

Supplementary Information for

**Zinc-specific intramolecular excimer formation in TQEN derivatives: fluorescence  
and zinc binding property of TPEN-based hexadentate ligands**

Yuji Mikata,<sup>a,b,\*</sup> Saaya Takeuchi,<sup>b</sup> Eri Higuchi,<sup>b</sup> Ayaka Ochi,<sup>b</sup> Hideo Konno,<sup>c</sup> Kazuma Yanai<sup>d</sup> and Shin-ichiro Sato<sup>e</sup>

<sup>a</sup>*KYOUSEI Science Center, Nara Women's University, Nara 630-8506, Japan, <sup>b</sup>Department of Chemistry, Faculty of Science, Nara Women's University, Nara 630-8506, Japan,*

<sup>c</sup>*National Institute of Advanced Industrial Science and Technology (AIST), 1-1-1 Higashi, Tsukuba, Ibaraki 305-8565, Japan, <sup>d</sup>Graduate School of Chemical Sciences and Engineering, Hokkaido University, Japan and <sup>e</sup>Graduate School of Engineering, Hokkaido University, Japan*

**Table S1** Crystallographic Data for *N,N*-1-isoBQBPEN (**1b**) and *N,N'*-1-isoBQBPEN·4H<sub>2</sub>O (**2b**·4H<sub>2</sub>O)

	<i>N,N</i> -1-isoBQBPEN ( <b>1b</b> )	<i>N,N'</i> -1-isoBQBPEN·4H <sub>2</sub> O ( <b>2b</b> ·4H <sub>2</sub> O)
Formula	C <sub>34</sub> H <sub>32</sub> N <sub>6</sub>	C <sub>34</sub> H <sub>40</sub> N <sub>6</sub> O <sub>4</sub>
FW	524.67	596.73
Crystal system	triclinic	monoclinic
Space group	P-1	P2 <sub>1</sub> /a
<i>a</i> , Å	10.183(6)	9.4162(8)
<i>b</i> , Å	11.355(7)	14.2496(9)
<i>c</i> , Å	13.428(7)	11.8720(12)
α, deg	107.863(10)	90
β, deg	92.769(9)	105.124(4)
γ, deg	102.293(7)	90
<i>V</i> , Å <sup>3</sup>	1433(2)	1537.8(3)
Z	2	2
<i>D</i> <sub>calc</sub> , g cm <sup>-3</sup>	1.216	1.289
μ, mm <sup>-1</sup>	0.0737	0.0862
2θ <sub>max</sub> , deg	62.8	55
temp, K	153	123
no. reflns collected	11895	11810
no. reflns used	6331	3501
no. of params	489	279
<i>R</i> <sub>int</sub>	0.0302	0.0255
Final <i>R</i> 1 ( <i>I</i> > 2σ( <i>I</i> )) <sup>a</sup>	0.0552	0.0463
<i>wR</i> 2 (all data) <sup>b</sup>	0.1424	0.1185
GOF	1.065	1.159

<sup>a</sup>*R*1 = Σ ||*F*<sub>o</sub>| - |*F*<sub>c</sub>| | / Σ |*F*<sub>o</sub>|.   <sup>b</sup>*wR*2 = [Σ*w*[(*F*<sub>o</sub><sup>2</sup> - *F*<sub>c</sub><sup>2</sup>)<sup>2</sup>] / Σ [*w*(*F*<sub>o</sub><sup>2</sup>)<sup>2</sup>]]<sup>1/2</sup>.

**Table S2** Crystallographic Data for [Zn(**1b**)](ClO<sub>4</sub>)<sub>2</sub>·2CH<sub>3</sub>OH and [Zn(**2b**)](ClO<sub>4</sub>)<sub>2</sub>·DMF

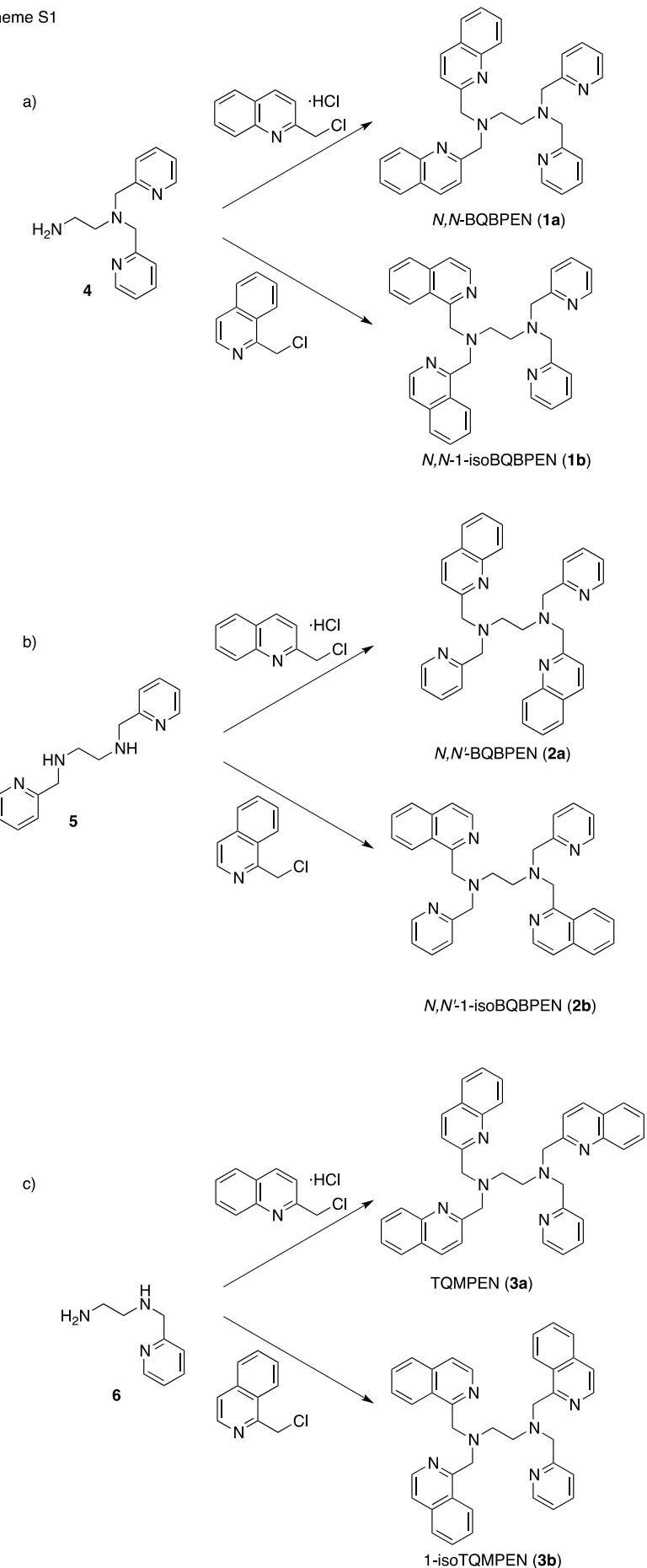
	[Zn( <b>1b</b> )](ClO <sub>4</sub> ) <sub>2</sub> · 2CH <sub>3</sub> OH	[Zn( <b>2b</b> )](ClO <sub>4</sub> ) <sub>2</sub> · ·DMF
Formula	C <sub>36</sub> H <sub>38</sub> Cl <sub>2</sub> N <sub>6</sub> O <sub>10</sub> Zn	C <sub>37</sub> H <sub>39</sub> Cl <sub>2</sub> N <sub>7</sub> O <sub>9</sub> Zn
FW	851.02	862.04
Crystal system	monoclinic	monoclinic
Space group	P2 <sub>1</sub> /c	P2 <sub>1</sub> /n
a, Å	12.0540(9)	13.4515(13)
b, Å	14.3530(8)	18.682(2)
c, Å	21.7345(12)	14.935(2)
β, deg	92.903(4)	97.4707(11)
V, Å <sup>3</sup>	3755.5(4)	3721.2(6)
Z	4	4
D <sub>calc</sub> , g cm <sup>-3</sup>	1.505	1.539
μ, mm <sup>-1</sup>	0.8622	0.08701
2θ <sub>max</sub> , deg	55	55
temp, K	123	153
no. reflns collected	28756	28520
no. reflns used	8511	8509
no. of params	541	633
R <sub>int</sub>	0.0210	0.0247
Final R1 ( <i>I</i> > 2σ( <i>I</i> )) <sup>a</sup>	0.0490	0.0667
wR2 (all data) <sup>b</sup>	0.1377	0.2058
GOF	1.047	1.173

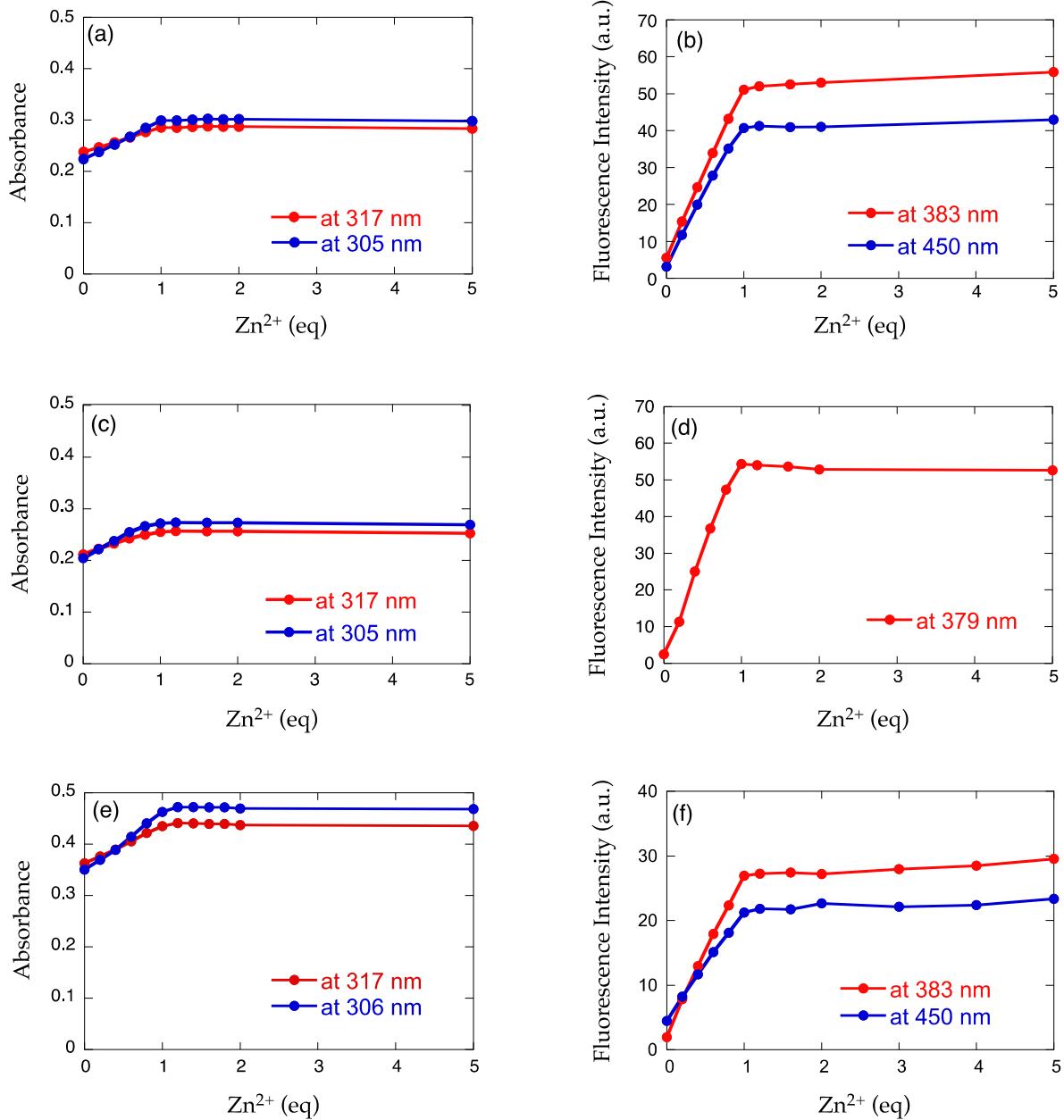
<sup>a</sup>R1 = Σ ||F<sub>o</sub>| - |F<sub>c</sub>| |/Σ |F<sub>o</sub>|.   <sup>b</sup>wR2 = [Σw[(F<sub>o</sub><sup>2</sup> - F<sub>c</sub><sup>2</sup>)<sup>2</sup>]/Σ[w(F<sub>o</sub><sup>2</sup>)<sup>2</sup>]]<sup>1/2</sup>.

**Table S3** Calculated Absorption Properties for  $[Zn(\mathbf{1b})]^{2+}$  and  $[Zn(\mathbf{2b})]^{2+}$ 

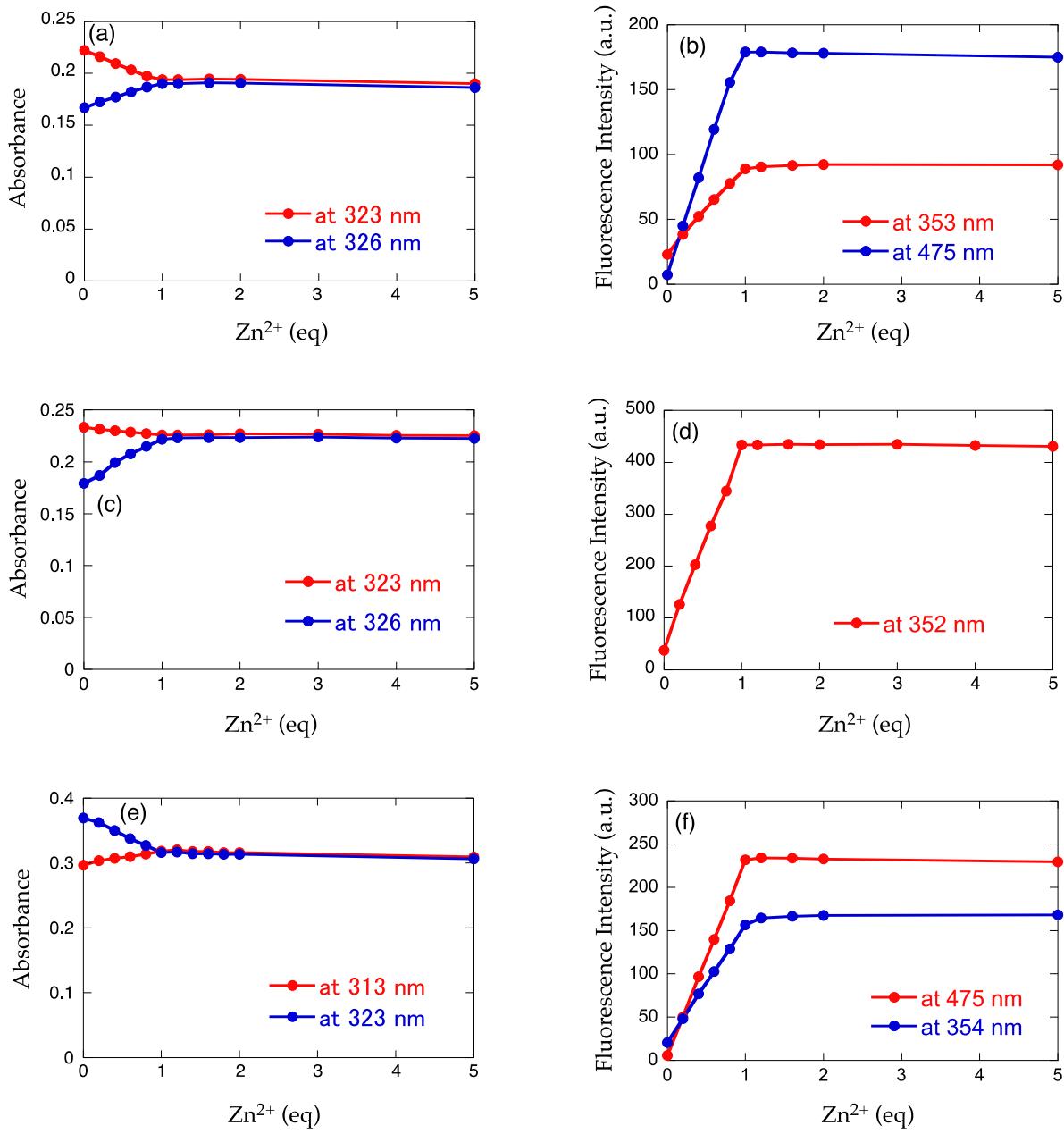
State	Wavelength (nm)	Oscillator Strength	Orbital Composition	CI (%)
$[Zn(\mathbf{1b})]^2$				
$S_1$	290	0.050	HOMO-1 -> LUMO	43
			HOMO -> LUMO+1	2
$S_2$	285	0.084	HOMO -> LUMO+1	39
			HOMO-1 -> LUMO	2
$S_3$	261	0.0040	HOMO-3 -> LUMO	21
			HOMO-1 -> LUMO+4	15
$[Zn(\mathbf{2b})]^{2+}$				
$S_1$	285	0.10	HOMO-1 -> LUMO+1	48
			HOMO -> LUMO	20
$S_2$	285	0.035	HOMO -> LUMO+1	22
			HOMO-1 -> LUMO	20
$S_3$	260	0.016	HOMO-3 -> LUMO+1	10
			HOMO-1 -> LUMO+6	10

Scheme S1

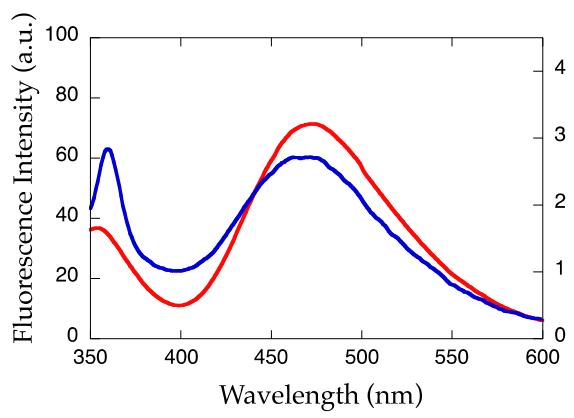




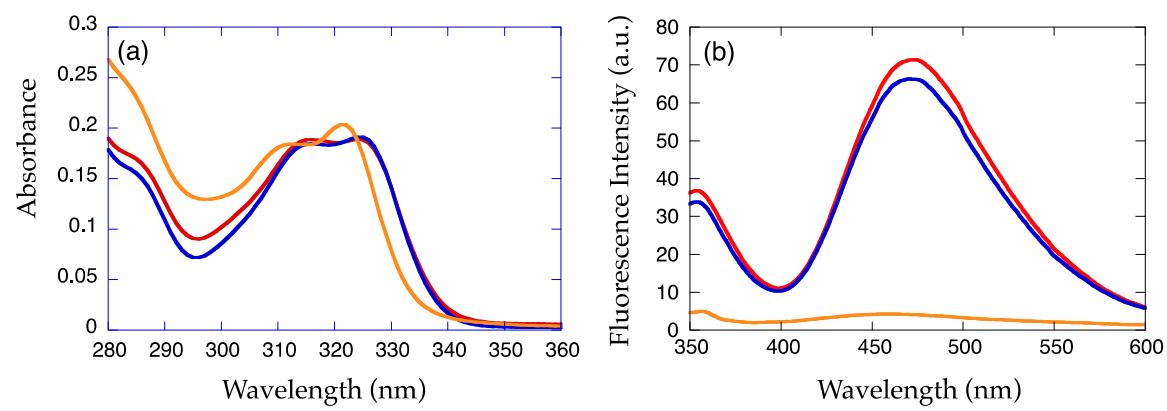
**Fig. S1** Zinc titration profile for  $34 \mu\text{M} N,N\text{-BQBPEN}$  (**1a**) (a, b),  $N,N'\text{-BQBPEN}$  (**2a**) (c, d) and  $\text{TQMPEN}$  (**3a**) (e, f) in  $\text{DMF}/\text{H}_2\text{O}$  (1:1) at  $25^\circ\text{C}$ . (a, c, e) Absorbance changes. (b, d, f) Fluorescence intensity changes.



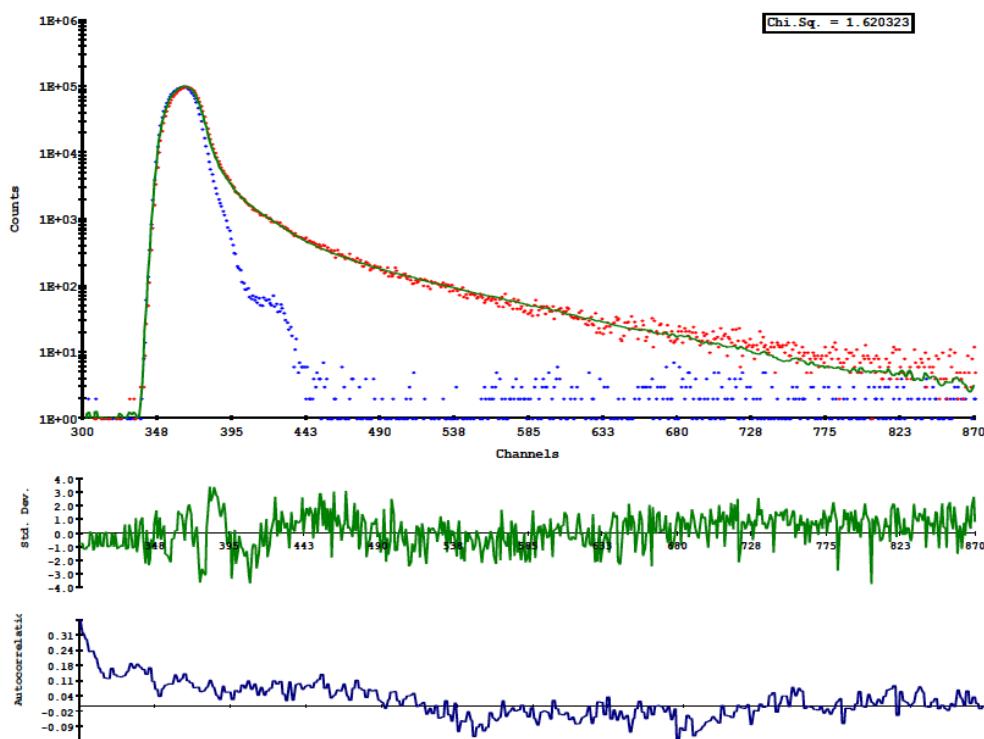
**Fig. S2** Zinc titration profile for 34  $\mu\text{M}$  *N,N*-1-isoBQBPEN (**1b**) (a, b), *N,N'*-1-isoBQBPEN (**2b**) (c, d) and 1-isoTQMPEPEN (**3b**) (e, f) in DMF/H<sub>2</sub>O (1:1) at 25 °C. (a, c, e) Absorbance changes. (b, d, f) Fluorescence intensity changes.



**Fig. S3** Fluorescence spectra of zinc complex of *N,N*-1-isoBQBPEN (**1b**) at 34  $\mu\text{M}$  (red, left Y axis) and 1  $\mu\text{M}$  (blue, right Y axis) in DMF/ $\text{H}_2\text{O}$  (1:1) at 25  $^{\circ}\text{C}$  ( $\lambda_{\text{ex}} = 326$  nm).



**Fig. S4** Fluorescence spectra of 34  $\mu\text{M}$  of zinc complex of *N,N*-1-isoBQBPEN (**1b**) in DMF/H<sub>2</sub>O (1:1) (red), DMF/H<sub>2</sub>O (4:1) (blue) and DMF (orange) at 25 °C ( $\lambda_{\text{ex}} = 326$  nm).



Calculated using 3 exponentials

Prompt data : Prompt  
Decay data : Decay

The initial parameters are:

Shift Value = 0	ch;	0	sec
Shift Limit = 30	ch;	1.677618E-09	sec
T1 Estimate = 2.669495	ch;	1.492797E-10	sec
T2 Estimate = 5.338989	ch;	2.985595E-10	sec
T3 Estimate = 10.67798	ch;	5.97119E-10	sec

A Free  
B1 Free  
B2 Free  
B3 Free

Prompt and decay LO = 300 ch; 1.677618E-08 sec  
Prompt and decay HI = 900 ch; 5.032854E-08 sec

Background on prompt = 0.4285714  
Time calibration = 5.59206E-11 sec/ch

The fitted parameters are:

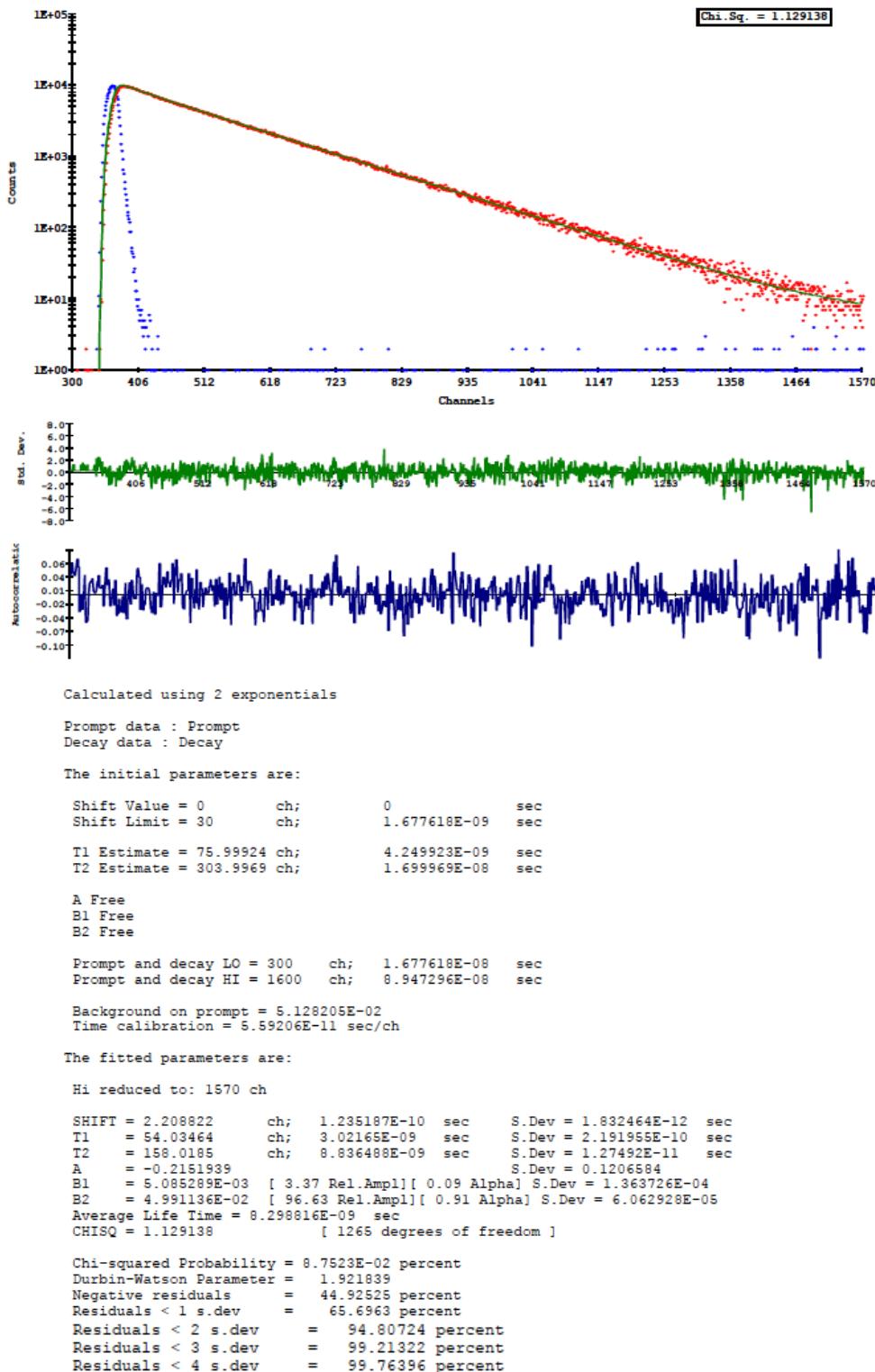
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Hi reduced to: 870 ch

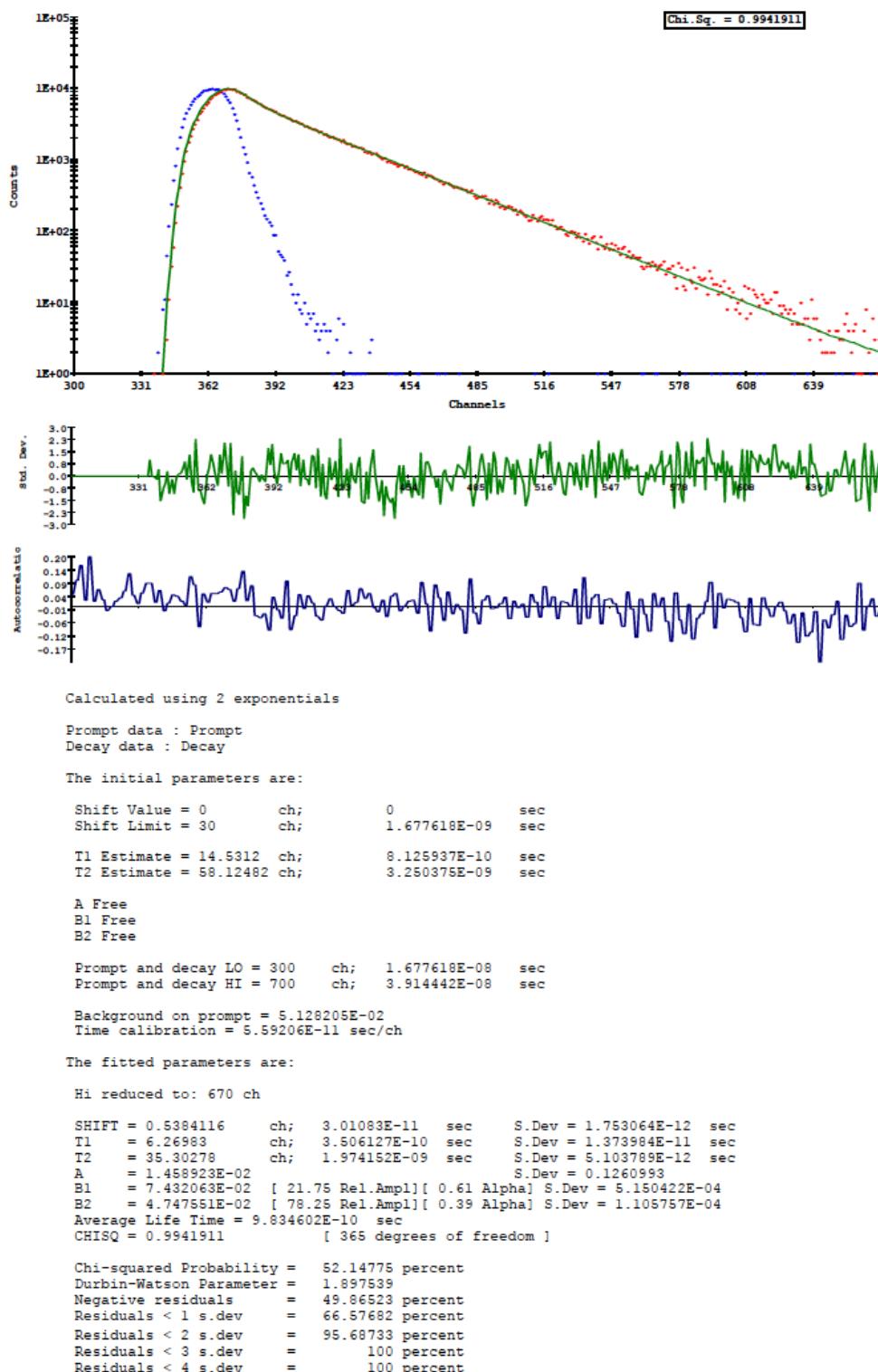
SHIFT = 0.3566583    ch;  1.994455E-11  sec   S.Dev = 4.304355E-13  sec
T1   = 2.029763     ch;  1.135055E-10  sec   S.Dev = 2.015821E-12  sec
T2   = 18.38545     ch;  1.028126E-09  sec   S.Dev = 1.78469E-11  sec
T3   = 77.43118     ch;  4.329998E-09  sec   S.Dev = 5.101761E-11  sec
A    = 0.7804149          S.Dev = 0.1262916
B1  = 0.6164292 [ 90.67 Rel.Ampl][ 0.99 Alpha] S.Dev = 5.023048E-04
B2  = 5.283901E-03 [ 7.04 Rel.Ampl][ 0.01 Alpha] S.Dev = 3.017124E-05
B3  = 4.082497E-04 [ 2.29 Rel.Ampl][ 0.00 Alpha] S.Dev = 2.869185E-06
Average Life Time = 1.240407E-10 sec
CHISQ = 1.620323 [ 563 degrees of freedom ]
Chi-squared Probability = 5.8224E-17 percent
Durbin-Watson Parameter = 1.248984
Negative residuals = 41.50613 percent
Residuals < 1 s.dev = 54.11559 percent
Residuals < 2 s.dev = 88.61646 percent
Residuals < 3 s.dev = 98.24869 percent
Residuals < 4 s.dev = 100 percent

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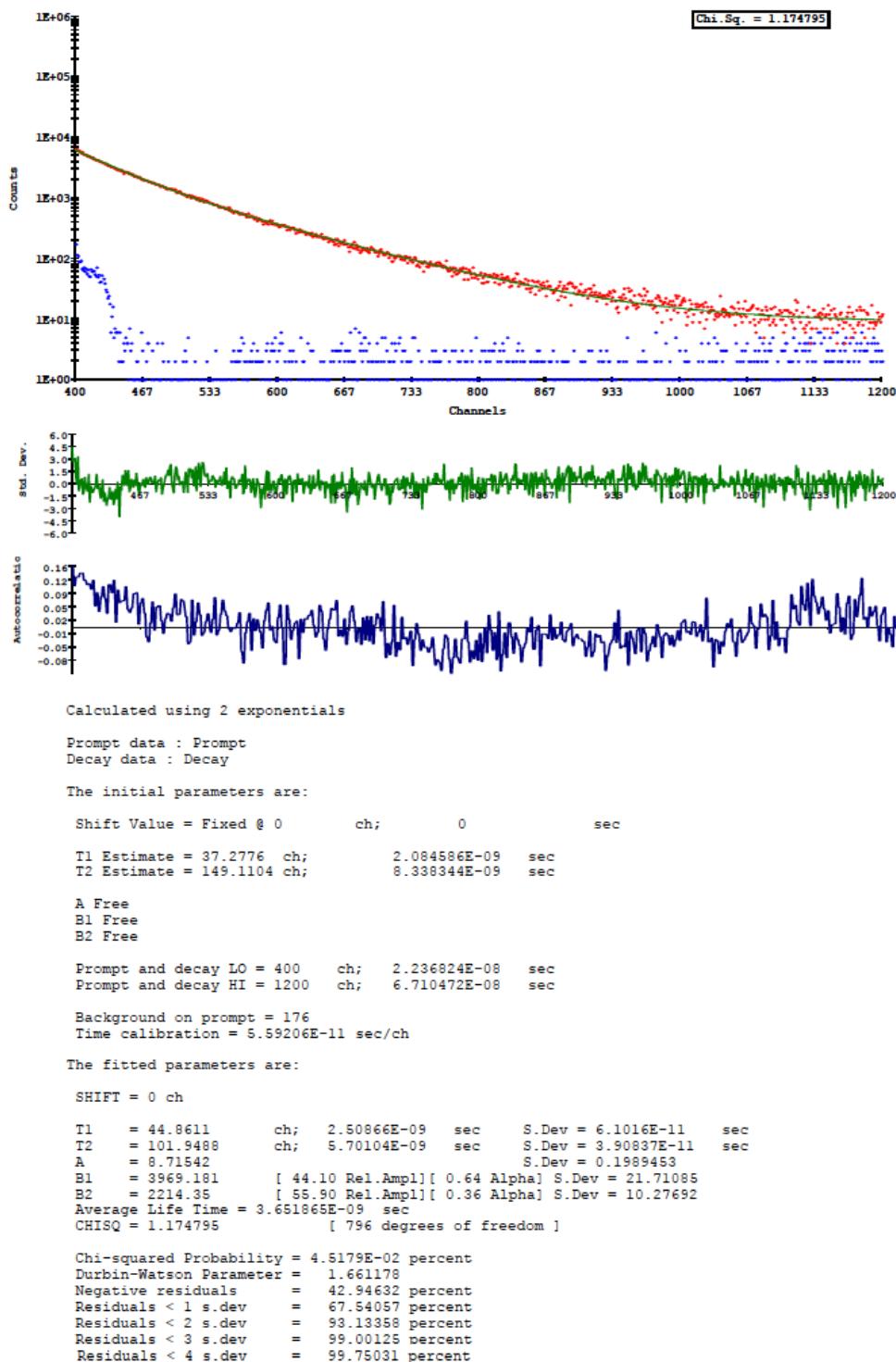
**Fig. S5** Fluorescence lifetime measurement for 34  $\mu$ M **1b** in DMF/H<sub>2</sub>O (1:1) at 353 nm ( $\lambda_{\text{ex}} = 333$  nm, 25 °C).



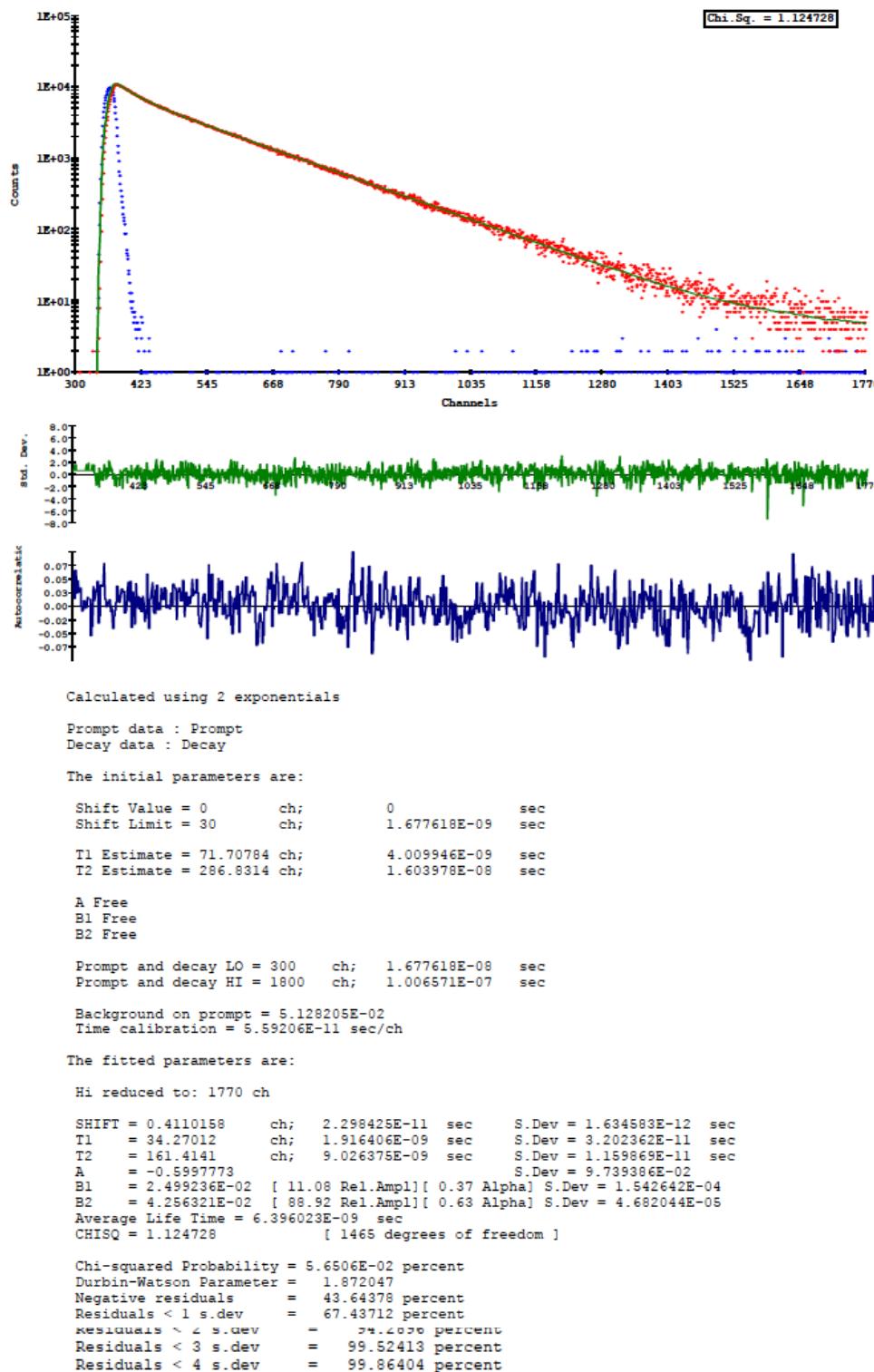
**Fig. S6** Fluorescence lifetime measurement for 34  $\mu$ M **1b** in DMF/H<sub>2</sub>O (1:1) at 475 nm ( $\lambda_{\text{ex}} = 333$  nm, 25 °C).



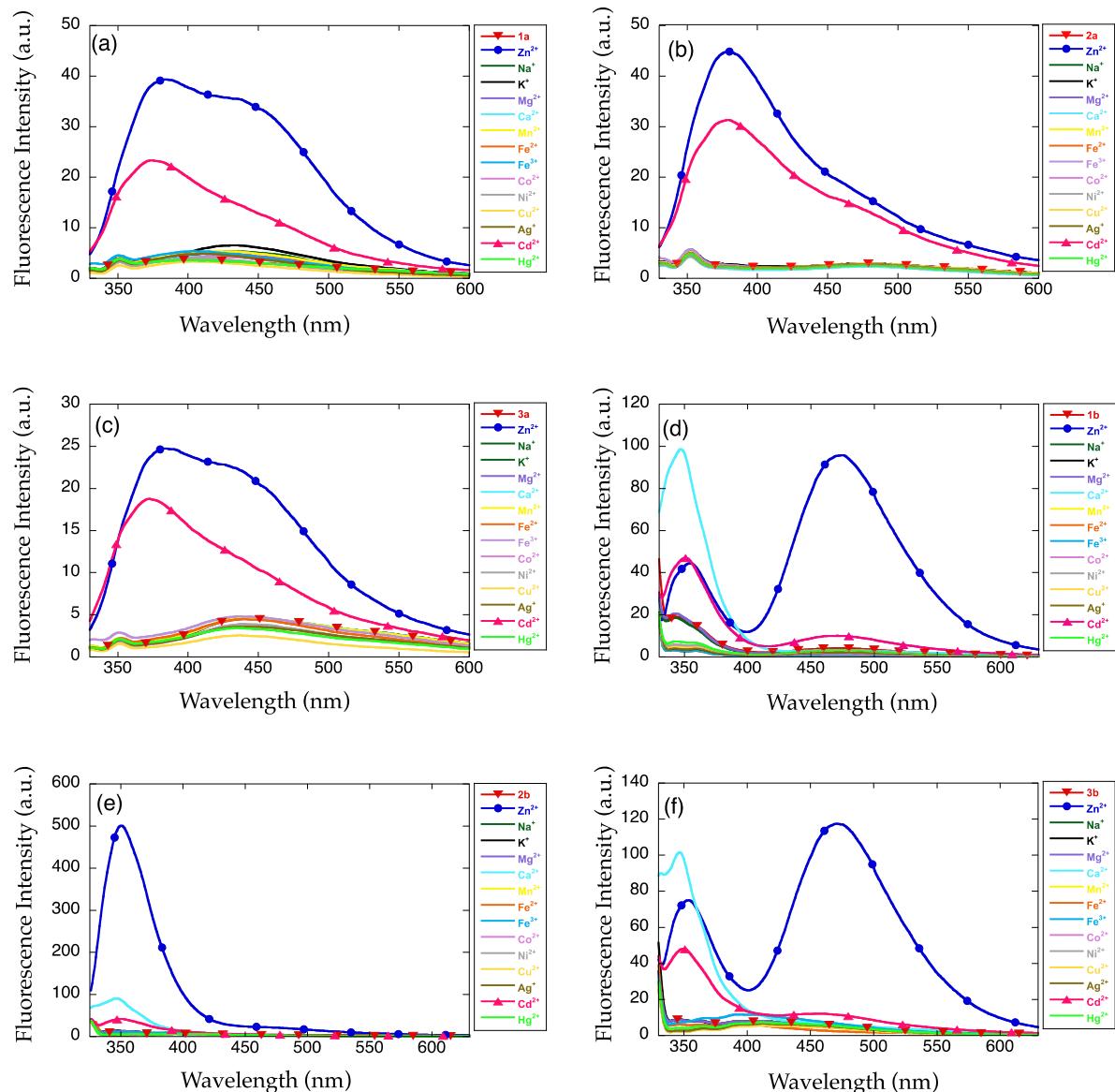
**Fig. S7** Fluorescence lifetime measurement for 34  $\mu$ M **2b** in DMF/H<sub>2</sub>O (1:1) at 352 nm ( $\lambda_{\text{ex}} = 333$  nm, 25 °C).



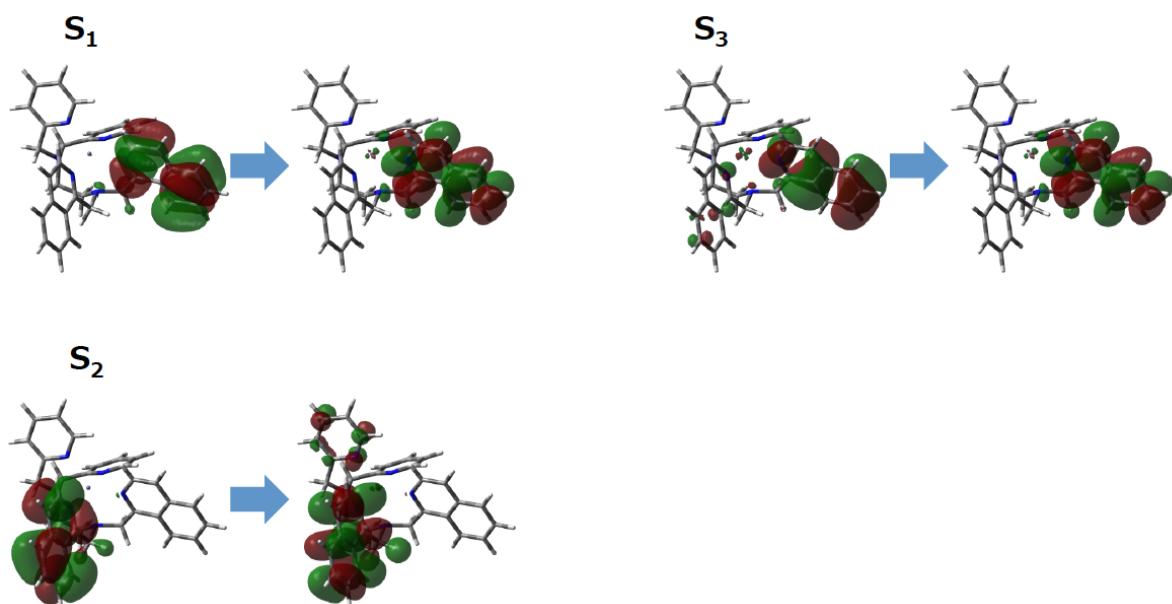
**Fig. S8** Fluorescence lifetime measurement for 34  $\mu\text{M}$  1-isoTQEN in DMF/H<sub>2</sub>O (1:1) at 357 nm ( $\lambda_{\text{ex}} = 333$  nm, 25 °C).



**Fig. S9** Fluorescence lifetime measurement for 34  $\mu$ M 1-isoTQEN in DMF/H<sub>2</sub>O (1:1) at 477 nm ( $\lambda_{\text{ex}} = 333$  nm, 25 °C).

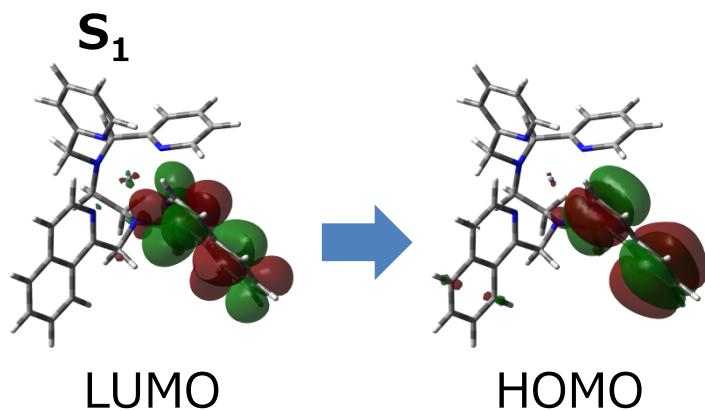


**Fig. S10** Comparison of fluorescence spectra of 34  $\mu\text{M}$  **1-3a** and **1-3b** in DMF/H<sub>2</sub>O (1:1) at 25 °C in the presence of 1 equivalent of various metal ions.

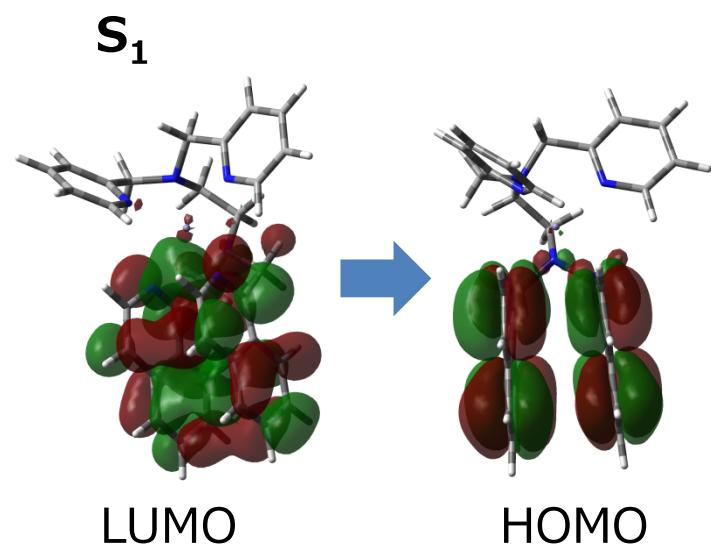


**Fig. S11** Schematic representation of molecular orbitals for absorption of  $[Zn(1b)]^{2+}$ .

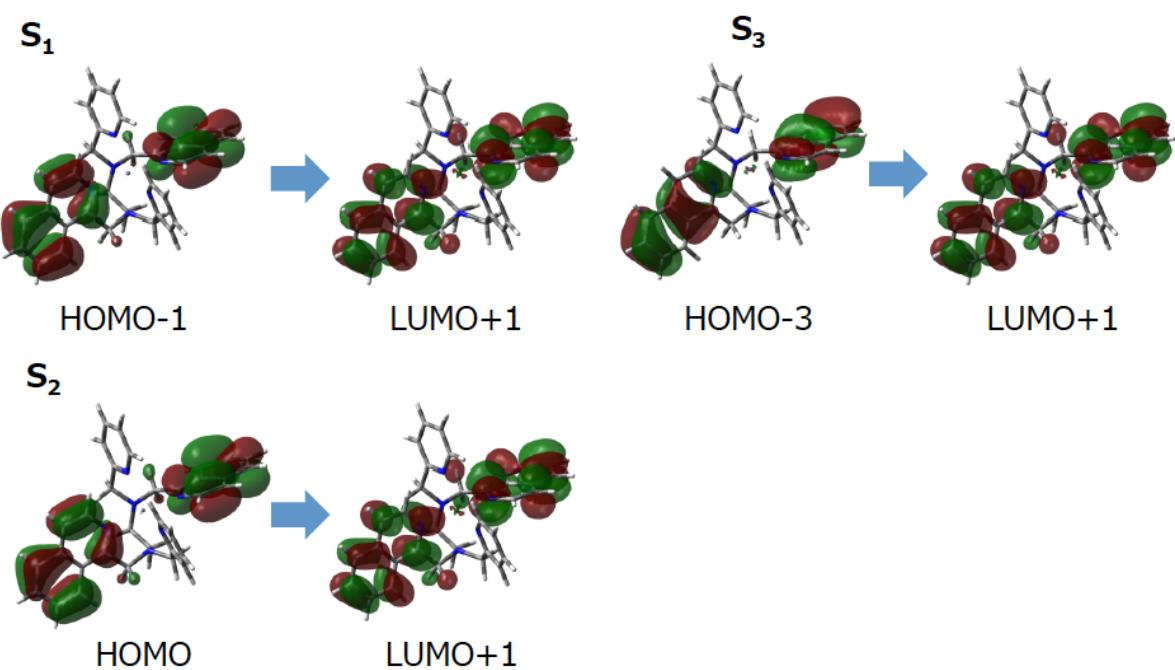
(a)



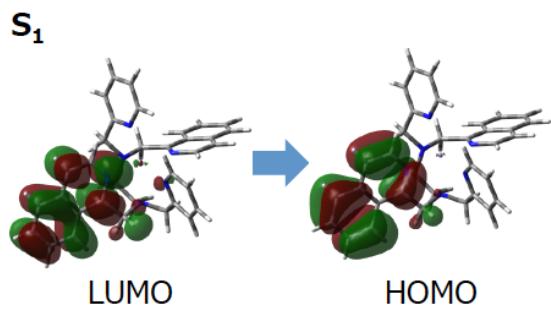
(b)



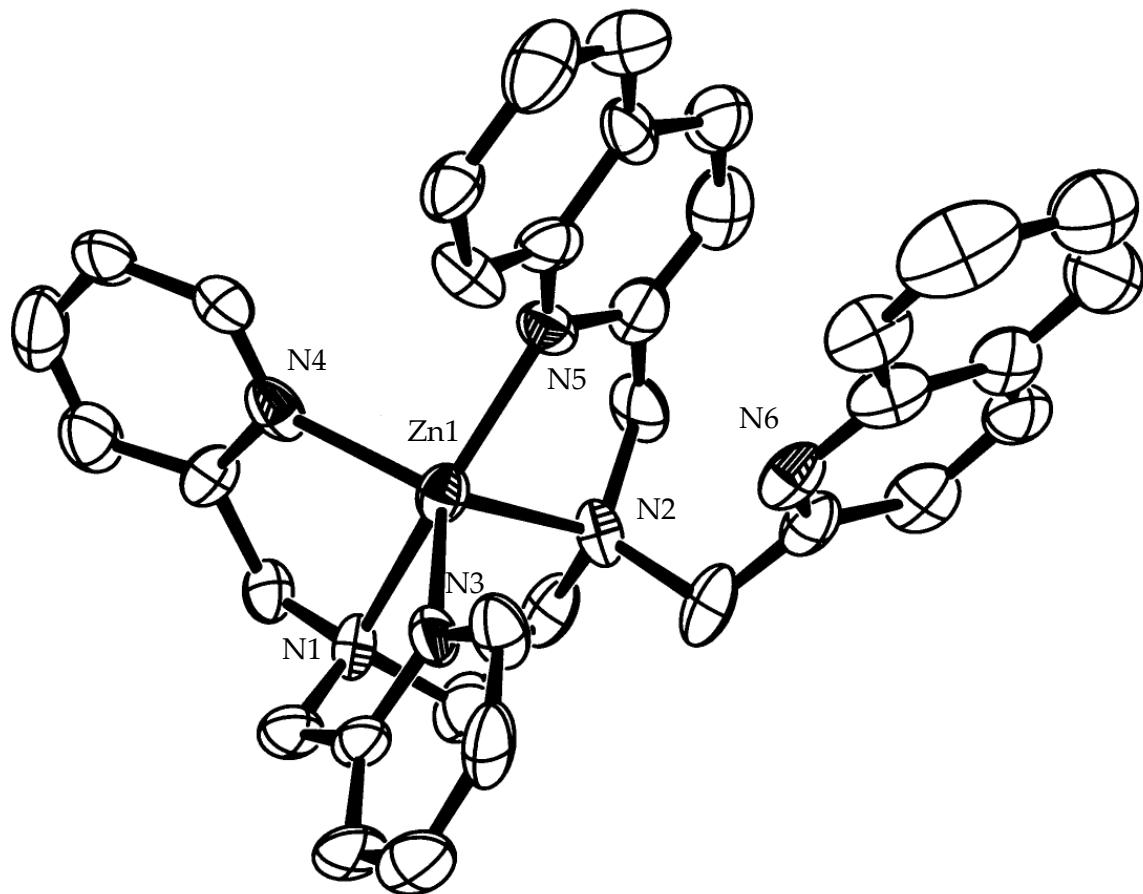
**Fig. S12** Schematic representation of molecular orbitals for emission of  $[\text{Zn}(\mathbf{1b})]^{2+}$  with (a) Franck Condon and (b) excimer conformation.



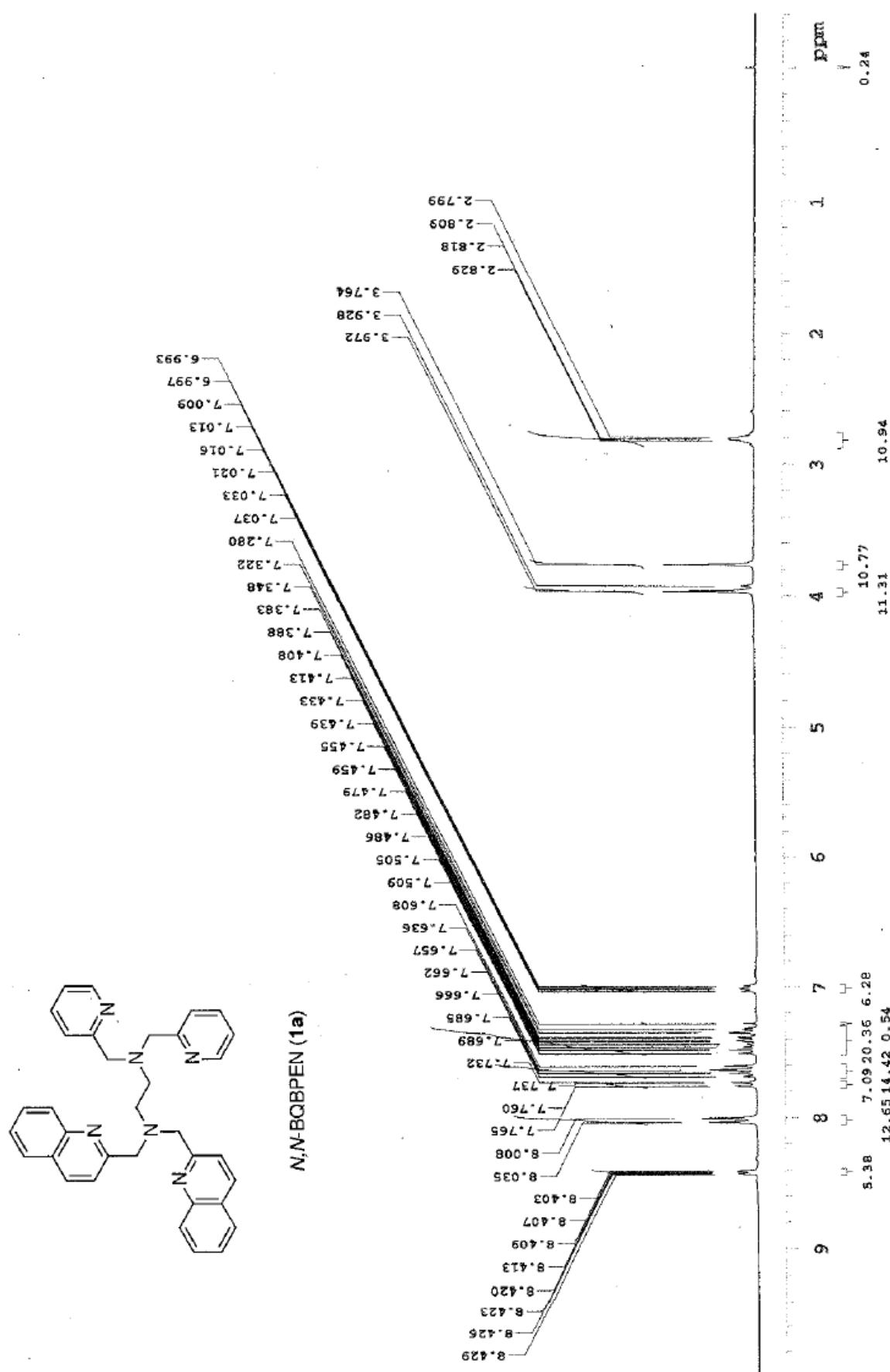
**Fig. S13** Schematic representation of molecular orbitals for absorption of  $[Zn(2b)]^{2+}$ .



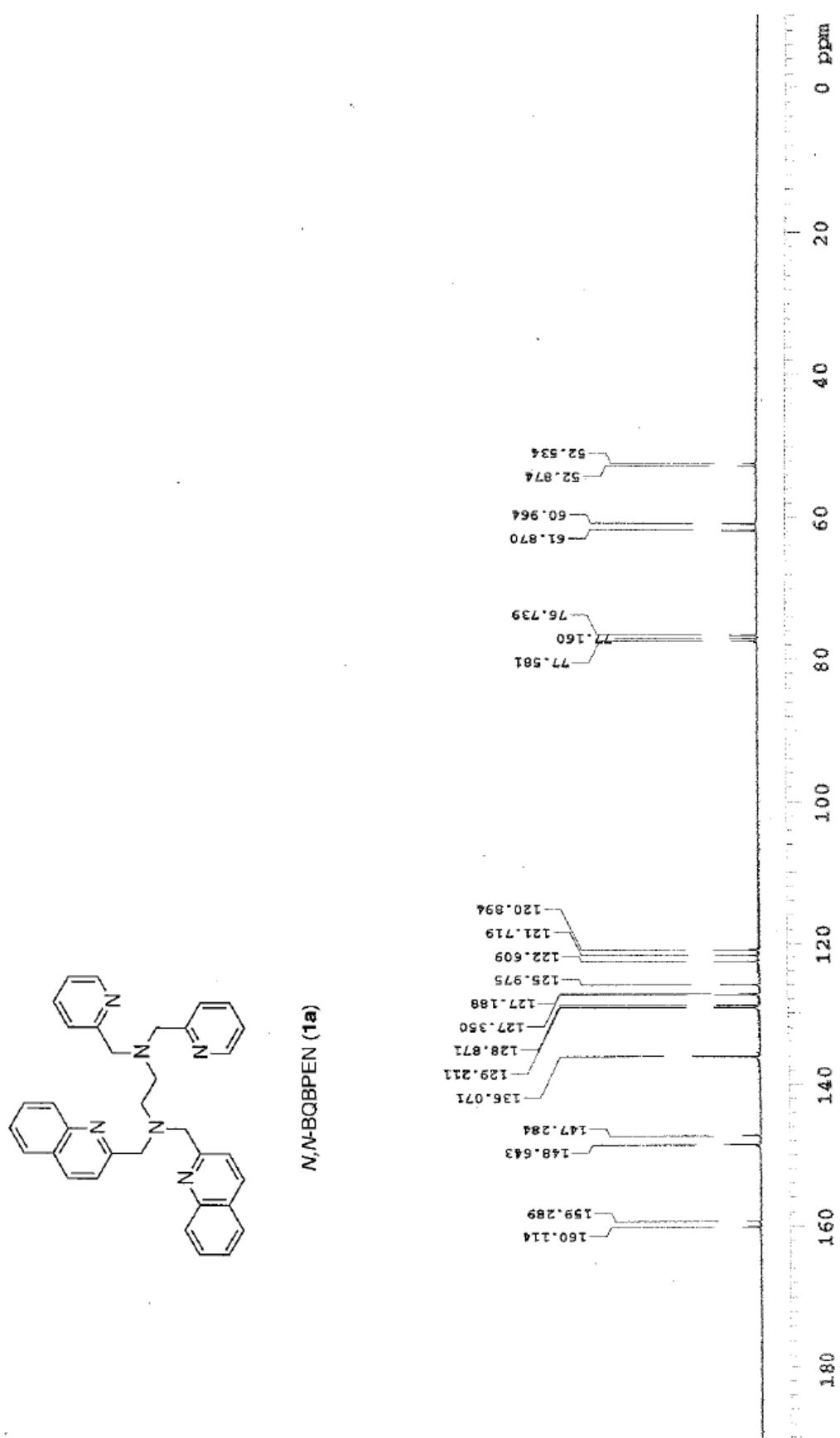
**Fig. S14** Schematic representation of molecular orbitals for emission of  $[\text{Zn}(\mathbf{2b})]^{2+}$ .



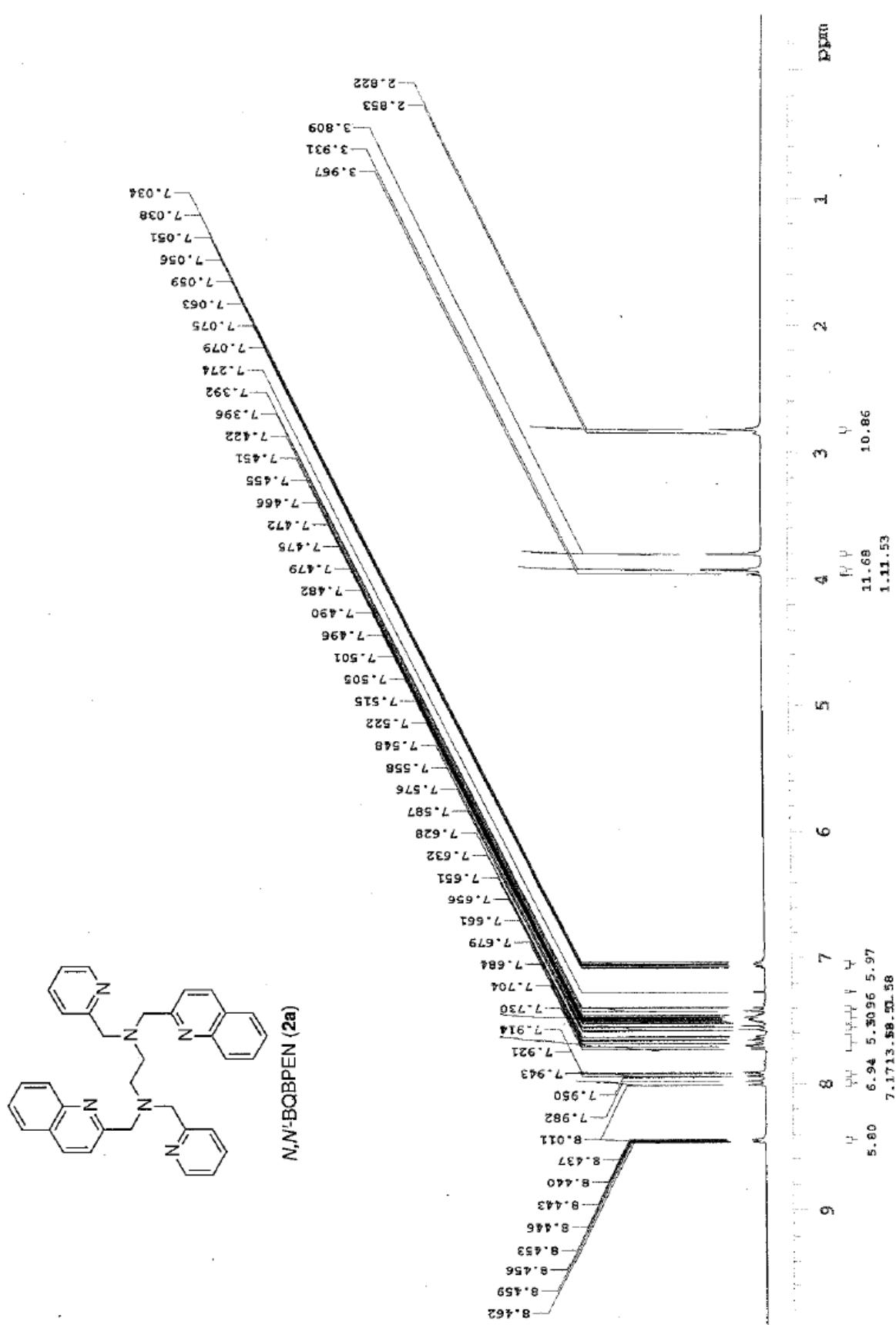
**Fig. S15** Preliminary X-ray analysis of  $[\text{Zn}(\mathbf{1a})](\text{ClO}_4)_2$  in 50% probability. One of the crystallographically independent molecules, hydrogen atoms and counter anions were omitted for clarity. Crystal data for  $\text{C}_{34}\text{H}_{32}\text{Cl}_2\text{N}_6\text{O}_8\text{Zn}$ : monoclinic, space group  $P2_1/c$ ,  $a = 12.450(11)$  Å,  $b = 13.933(13)$  Å,  $c = 38.54(3)$  Å,  $\beta = 92.234(10)$  °,  $V = 6681(10)$  Å<sup>3</sup>,  $Z = 8$ ,  $R1 = 0.250$ ,  $wR2 = 0.598$ , GOF = 1.937.



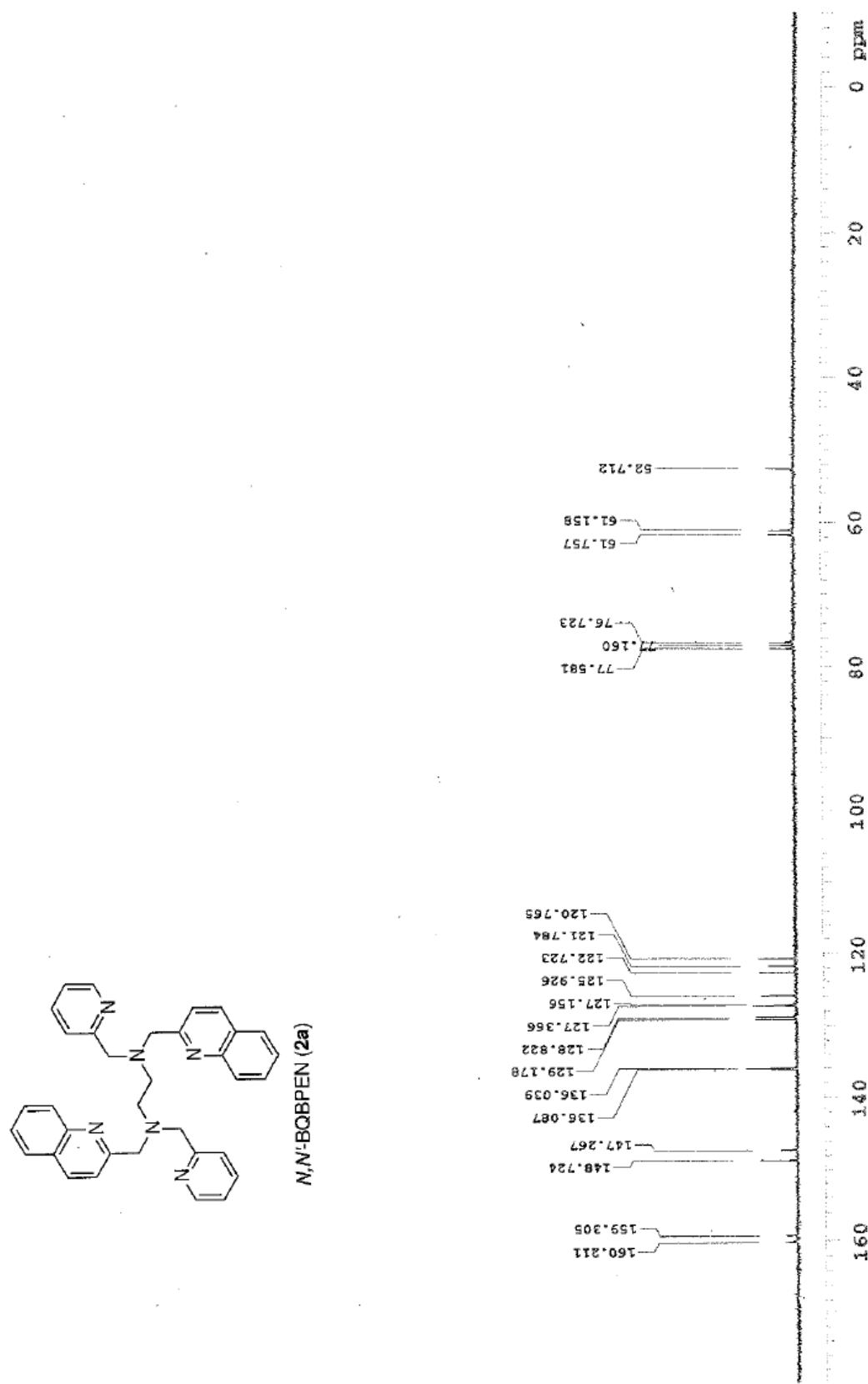
**Fig. S16**  $^1\text{H}$  NMR spectrum of *N,N*-BQBPEN (**1a**) in  $\text{CDCl}_3$ .



**Fig. S17**  $^{13}\text{C}$  NMR spectrum of *N,N*-BQBPEN (**1a**) in  $\text{CDCl}_3$ .



**Fig. S18** <sup>1</sup>H NMR spectrum of *N,N'*-BQBPEN (**2a**) in CDCl<sub>3</sub>.



**Fig. S19**  $^{13}\text{C}$  NMR spectrum of *N,N'*-BQBPEN (**2a**) in  $\text{CDCl}_3$ .

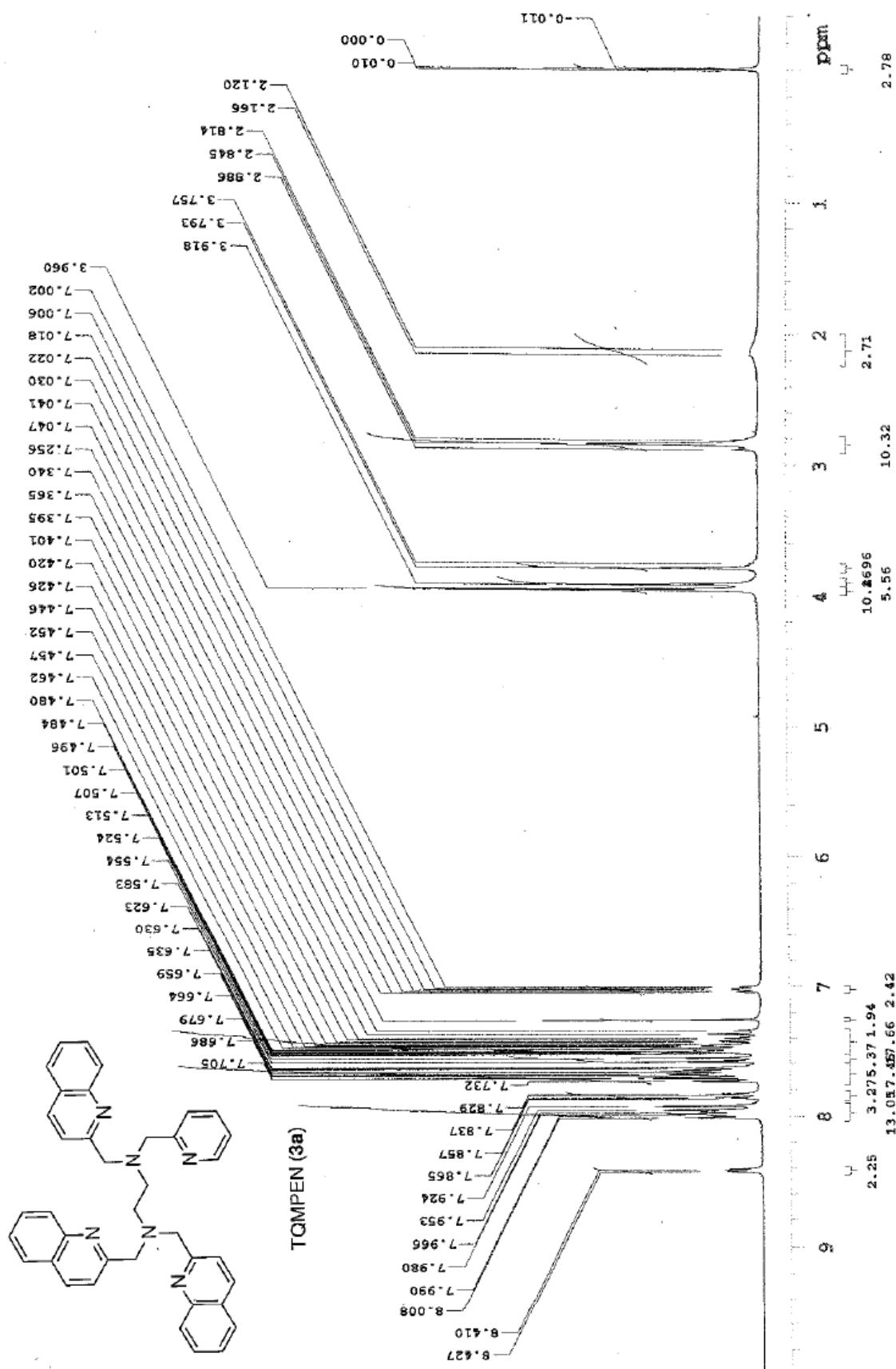
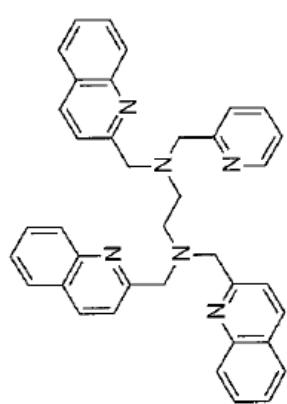
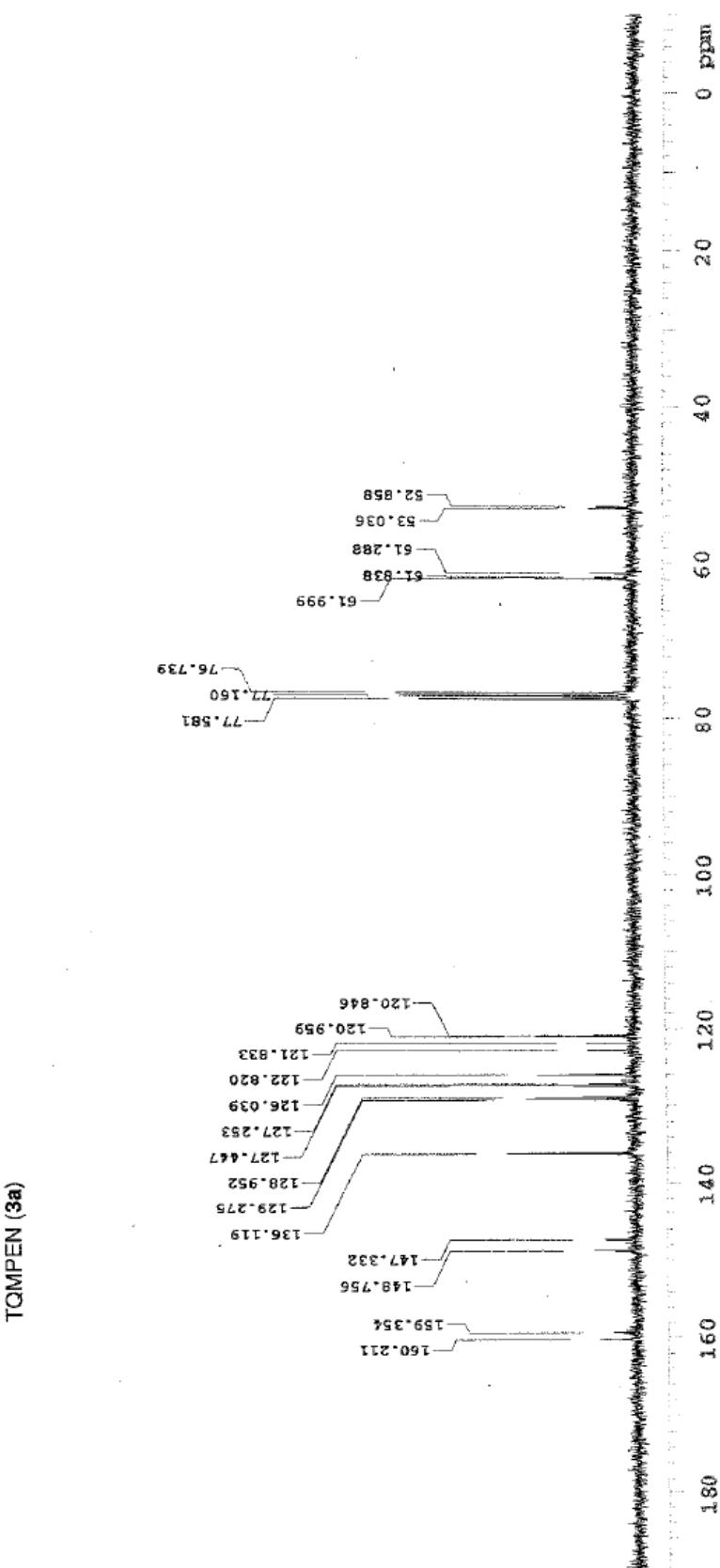
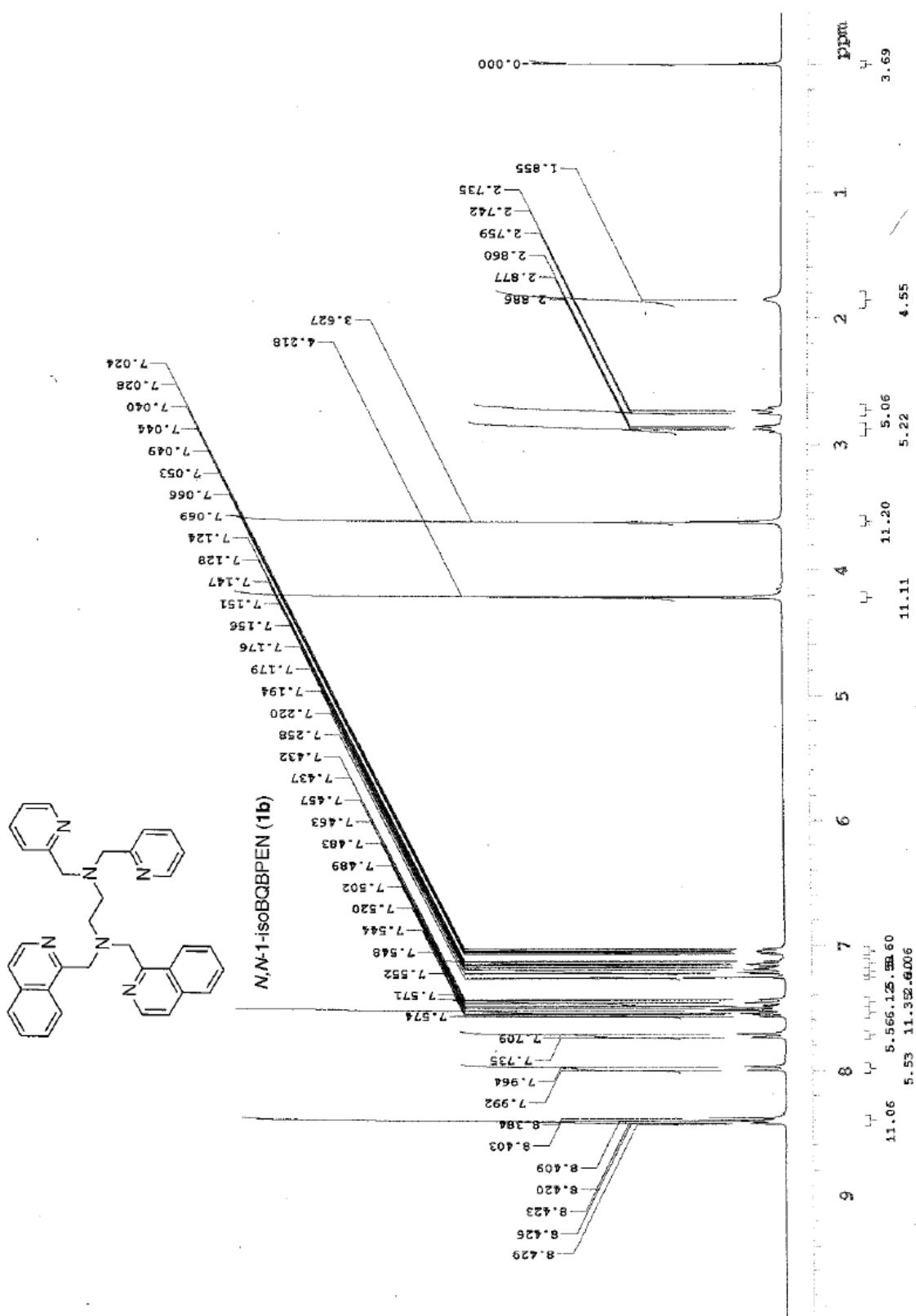


Fig. S20  $^1\text{H}$  NMR spectrum of TQMPEN (3a) in  $\text{CDCl}_3$ .

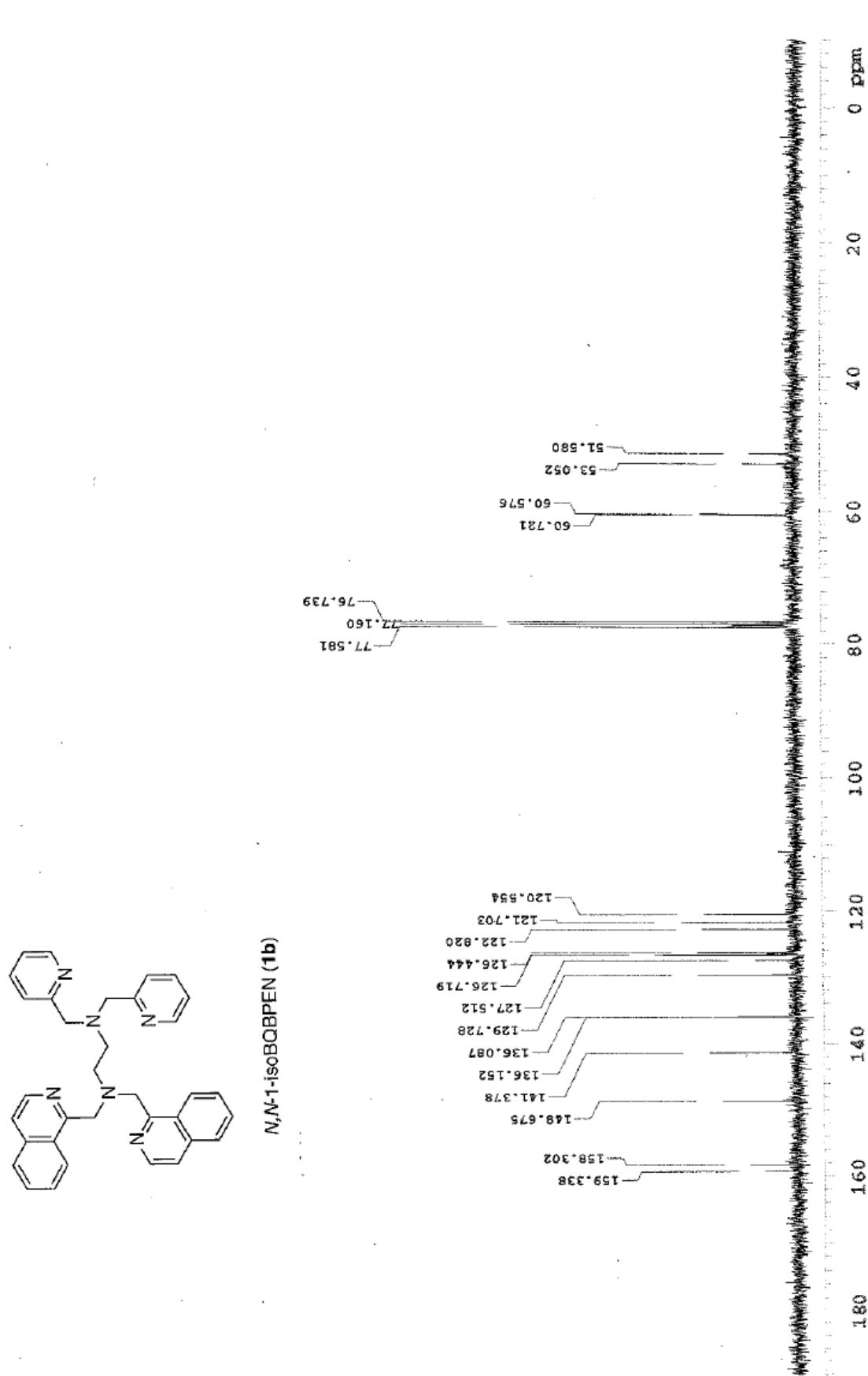


**Fig. S21**  $^{13}\text{C}$  NMR spectrum of TQMPEN (**3a**) in  $\text{CDCl}_3$ .





**Fig. S22**  $^1\text{H}$  NMR spectrum of *N,N*-1-isoBQBPEN (**1b**) in  $\text{CDCl}_3$ .



**Fig. S23**  $^{13}\text{C}$  NMR spectrum of *N,N*-1-isoBQBPEN (**1b**) in  $\text{CDCl}_3$ .

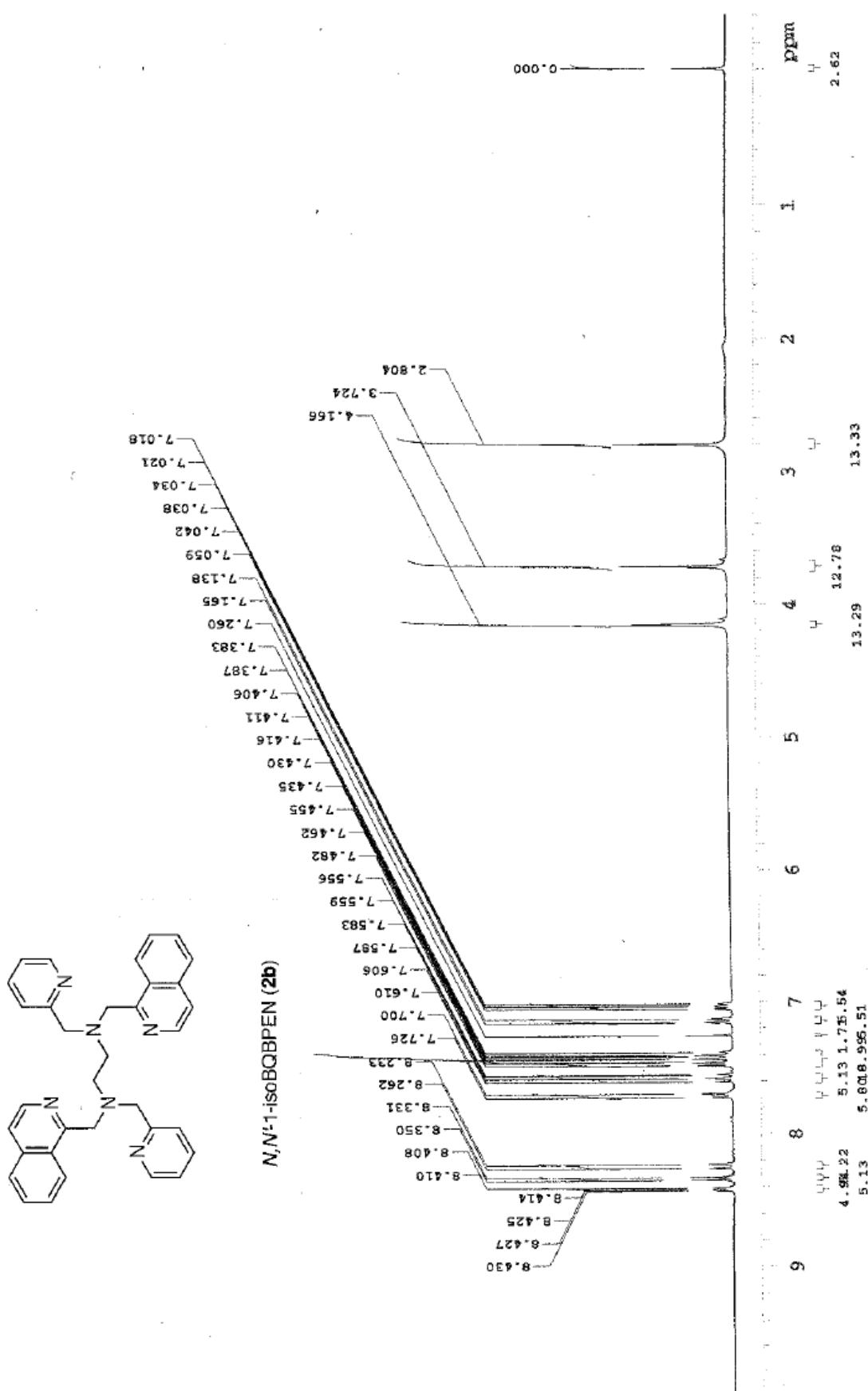
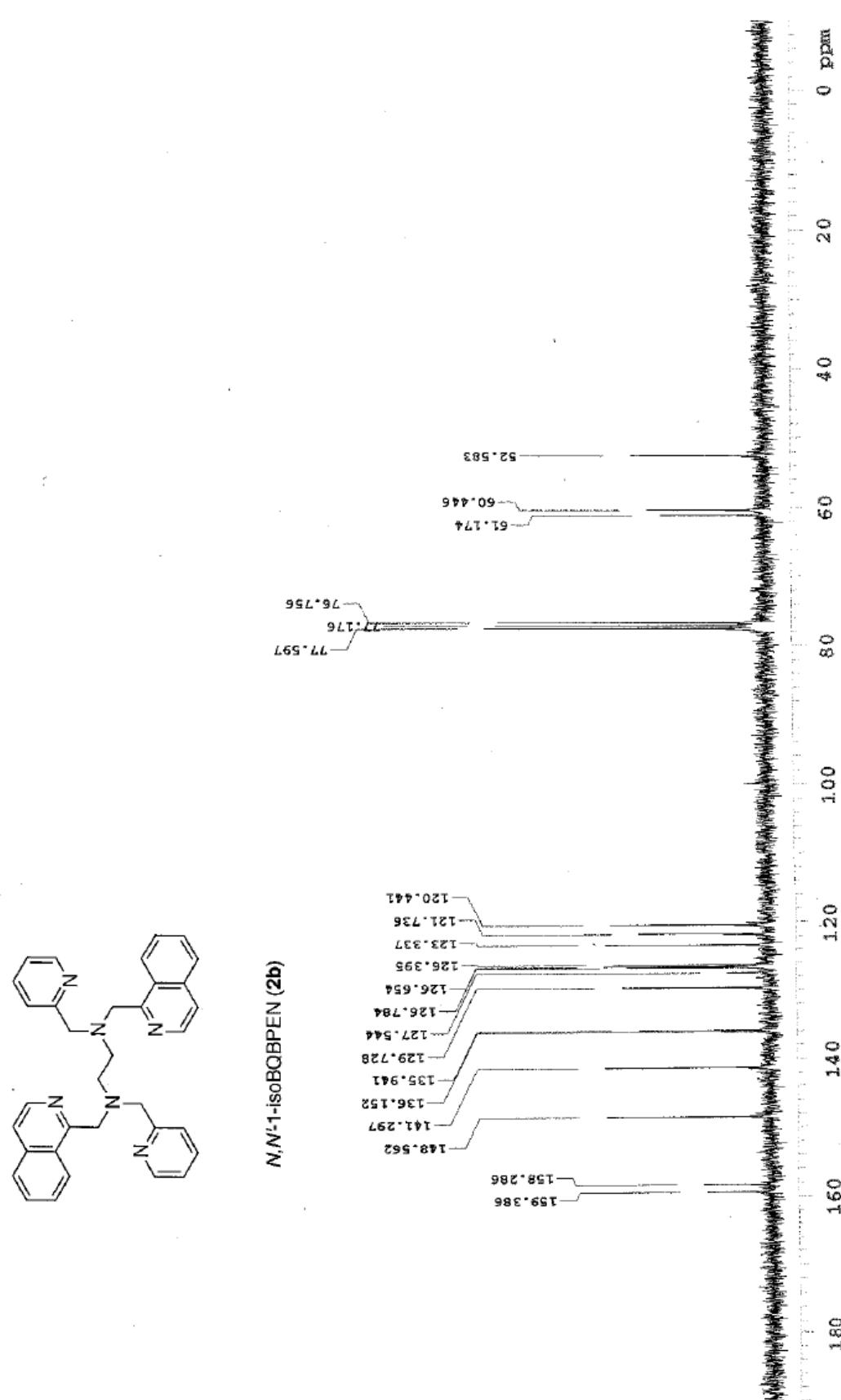


Fig. S24  $^1\text{H}$  NMR spectrum of  $N,N'$ -1-isoBQBPEN (**2b**) in  $\text{CDCl}_3$ .



**Fig. S25**  $^{13}\text{C}$  NMR spectrum of *N,N'*-1-isoBQBPEN (**2b**) in  $\text{CDCl}_3$ .

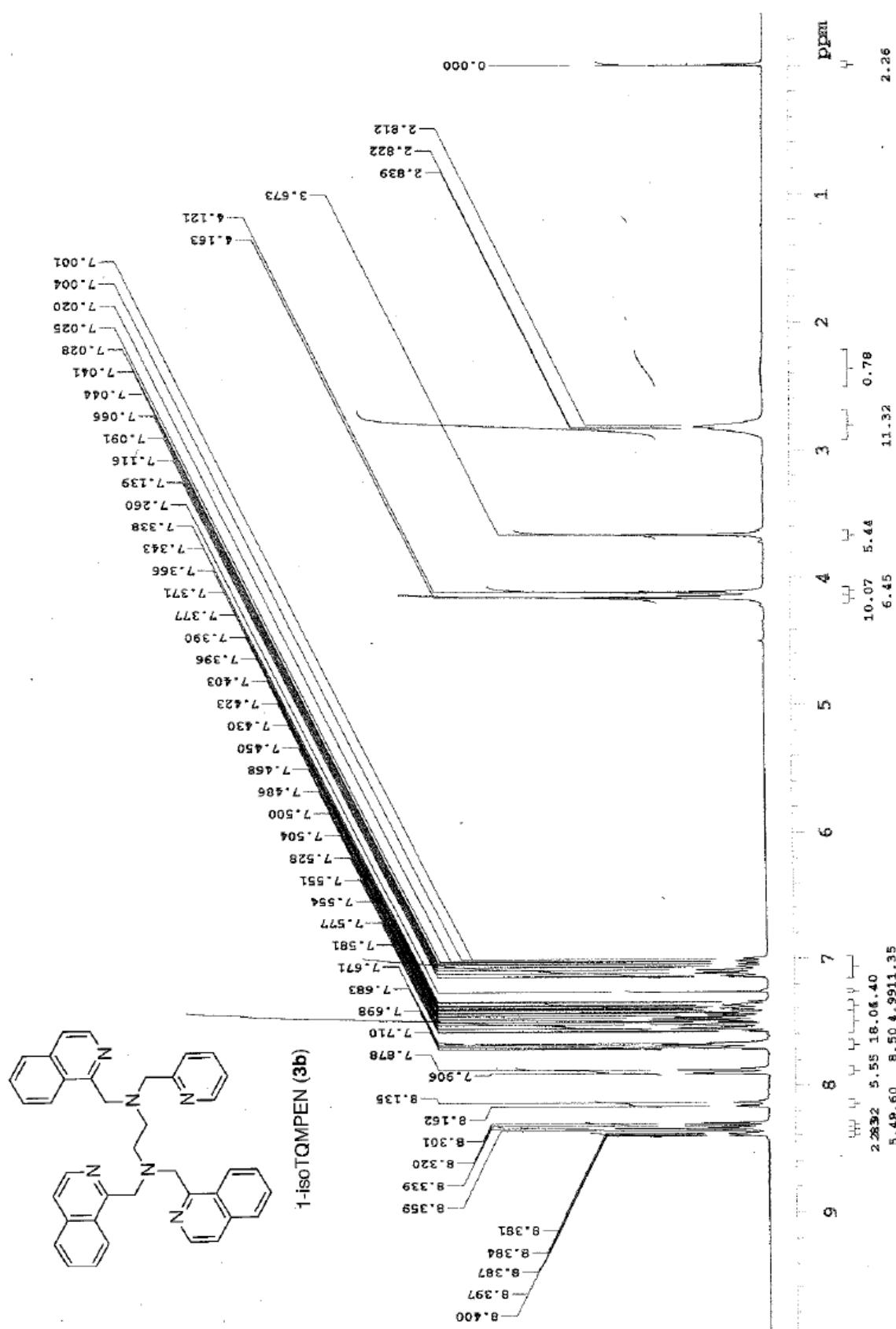
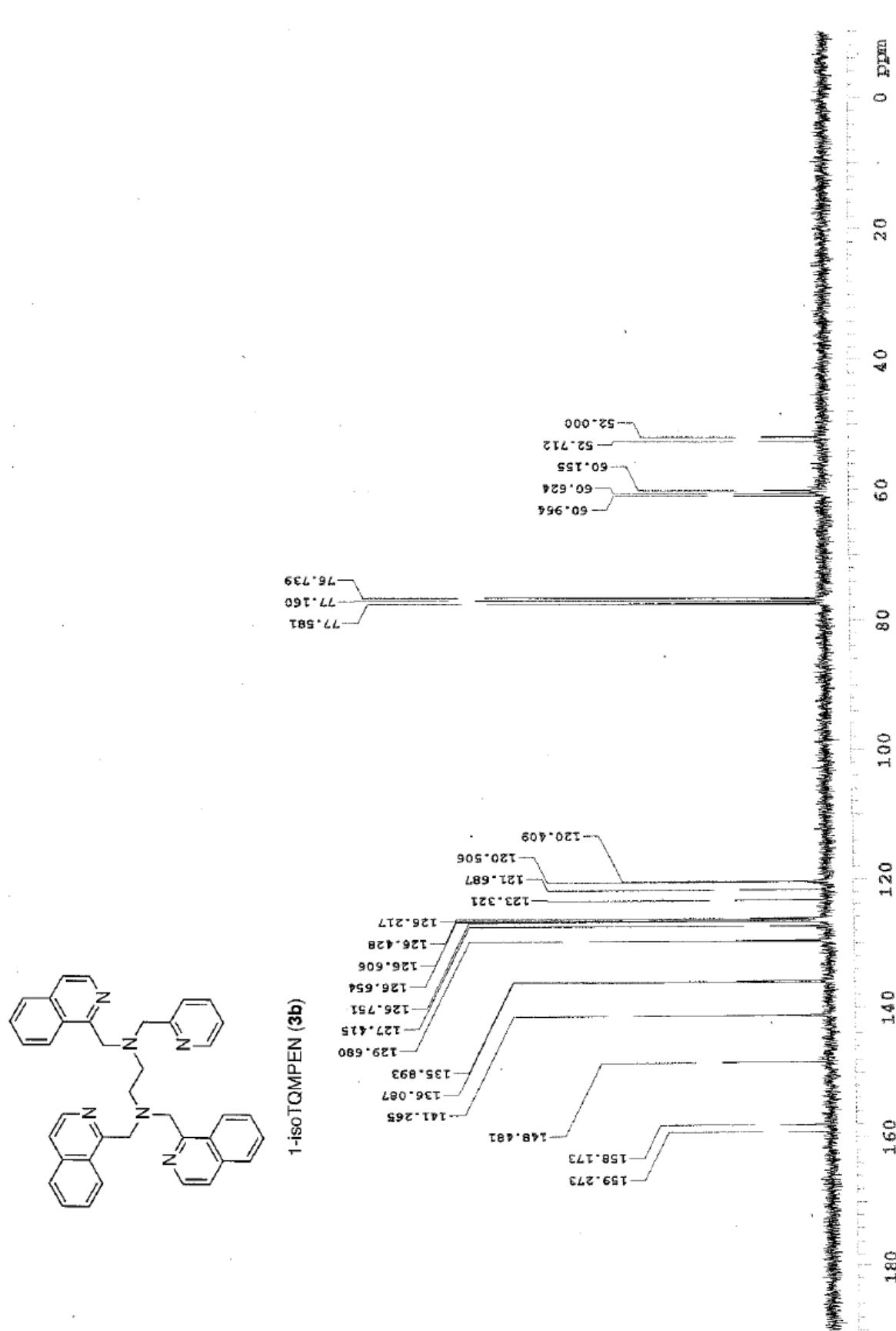
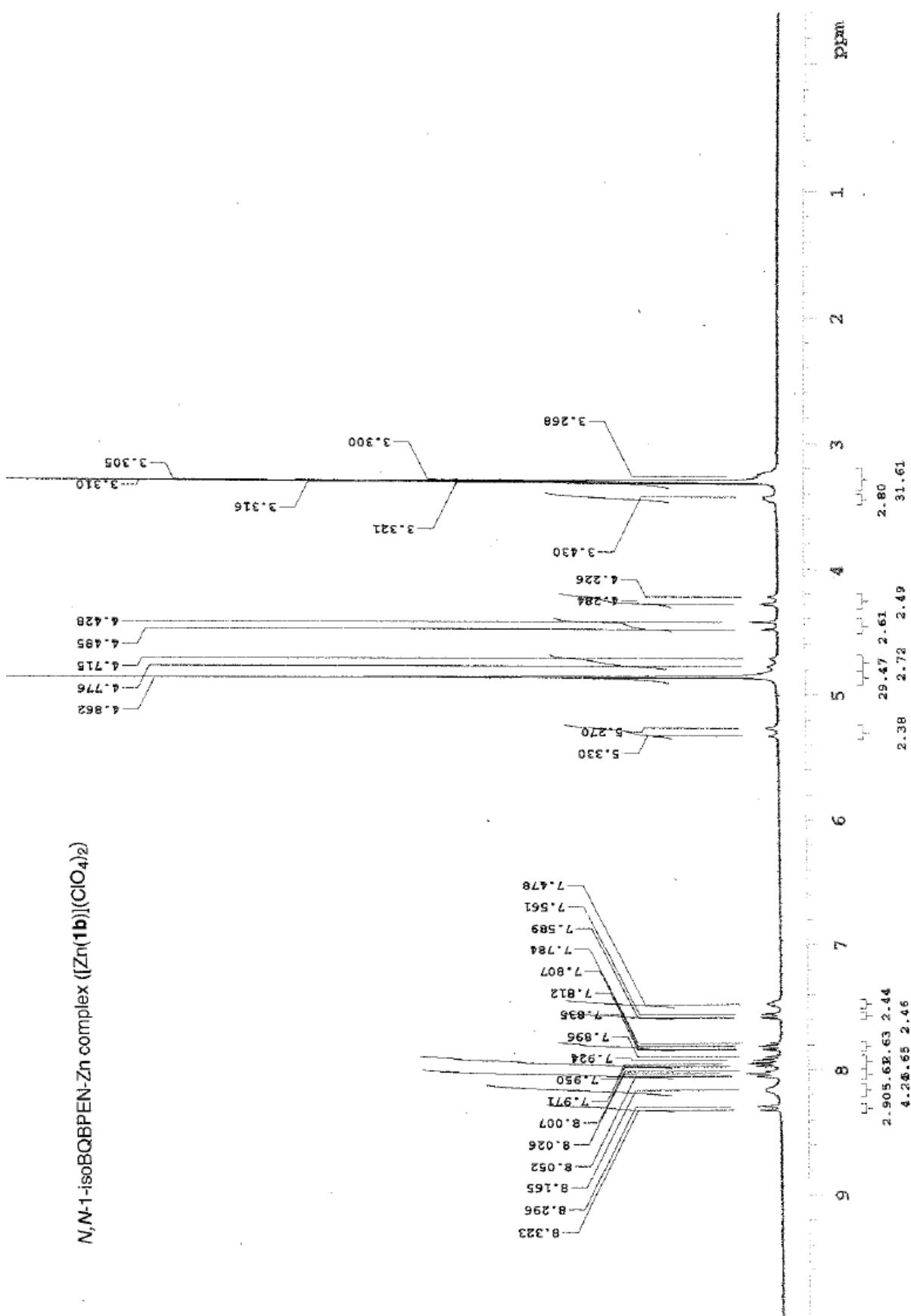


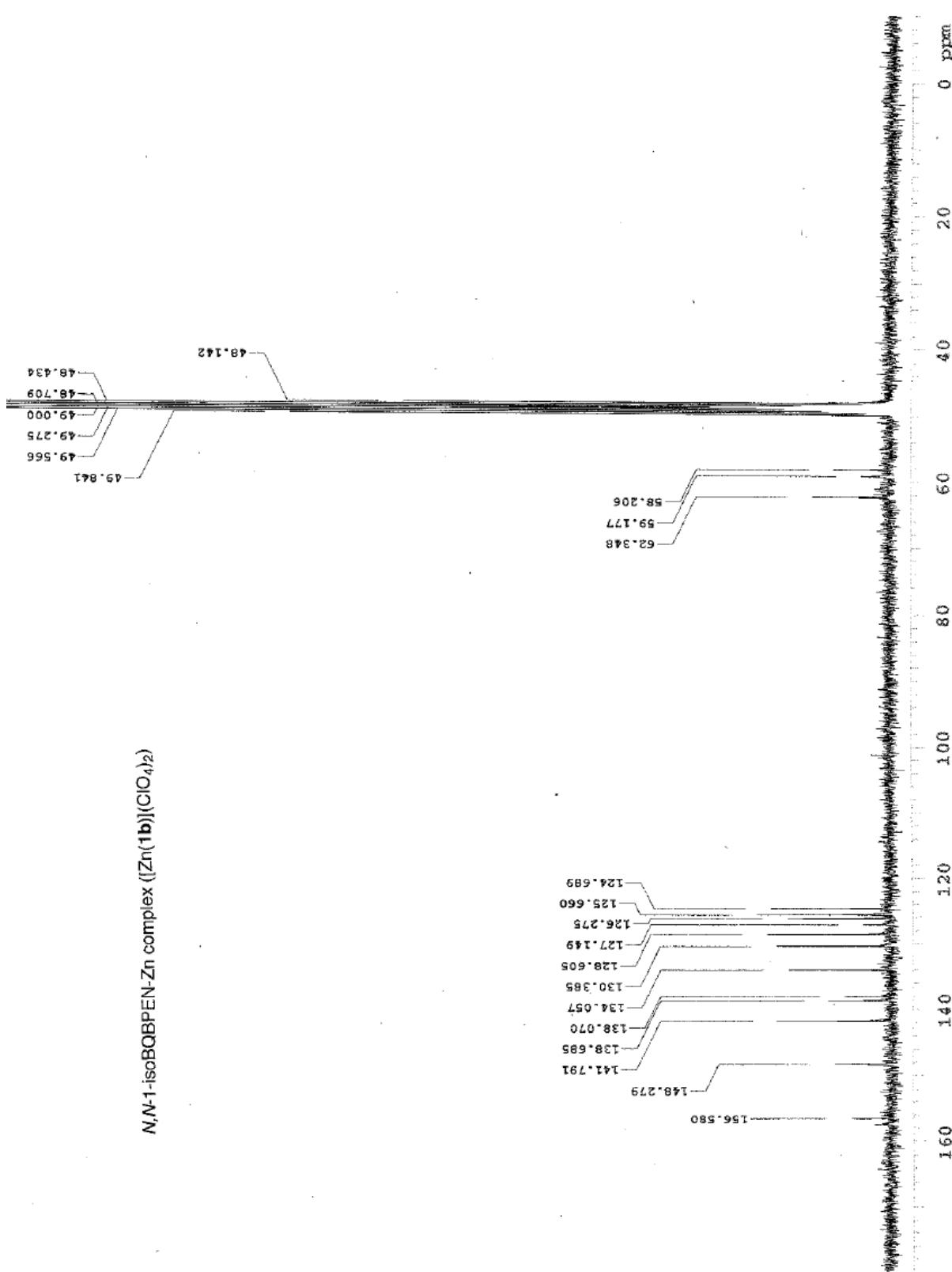
Fig. S26  $^1\text{H}$  NMR spectrum of 1-isoTQMPEN (**3b**) in  $\text{CDCl}_3$ .



**Fig. S27**  $^{13}\text{C}$  NMR spectrum of 1-isoTQMPEN (**3b**) in  $\text{CDCl}_3$ .



**Fig. S28** <sup>1</sup>H NMR spectrum of [Zn(**1b**)][ClO<sub>4</sub>)<sub>2</sub> in CD<sub>3</sub>OD.



**Fig. S29** <sup>13</sup>C NMR spectrum of [Zn(**1b**)](ClO<sub>4</sub>)<sub>2</sub> in CD<sub>3</sub>OD.

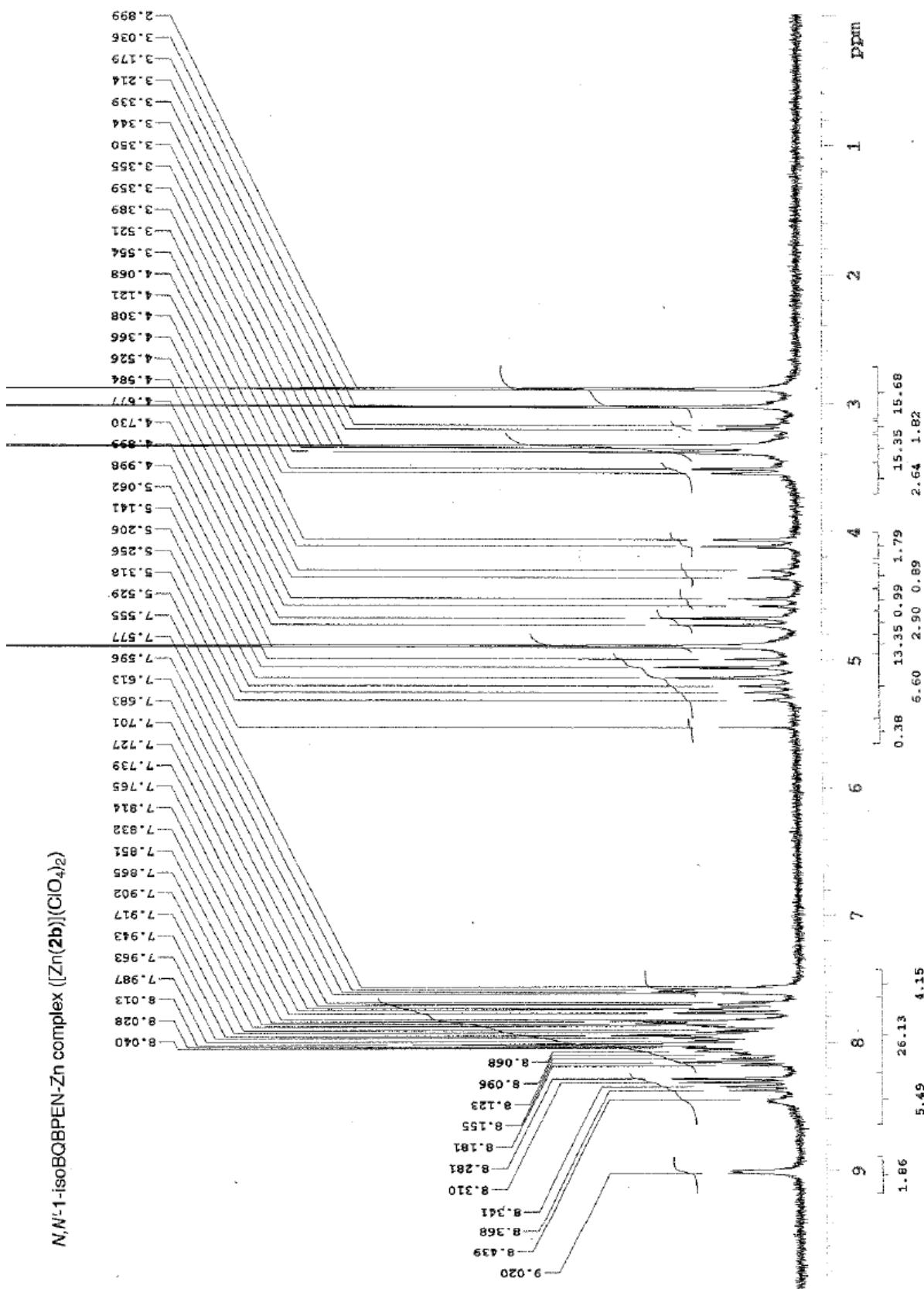


Fig. S30 <sup>1</sup>H NMR spectrum of [Zn(2b)][ClO<sub>4</sub>]<sub>2</sub> in CD<sub>3</sub>OD.

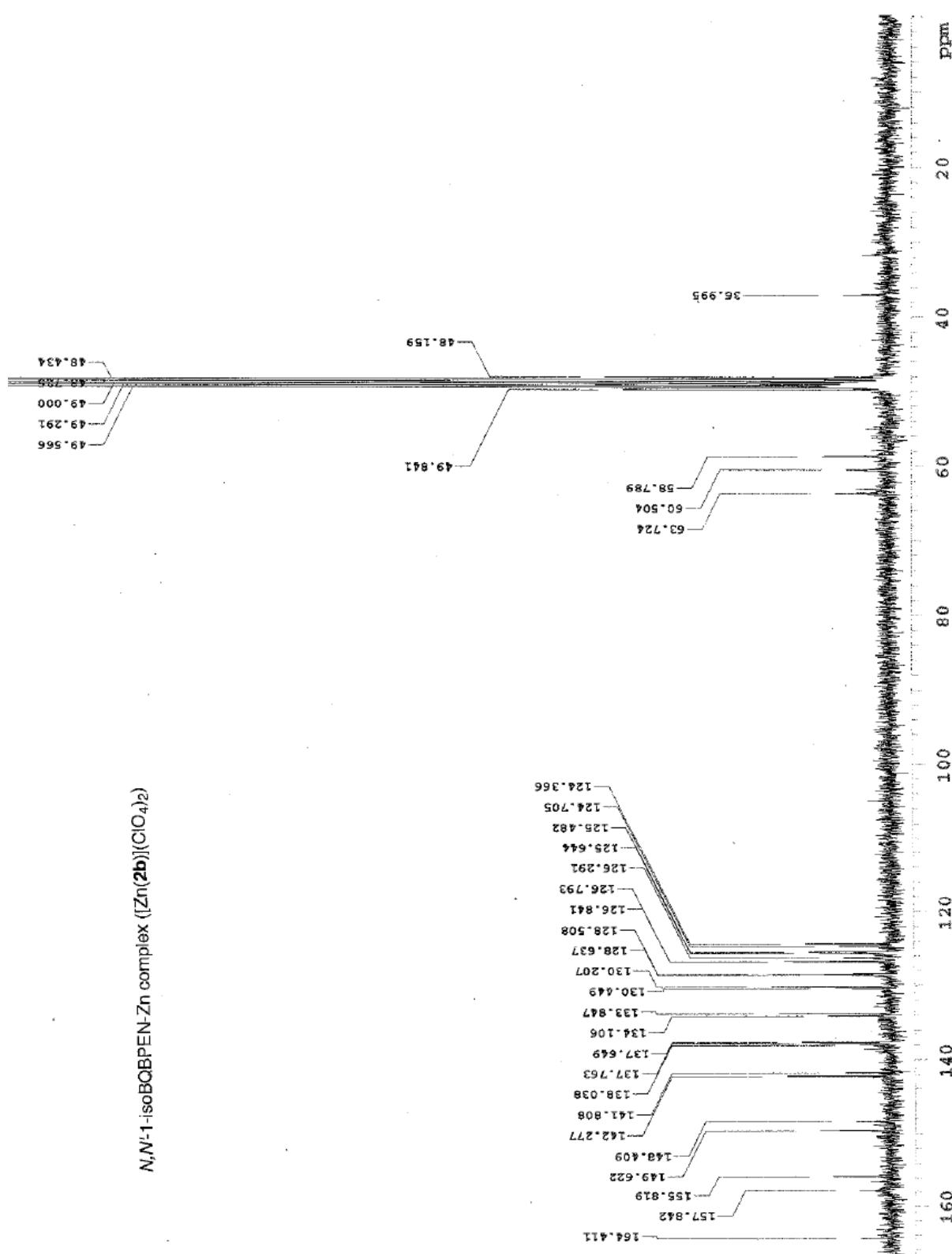


Fig. S31  $^{13}\text{C}$  NMR spectrum of  $[\text{Zn}(\textbf{2b})](\text{ClO}_4)_2$  in  $\text{CD}_3\text{OD}$ .