KappaCCD
λ (MoK _α) / Å	0.71073	0.71073
θ-range / °	4.22–26.14	3.24–25.19
reflns. collected	12551	10103
indep. reflns. (<i>R</i> _{int})	5806 (0.0487)	5799 (0.072)
mean σ(<i>I</i>) / <i>I</i>	0.1784	0.0993
reflns. with <i>I</i> ≥ 2σ(<i>I</i>)	2722	3892
x, y (weighting scheme)	0.0082, 0	0.0558, 4.6271
parameters	448	474
restraints	0	31
<i>R</i> (<i>F</i>) (all data) ^[a]	0.0411 (0.1101)	0.0699 (0.1140)
w <i>R</i> (<i>F</i> ²) ^[b]	0.0518	0.1853
GooF	0.690	1.064
shift/error _{max}	0.001	0.000
max., min. resid. dens. / e Å ⁻³	0.42, -0.38	0.60, -0.73

[a] $R(F) = \sum ||F_o| - |F_c|| / \sum |F_o|$. [b] $wR(F^2) = [\sum [w(F_o^2 - F_c^2)^2] / \sum w(F_o^2)^2]^{1/2}$, $w = 1 / [\sigma^2(F_o^2) + (aP)^2 + bP]$, where $P = [F_o^2 + 2(F_c^2)]/3$.

Figure S1. Plots of the $\chi_M T$ product vs. T for the compounds **1**, **2**·(0.3 bpea), **4**·(0.5 bipy)(MeOH) and **2**·(0.5 bpea)(0.5 tol). The green curve of **2**·(0.5 bpea)(0.5 tol) was obtained after heating of the sample to 400 K for one hour.

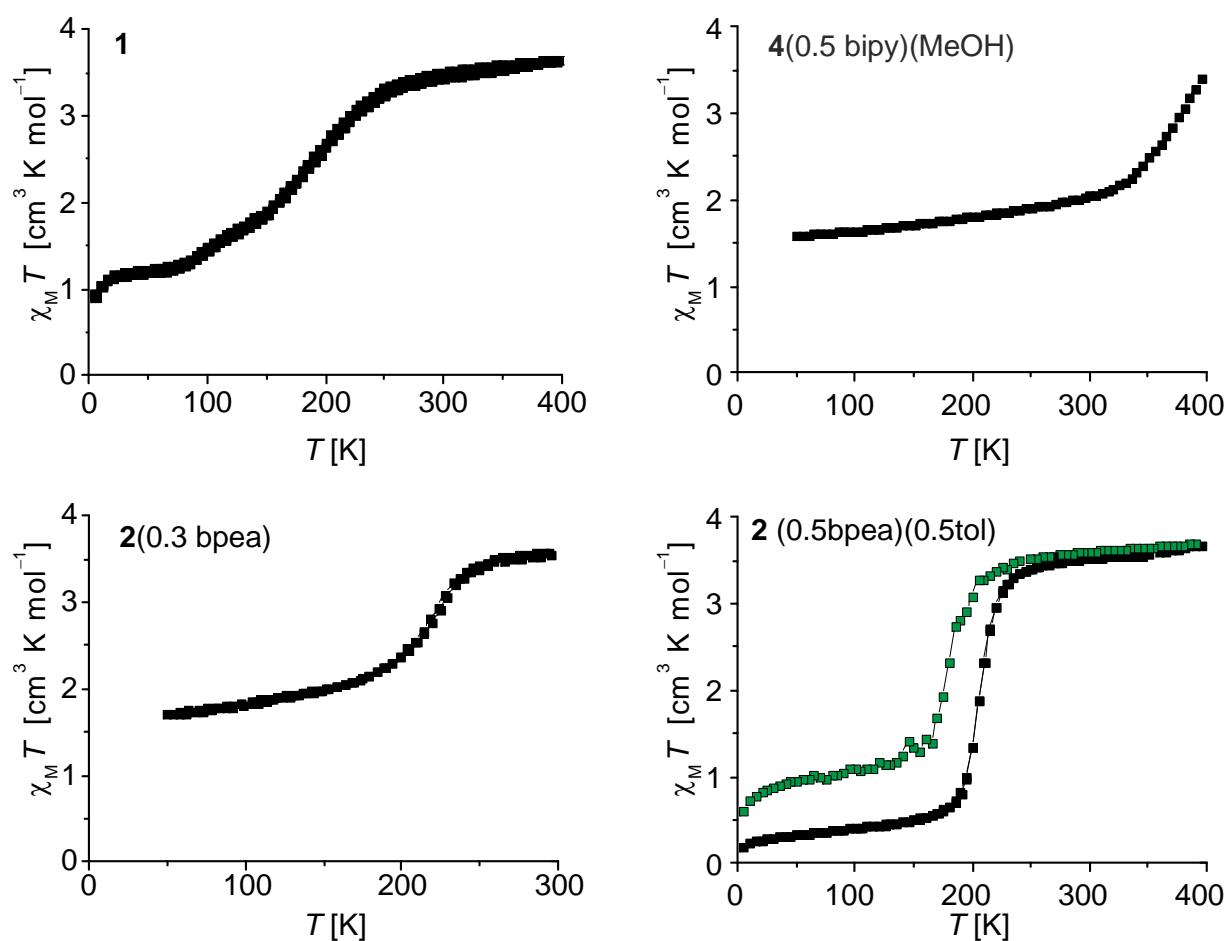


Figure S2. Mössbauer spectrum of the partly oxidised starting material $[\text{FeL1a}(\text{MeOH})_2]$. The oxidation product is the corresponding iron(III)- μ -oxido complex. The Mössbauer parameters are $\delta = 1.004(2)$ mm/s and $\Delta E_Q = 2.328(4)$ mm/s for the iron(II) HS complex and $\delta = 0.397(3)$ mm/s and $\Delta E_Q = 0.768(6)$ mm/s for the iron(III)- μ -oxido complex.

