ESI to accompany:

Red emitting $[Ir(C^N)_2(N^N)]^+$ complexes employing bidentate 2,2':6,2''-terpyridine ligands for light-emitting electrochemical cells

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Fig. S1. Structure of one of two independent $[Ir(ppy)_2(tpy)]^+$ cations in $4\{[Ir(ppy)_2(tpy)][PF_6]\}\cdot 2.5Et_2O\cdot 1.5MeCN\cdot 3H_2O$. Ellipsoids plotted at 30% probability level; H atoms omitted. Selected bond lengths: Ir1a–N1a = 2.130(5), Ir1a–N2a = 2.214(5), Ir1a–C27a = 1.994(6), Ir1a–C16a = 2.018(6), Ir1a–N5a = 2.042(5), Ir1a–N4a = 2.049(5) Å.



Fig. S2. Structure of one of two independent $[Ir(ppy)_2(2)]^+$ cations in $4\{[Ir(ppy)_2(2)][PF_6]\} \cdot Et_2O \cdot 2MeCN$. Ellipsoids plotted at 30% probability level; H atoms omitted. Selected bond lengths: Ir1a-C23a = 1.996(3), Ir1a-C34a = 2.015(3), Ir1a-N4a = 2.039(2), Ir1a-N5A = 2.058(2), Ir1a-N1a = 2.122(2), Ir1a-N2a = 2.206(2) Å.



Fig. S3. Structure of the $[Ir(ppy)_2(3)]^+$ cation in $2\{[Ir(ppy)_2(3)][PF_6]\}$ ·MeCN·H₂O with ellipsoids plotted at 30% probability level; H atoms omitted. Selected bond lengths: Ir1-C23 = 2.0037(12), Ir1-C34 = 2.0140(14), Ir1-N5 = 2.0504(12), Ir1-N4 = 2.0533(11), Ir1-N1 = 2.1291(11), Ir1-N2 = 2.2300(10) Å.



Fig. S4. Structure of the $[Ir(ppy)_2(4)]^+$ cation in $[Ir(ppy)_2(4)][PF_6] \cdot 2CH_2Cl_2$ with ellipsoids plotted at 30% probability level; H atoms except those in the NH₂ unit are omitted. Selected bond lengths: Ir1–C33 = 2.012(5), Ir1–C22 = 2.013(5), Ir1–N6 = 2.039(4), Ir1–N5 = 2.057(4), Ir1–N1 = 2.153(4), Ir1–N2 = 2.204(4) Å.



Fig. S5. Structure of one of two independent $[Ir(ppy)_2(pytpy)]^+$ cations in $2\{[Ir(ppy)_2(pytpy)][PF_6]\}\cdot Et_2O\cdot CH_2Cl_2$. Ellipsoids plotted at 30% probability level and H atoms omitted. Selected bond lengths: Ir1b-C32b = 1.929(17), Ir1b-C21b = 1.978(15), Ir1b-N5b = 2.034(13), Ir1b-N6b = 2.043(13), Ir1b-N1b = 2.115(12), Ir1b-N2b = 2.214(14) Å.



Fig. S6. Structure of the $[Ir(dmppz)_2(tpy)]^+$ cation in $[Ir(dmppz)_2(tpy)][PF_6]$ with ellipsoids plotted at 30% probability level and H atoms omitted. Selected bond lengths: Ir1–C16 = 2.0030(19), Ir1–C27 = 2.011(2), Ir1–N6 = 2.0250(17), Ir1–N4 = 2.0281(18), Ir1–N1 = 2.1402(17), Ir1–N2 = 2.2212(17) Å.



Fig. S7. Structure of the $[Ir(dmppz)_2(pytpy)]^+$ cation in $[Ir(dmppz)_2(pytpy)][PF_6] \cdot CH_2 Cl_2$; ellipsoids plotted at 30% probability level and H atoms omitted. Selected bond lengths: Ir1-C21 = 2.008(3), Ir1-C32 = 2.015(2), Ir1-N7 = 2.031(2), Ir1-N5 = 2.051(2), Ir1-N1 = 2.134(2), Ir1-N2 = 2.206(2) Å.



6.0 4.5 9.0 6.5 5.5 5.0 4.0 3.5 8.5 8.0 7.5 7.0 3.0 2.5 2.0 1.5 Fig. S8 500 MHz ¹H NMR spectrum of [Ir(dmppz)₂(Mepytpy)][PF₆]₂ (CDCl₃, 295 K). Peaks marked ** and * are residual CHCl₃ and H₂O respectively.



Fig. S9. Structure of the $[Ir(dmppz)_2(Mepytpy)]^{2+}$ cation in $[Ir(dmppz)_2(Mepytpy)][PF_6]_2 \cdot 2CH_2Cl_2$; ellipsoids plotted at 30% probability level and H atoms omitted. Selected bond lengths: Ir1–C33 1.991(5), Ir1–N5 = 2.018(5), Ir1–C22 = 2.024(5), Ir1–N7 = 2.044(4), Ir1–N1 = 2.152(4), Ir1–N2 = 2.214(4) Å.



Fig. S10 Sweep to positive potential in the cyclic voltammograms of (a) $[Ir(ppy)_2(2)][PF_6]$ and (b) $[Ir(dmppz)_2(tpy)][PF_6]$ (with respect to Fc/Fc⁺; CH₂Cl₂ solution) showing the quasi-reversible process.



Fig S11. The device performance of the $[Ir(ppy)_2(6-Phbpy)][PF_6]$ under the pulsed driving conditions, analogous to those used for the best preforming devices in this paper. (a) Luminance, and (b) average voltage versus time.



Fig. S12. Photoluminescence spectra in thin films (iTMC:IL 4:1).