

ESI

for

Phosphine Complexes of Aluminium(III) Halides – Preparation and Structural and Spectroscopic Systematics

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Table S1. X-ray crystallographic data

Compound	$[o\text{-C}_6\text{H}_4(\text{PMe}_2)(\text{PHMe}_2)]_2$ [AlCl ₄][Cl]	$[o\text{-C}_6\text{H}_4(\text{PMe}_2)(\text{PHMe}_2)]_2$ [AlBr ₄][Br]	[Cy ₂ HP(CH ₂) ₂ PHCy ₂] [AlCl ₄] ₂
Formula	C ₂₀ H ₃₄ AlCl ₅ P ₄	C ₂₀ H ₃₄ AlBr ₅ P ₄	C ₂₆ H ₅₀ Al ₂ Cl ₈ P ₂
<i>M</i>	602.58	824.88	762.16
Crystal system	monoclinic	monoclinic	monoclinic
Space group	C2/c (no. 15)	C2/c (no. 15)	P2 ₁ /c (no. 14)
<i>a</i> /Å	14.599(5)	14.834(7)	7.073(6)
<i>b</i> /Å	20.672(7)	21.144(8)	17.551(13)
<i>c</i> /Å	11.006(5)	11.274(5)	14.924(13)
α /°	90	90	90
β /°	115.061(5)	115.646(6)	101.682(15)
γ /°	90	90	90
<i>U</i> /Å ³	3009(2)	3188(2)	1814(3)
<i>Z</i>	4	4	2
μ (Mo-K α)/mm ⁻¹	0.733	6.540	0.775
<i>F</i> (000)	1248	1608	796
Total number reflns	18200	13640	10205
Unique reflns	3461	3631	3545
<i>R</i> _{int}	0.0930	0.0306	0.0741
No. of params, restraints	146, 0	146, 0	175, 0
<i>R</i> ₁ , <i>wR</i> ₂ [<i>I</i> > 2σ(<i>I</i>)] ^b	0.0640, 0.1041	0.0321, 0.0536	0.0884, 0.1658
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0859, 0.1111	0.0382, 0.0555	0.1146, 0.1786

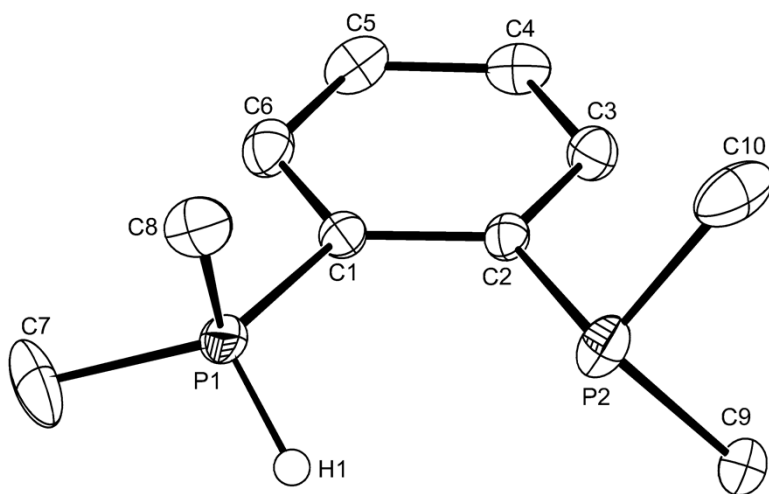


Fig. S1. The structure of the cation in $[o\text{-C}_6\text{H}_4(\text{PMe}_2)(\text{PHMe}_2)]_2[\text{AlCl}_4][\text{Cl}]$ showing the atom labelling scheme. Ellipsoids are drawn at the 50% probability level and H atoms bonded to C are omitted for clarity. Selected bond lengths (\AA): P1—C7 = 1.777(4), P1—C8 = 1.783(3), P1—C1 = 1.799(3), P1—H1 = 1.29(3), P2—C10 = 1.836(4), P2—C9 = 1.838(3), P2—C2 = 1.845(3). The cation in the isomorphous $[o\text{-C}_6\text{H}_4(\text{PMe}_2)(\text{PHMe}_2)]_2[\text{AlBr}_4][\text{Br}]$ has effectively identical dimensions.

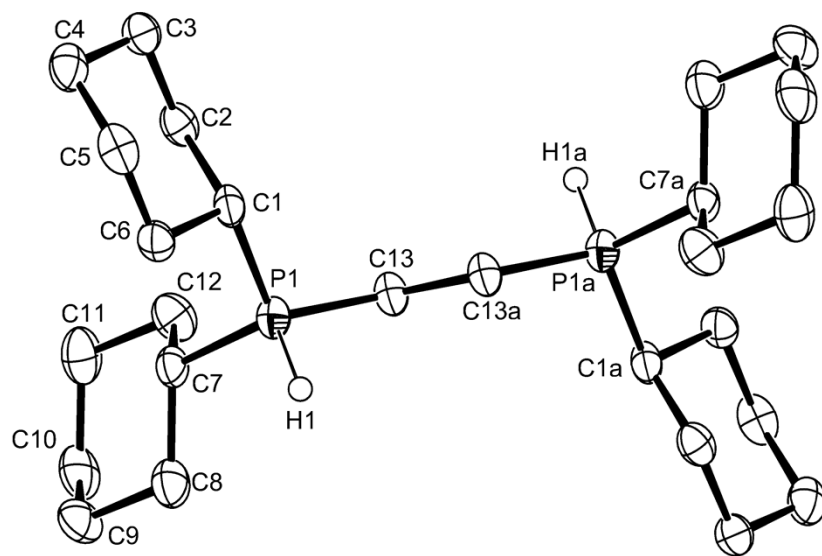


Fig. S2. The structure of the centrosymmetric cation in $[\text{Cy}_2\text{HP}(\text{CH}_2)_2\text{PHCy}_2][\text{AlCl}_4]_2$ showing the atom labelling scheme. Ellipsoids are drawn at the 50% probability level and H atoms bonded to C are omitted for clarity. Symmetry operation: $a = 1 - x, 1 - y, 2 - z$. Selected bond lengths (\AA): P1—C13 = 1.811(6), P1—C1 = 1.812(6), P1—C7 = 1.822(6), P1—H1 = 1.36(6).

These phosphonium salts were formed either by trace hydrolysis or as by-products (through reaction with the solvent) of the reactions of the diphosphanes with AlX_3 as described in the main text. The spectroscopic data are listed below:-

$[\text{Cy}_2\text{HP}(\text{CH}_2)_2\text{PHCy}_2][\text{AlCl}_4]_2$: ^{31}P NMR (CD_2Cl_2 , 295 K): $\delta = 26.8$ (d, $^1J_{\text{PH}} = 462$ Hz).

$[\text{o-C}_6\text{H}_4(\text{PMe}_2)(\text{PHMe}_2)]_2[\text{AlCl}_4][\text{Cl}]$: $^{31}\text{P}\{^1\text{H}\}$ NMR (CD_2Cl_2 , 295 K): $\delta = -4.2$ (d, $^4J_{\text{PP}} = 56$ Hz), -51.9 (d, $^4J_{\text{PP}} = 56$ Hz). ^{31}P NMR: $\delta = -4.2$ (d, $^1J_{\text{PH}} = 540$ Hz).

Analysis of M–X Bond Lengths in $[\text{MX}_4]^-$ (M = Al, Ga, I): Data taken from the Cambridge Structural Database (CSD version 5.34; latest update = May 2013) on the 21st January 2014. The data were analysed in Legacy Vista (version 2.1). The REFCodes used all contained discrete $[\text{MX}_4]^-$ molecules – those containing bonds to other atoms and structures containing any disorder were omitted.

Figure S3. Al–Cl Bond Length Distribution in $[\text{AlCl}_4]^-$

Data points = 915 from 184 REFCodes

Range = 2.055 – 2.216 Å

Mean Al–Cl bond length = 2.126 ± 0.018 Å

Al-Cl Bond Length Distribution in $[\text{AlCl}_4]^-$

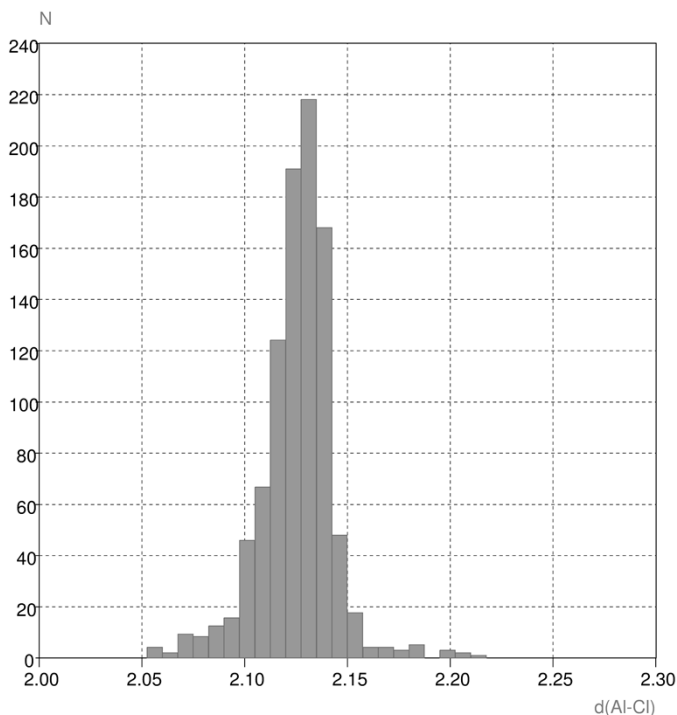


Figure S4. Al–Br Bond Length Distribution in $[\text{AlBr}_4]^-$

Data points = 64 from 13 REFcodes

Range = 2.244 – 2.320 Å

Mean Al–Br bond length = 2.289 ± 0.015 Å

Al-Br Bond Length Distribution in AlBr4-

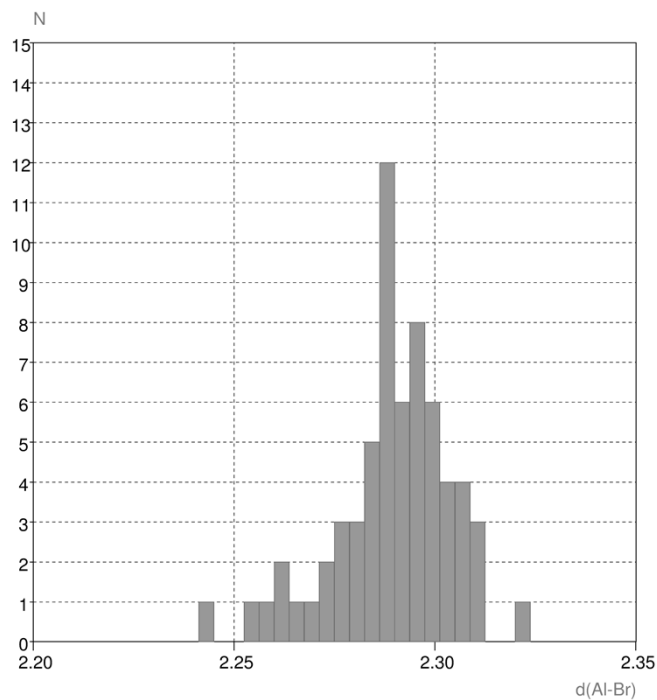


Figure S5. Al–I Bond Length Distribution in $[\text{AlI}_4]^-$

Data points = 12 from 3 REFcodes

Range = 2.506 – 2.555 Å

Mean Al–I bond length = 2.527 ± 0.012 Å

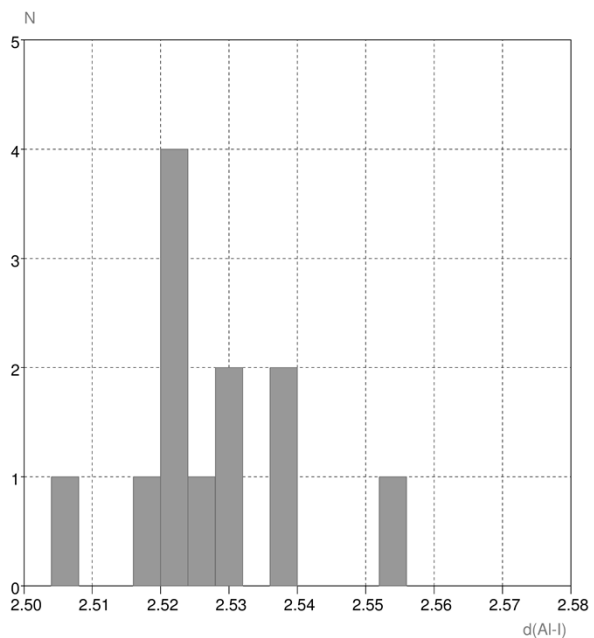


Figure S6. Ga-Cl Bond Length Distribution in $[\text{GaCl}_4]^-$

Data points = 705 from 157 REFcodes

Range = 2.036 – 2.258 Å

Mean Ga-Cl bond length = 2.166 ± 0.018 Å

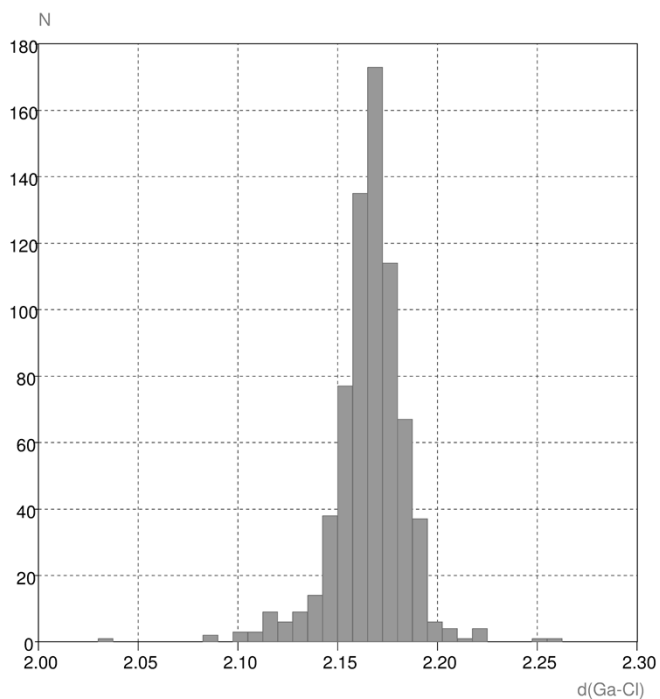


Figure S7. Ga–Br Bond Length Distribution in $[\text{GaBr}_4]^-$

Data points = 91 from 27 REFCodes

Range = 2.277 – 2.361 Å

Mean Ga–Br bond length = 2.321 ± 0.016 Å

Ga-Br Bond Length Distribution in GaBr4-

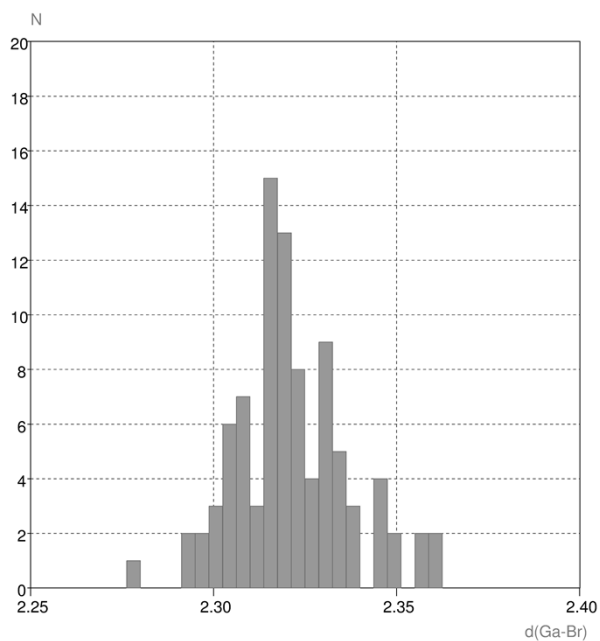


Figure S8. Ga–I Bond Length Distribution in $[\text{GaI}_4]^-$

Data points = 31 from 8 REFCodes

Range = 2.506 – 2.586 Å

Mean Ga–I bond length = 2.544 ± 0.018 Å

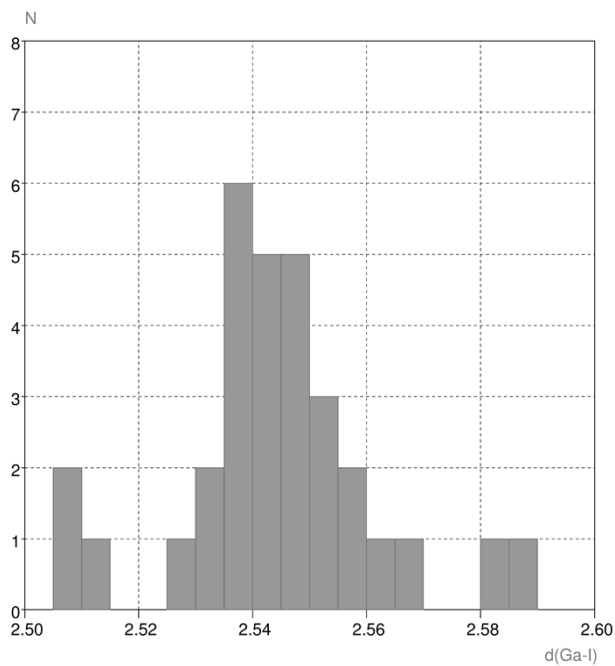


Figure S9. In-Cl Bond Length Distribution in $[\text{InCl}_4]^-$

Data points = 26 from 7 REFcodes

Range = 2.295 – 2.368 Å

Mean In-Cl bond length = 2.337 ± 0.022 Å

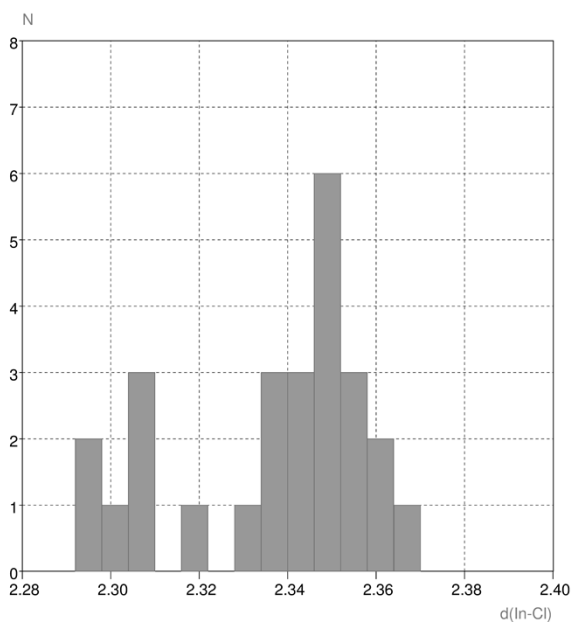


Figure S10. In–Br Bond Length Distribution in $[\text{InBr}_4]^-$

Data points = 50 from 11 REFcodes

Range = 2.438 – 2.510 Å

Mean In–Br bond length = 2.484 ± 0.018 Å

In-Br Bond Length Distribution for InBr4-

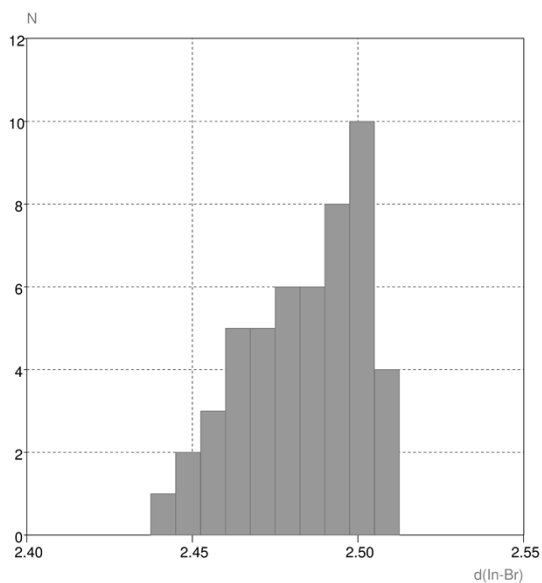


Figure S11. In–I Bond Length Distribution in $[\text{InI}_4]^-$

Data points = 35 from 8 REFcodes

Range = 2.663 – 2.731 Å

Mean In–I bond length = 2.702 ± 0.014 Å

