

**Supporting Information for:**

**Structure and photoluminescence of  
silver(I) trinuclear halopyrazolato complexes**

**Yui Morishima<sup>a</sup>, David James Young<sup>b</sup>, Kiyoshi Fujisawa<sup>a\*</sup>**

<sup>a</sup>Department of Chemistry, Ibaraki University, Mito  
310-8512, Japan

<sup>b</sup>School of Science, Monash University, Jalan Lagoon Selatan, 47500 Bandar  
Sunway, Selangor D.E., Malaysia

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**Table S1**X-ray crystallographic data for **L1** and **L5** complexes.

	[Ag(L1Cl)] <sub>2</sub> · 0.5C <sub>6</sub> H <sub>14</sub> · 3 × 0.25H <sub>2</sub> O	[Ag(L1I)]	[Ag(L1I)] <sub>2</sub> · C <sub>6</sub> H <sub>14</sub>	[Ag(L5Cl)] · 2CH <sub>2</sub> Cl <sub>2</sub>
Formula	C <sub>57</sub> H <sub>92.5</sub> Ag <sub>6</sub> Cl <sub>6</sub> N <sub>12</sub> O <sub>0.75</sub>	C <sub>27</sub> H <sub>42</sub> Ag <sub>3</sub> I <sub>3</sub> N <sub>6</sub>	C <sub>60</sub> H <sub>98</sub> Ag <sub>6</sub> I <sub>6</sub> N <sub>12</sub>	C <sub>47</sub> H <sub>34</sub> Ag <sub>3</sub> Cl <sub>7</sub> N <sub>6</sub>
Formula weight	1817.86	1154.99	2396.15	1254.60
Crystal system	Monoclinic	Orthorhombic	Monoclinic	Orthorhombic
Space group	<i>P2</i> <sub>1</sub> / <i>c</i> (No. 14)	<i>Pna</i> 2 <sub>1</sub> (No. 33)	<i>P2</i> <sub>1</sub> / <i>c</i> (No. 14)	<i>Pna</i> 2 <sub>1</sub> (No. 33)
<i>a</i> / Å	10.2520(11)	11.5188(10)	14.823(2)	9.544(2)
<i>b</i> / Å	29.412(2)	23.8743(13)	12.7462(11)	20.286(4)
<i>c</i> / Å	27.446(3)	14.0966(7)	20.735(3)	24.142(4)
$\beta$ / °	99.3800(11)	---	90.874(2)	---
<i>V</i> / Å <sup>3</sup>	8165.2(12)	3876.6(5)	3917.1(8)	4674(2)
<i>Z</i>	4	4	2	4
<i>D</i> <sub>calc</sub> / gcm <sup>-3</sup>	1.479	1.979	2.031	1.783
<i>R</i> (int)	0.0512	0.0281	0.0456	0.0780
$\mu$ (Mo <i>K</i> $\alpha$ ) / cm <sup>-1</sup>	16.417	39.099	38.734	16.821
Reflections collected	63254	30308	28371	37740
Unique reflections	18485	4590	8762	8670
No. of observations	7487 ( <i>I</i> > 3 $\sigma$ ( <i>I</i> ))	4145 ( <i>I</i> > 3 $\sigma$ ( <i>I</i> ))	6953 ( <i>I</i> > 3 $\sigma$ ( <i>I</i> ))	5024 ( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))
No. of variables	823	395	428	603
Reflections/para. ratio	9.10	10.49	16.25	8.33
<i>R</i> ( <i>I</i> > 3 $\sigma$ ( <i>I</i> ))	0.0603	0.0306	0.0517	0.0390( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))
<i>R</i> <sub>w</sub> ( <i>I</i> > 3 $\sigma$ ( <i>I</i> ))	0.1335	0.0755	0.1282	0.0530 ( <i>I</i> > 2 $\sigma$ ( <i>I</i> ))
Good. of fit indicator	1.854	1.628	2.546	0.567
Flack parameter	---	0.25(8)		0.06(3)
		(Friedel pairs = 3956)		(Friedel pairs = 3170)
Max/min peak, /e Å <sup>3</sup>	1.68 / -0.69	0.89 / -0.59	2.51 / -1.73	1.29 / -1.12

$$R = \sum ||F_o| - |F_c|| / \sum |F_o|; R_w = [(\sum w((F_o^2 - F_c^2)^2) / \sum w(F_o^2)^2)]^{1/2}$$

## Experimental section

### Preparation of Complexes

#### $\{[\text{Ag}(\mu\text{-4-Br-3,5-}i\text{Pr}_2\text{pz})]_3\}_2$ ( $[\text{Ag}(\text{L1Br})]_2$ )<sup>1</sup>

This complex was prepared by a procedure analogous to that used for  $[\text{Ag}(\text{L1Cl})]_2$ , substituting  $\text{NaL1Br}$  (742 mg, 2.93 mmol). Colorless crystals (753 mg, 0.371 mmol, yield 76%) were isolated from hexane at  $-15$  °C. Far-IR (CsI,  $\text{cm}^{-1}$ ): 670w, 658w, 646w, 585m, 555m, 545w (C–Br), 505m (Ag–N), 405s, 366s, 344w, 242vs, 211s. Raman (solid,  $\text{cm}^{-1}$ ): 2970vs (C–H, aliphatic), 2910vs (C–H, aliphatic), 2868vs (C–H, aliphatic), 2754w, 2713w, 1499m (C=N), 1486m, 1443s, 1431m, 1367vs, 1303m, 1280m, 1181m, 1145m, 1109m, 1091m, 1045m, 957m, 881s, 708w, 647m, 586w, 533s (C–Br), 506m (Ag–N), 401w, 331w, 239m, 186w (Ag...Ag). UV–Vis (solution, cyclohexane,  $\lambda_{\text{max}}/\text{nm}$  ( $\epsilon/\text{cm}^{-1}\text{mol}^{-1}\text{dm}^3$ )): 227 (38100). UV–Vis (solid, nujol, nm): 227.5. Emission (solid, 299 K,  $\lambda_{\text{max}}/\text{nm}$ ): 312, 379, 620 (ex. 280 nm).

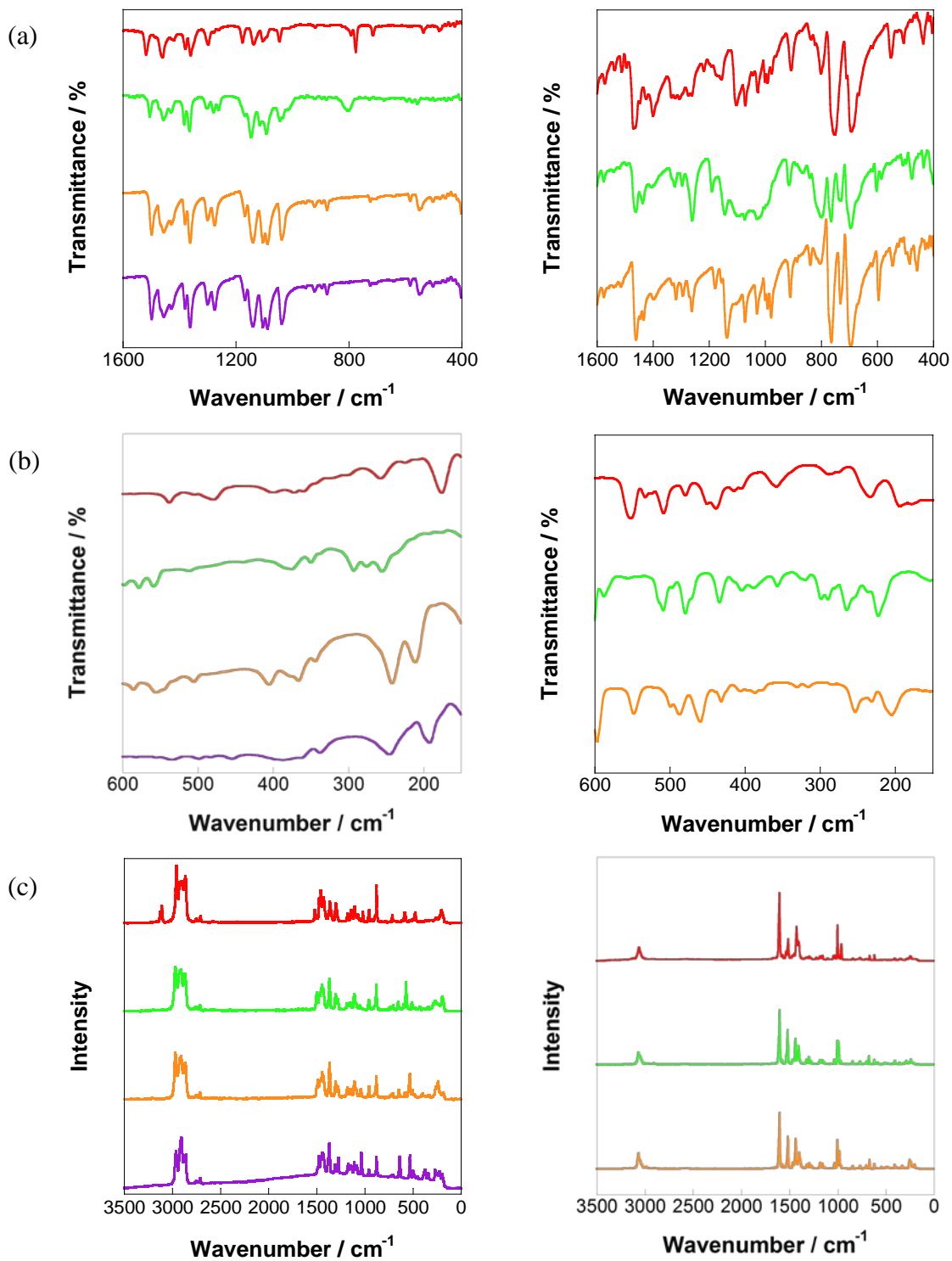
#### $\{[\text{Ag}(\mu\text{-3,5-Ph}_2\text{pz})]_3\}_2$ ( $[\text{Ag}(\text{L5H})]_2$ )<sup>2-3</sup>

A solution of  $\text{NaL5H}$  (688 mg, 2.84 mmol) in tetrahydrofuran ( $20 \text{ cm}^3$ ) was added dropwise to a suspension of silver(I) nitrate (484 mg, 2.85 mmol) in tetrahydrofuran ( $10 \text{ cm}^3$ ) over 15 min. The reaction was protected from light and after stirring for a few days, the solvent was evaporated under vacuum. The resulting solid was extracted with dichloromethane ( $40 \text{ cm}^3$ ) to remove sodium nitrate. The filtrate was concentrated under vacuum to provide a white powder that was recrystallized from dichloromethane at  $-15$  °C. These colorless crystals (615 mg, 0.313 mmol, yield 66%) were filtered and dried under vacuum. IR (KBr,  $\text{cm}^{-1}$ ): 3061m (C–H, aliphatic), 3028m (C–H, aliphatic), 1943w, 1870w, 1799w, 1748w, 1603m, 1573m, 1538w, 1512w, 1496w, 1470vs (C=N), 1425m, 1400s, 1335m, 1322m, 1306m, 1275m, 1220w, 1156m, 1103s, 1071s, 1028m, 1002m, 992m, 978w, 908m, 801m, 752vs, 693vs, 553m, 507w, 437w. Far-IR (CsI,  $\text{cm}^{-1}$ ): 618m, 553vs, 533m, 526w, 508vs (Ag–N), 479m, 450m, 439s, 415m, 405m, 359m, 288m, 233m, 194s, 178ws, 159w, 152w. Raman (solid,  $\text{cm}^{-1}$ ): 3124m (C–H, aromatic), 3061m (C–H, aromatic), 1606vs, 1574w, 1537w, 1516m, 1464w (C=N), 1451w, 1428s, 1413m, 1404m, 1322w, 1296w, 1223w, 1182w, 1158w, 1102w, 1046w, 1030w, 1002s, 963m, 843w, 768w, 697w, 669w, 619w, 510w (Ag–N), 407w, 350w, 276w, 244w (Ag...Ag). <sup>1</sup>H-NMR ( $\text{CDCl}_3$ , 500 MHz):  $\delta$  6.77 (s, 6H, pz-*H*), 7.15 (t,  $J = 7$  Hz, 24H, *m*-Ph), 7.26 (t,  $J = 7$  Hz, 12H,

*p*-Ph), 7.65 (d,  $J = 3.5$  Hz, 24H, *o*-Ph).  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 125 MHz):  $\delta$  101.7 (*pz-C*<sub>4</sub>), 126.7 (*o*-Ph), 128.2 (*p*-Ph), 129.2 (*m*-Ph), 133.8 (Ph-*C*<sub>1</sub>), 155.5 (*pz-C*<sub>3,5</sub>). UV–Vis (solution,  $\text{CH}_2\text{Cl}_2$ ,  $\lambda_{\text{max}}/\text{nm}$  ( $\epsilon/\text{cm}^{-1}\text{mol}^{-1}\text{dm}^3$ )): 253.5 (103000). UV–Vis (solid, nujol, nm): 257. Emission (solid, 299 K,  $\lambda_{\text{max}}/\text{nm}$ ): 329, 649 (ex. 280 nm).

#### **[Ag( $\mu$ -4-Br-3,5-Ph<sub>2</sub>pz)]<sub>3</sub> ([Ag(L5Br)]<sup>4</sup>)**

This complex was prepared by a procedure analogous to that used for [Ag(L5H)], substituting NaL5Br (198 mg, 0.617 mmol). Colorless crystals (102.8 mg, 0.0422 mmol, yield 41%) were isolated from dichloromethane at  $-15$  °C. IR (KBr,  $\text{cm}^{-1}$ ): 3063m (C–H, aromatic), 3023m (C–H, aromatic), 2963w, 1948w, 1876w, 1802w, 1750w, 1603m, 1577w, 1462s (C=N), 1435m, 1319w, 1294w, 1262m, 1179w, 1137s, 1072m, 1030m, 992m, 980m, 911m, 840w, 805w, 765vs, 733m, 695vs, 596m, 547w, 486w, 459m, 429w, 409w. Far-IR (CsI,  $\text{cm}^{-1}$ ): 669w, 618m, 597vs, 549s (C–Br), 499w, 488s (Ag–N), 460s, 432m, 406w, 387w, 331w, 316w, 284w, 254s, 231w, 205s, 163w, 153w. Raman (solid,  $\text{cm}^{-1}$ ): 3066m (C–H, aromatic), 2986w, 1605vs, 1577w, 1553w, 1518s, 1465w (C=N), 1437s, 1402m, 1322w, 1296w, 1274w, 1181w, 1158w, 1034w, 1003s, 982m, 843w, 767w, 703w, 670m, 619w, 546w (C–Br), 488w (Ag–N), 408w, 331w, 253m, 241w, 221w, 202w.  $^{13}\text{C}$ -NMR ( $\text{CDCl}_3$ , 125 MHz):  $\delta$  128.2 (*o*-Ph), 128.9 (*m*-Ph). UV–Vis (solution,  $\text{CH}_2\text{Cl}_2$ ,  $\lambda_{\text{max}}/\text{nm}$  ( $\epsilon/\text{cm}^{-1}\text{mol}^{-1}\text{dm}^3$ )): 242 (86000). UV–Vis (solid, nujol, nm): 252.5. Emission (solid, 299 K,  $\lambda_{\text{max}}/\text{nm}$ ): 326.5, 449, 647 (ex. 280 nm).



**Fig. S1** IR, far-IR, Raman spectra (left: **L1** complexes, right: **L5** complexes; (a) IR, (b) far-IR, (c) Raman; red: H, green: Cl, orange: Br, purple: I).

**Table S2**

Emission wavelengths (nm) at variable temperatures and 280 nm irradiation.

<b>[Ag(L1H)]<sub>2</sub></b>	299 K	315.5, 493, 626.5
	173 K	314.5, 507, 623
	83 K	315, 329, 344.5, 361, 379, 397.5, 417, 620, 657, 687, 719.5
<b>[Ag(L1Cl)]<sub>2</sub></b>	299 K	312, 373.5, 620, 721
	173 K	313, 365, 620.5, 705
	83 K	310.5, 358.5, 469.5, 618.5, 693.5
<b>[Ag(L1Br)]<sub>2</sub></b>	299 K	312, 379, 620
	173 K	313.5, 371, 469, 620.5, 723
	83 K	313, 362.5, 479, 618.5, 712.5
<b>[Ag(L1I)]<sub>n</sub></b>	299 K	314, 363.5, 621, 700
	173 K	313.5, 358, 620, 688
	83 K	313, 354, 481, 535, 616.5, 682
<b>[Ag(L5H)]<sub>2</sub></b>	299 K	329, 649
	173 K	323.5, 442, 449.5, 469.5, 639
	83 K	322.5, 423.5, 450, 470.5, 633
<b>[Ag(L5Cl)]</b>	299 K	320.5, 634.5
	173 K	315, 455, 475, 626
	83 K	315, 427.5, 454, 474, 623.5
<b>[Ag(L5Br)]</b>	299 K	326.5, 449, 647
	173 K	319, 427, 455, 478.5, 631.5
	83 K	320, 427.5, 454.5, 483, 624

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