

Supporting Information

Synthesis and Characterization of Germa[*n*]pericyclines

Hiroki Tanimoto,^a Tomohiko Nagao,^a Yasuhiro Nishiyama,^a Tsumoru Morimoto,^a Fumiyasu Iseda,^a
Yuko Nagato,^a Toshimasa Suzuka,^b Ken Tsutsumi,^{*c} and Kiyomi Kakiuchi^{*a}

^a Graduate School of Materials Science, Nara Institute of Science and Technology (NAIST), 8916-5
Takayamacho, Ikoma, Nara 630-0192, Japan

^b Department of Chemistry, University of the Ryukyus, Nishihara, Okinawa 903-0213, Japan.

^c Department of Applied Chemistry, Faculty of Engineering, Miyazaki University, 1-1 Gakuen
Kibanadai-Nishi, Miyazaki 889-2155, Japan.

tsutsumi@cc.miyazaki-u.ac.jp; kakiuchi@ms.naist.jp

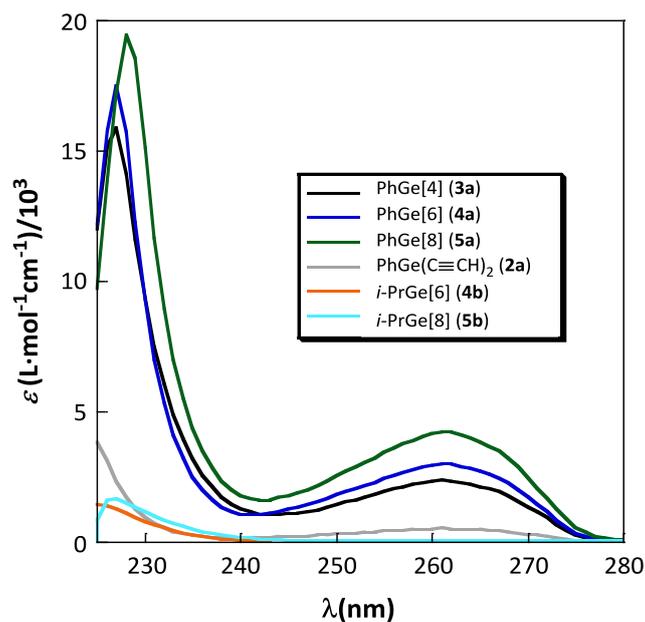
General Information	S1
UV-visible and fluorescence emission spectra	S2
Cyclic voltammetry and differential pulse voltammetry	S4
¹ H and ¹³ C NMR spectra	S5

General information

^1H and ^{13}C NMR spectra were recorded using a Jeol JNM-ECP500 spectrometer (500 MHz for ^1H NMR and 125 MHz for ^{13}C NMR). Chemical shifts are reported as δ values in ppm and calibrated with respect to the residual solvent peak (CDCl_3 , δ 7.26 for ^1H NMR and δ 77.00 for ^{13}C NMR) or tetramethylsilane (δ 0 for ^1H NMR). The abbreviations used are as follows: s (singlet), d (doublet), t (triplet), q (quartet), sept (septet), br (broad peak), and m (complex multiplet). Melting points were measured using a Yanaco Micro melting point apparatus. Infrared spectra were measured using a Jasco FT-IR-4200 spectrometer. Mass spectra were recorded using a Jeol JMS-700 MStation [EI (70 eV), CI, FAB, and ESI]. X-ray diffraction (XRD) analyses were performed using a Rigaku R-AXIS RAPID/S imaging plate diffractometer.

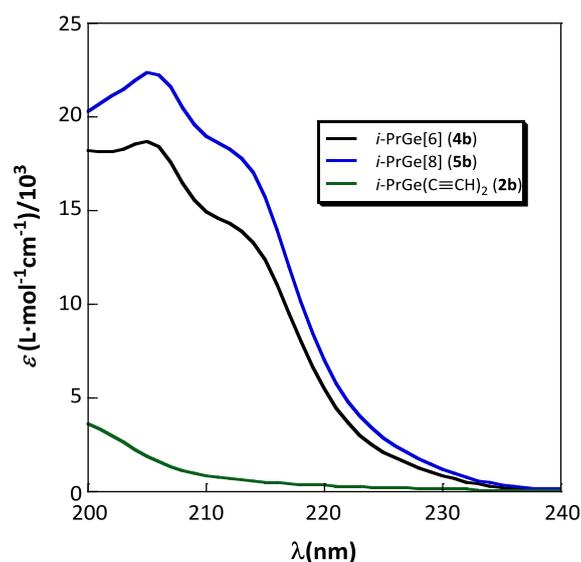
UV/visible spectra were recorded using Jasco V-630. Fluorescence spectra were collected using Jasco FP-6500. Raman spectra were obtained using a Jasco laser Raman spectrophotometer, NRS-2100. The cyclic voltammetry measurements of the compounds were performed using a BAS electrochemical analyser ALS612D in dichloromethane containing *n*-Bu₄NPF₆ as the supporting electrolyte at 298 K (100 mV s⁻¹). The glassy carbon working electrode was polished using BAS polishing alumina suspension and rinsed with water before use. The counter electrode was a platinum wire. The measured potentials were recorded with respect to Ag/AgNO₃ and normalized with respect to Fc/Fc⁺. Flash column chromatography was performed using Merck Silica gel 60. The progress of the reactions was monitored by silica gel thin layer chromatography (TLC) (Merck TLC Silica gel 60 F₂₅₄). The purification of the mixture of germapericyclines was performed using a LC-908 recycling preparative high-performance liquid chromatography (HPLC) equipped with a JAIGEL 2H-40 column made by Japan Analytical Industry Co., Ltd. Ethanol solutions of phosphomolybdic acid and anisaldehyde-acetic acid-sulfuric acid were used as the TLC stains. All the reagents were purchased from Sigma-Aldrich, Wako Pure Chemical Industries, Ltd, TCI (Tokyo Chemical Industry, Co. Ltd), Kanto Chemical Co. Inc., and Nakalai Tesque. Anhydrous tetrahydrofuran (THF) was purchased from Kanto Chemical. Density Functional Theory (DFT) calculations were performed using the Gaussian09, and the geometries of the molecules were optimized by employing the B3LYP density functionals and the 6-311G* basis set in this series of calculations.

UV-visible and fluorescence emission spectra



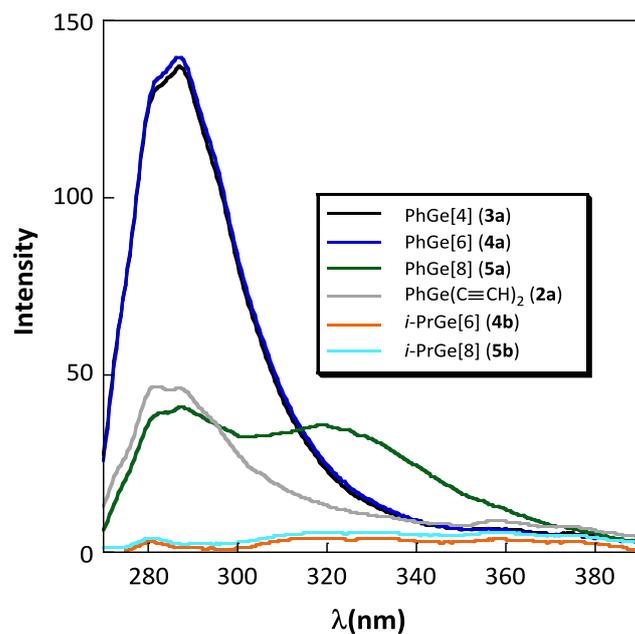
Compounds	Absorption maxima / nm (ϵ)
2a	225 (3840), 261 (531)
3a	227 (15900), 261 (2375)
4a	227 (17500), 261 (3010)
5a	228 (19440), 261 (4240)
4b	225 (1446)
5b	227 (1648)

Fig S1. UV-visible spectra of germa[*n*]pericyclines (0.1 mM in CH₂Cl₂)



Compounds	Absorption maxima / nm (ϵ)
2b	—
4b	205 (18683), 213 (13919)
5b	205 (22377), 213 (17798)

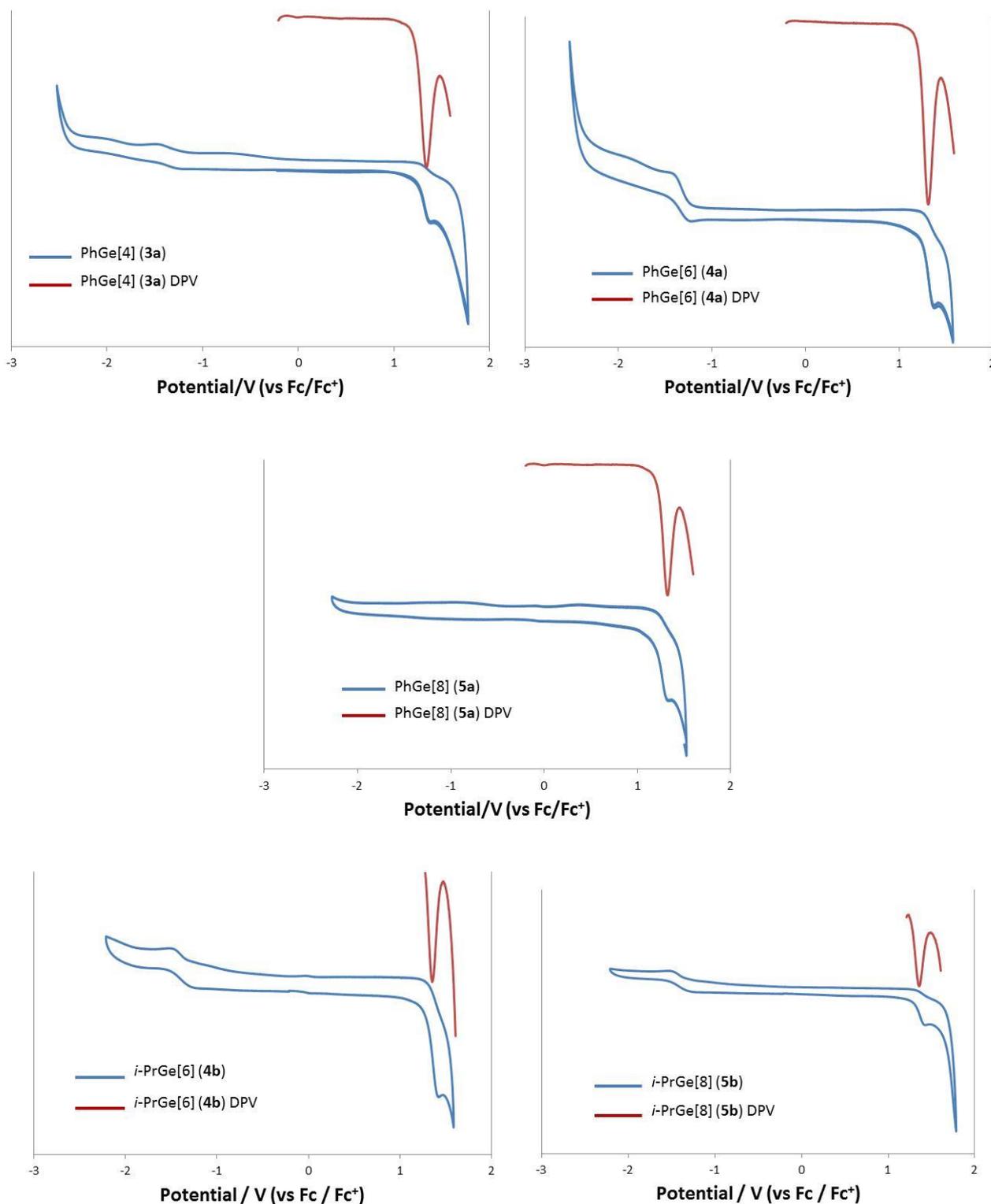
Fig S2. UV-visible spectra of isopropylgerma[*n*]pericyclines (0.1 mM in hexane)



Compounds	Emission maxima / nm
2a	282, 288
3a	281, 288
4a	281, 288
5a	281, 288, 318
4b	—
5b	—

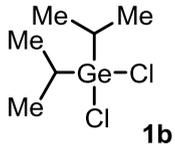
Fig S3. Fluorescence emission spectra of germa[*n*]pericyclines (0.1 mM in CH₂Cl₂, λ_{EXT} = 260 nm)

Cyclic voltammetry (CV) and differential pulse voltammetry (DPV)

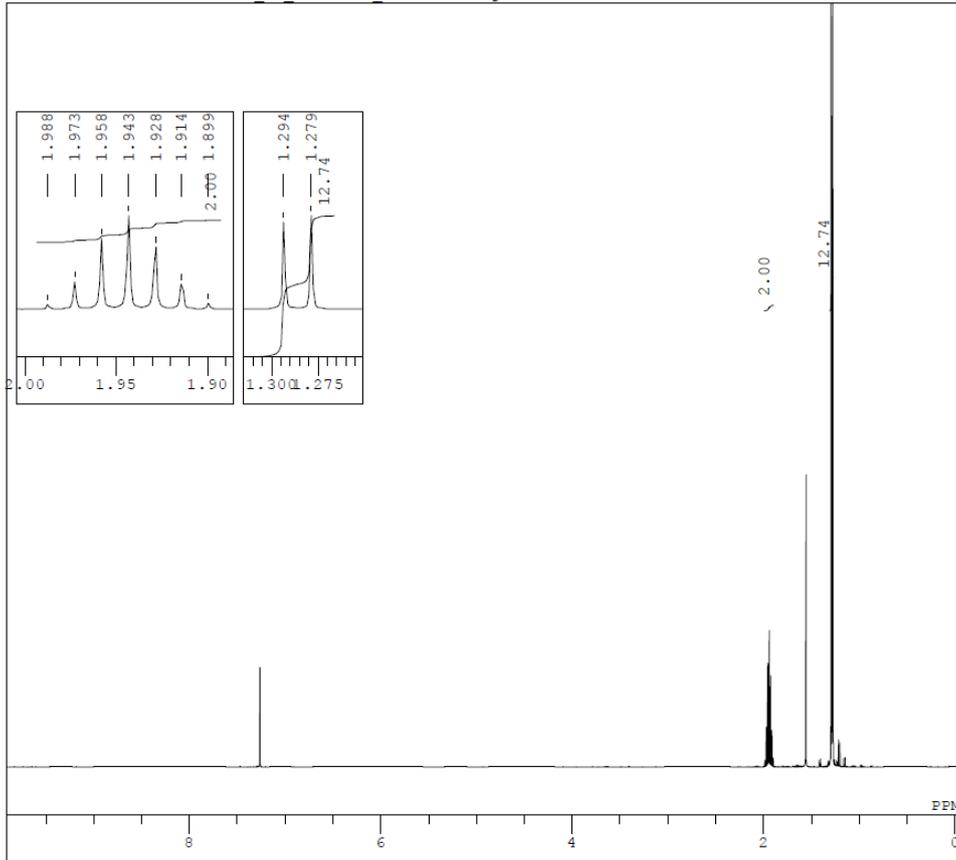


Compounds	Oxidation potential from DPV / V
3a	1.336
4a	1.312
5a	1.320
4b	1.352
5b	1.356

Fig S3. Cyclic and differential pulse voltammograms of germa[*n*]pericyclines (1.0 mM (3a, 4a, 4b, and 5b); 0.5 mM (5a) in 0.1 M *n*-Bu₄NPF₆/CH₂Cl₂ solution; Scan rate = 0.1 V/s)



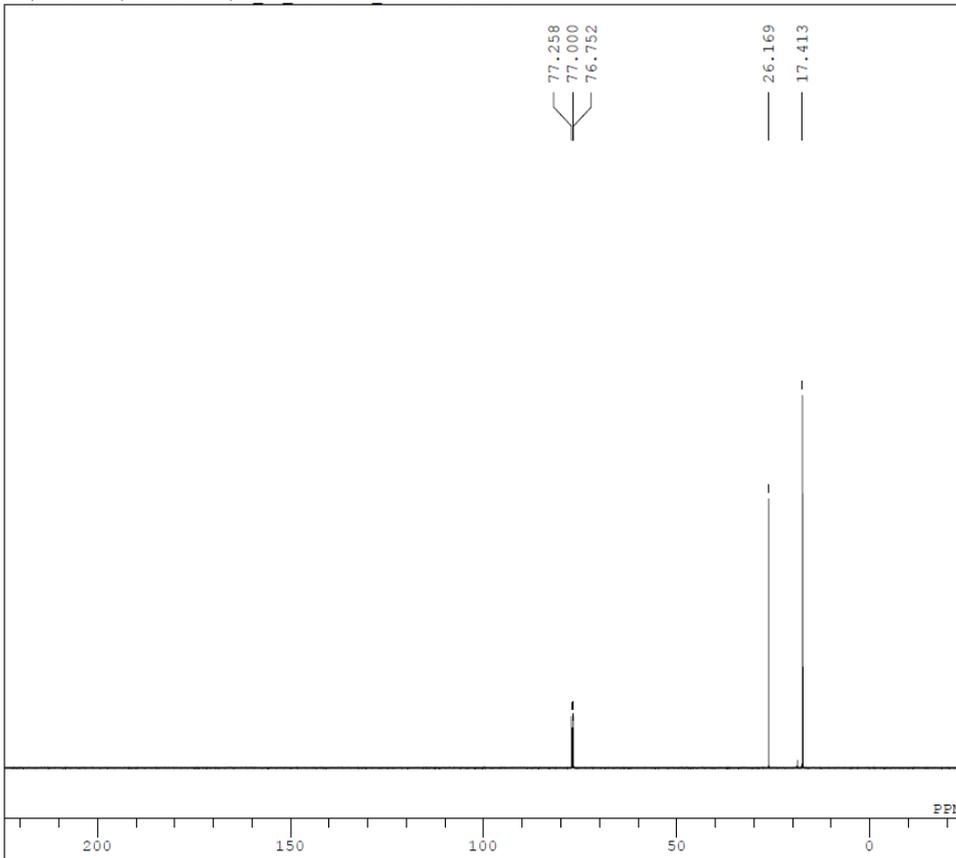
Z:\NMR-data\no.23 data\tn_01_023 data Proton-1-1.jdf



```

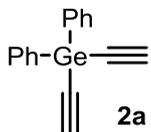
DFILE  tn_01_023 data_Proton-1-1.jdf
COMNT  single_pulse
DATIM  2013-09-21 11:29:39
OBNUC  1H
EXMOD  proton.jxp
OBFRQ  500.16 MHz
OBSET  2.41 KHz
OBFIN  6.01 Hz
POINT  16384
FREQU  9384.38 Hz
SCANS  8
ACQTM  1.7459 sec
PD      5.0000 sec
PW1     6.22 usec
IRNUC  1H
CTEMP  18.8 c
SLVNT  CDCL3
EXREF  7.26 ppm
BF      0.10 Hz
RGAIN  44
  
```

Z:\NMR-data\no.23 data\tn_01_023 data Carbon-1-1.als

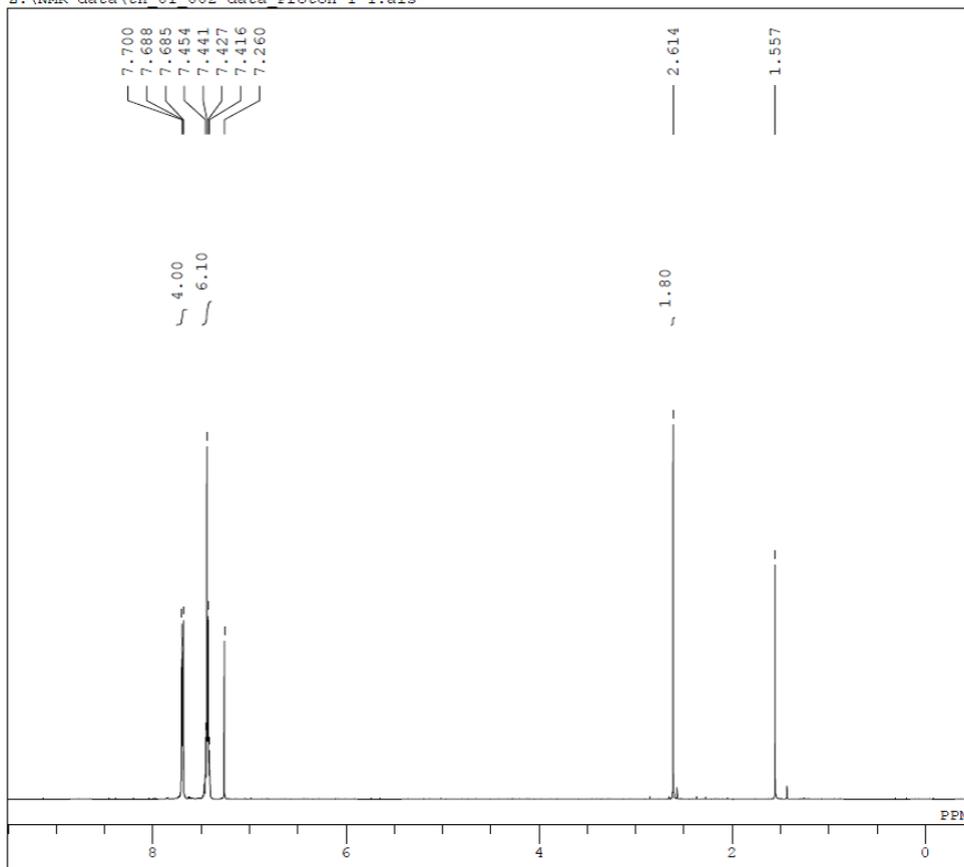


```

DFILE  tn_01_023 data_Carbon-1-1.als
COMNT  single pulse decoupled gated NOE
DATIM  2013-09-21 11:34:54
OBNUC  13C
EXMOD  carbon.jxp
OBFRQ  125.77 MHz
OBSET  7.87 KHz
OBFIN  4.21 Hz
POINT  26214
FREQU  31446.54 Hz
SCANS  186
ACQTM  0.8336 sec
PD      2.0000 sec
PW1     3.12 usec
IRNUC  1H
CTEMP  19.0 c
SLVNT  CDCL3
EXREF  77.00 ppm
BF      0.10 Hz
RGAIN  58
  
```

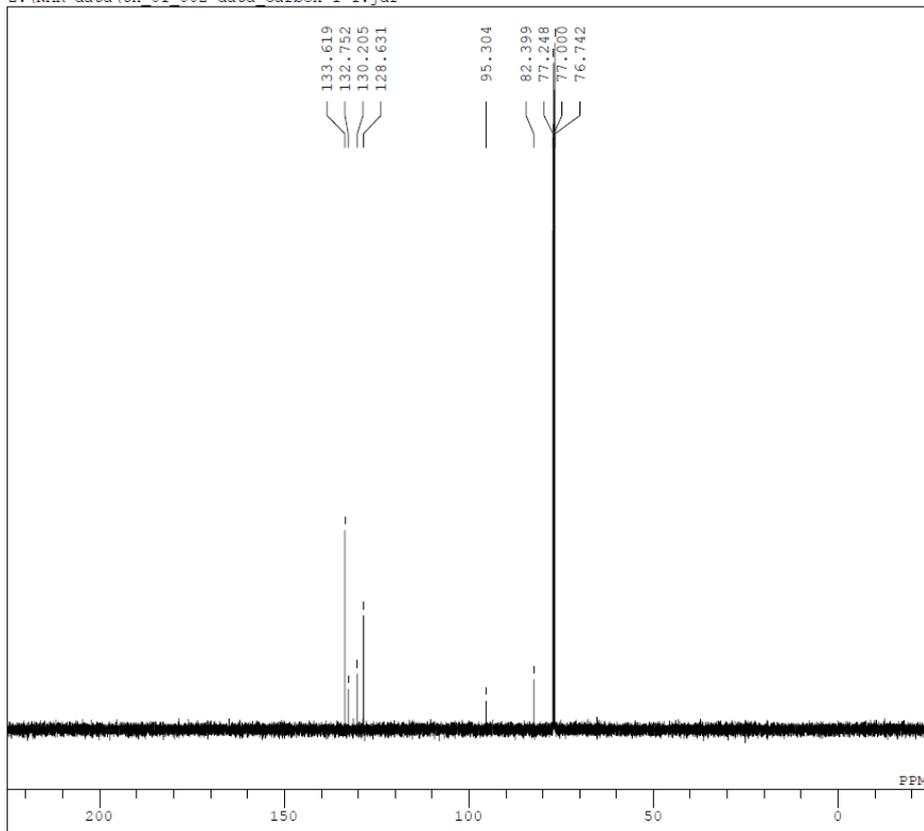


2:\NMR-data\tn_01_002 data Proton-1-1.als

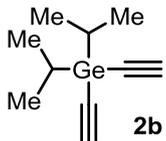


DFILE tn_01_002 data_Proton-1-1.als
 COMNT single pulse
 DATIM 2013-06-04 10:51:06
 OBNUC 1H
 EXMOD proton.jxp
 OBFREQ 500.16 MHz
 OBSET 2.41 KHz
 OBFIN 6.01 Hz
 POINT 13107
 FREQU 7507.51 Hz
 SCANS 8
 ACQTM 1.7459 sec
 PD 5.0000 sec
 FW1 6.22 usec
 IRNUC 1H
 CTEMP 19.6 c
 SLVNT CDCL3
 EXREF 7.26 ppm
 BF 0.10 Hz
 RGAIN 44

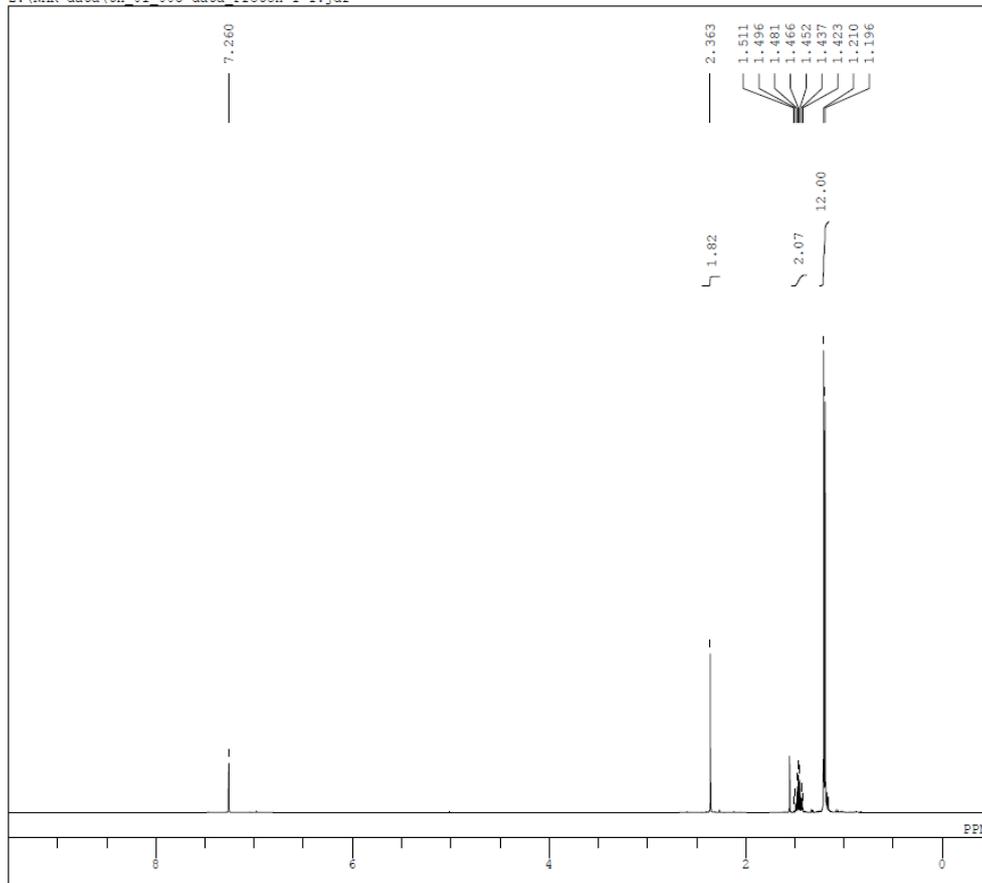
2:\NMR-data\tn_01_002 data Carbon-1-1.jdf



DFILE tn_01_002 data_Carbon-1-1.jdf
 COMNT single pulse decoupled gated NOE
 DATIM 2013-06-04 10:54:54
 OBNUC 13C
 EXMOD carbon.jxp
 OBFREQ 125.77 MHz
 OBSET 7.87 KHz
 OBFIN 4.21 Hz
 POINT 32767
 FREQU 39308.18 Hz
 SCANS 503
 ACQTM 0.8336 sec
 PD 2.0000 sec
 FW1 3.12 usec
 IRNUC 1H
 CTEMP 20.4 c
 SLVNT CDCL3
 EXREF 77.00 ppm
 BF 0.10 Hz
 RGAIN 56



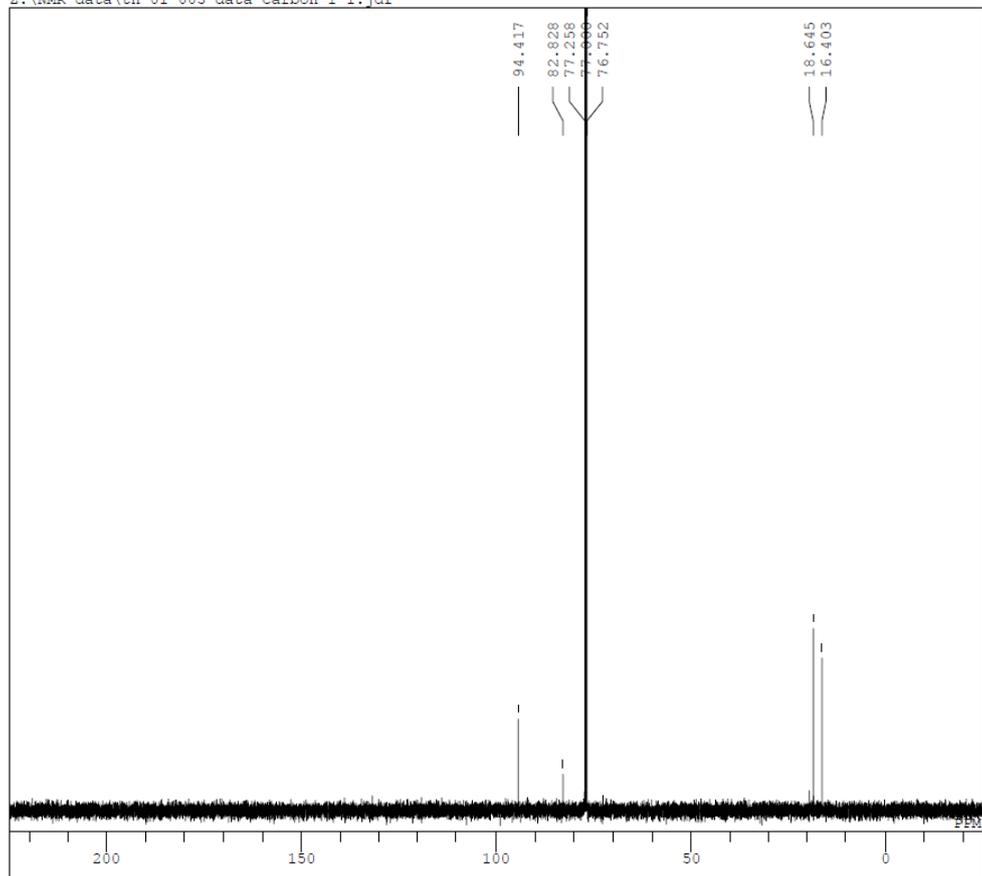
Z:\NMR-data\tn_01_003 data Proton-1-1.jdf



```

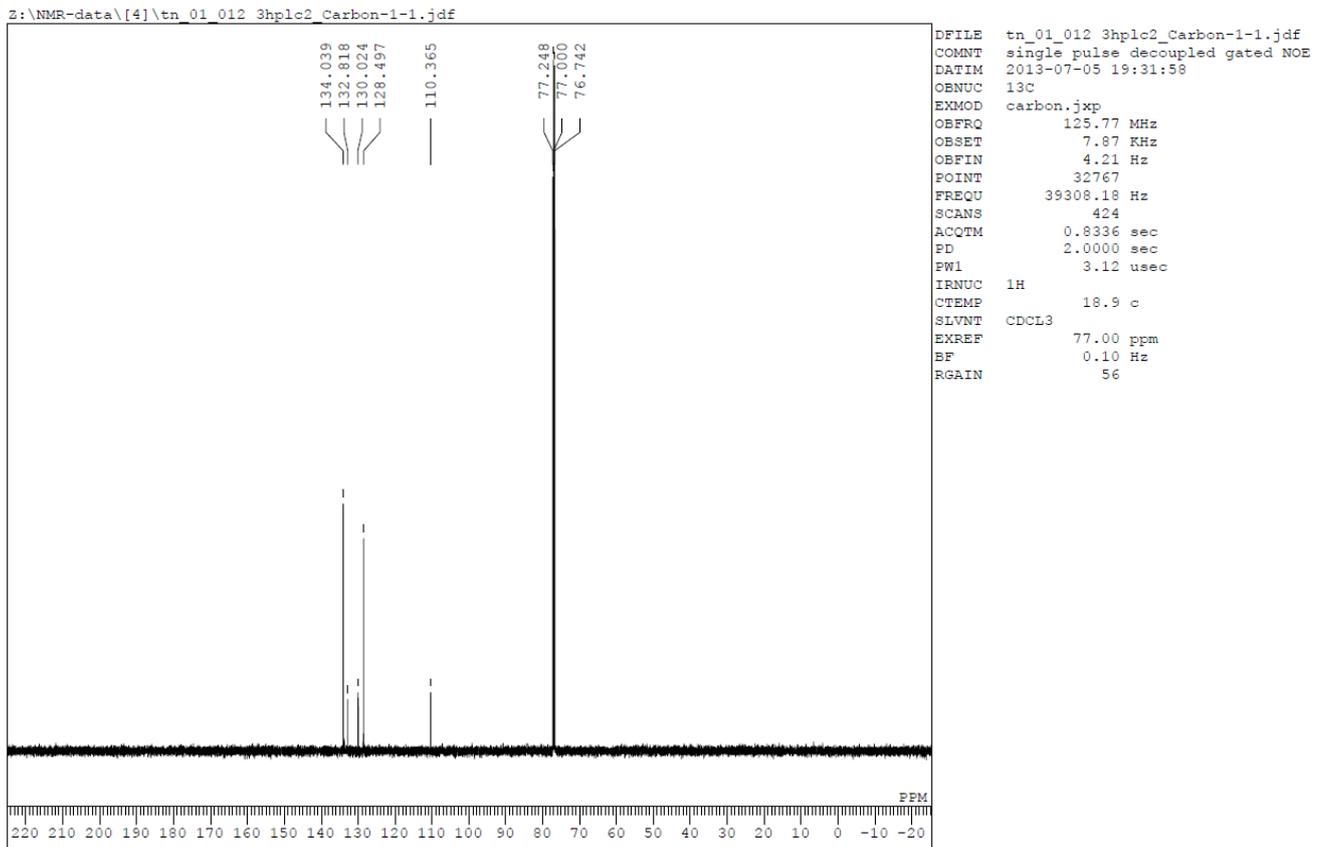
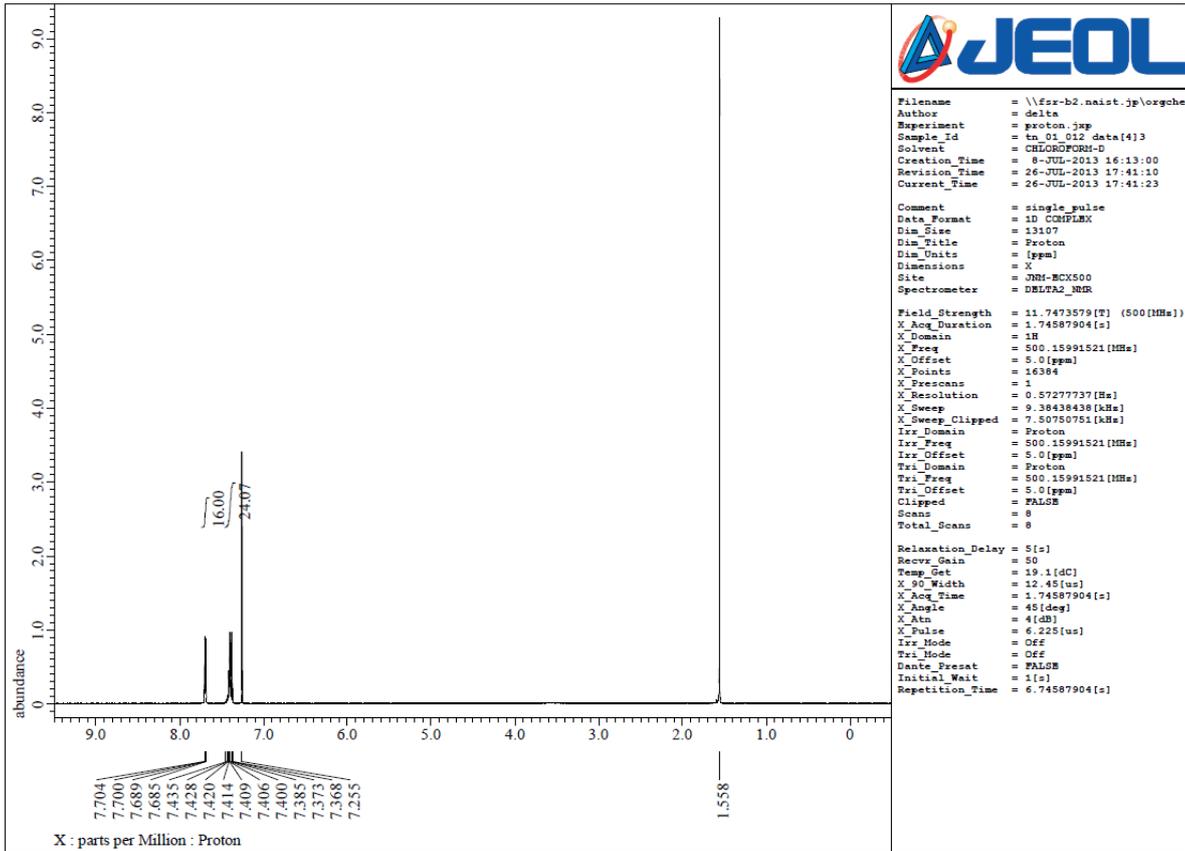
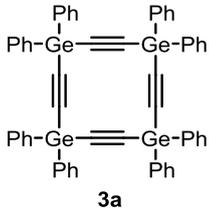
DFILE tn_01_003 data Proton-1-1.jdf
COMNT single_pulse
DATIM 2013-05-31 18:41:46
OBNUC 1H
EXMOD proton.jxp
OBFRQ 500.16 MHz
OBSEI 2.41 KHz
OBFIN 6.01 Hz
POINT 16384
FREQU 9384.38 Hz
SCANS 8
ACQTM 1.7459 sec
PD 5.0000 sec
PWL 6.22 usec
IRNUC 1H
CTEMP 19.1 c
SLVNT CDCL3
EXREF 7.26 ppm
BF 0.10 Hz
RGAIN 42
  
```

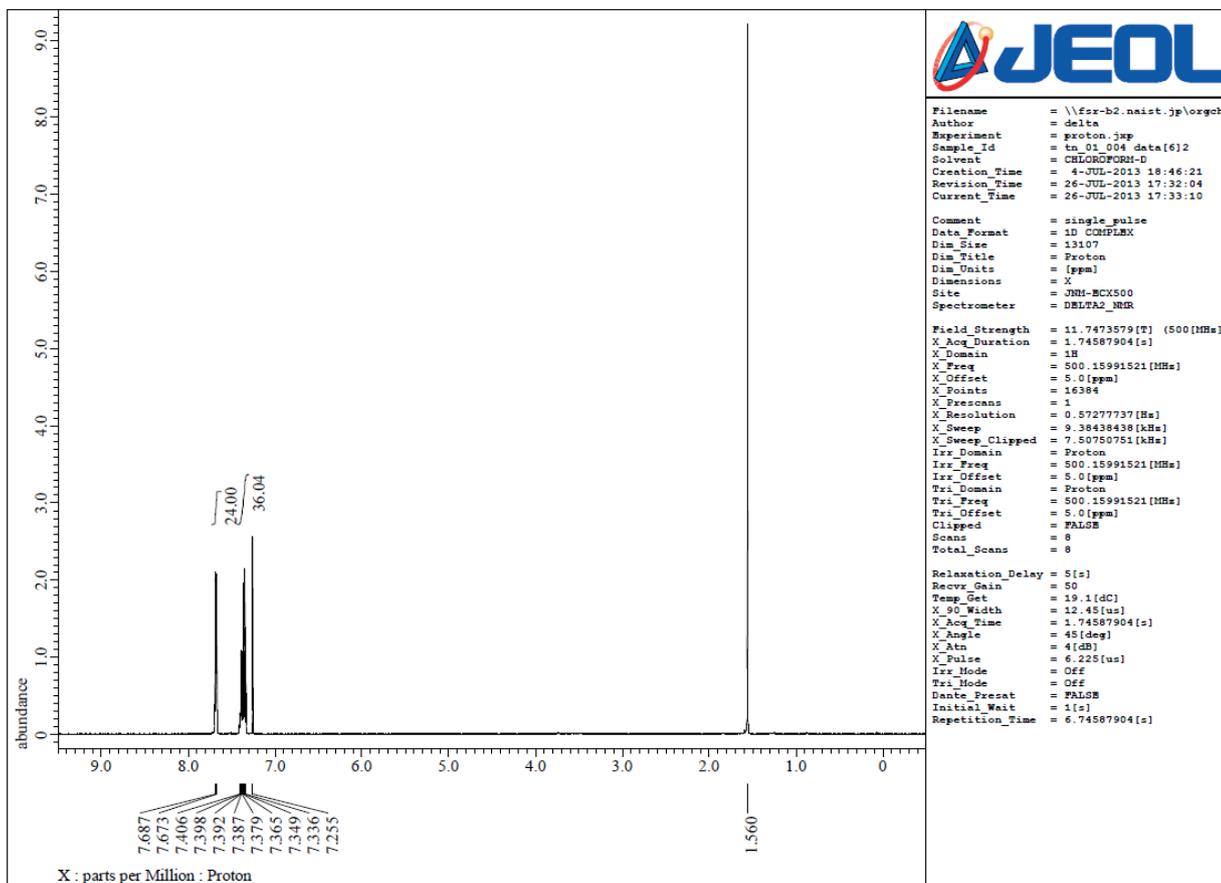
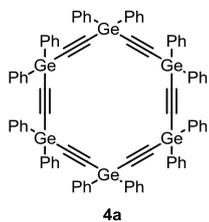
Z:\NMR-data\tn_01_003 data Carbon-1-1.jdf



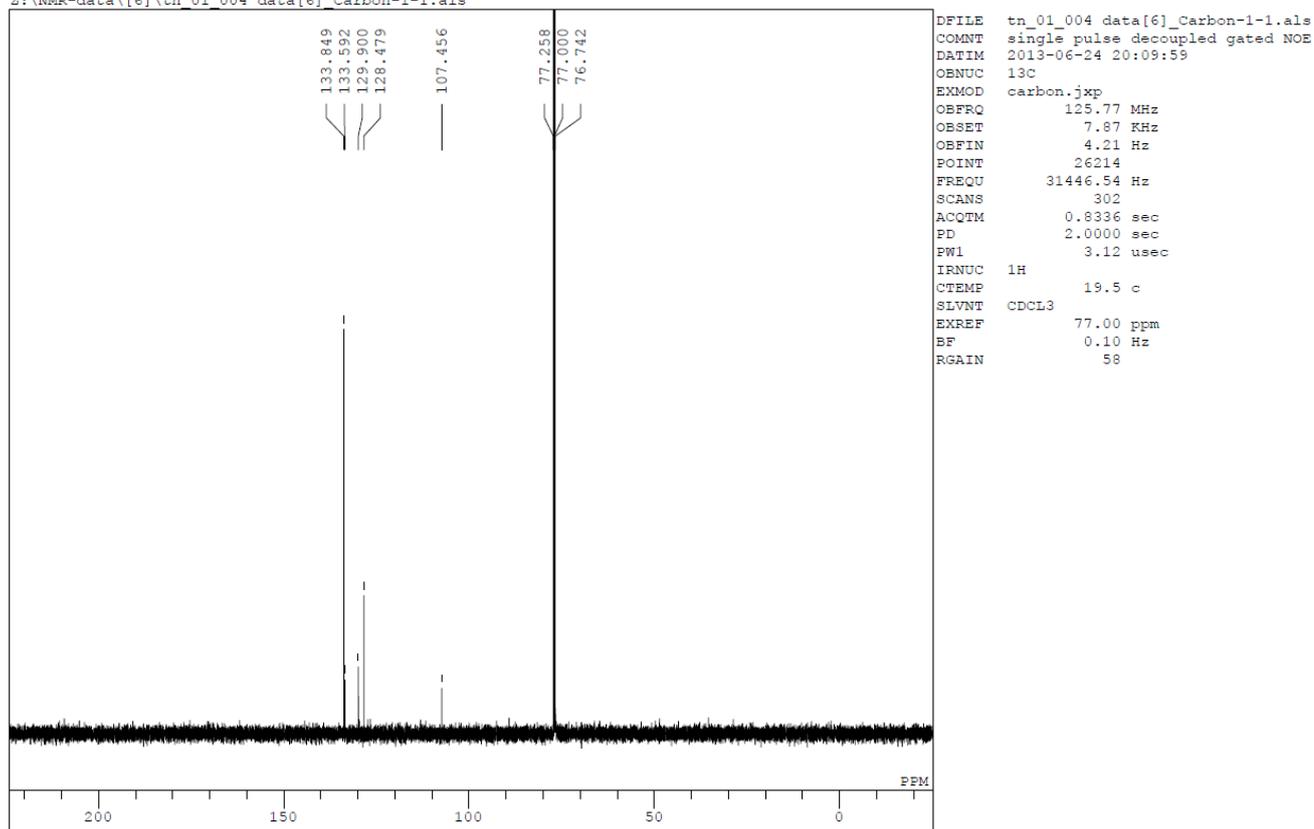
```

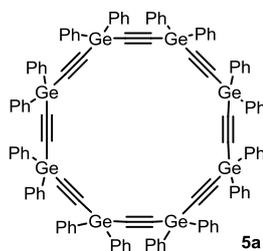
DFILE tn_01_003 data Carbon-1-1.jdf
COMNT single_pulse decoupled gated NOE
DATIM 2013-05-31 18:43:16
OBNUC 13C
EXMOD carbon.jxp
OBFRQ 125.77 MHz
OBSEI 7.87 KHz
OBFIN 4.21 Hz
POINT 32767
FREQU 39308.18 Hz
SCANS 257
ACQTM 0.8336 sec
PD 2.0000 sec
PWL 3.12 usec
IRNUC 1H
CTEMP 19.3 c
SLVNT CDCL3
EXREF 77.00 ppm
BF 0.10 Hz
RGAIN 58
  
```



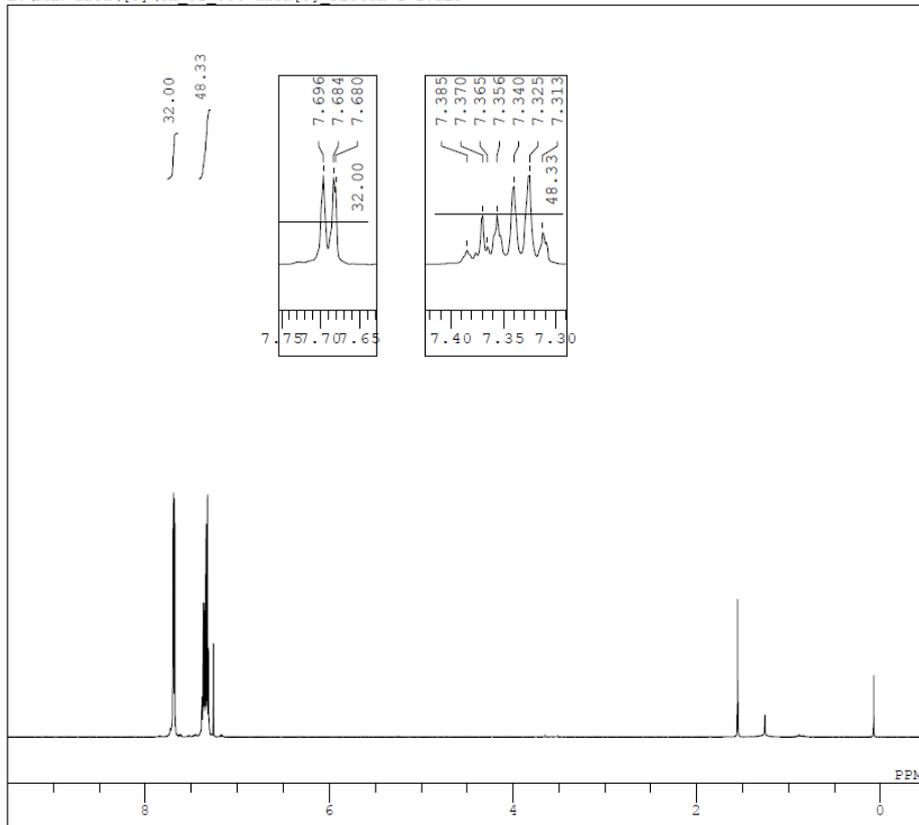


z:\NMR-data\[6]\tn_01_004_data[6] Carbon-1-1.als





Z:\NMR-data\[8]\tn_01_004 data[8] Proton-1-1.als

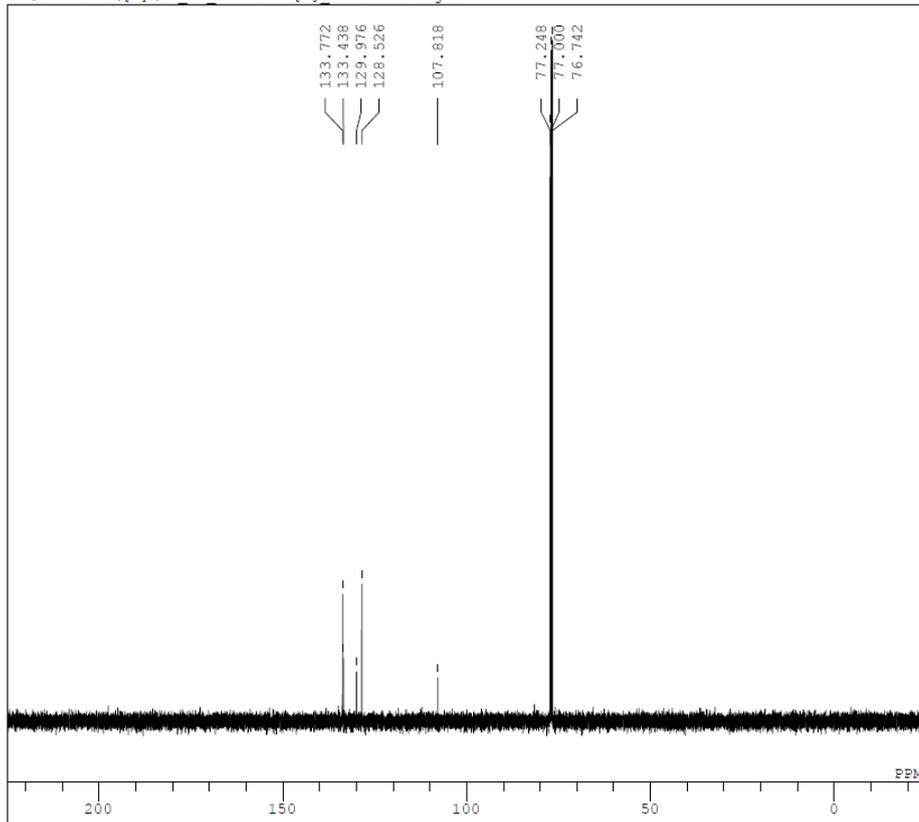


```

DFILE tn_01_004 data[8] Proton-1-1.als
COMNT single_pulse
DATIM 2013-06-24 20:33:59
OBNUC 1H
EXMOD proton.jxp
OBFRQ 500.16 MHz
OBSET 2.41 KHz
OBFIN 6.01 Hz
POINT 13107
FREQU 7507.51 Hz
SCANS 8
AQTM 1.7459 sec
PD 5.0000 sec
FW1 6.22 usec
IRNUC 1H
CTEMP 19.1 c
SLVNT CDCL3
EXREF 7.26 ppm
BF 0.10 Hz
RGAIN 46

```

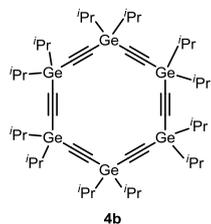
Z:\NMR-data\[8]\tn_01_004 data[8] Carbon-1-1.jdf



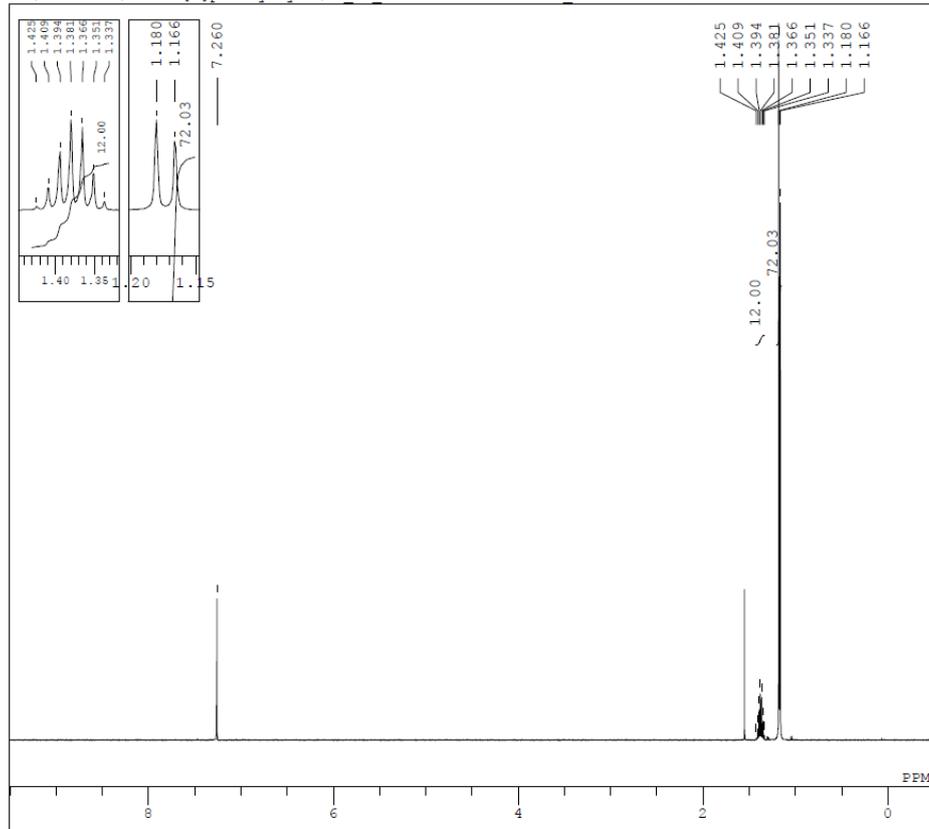
```

DFILE tn_01_004 data[8] Carbon-1-1.jdf
COMNT single_pulse decoupled gated NOE
DATIM 2013-06-24 20:35:40
OBNUC 13C
EXMOD carbon.jxp
OBFRQ 125.77 MHz
OBSET 7.87 KHz
OBFIN 4.21 Hz
POINT 32767
FREQU 39308.18 Hz
SCANS 353
AQTM 0.8336 sec
PD 2.0000 sec
FW1 3.12 usec
IRNUC 1H
CTEMP 19.0 c
SLVNT CDCL3
EXREF 77.00 ppm
BF 0.10 Hz
RGAIN 58

```



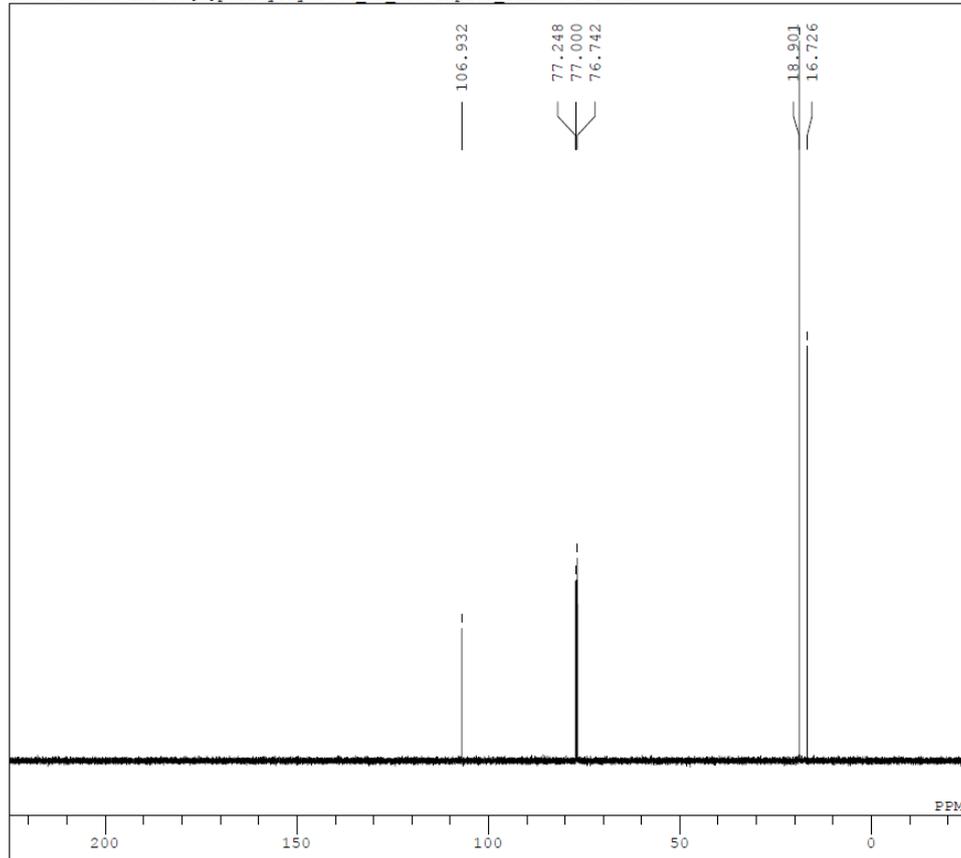
Z:\NMR-data\iPr-Ge[6]pericyclyne\tn_01_039 datakai iProGe6_Proton-1-1.als



```

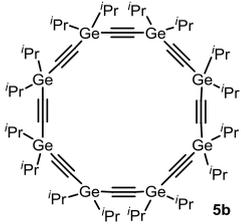
DFFILE  tn_01_039 datakai iProGe6_Proton-1-1.a
COMNT   single_pulse
DATIM   2013-11-14 19:16:04
OBNUC   1H
EXMOD   proton.jxp
OBFRQ   500.16 MHz
OBSET   2.41 KHz
OBFIN   6.01 Hz
POINT   13107
FREQU   7507.51 Hz
SCANS   8
ACQTM   1.7459 sec
PD      5.0000 sec
FWL     6.22 usec
IRNUC   1H
CTEMP   16.6 c
SLVNT   CDCL3
EXREF   7.26 ppm
BF      0.10 Hz
RGAIN   50
  
```

Z:\NMR-data\iPr-Ge[6]pericyclyne\tn_01_039 1hplc3_Carbon-1-1.als

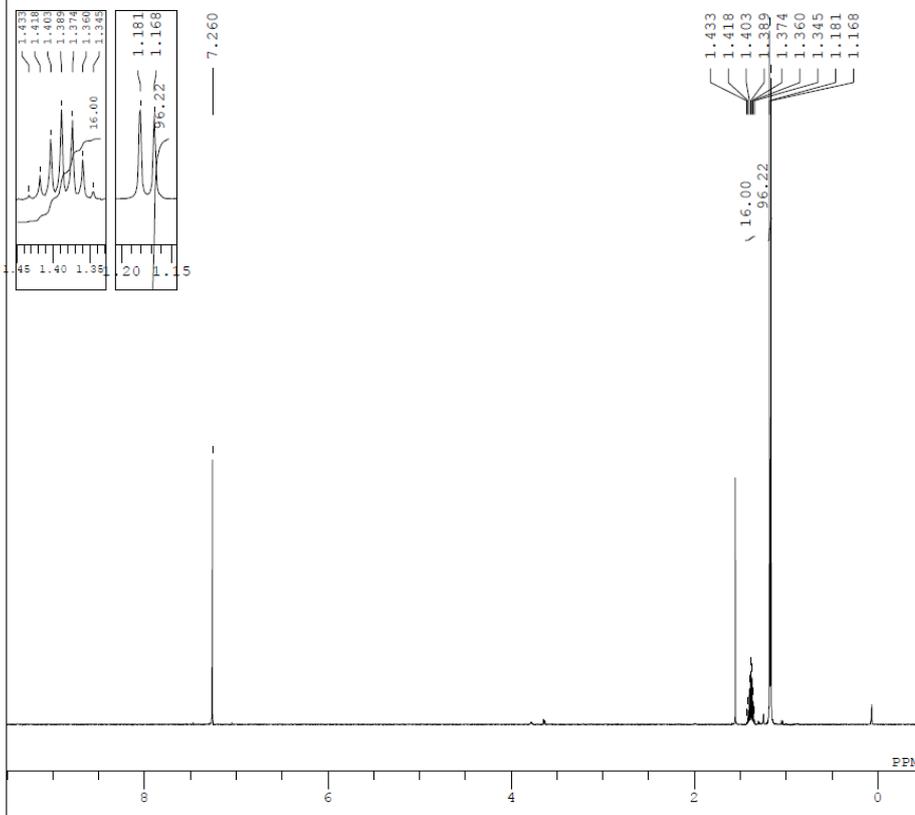


```

DFFILE  tn_01_039 1hplc3_Carbon-1-1.als
COMNT   single pulse decoupled gated NOE
DATIM   2013-11-09 04:42:20
OBNUC   13C
EXMOD   carbon.jxp
OBFRQ   125.77 MHz
OBSET   7.87 KHz
OBFIN   4.21 Hz
POINT   26214
FREQU   31446.54 Hz
SCANS   52
ACQTM   0.8336 sec
PD      2.0000 sec
FWL     3.12 usec
IRNUC   1H
CTEMP   16.3 c
SLVNT   CDCL3
EXREF   77.00 ppm
BF      0.10 Hz
RGAIN   56
  
```



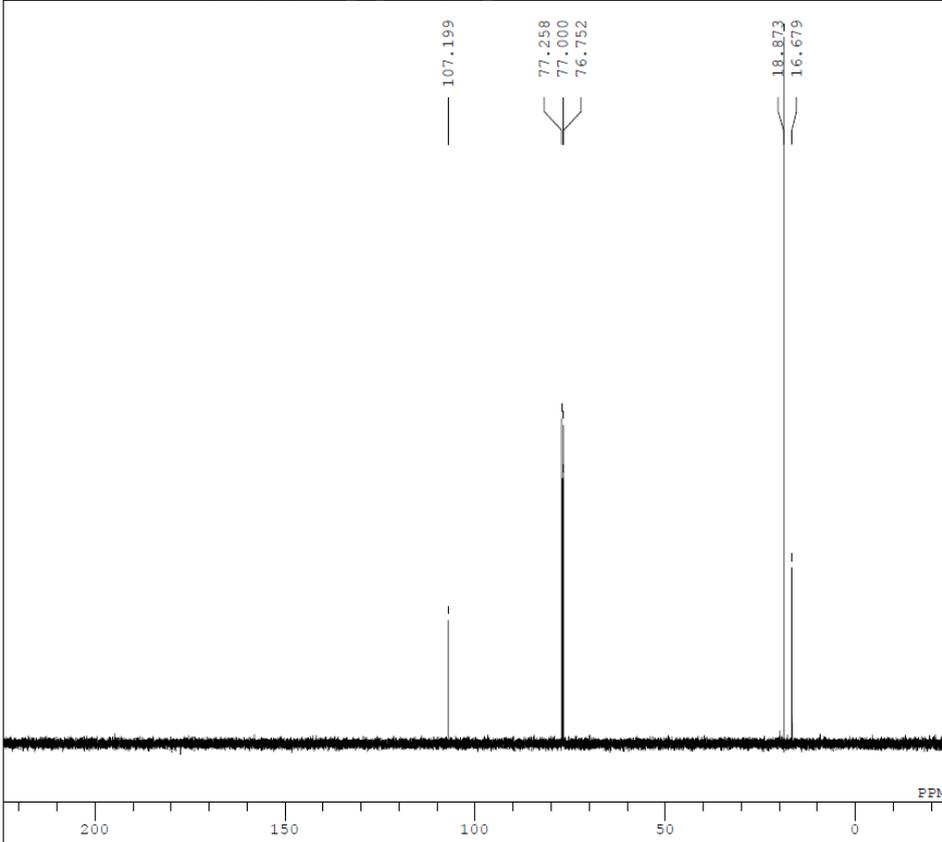
2:\NMR-data\iPr-Ge[8]pericyclyne\tn_01_039_datakai iProGe8_Proton-1-1.als



```

DFILE tn_01_039_datakai iProGe8_Proton-1-1.a
COMNT single_pulse
DATIM 2013-11-14 19:24:09
OBNUC 1H
EXMOD proton.jxp
OBFRQ 500.16 MHz
OBSET 2.41 KHz
OBFIN 6.01 Hz
POINT 13107
FREQU 7507.51 Hz
SCANS 8
ACQTM 1.7459 sec
PD 5.0000 sec
PWL 6.22 usec
IRNUC 1H
CTEMP 16.5 c
SLVNT CDCL3
EXREF 7.26 ppm
BF 0.10 Hz
RGAIN 50
  
```

2:\NMR-data\iPr-Ge[8]pericyclyne\tn_01_039_1hplc2_Carbon-1-1.als



```

DFILE tn_01_039_1hplc2_Carbon-1-1.als
COMNT single_pulse decoupled gated NOE
DATIM 2013-11-09 04:31:55
OBNUC 13C
EXMOD carbon.jxp
OBFRQ 125.77 MHz
OBSET 7.87 KHz
OBFIN 4.21 Hz
POINT 26214
FREQU 31446.54 Hz
SCANS 45
ACQTM 0.8336 sec
PD 2.0000 sec
PWL 3.12 usec
IRNUC 1H
CTEMP 16.2 c
SLVNT CDCL3
EXREF 77.00 ppm
BF 0.10 Hz
RGAIN 56
  
```