## Supporting information for

# Generation of low-valent tantalum species by reversible C-H activation in a cyclometallated tantalum hydride complex

- A. NMR spectrospopic analysis
  - A.1 Representative <sup>1</sup>H NMR spectrum of a mixture of 2a/2b
  - A.2 Deuterium scrambling observed by <sup>2</sup>H NMR spectroscopy
  - A.3 Observation of intermediate A by <sup>1</sup>H NMR spectroscopy
- B. X-ray crystallographic analysis
  - B.1 X-Ray crystal structure of 3
  - B.2 Determination of  $\tau$  parameter for 2a

## A. NMR spectroscopic analysis



#### Figure S1 Representative <sup>1</sup>H NMR spectrum of an isolated 7:1 mixture of 2a to 2b showing hydride and BDI backbone methine resonances.

# A.1 Representative <sup>1</sup>H NMR spectrum of a mixture of 2a/2b



Figure S2 Zoom in on the region shown in Figure S1 showing the methine and diastereotopic methylene resonances for 2a and 2b.

A.2 Deuterium scrambling observed by <sup>2</sup>H NMR spectroscopy



Figure S3 <sup>2</sup>H NMR spectrum of product isolated from reaction of 1 with 1 atm  $D_2$  showing scrambling of deuterium into the methyl and methine groups of the BDI ligand.

A.3 Observation of intermediate A by <sup>1</sup>H NMR Spectroscopy



Figure S4 <sup>1</sup>H NMR spectrum taken after 90 minutes of introducing 1 atm of  $H_2$  to an NMR tube containing 1. Compound 1 is partially converted to a mixture of 2a, 2b, and 3. Resonances attributed to intermediate A are shown.

### B. X-ray crystallographic analysis

B.1 X-Ray crystal structure of 3



Figure S5 Molecular structure of 3 as determined by a single crystal X-ray diffraction study. The hydrogen atoms and the co-crystallized benzene molecule are omitted for clarity; the thermal ellipsoids are set at the 50% probability level. Selected bond lengths (Å): Ta(1)-C(34) 2.172(2), Ta(1)-N(1) 1.785(2), Ta(1)-N(2) 2.300(2), Ta(1)-N(3) 2.102(2), Ta(1)-C(17) 2.184(2). Selected bond angles (°): N(3)-Ta(1)-C(34) 120.5(9), C(17)-Ta(1)-C(34) 115.7(1), N(3)-Ta(1)-C(17) 119.8(8), N(1)-Ta(1)-N(2) 173.1(1), C(1)-N(1)-Ta(1) 170.2(2).

# B.2 Determination of $\tau$ parameter for **2a**



**Figure S6** Molecular structure of **2a** showing the angles  $\alpha$  and  $\beta$  used for determining the parameter  $\tau$  using the equation  $\tau = (\alpha - \beta)/60^\circ = 52^\circ/60^\circ = 0.87$ . The *tert*-butyl group and aryl groups have been truncated for clarity. A value of  $\tau$  close to 1.0 indicates a nearly trigonal bipyramidal structure, whereas a value close to 0.0 indicates a nearly square pyramidal structure.