

Supporting Information

for

A Theoretical Analysis of the Phosphorescence Efficiencies of Cu(I) Complexes

Lu-Yi Zou,[†] Yan-Xiang Cheng,[‡] Yan Li,[†] Hui Li,[‡] Hong-Xing Zhang,[†] Ai-Min Ren^{*†}

[†]State Key Laboratory of Theoretical and Computational Chemistry, Institute of Theoretical Chemistry, Jilin University, Changchun 130023, P. R. China

[‡]State Key Laboratory of Polymer Physics and Chemistry, Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, P. R. China.

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Table S1. Molecular orbital compositions in S₀ States for these complexes (%)

	[Cu(phen)(PPh ₃) ₂] ⁺			[Cu(dmp)(PPh ₃) ₂] ⁺			[Cu(dbp)(PPh ₃) ₂] ⁺			[Cu(phen)(POP)] ⁺			[Cu(dmp)(POP)] ⁺			[Cu(dbp)(POP)] ⁺		
	Cu	PPh ₃	NN	Cu	PPh ₃	NN	Cu	PPh ₃	NN	Cu	POP	NN	Cu	POP	NN	Cu	POP	NN
H-2	69.0	4.9	26.1	69.0	3.7	27.4	68.8	3.5	27.7	50.0	32.1	17.9	51.6	30.4	18.0	66.1	8.6	25.3
H-1	38.6	53.2	8.2	44.3	45.9	9.8	45.3	44.3	10.4	43.0	49.0	8.0	50.3	38.1	11.6	45.5	44.2	10.2
H	32.9	63.6	3.5	34.2	61.2	4.6	34.9	59.8	5.4	34.2	62.2	3.6	36.1	59.0	4.9	37.7	57.2	5.0
L	3.1	3.4	93.6	3.1	3.5	93.4	2.7	3.6	93.8	3.1	3.3	93.6	3.4	3.4	93.3	2.9	3.4	93.7
L+1	0.2	0.8	99.0	0.2	0.8	99.0	0.2	0.8	99.0	0.1	0.7	99.2	0.3	1.5	98.2	0.3	0.8	98.9
L+2	8.2	14.5	77.3	3.9	93.9	2.2	4.2	92.8	3.0	1.6	96.2	2.2	1.4	94.3	4.3	1.4	95.8	2.8

Table S2. Molecular orbital compositions of Cu at S₀ State in Solvent DCM (c_d² > 1%)

	MO	Energy (eV)	Cu contribution (%)
[Cu(phen)(PPh ₃) ₂] ⁺	L	-2.32	3.46 (1.25 d_{xy})
	H	-6.13	35.74 (27.00 d_{xy})
	H-1	-6.47	43.67 (21.34 $d_{x^2-y^2}$ +13.47 d_{z^2})
[Cu(dmp)(PPh ₃) ₂] ⁺	L	-2.25	3.35 (1.15 d_{xy})
	H	-6.13	35.98(27.81 d_{xy})
	H-1	-6.45	47.71 (23.58 $d_{x^2-y^2}$ +15.10 d_{z^2})
[Cu(dbp)(PPh ₃) ₂] ⁺	L	-2.23	2.80
	H	-6.14	36.22 (18.50 d_{yz} +10.11 d_{xz})
	H-1	-6.46	47.85 (25.26 d_{xy} +8.95 $d_{x^2-y^2}$ +4.69 d_{z^2})
[Cu(phen)(POP)] ⁺	L	-2.30	3.48 (1.40 $d_{x^2-y^2}$)
	H	-5.99	37.03 (29.02 $d_{x^2-y^2}$)
	H-1	-6.45	48.26 (19.85 d_{xy} +17.87 d_{z^2} +1.94 d_{yz})
[Cu(dmp)(POP)] ⁺	L	-2.23	3.57 (1.08 $d_{x^2-y^2}$)
	H	-6.00	38.28 (31.36 $d_{x^2-y^2}$)
	H-1	-6.53	66.01 (34.30 d_{yz} +25.00 d_{xz} +3.40 d_{xy})
[Cu(dbp)(POP)] ⁺	L	-2.19	3.06
	H	-6.07	39.06 (21.37 d_{z^2} +5.20 $d_{x^2-y^2}$ +4.82 d_{yz} +1.44 d_{xy})
	H-1	-6.43	49.63 (21.41 d_{yz} +12.84 $d_{x^2-y^2}$ +4.48 d_{xy} +3.73 d_{xz})

Table S3. Emission Spectra Obtained by TDDFT Method for these Complexes, Together with Experimental Values

	Electronic transitions	λ^{em} (nm)	Excitation energies (eV)	Main configurations	
[Cu(phen)(PPh ₃) ₂] ⁺	T ₁ →S ₀	834.4/680 ^a	1.49	HOMO→LUMO	0.69
				HOMO-2→LUMO	0.11
[Cu(dmp)(PPh ₃) ₂] ⁺	T ₁ →S ₀	607.1/560 ^a	2.04	HOMO-2→LUMO	0.60
				HOMO→LUMO	-0.33
[Cu(dbp)(PPh ₃) ₂] ⁺	T ₁ →S ₀	607.2	2.04	HOMO-2→LUMO	0.57
				HOMO→LUMO	-0.36
[Cu(phen)(POP)] ⁺	T ₁ →S ₀	812.6/700 ^a	1.53	HOMO→LUMO	0.69
[Cu(dmp)(POP)] ⁺	T ₁ →S ₀	658.0/570 ^a	1.88	HOMO→LUMO	0.68
[Cu(dbp)(POP)] ⁺	T ₁ →S ₀	611.9/560 ^a	2.03	HOMO-2→LUMO	0.50
				HOMO→LUMO	0.36

^a Measured in DCM in reference [13].

Table S4. Molecular orbital compositions of Cu at T₁ State in Solvent DCM (c_d² > 1%)

	MO	Energy (eV)	Cu contribution (%)
[Cu(phen)(PPh ₃) ₂] ⁺	L	-2.73	4.55 (1.98 d_{xy})
	H	-5.38	41.77 (28.09 d_{xy} + 5.61 d_{yz} + 3.06 $d_{x^2-y^2}$)
	H-1	-6.44	36.05 (11.15 d_{z^2} + 8.63 $d_{x^2-y^2}$ + 7.21 d_{xz})
[Cu(dmp)(PPh ₃) ₂] ⁺	L+1	-2.19	4.00 (1.33 d_{xy})
	H	-6.08	32.23 (25.65 d_{xy})
	H-1	-6.45	6.39 (3.49 d_{xy})
[Cu(dbp)(PPh ₃) ₂] ⁺	L+1	-2.16	3.32
	H	-6.08	31.21 (17.12 d_{yz} + 8.10 d_{xz})
	H-1	-6.41	9.58 (3.82 d_{yz} + 1.21 d_{xz})
[Cu(phen)(POP)] ⁺	L	-2.64	4.86 (1.39 $d_{x^2-y^2}$)
	H	-5.34	42.64 (20.27 $d_{x^2-y^2}$ + 12.60 d_{xy} + 4.53 d_{yz} + 1.88 d_{xz})
	H-1	-6.47	49.61 (30.82 d_{yz} + 6.22 d_{z^2} + 4.10 $d_{x^2-y^2}$ + 2.05 d_{xy})
[Cu(dmp)(POP)] ⁺	L	-2.59	4.44 (1.82 $d_{x^2-y^2}$)

	H	-5.57	42.59 (29.88 $d_{x^2-y^2}$ +6.97 d_{xz} +1.15 d_{yz})
	H-1	-6.32	59.75 (30.17 d_{yz} +19.80 d_{xz} +5.01 d_{xy})
[Cu(dbp)(POP)]⁺	L	-2.44	3.86
	H	-6.02	34.38 (17.05 d_{z^2} +6.34 d_{yz} +5.73 $d_{x^2-y^2}$)
	H-1	-6.38	27.83 (11.36 d_{xz} +7.23 $d_{x^2-y^2}$ +2.59 d_{z^2} +1.99 d_{xy} +1.03 d_{yz})

Table S5. SOC matrix elements between the $^3(d_k\pi^*)$ and $^1(d_l\pi^*)$ states.

	$^1(d_{xz}\pi^*)$	$^1(d_{yz}\pi^*)$	$^1(d_{z^2}\pi^*)$	$^1(d_{xy}\pi^*)$	$^1(d_{x^2-y^2}\pi^*)$
$^3(d_{xz}\pi^*)$	0	$\frac{c_{xz}c_{yz}}{2}$	$-\frac{\sqrt{3}}{2}c_{xz}c_{z^2}$	$-\frac{c_{xz}c_{xy}}{2}$	$-\frac{c_{xz}c_{x^2-y^2}}{2}$
$^3(d_{yz}\pi^*)$	$-\frac{c_{yz}c_{xz}}{2}$	0	$-\frac{\sqrt{3}}{2}c_{yz}c_{z^2}$	$-\frac{c_{yz}c_{xy}}{2}$	$\frac{c_{yz}c_{x^2-y^2}}{2}$
$^3(d_{z^2}\pi^*)$	$\frac{\sqrt{3}}{2}c_{z^2}c_{xz}$	$\frac{\sqrt{3}}{2}c_{z^2}c_{yz}$	0	0	0
$^3(d_{xy}\pi^*)$	$\frac{c_{xy}c_{xz}}{2}$	$\frac{c_{xy}c_{yz}}{2}$	0	0	$-c_{xy}c_{x^2-y^2}$
$^3(d_{x^2-y^2}\pi^*)$	$\frac{c_{x^2-y^2}c_{xz}}{2}$	$-\frac{c_{x^2-y^2}c_{yz}}{2}$	0	$c_{x^2-y^2}c_{xy}$	0

Table S6. Electronic transitions of phen-pop calculated at the T₁ optimized geometry by TDDFT in Solvent DCM (CI > 0.20)

S_n/T_m	Excitation energies (eV)	λ (nm)	Energy (cm⁻¹)	f	Main configurations	
T₁	1.66	746.4	13398	0.0000	HOMO→LUMO	0.69
S₁	1.91	648.7	15416	0.0374	HOMO→LUMO	0.69
S₂	2.44	507.6	19701	0.0048	HOMO→LUMO+1	0.70

S₃	3.03	409.3	24433	0.0530	HOMO-1→LUMO	0.68
S₄	3.15	393.2	25433	0.0352	HOMO-2→LUMO	0.68
S₅	3.39	366.1	27316	0.0094	HOMO→LUMO+2	0.69
S₆	3.44	360.0	27779	0.0505	HOMO→LUMO+3	0.69
S₇	3.54	350.7	28515	0.0176	HOMO→LUMO+4	0.67
S₈	3.59	345.7	28928	0.0026	HOMO-1→LUMO+1	0.59
S₉	3.59	345.1	28978	0.0117	HOMO-7→LUMO	0.45
S₁₀	3.64	340.3	29387	0.0158	HOMO-1→LUMO+1	0.24
					HOMO-6→LUMO	0.23
					HOMO→LUMO+5	0.45
					HOMO-11→LUMO	0.33
					HOMO→LUMO+6	0.21

Table S7. Electronic transitions of dmp-pop calculated at the T₁ optimized geometry by TDDFT in Solvent DCM (CI > 0.20)

S_n/T_m	Excitation energies (eV)	λ (nm)	Energy (cm⁻¹)	<i>f</i>	Main configurations	
T₁	1.99	623.3	16044	0.0000	HOMO→LUMO	0.68
S₁	2.22	559.1	17886	0.0348	HOMO→LUMO	0.68
S₂	2.81	442.0	22625	0.0100	HOMO→LUMO+1	0.69
S₃	2.90	427.0	23420	0.0611	HOMO-1→LUMO	0.67
S₄	3.13	396.8	25202	0.0067	HOMO-2→LUMO	0.68
S₅	3.51	353.1	28321	0.0064	HOMO-1→LUMO+1	0.68
S₆	3.56	348.5	28695	0.0178	HOMO-4→LUMO	0.46
					HOMO-5→LUMO	-0.38
S₇	3.61	343.4	29121	0.0094	HOMO-5→LUMO	0.38
					HOMO-6→LUMO	-0.32
					HOMO-4→LUMO	0.29
					HOMO-10→LUMO	0.20
S₈	3.64	340.5	29369	0.0057	HOMO→LUMO+2	0.68
S₉	3.68	336.7	29701	0.0002	HOMO-3→LUMO	0.64
S₁₀	3.72	332.9	30040	0.0567	HOMO→LUMO+3	0.69

Table S8. Electronic transitions of dbp-pop calculated at the T₁ optimized geometry by TDDFT in Solvent DCM (CI > 0.20)

S _n /T _m	Excitation energies (eV)	λ (nm)	Energy (cm ⁻¹)	<i>f</i>	Main configurations	
T ₁	2.04	607.0	16475	0.0000	HOMO-1→LUMO	0.46
					HOMO→LUMO	0.40
					HOMO-2→LUMO	0.26
T ₂	2.79	444.7	22488	0.0000	HOMO→LUMO+1	0.54
					HOMO-1→LUMO+1	0.35
S ₁	3.03	408.6	24475	0.0285	HOMO→LUMO	0.67
S ₂	3.22	385.2	25961	0.0037	HOMO-3→LUMO	0.58
					HOMO-1→LUMO	0.37
S ₃	3.25	381.7	26199	0.0678	HOMO→LUMO+1	0.68
S ₄	3.31	374.8	26682	0.0028	HOMO-2→LUMO	0.53
					HOMO-3→LUMO	0.27
					HOMO-1→LUMO	-0.23
					HOMO-3→LUMO+1	0.22
S ₅	3.36	369.1	27094	0.0005	HOMO-3→LUMO+1	0.54
					HOMO-1→LUMO+1	0.34
					HOMO-2→LUMO	-0.23
S ₆	3.49	355.4	28138	0.0105	HOMO-2→LUMO+1	0.62
					HOMO-1→LUMO+1	-0.21
					HOMO-3→LUMO+1	0.20
S ₇	3.55	349.4	28621	0.2504	HOMO-1→LUMO	0.50
					HOMO-2→LUMO	0.34
					HOMO-3→LUMO	-0.27
S ₈	3.64	340.9	29335	0.0014	HOMO-1→LUMO+1	0.53
					HOMO-3→LUMO+1	-0.31
					HOMO-2→LUMO+1	0.28
S ₉	3.97	312.6	31991	0.0000	HOMO-4→LUMO	0.70
S ₁₀	3.97	311.9	32063	0.0059	HOMO→LUMO+2	0.68

Table S9. SOC matrix elements $\langle T_m | H_{\text{SOC}} | S_n \rangle$ (cm⁻¹), the oscillator strengths f_n of [Cu(phen)(POP)]⁺ calculated at the T₁^{opt} optimized geometry with solvent CH₂Cl₂ included. The radiative decay rate constant (k_r /s⁻¹) is also given.

E(T₁)=13398cm⁻¹

	SOC	E(S _n)	f(S _n)	k _n
S ₃	-56.8445	24433	0.0530	266.4958

S ₄	-7.3201	25433	0.0352	2.3705
S ₉	21.8027	28978	0.0117	3.6607
S ₁₀	-55.9326	29387	0.0158	30.4613
k _r				302.9883

Table S10. SOC matrix elements $\langle T_m | H_{SOC} | S_n \rangle$ (cm⁻¹), the oscillator strengths f_n of [Cu(dmp)(POP)]⁺ calculated at the T₁^{opt} optimized geometry with solvent CH₂Cl₂ included. The radiative decay rate constant (k_r/s^{-1}) is also given.

E(T₁)=16044cm⁻¹

	SOC	E(S _n)	f(S _n)	k _n
S ₃	15.9641	23420	0.0611	97.1573
S ₄	-49.4996	25202	0.0067	61.7474
S ₆	-5.1149	28695	0.0178	0.8062
S ₇	-38.6065	29121	0.0094	22.3668
S ₉	-16.2070	29701	0.0002	182.0776
k _r				364.1553

Table S11. SOC matrix elements $\langle T_m | H_{SOC} | S_n \rangle$ (cm⁻¹), the oscillator strengths f_n of [Cu(dbp)(POP)]⁺ calculated at the T₁^{opt} optimized geometry with solvent CH₂Cl₂ included. The radiative decay rate constant (k_r/s^{-1}) is also given.

E(T₁)=16475cm⁻¹

	SOC	E(S _n)	f(S _n)	k _n
S ₁	-65.0599	24475	0.0285	662.9552
S ₂	35.6793	25961	0.0037	17.3565
S ₄	36.9834	26682	0.0028	11.8596
S ₅	-12.1697	27094	0.0005	0.2293
S ₇	14.2107	28621	0.2504	103.0932
k _r				795.4939

E(T₂)=22488cm⁻¹

	SOC	E(S _n)	f(S _n)	k _n
S ₃	-28.9404	26199	0.0678	93.8309
S ₄	18.8879	26682	0.0028	1.6207
S ₅	63.0372	27094	0.0005	3.1745
S ₆	51.1593	28138	0.0105	43.9091
S ₈	19.3804	29335	0.0014	0.7760
k _r				143.3112

Table S12. Electronic transitions of dmp-pph3 calculated at the T₁ optimized geometry by TDDFT in Solvent DCM (CI > 0.20)

S _n /T _m	Excitation energies (eV)	λ (nm)	Energy (cm ⁻¹)	<i>f</i>	Main configurations	
T ₁	2.06	603.1	16582	0.0000	HOMO-1→LUMO	0.56
					HOMO→LUMO	-0.40
T ₂	2.78	446.0	22422	0.0000	HOMO→LUMO+1	0.52
					HOMO-1→LUMO+1	-0.45
S ₁	3.07	403.7	24772	0.0281	HOMO→LUMO	0.68
S ₂	3.25	380.9	26254	0.0728	HOMO→LUMO+1	0.70
S ₃	3.30	375.6	26625	0.0012	HOMO-3→LUMO	0.70
S ₄	3.33	371.9	26890	0.0000	HOMO-2→LUMO	0.68
S ₅	3.41	363.1	27541	0.0011	HOMO-3→LUMO+1	0.68
S ₆	3.50	354.7	28194	0.0112	HOMO-2→LUMO+1	0.69
S ₇	3.57	347.3	28794	0.2169	HOMO-1→LUMO	0.66
S ₈	3.63	341.7	29266	0.0004	HOMO-1→LUMO+1	0.68
S ₉	4.16	298.4	33513	0.0536	HOMO→LUMO+2	0.68
S ₁₀	4.19	295.9	33796	0.0029	HOMO-8→LUMO	0.52
					HOMO-12→LUMO	0.25
					HOMO-14→LUMO	0.25
					HOMO-13→LUMO	0.20

Table S13. SOC matrix elements $\langle T_m | H_{\text{SOC}} | S_n \rangle (\text{cm}^{-1})$, the oscillator strengths f_n of $[\text{Cu}(\text{dmp})(\text{PPh}_3)_2]^+$ calculated at the T_1^{opt} optimized geometry with solvent CH_2Cl_2 included. The radiative decay rate constant (k_r/s^{-1}) is also given.

E(T_1)=16582cm ⁻¹				
	SOC	E(S_n)	f(S_n)	k_n
S_1	0.5513	24772	0.0281	0.0451
S_3	-22.9633	26625	0.0012	2.0682
S_7	0.4452	28794	0.2169	0.0879
S_{10}	27.2302	33796	0.0029	1.8846
k_r				4.0858
E(T_2)=22422cm ⁻¹				
	SOC	E(S_n)	f(S_n)	k_n
S_2	0.5513	26254	0.0728	1.2454
S_5	-22.9633	27541	0.0011	17.4411
S_6	0.4452	28194	0.0112	0.0513
S_8	27.2302	29266	0.0004	4.6950
k_r				23.4328