## **Electronic Supplementary Information**

Synthesis and structure of a ferric complex of 2,6-di(1H-pyrazol-3-yl)pyridine and its excellent performance in the redox-controlled living ring-opening polymerization of ε-caprolactone

Yang-Yang Fang,<sup>a</sup> Wei-Jie Gong,<sup>a</sup> Xiu-Juan Shang,<sup>a</sup> Hong-Xi Li,<sup>a,\*</sup> Jun Gao,<sup>a</sup> Jian-Ping Lang<sup>a,b\*</sup>

<sup>a</sup> College of Chemistry, Chemical Engineering and Material Science, Soochow University, Suzhou

215123, People's Republic of China

<sup>b</sup> State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai 210032, People's Republic of China

## **Table of Contents**

<b>Fig. S1</b> . Variation of $\chi_M T$ with temperature for 1 and 2
Fig. S2. Cyclic voltammogram of complex 1. Conditions: 0.005 M acetonitrile solution of the
analyte, 0.1M Et <sub>4</sub> NClO <sub>4</sub> as supporting electrolyte, platinum electrode as working electrode,
platinum wire as counter electrode, scan rate of 50 mV/sS3
Fig. S3. The <sup>1</sup> H NMR spectrum of PCL initiated by $1$ /isopropanol with the molar ratio of $1/$
isopropanol /CL being 1/2/300 (Table 2, entry 5)S4
Fig. S4. The <sup>1</sup> H NMR spectrum of PCL initiated by $1/$ benzyl alcohol with the molar ratio of $1/$
benzyl alcohol/CL being 1/2/300 (Table 2, entry 7)S4
Fig. S5. Kinetic plot for the polymerization of $\epsilon$ -CL initiated by 1/isopropanol without switching
at 100 °CS5
Fig. S6. Comparison of the experimental and theoretical molecular weights with two initiating
alkoxides per iron centerS5
Fig. S7. Kinetic plot for the polymerization of $\epsilon$ -CL initiated by 1/isopropanol with switching at
100 °CS6
Fig. S8. PXRD patterns for 1. Red line: simulated; Black line: a single-phase polycrystalline
sample of 1S7
Fig. S9. PXRD patterns for 2. Red line: simulated; Black line: a single-phase polycrystalline
sample of 2S7



**Fig. S1.** Variation of  $\chi_M T$  with temperature for **1** (a) and **2** (b).



**Fig. S2.** Cyclic voltammogram of complex **1**. Conditions: 0.005 M acetonitrile solution of the analyte, 0.1M Et<sub>4</sub>NClO<sub>4</sub> as supporting electrolyte, platinum electrode as working electrode, platinum wire as counter electrode, scan rate of 50 mV/s.



**Fig. S3.** The <sup>1</sup>H NMR spectrum of PCL initiated by 1/isopropanol with the molar ratio of 1/ isopropanol /CL being 1/2/300 (Table 2, entry 5).



**Fig. S4.** The <sup>1</sup>H NMR spectrum of PCL initiated by 1/benzyl alcohol with the molar ratio of 1/benzyl alcohol/CL being 1/2/300 (Table 2, entry 7).



Fig. S5. Kinetic plot for the polymerization of  $\epsilon$ -CL initiated by 1/isopropanol without switching at 100 °C.



**Fig. S6.** Comparison of the experimental and theoretical molecular weights with two initiating alkoxides per iron center.



Fig. S7. Kinetic plot for the polymerization of  $\epsilon$ -CL initiated by 1/isopropanol with switching at 100 °C.



Fig. S8. PXRD patterns for 1. Red line: simulated; Black line: a single-phase polycrystalline sample of 1.



Fig. S9. PXRD patterns for 2. Red line: simulated; Black line: a single-phase polycrystalline sample of 2.