Helicity inversion and redox chemistry of chiral manganese(II) cubanes

Claire Deville, ^a Matteo Granelli, ^a Alan M. Downward, ^a Céline Besnard, ^b Laure Guenée^b and Alan F. Williams^{a,*}

Supplementary material.

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Figure S1.



a) UV-visible spectra of $[Mn_4(1-H)_4(OH)_2(MeCN)_2](ClO_4)_4$.MeCN(EtOH)₂, **4**, brown, and $[Mn_4(5-H)_4(DPP)_2](DPP)_2$.5EtOH, **6**, orange. **4** was measured in MeCN at $3*10^{-5}$ M, **6** $1*10^{-5}$ M in EtOH/CH₂Cl₂ 50/50.



b) Measurements at higher concentrations: **4** in MeCN at $3*10^{-4}$ M, **6** at $3*10^{-3}$ M in EtOH/CH₂Cl₂ 50/50.





CD spectra of **4** ($3*10^{-5}$ M in MeCN, brown) and **6** ($1*10^{-5}$ M in EtOH/CH₂Cl₂ 50/50, orange).



Figure S3. IR-spectra for $Mn_4(S-2-H)_4$ cubanes with various capping ligands: DPP (green), $PhCO_2^-$ (violet), $m-(NO_2)C_6H_4CO_2^-$ (blue), $p-(NO_2)C_6H_4CO_2^-$ (red).

Figure S4 Comparison of $[Mn_4(S-3-H)_4]^{4+}$ cores in 8 (left) and 9 (right).



Table S1. Bond distances in $[Mn_4(1-H)_4(OH)_2(MeCN)_2](CIO_4)_4$. MeCN(EtOH)₂, 4.

Bond to Mn(III)	Distance, Å	Bond to Mn(II)	Distance, Å
Mn1-O1A	1.888(3)	Mn(3)-013	2.040(4)
Mn1-O1B	2.102(4)	Mn3-N4B	2.146(4)
Mn1-N2A	2.160(4)	Mn3-N4A	2.152(4)
Mn2-013	1.855(3)	Mn3-O1A	2.177(3)
Mn2-N1B	2.101(4)	Mn3-N1S	2.388(5)
Mn2-O1B	2.166(4)	Mn3-O1B	2.433(3)

[Mn ₄ (1 -H) ₄ (OH) ₂ (MeCN) ₂](ClO ₄) ₄ .N	/IeCN(EtOH)2, 4	•					
D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H··· A			
N1 <i>A</i> —H1 <i>C</i> ···O8 <i>PB</i>	0.88	2.21	2.93 (2)	139			
$N3A - H3C \cdots O1S^{iv}$	0.88	1.89	2.731 (6)	160			
N2B—H2D···O4 E^{v}	0.88	2.20	3.029 (13)	157			
N3B—H3D···O2 P^{v}	0.88	2.06	2.909 (6)	161			
O13—H19···O3 <i>D</i> ^{iv}	0.86	2.00	2.835 (17)	162.3			
[Mn ₄ (5 -H) ₄ (DPP) ₂](DPP) ₂ .5EtOH, 6							
D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A			
N21 B —H21 B ···O2 D^{ii}	0.86	1.95	2.763 (5)	158			
N11 <i>B</i> —H11 <i>B</i> ⋯O1	0.86	1.95	2.797 (6)	170			
N11 <i>A</i> —H11 <i>A</i> ····O3 <i>D</i> ⁱⁱⁱ	0.86	1.89	2.739 (5)	167			
$O1$ — $H1A$ ··· $O2D^{ii}$	0.82	1.91	2.732 (5)	176			
Symmetry codes: (ii) $-x+3/2$, $-y+1/2$, $-z$; (iii) $-x+3/2$, $y+1/2$, $-z+1/2$.							
[Mn ₄ (S- 3 -H) ₄ (DPP) ₂](ClO ₄) ₂ , 8							
D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A			
N11 <i>A</i> —H11 <i>A</i> …O4	0.86	2.21	3.062 (5)	170			
N11 <i>B</i> —H11 <i>B</i> …O5	0.91	2.30	3.026 (5)	137			
[Mn ₄ (S- 3 -H) ₄ (m-C ₆ H ₄ (NO ₂)CO ₂) ₂](ClO ₄) ₂ , 9							
D—H···A	<i>D</i> —Н	$H \cdots A$	$D \cdots A$	D—H···A			
N11A—H11A…O7	0.84 (3)	2.25 (4)	3.074 (6)	168 (5)			
N11 <i>B</i> —H11 <i>B</i> …O1	0.85 (3)	2.13 (4)	2.894 (6)	148 (5)			
N11C—H11C…O3	0.85 (3)	2.25 (4)	3.092 (6)	170 (5)			
N11 <i>D</i> —H11 <i>D</i> …O5	0.85 (3)	2.11 (4)	2.851 (6)	145 (5)			
Symmetry codes: (iv) $x-1/2$, $y+1/2$, z ; (v) $x-1/2$, $y+1/2$, $z+1$.							

 Table S2
 Hydrogen Bonds.

 [Mp.(1-H).(OH).(MeCN).](CIO.). MeCN(EtOH).