

ELECTRONIC SUPPLEMENTARY MATERIALS

Efficient hydrolytic cleavage of plasmid DNA by chloro-cobalt(II) complexes based on sterically hindered pyridyl tripod tetraamine ligands: synthesis, crystal structure and DNA cleavage activity[†]

Salah S. Massoud,^{*a} Richard S. Perkins,^a Febbee R. Louka,^a Wu Xu,^a Anne Le Roux,^a Quentin Dutercq,^a Roland C. Fischer,^b Franz A. Mautner,^c Makoto Handa,^d Yuya Hiraoka,^d Gabriel L. Kreft,^e Tiago Bortolotto,^e Hernán Terenzi,^{*e}

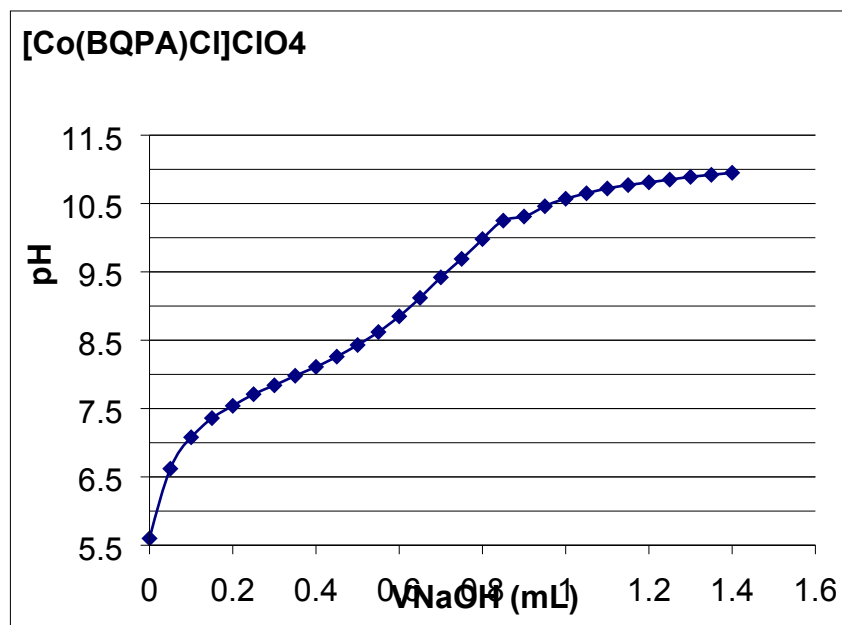


Fig. S1. Potentiometric pH titration of [Co(BQPA)(H₂O)]²⁺ (4.0×10^{-3} M) with standard 0.05 M NaOH at 37 °C .

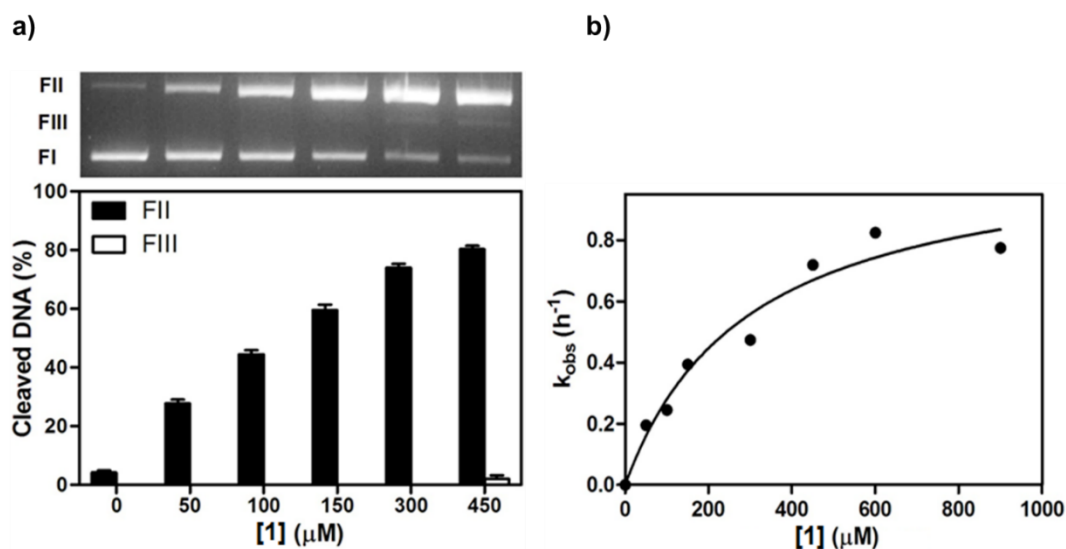


Fig. S2. (a) Agarose gel electrophoresis pattern and the corresponding plots for the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by complex **1** at different complex concentrations (50-450 μM) at pH 7.0 Tris-HCl buffer (10 mM) and 37 °C; $t = 2$ h. Results are expressed as mean standard deviation ($n = 3$). **(b)** *Pseudo* Michaelis-Menten kinetics of the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$), $[1] = 50\text{-}900 \mu\text{M}$, pH 7.0 Tris-HCl buffer (10 mM) and 37 °C.

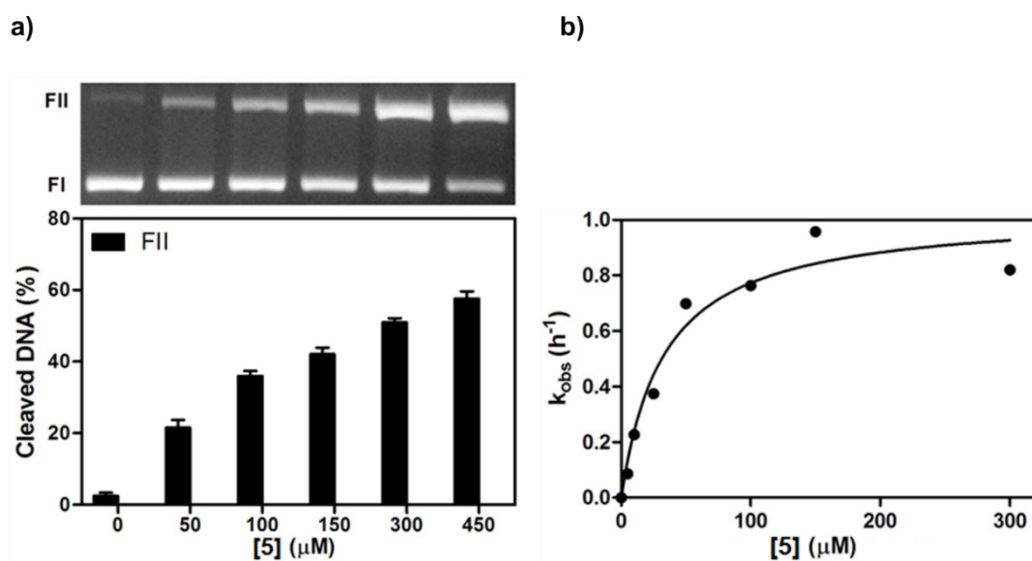


Fig. S3. (a) Agarose gel electrophoresis pattern and the corresponding plots for the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by complex **5** at different complex concentrations (50-450 μM) at pH 7.0 Tris-HCl buffer (10 mM) and 37 °C; $t = 2$ h. Results are expressed as mean standard deviation ($n = 3$). **(b)** *Pseudo* Michaelis-Menten kinetics of the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$), [5] = 5-300 μM , pH 7.0 Tris-HCl buffer (10 mM) and 37 °C.

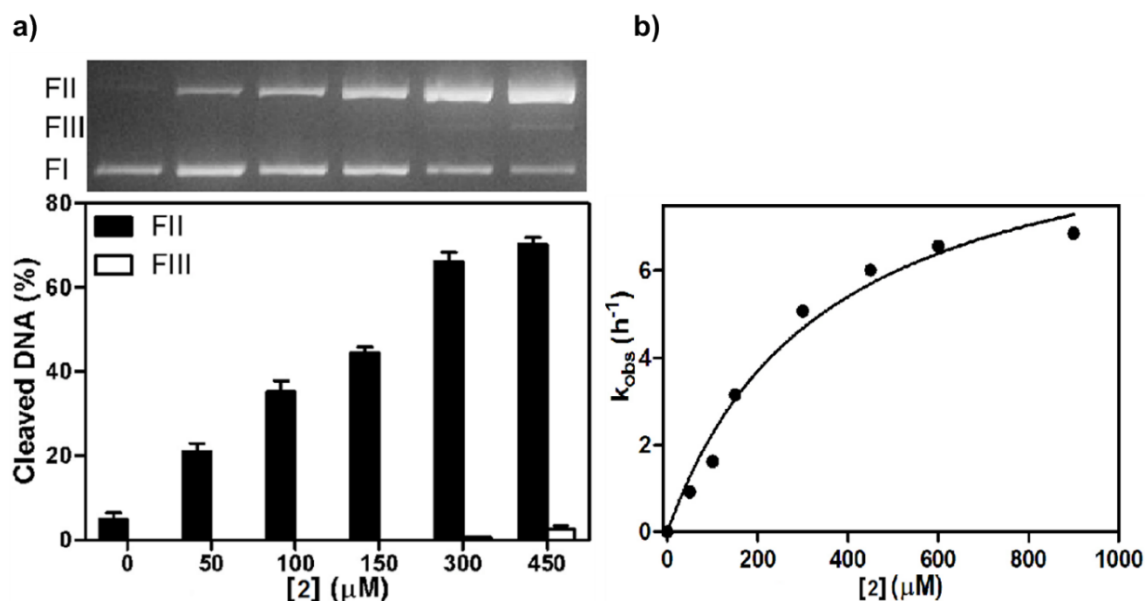


Fig. S4. (a) Agarose gel electrophoresis pattern and the corresponding plots for the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by complex **2** at different complex concentrations (50-450 μM) at pH 9.0 Tris-HCl buffer (10 mM) and 37 $^{\circ}\text{C}$; $t = 15$ min. Results are expressed as mean standard deviation ($n = 3$). (b) *Pseudo* Michaelis-Menten kinetics of the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$), [2] = 50-900 μM , pH 9.0 Tris-HCl buffer (10 mM) and 37 $^{\circ}\text{C}$.

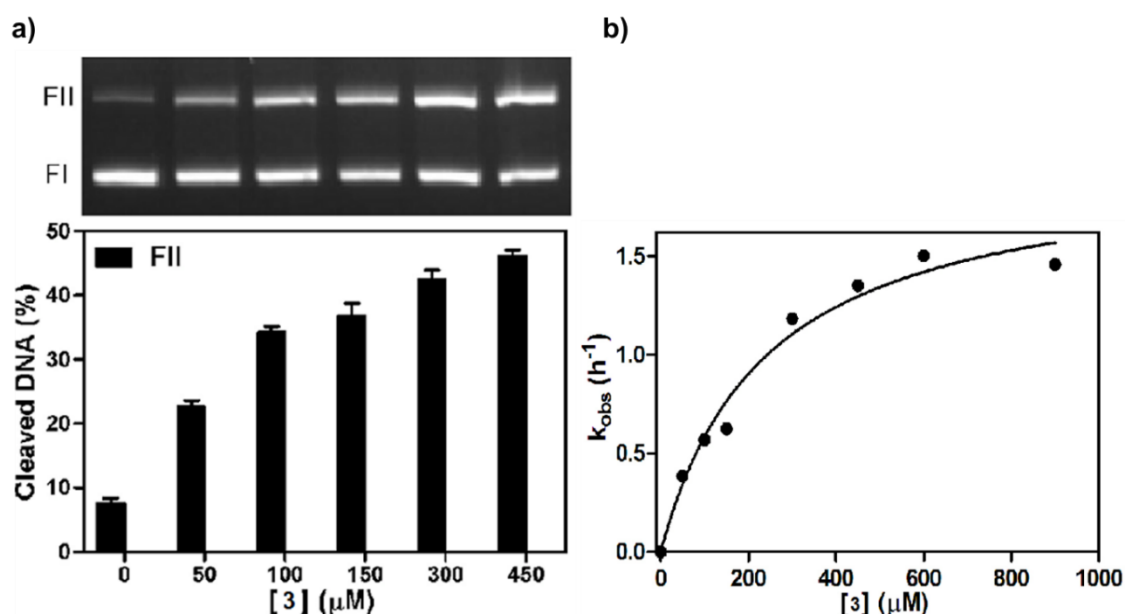


Fig. S5. (a) Agarose gel electrophoresis pattern and the corresponding plots for the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by complex **3** at different complex concentrations (50-450 μM) at pH 9.0 Tris-HCl buffer (10 mM) and 37 $^{\circ}\text{C}$; time = 30 min. Results are expressed as mean standard deviation ($n = 3$). (b) *Pseudo* Michaelis-Menten kinetics of the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$), [2] = 50-900 μM , pH 9.0 Tris-HCl buffer (10 mM) and 37 $^{\circ}\text{C}$.

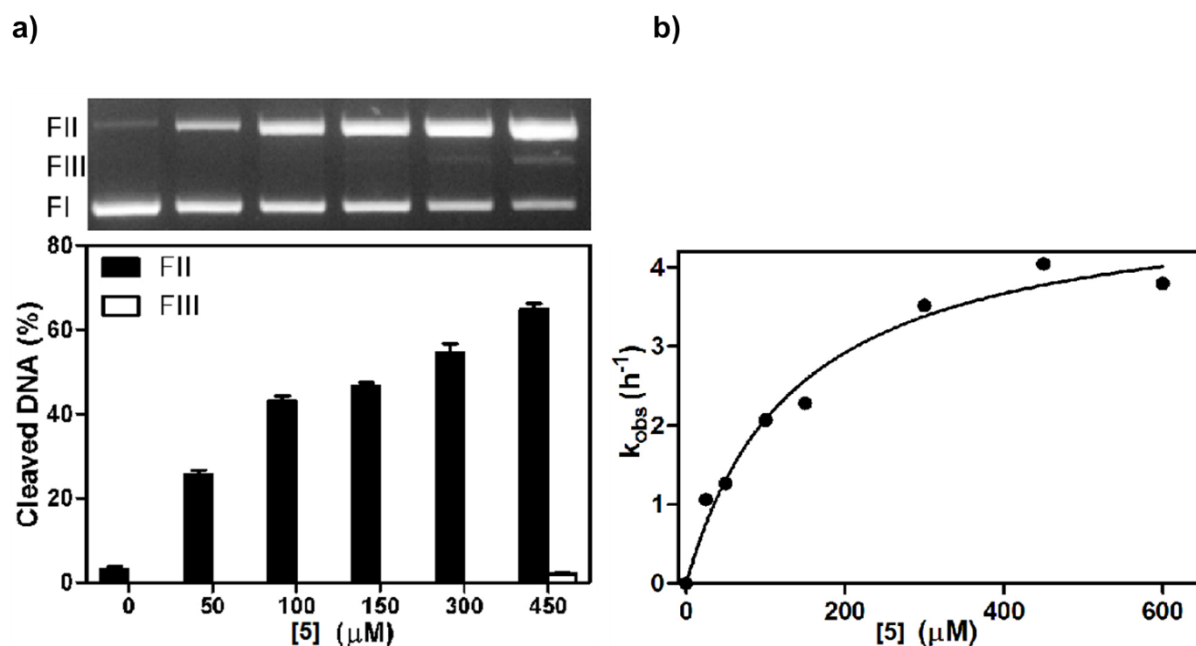


Fig. S6. (a) Agarose gel electrophoresis pattern and the corresponding plots for the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by complex **5** at different complex concentrations (50-450 μM) at pH 9.0 Tris-HCl buffer (10 mM) and 37 °C; time = 30 min. Results are expressed as mean \pm standard deviation ($n = 3$). **(b)** *Pseudo* Michaelis-Menten kinetics of the cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$), [5] = 50-600 μM , pH 9.0 Tris-HCl buffer (10 mM) and 37 °C.

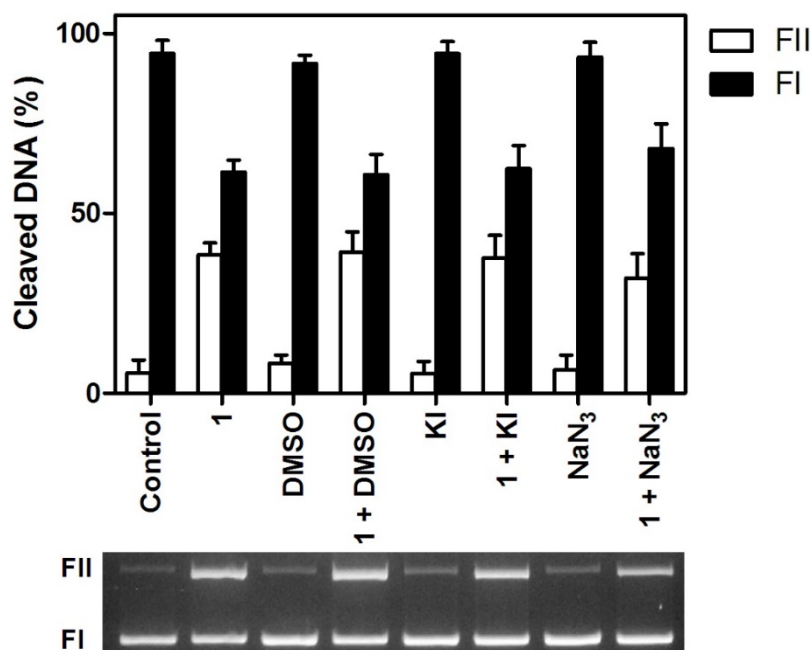


Fig. S7. Cleavage of pBSK II plasmid DNA ($[DNA] = \sim 25 \mu M$) by **1** ($[1] = 450 \mu M$) in the presence and absence of ROS scavengers. Reaction Conditions: $[Buffer] = 10 \text{ mM Tris-HCl pH } 7.0$, 37°C , $t = 30 \text{ min.}$, $[DMSO] = 0.5 \text{ M}$, $[KI] = 0.4 \text{ mM}$, $[NaN_3] = 0.5 \text{ mM}$. Results are expressed as mean \pm standard deviation ($n = 3$).

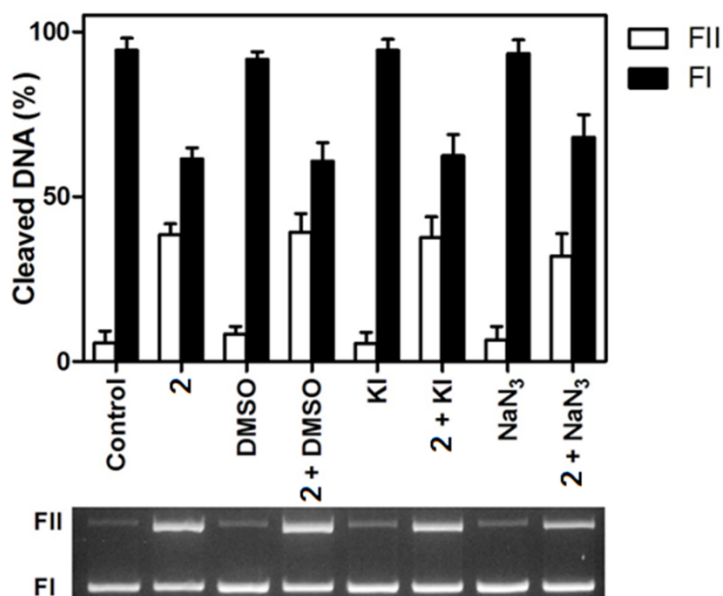


Fig. S8. Cleavage of pBSK II plasmid DNA ($\sim 25 \mu M$) by **2** ($450 \mu M$) in the presence and absence of ROS scavengers. Reaction Conditions: $[Buffer] = 10 \text{ mM Tris-HCl pH } 7.0$, 37°C , $t = 30 \text{ min.}$, $[DMSO] = 0.5 \text{ M}$, $[KI] = 0.4 \text{ mM}$, $[NaN_3] = 0.5 \text{ mM}$. Results are expressed as mean \pm standard deviation ($n = 3$).

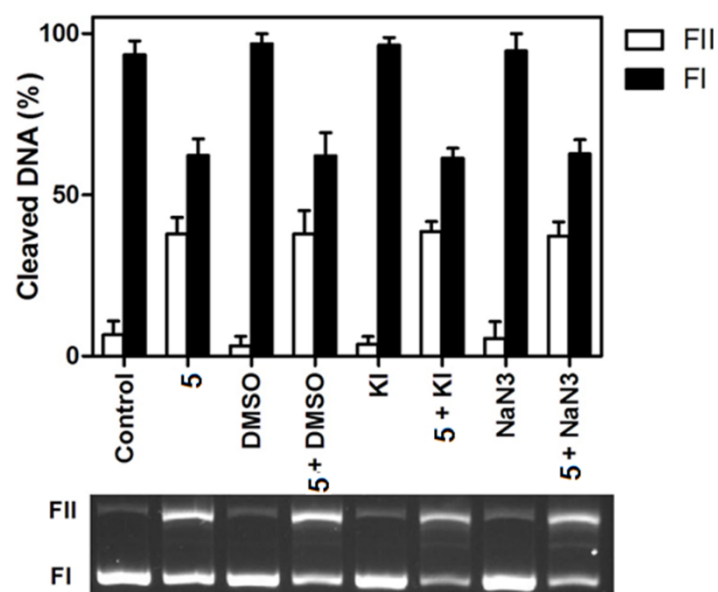


Fig. S9. Cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by **5** ($450 \mu\text{M}$) in the presence and absence of ROS scavengers. Reaction Conditions: [Buffer] = 10 mM Tris-HCl pH 7.0, 37°C , $t = 30 \text{ min}$, [DMSO] = 0.5 M, [KI] = 0.4 mM, [NaN₃] = 0.5 mM. Results are expressed as mean \pm standard deviation ($n = 3$).

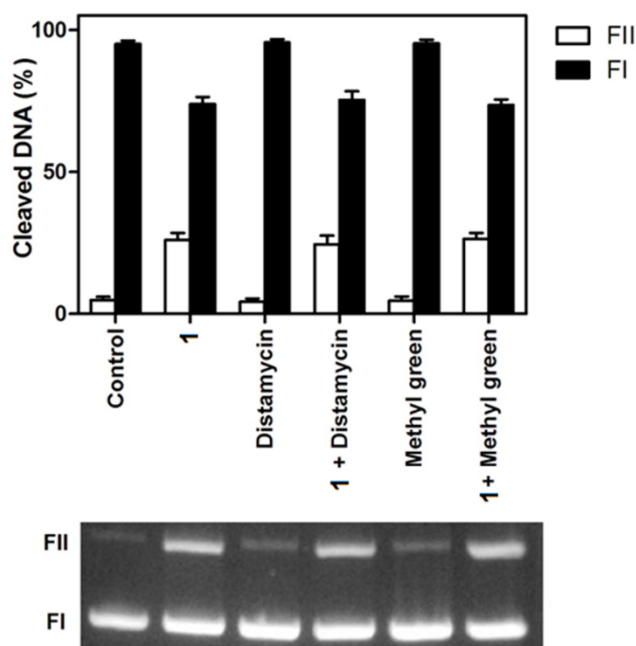


Fig. S11. Cleavage of pBSK II plasmid DNA (~25 μM) by **1** (450 μM) in the presence of DNA groove binders distamycin and methyl green (MG) [Distamycin or MG] = 50 μM. Reaction Conditions: [Buffer] = 10 mM Tris-HCl pH 7.0, 37 °C, t = 30 min. Results are expressed as mean ± standard deviation (n = 3).

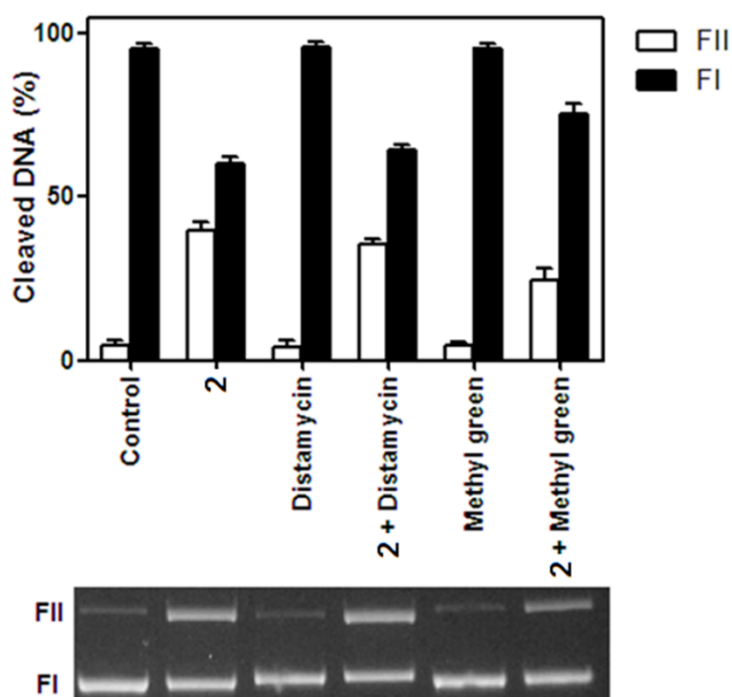


Fig. S12. Cleavage of pBSK II plasmid DNA (~25 μM) by **2** (450 μM) in the presence of DNA groove binders distamycin and methyl green (MG) [Distamycin or MG] = 50 μM. Reaction Conditions: [Buffer] = 10 mM Tris-HCl pH 7.0, 37 °C, t = 30 min. Results are expressed as mean ± standard deviation (n = 3).

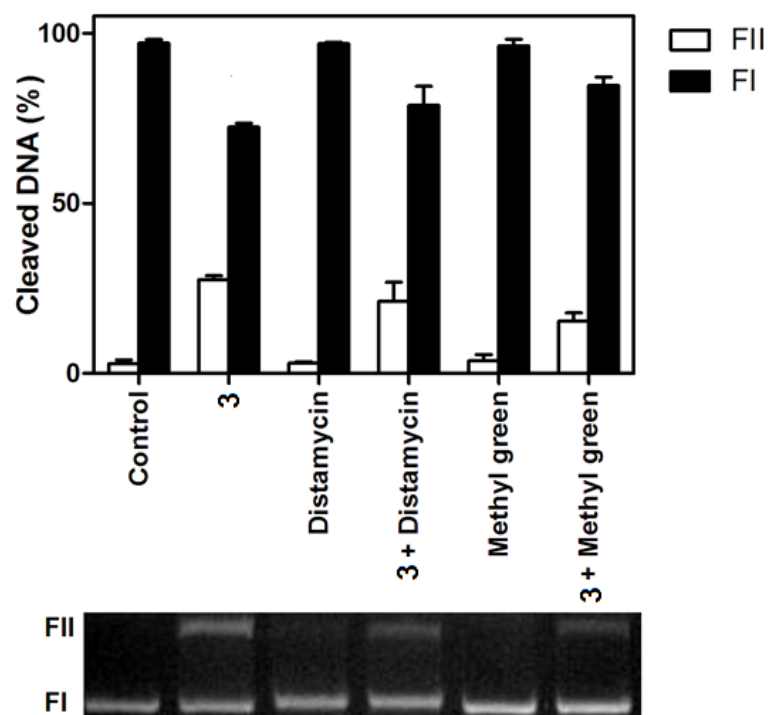


Fig. S13. Cleavage of pBSK II plasmid DNA ($\sim 25 \mu\text{M}$) by **3** ($450 \mu\text{M}$) in the presence of DNA groove binders distamycin and methyl green (MG) [Distamycin or MG] = $50 \mu\text{M}$). Reaction Conditions: [Buffer] = 10 mM Tris-HCl pH 7.0, 37°C , $t = 30 \text{ min}$. Results are expressed as mean \pm standard deviation ($n = 3$).

Table S1. *Pseudo*-Michael-Menten Kinetics of pBSK II Plasmid DNA Cleavage by Chloro-cobalt(II) Complexes **1-5** at Different Complex Concentrations, [DNA] \approx 25 μ M, [buffer] = 10 mM Tris-HCl pH 9.0 and 37 $^{\circ}$ C.

Complex	[Co(II)] (μ M)	k_{obs} (h^{-1})	k_{cat} (h^{-1}) ^{a)}	K_M (M)
[Co(TPA)Cl]ClO ₄ (1)	5	0.232	3.02	3.64×10^{-5}
	10	0.732		
	15	0.810		
	30	1.41		
	50	1.78		
	100	2.17		
	150	2.43		
[Co(6-MeTPA)Cl]ClO ₄ (2)	50	0.924	10.1	3.47×10^{-4}
	100	1.62		
	150	3.15		
	300	5.08		
	450	6.02		
	600	6.56		
	900	6.86		
[Co(6-Me ₂ TPA)Cl]ClO ₄ (3)	50	0.385	1.99	2.40×10^{-4}
	100	0.569		
	150	0.624		
	300	1.18		
	450	1.35		
	600	1.50		
	900	1.46		
[Co(BPQA)Cl]ClO ₄ (4)	50	2.61	16.8	3.65×10^{-4}
	100	3.89		
	150	5.09		
	300	6.66		
	450	9.72		
	600	10.2		
	900	12.3		
[Co(BQPA)Cl]ClO ₄ (5)	25	1.06	4.92	1.37×10^{-4}
	50	1.27		
	100	2.07		
	150	2.28		
	300	3.52		
	450	4.04		
	600	3.79		

a) k_{cat} represents the maximum rate of cleavage at the saturation ($V_{max} = k_{cat}$)