Supplementary material

N-Benzoylbenzamidinate Complexes of Aluminium: Highly Efficient Initiators for the Ring-Opening Polymerization of ε -Caprolactone

K. Bakthavachalam, Ashwene Rajagopal and N. Dastagiri Reddy*



Figure S1. ¹H NMR spectra of a reaction between **1** and ε -CL in 1:1 ratio in CDCl₃: A: ε -CL at room temperature. B: Complex **2** at room temperature. C: ε -CL and **2** at room temperature. D and E: ε -CL and **2** at 80 °C after 15 and 30 minutes respectively.



Figure S2. ¹H NMR spectra of a reaction between ketiminate-Al complex and ε -CL in 1:1 ratio in CDCl₃: A: ε -CL at room temperature. B: Ketiminate-Al complex at room temperature. C: ε -CL and ketiminate-Al complex at room temperature. D and E: ε -CL and ketiminate-Al complex at 80 °C after 1 and 2 h respectively.



Figure S3. Plot of number-averaged molecular weight (M_n) and polydispersity index, (PDI) vs [CHO]/[A1] for the polymerization of cyclohexene oxide using complex 1 at room temperature (entries 1 – 4, Table 2). Red squares (**•**) represent M_n (uncorrected) values and bule triangles (**•**) represent PDI values.



Figure S4. Plot of number-averaged molecular weight (M_n) and polydispersity index, (PDI) vs [CHO]/[A1] for the polymerization of cyclohexene oxide using complex **2** at room temperature (entries 4 – 8, Table 2). Red squares (**•**) represent M_n (uncorrected) values and bule triangles (**•**) represent PDI values.

	1	3	4
empirical formula	C ₂₈ H ₃₃ N ₂ OA1	$C_{53}H_{57}N_4O_2Al$	$C_{60}H_{45}AlN_6O_3$
formula wt	440.54	809.01	925.00
temp (K)	150(2)	150(2)	150(2)
cryst syst	monoclinic	Orthorhombic	triclinic
space group	<i>I2/a</i>	$P2_{1}2_{1}2_{1}$	$P\overline{1}$
<i>a</i> (Å)	25.2618(13)	13.1875(5)	11.3761(4)
<i>b</i> (Å)	11.2363(7)	17.1684(9)	12.4523(5)
<i>c</i> (Å)	22.8425(16)	20.1357(10)	19.8133(7)
α (deg)	90.00	90.00	99.277(3)
β (deg)	106.040(7)	90.00	96.531(3)
γ (deg)	90.00	90.00	109.381(3)
$V(Å^3)$	4998.0(6)	4558.9(4)	2570.54(16)
Ζ	8	4	2
$ ho_{ m calcd}$ (Mg m ⁻³)	1.171	1.179	1.195
μ (mm ⁻¹)	0.103	0.089	0.090
<i>F</i> (000)	1888.0	1728.0	968.0
cryst size (mm)	$0.3\times0.3\times0.1$	$0.42 \times 0.21 \times 0.2$	$0.35 \times 0.31 \times 0.3$
θ range (deg)	2.73-29.22	2.55–25	2.96–25
no.of collected/unique rflns	17036 /5948	14429 /7812	23555/9060
	(R(int) = 0.0477)	(R(int) = 0.0610)	[R(int) = 0.0584]
no.ofdata/restraints/ params	5948/0/295	7812/0/550	9060/0/631
$R1, wR2 (I > 2\sigma(I))^a$	0.0532, 0.1386	0.0632, 0.0784	0.0525, 0.11136
<i>R</i> 1, <i>wR</i> 2 (all data) ^{<i>a</i>}	0.0702, 0.1537	0.1186, 0.0952	0.0937, 0.1339
GOF	1.040	0.991	0.905
$\Delta ho_{\rm max}/\Delta ho_{\rm min}~({\rm e~\AA^{-3}})$	0.53/-0.47	0.24/-0.31	0.22/-0.32

 Table S1. Crystal Data for Compounds 1, 3 and 4

 $\overline{{}^{a}R1 = \Sigma ||F_{o}| - |F_{c}|| / \Sigma |F_{o}|; wR2} = [\Sigma w (F_{o}^{2} - F_{c}^{2})^{2} / \Sigma w (F_{o}^{2})^{2}]^{0.5}.$