

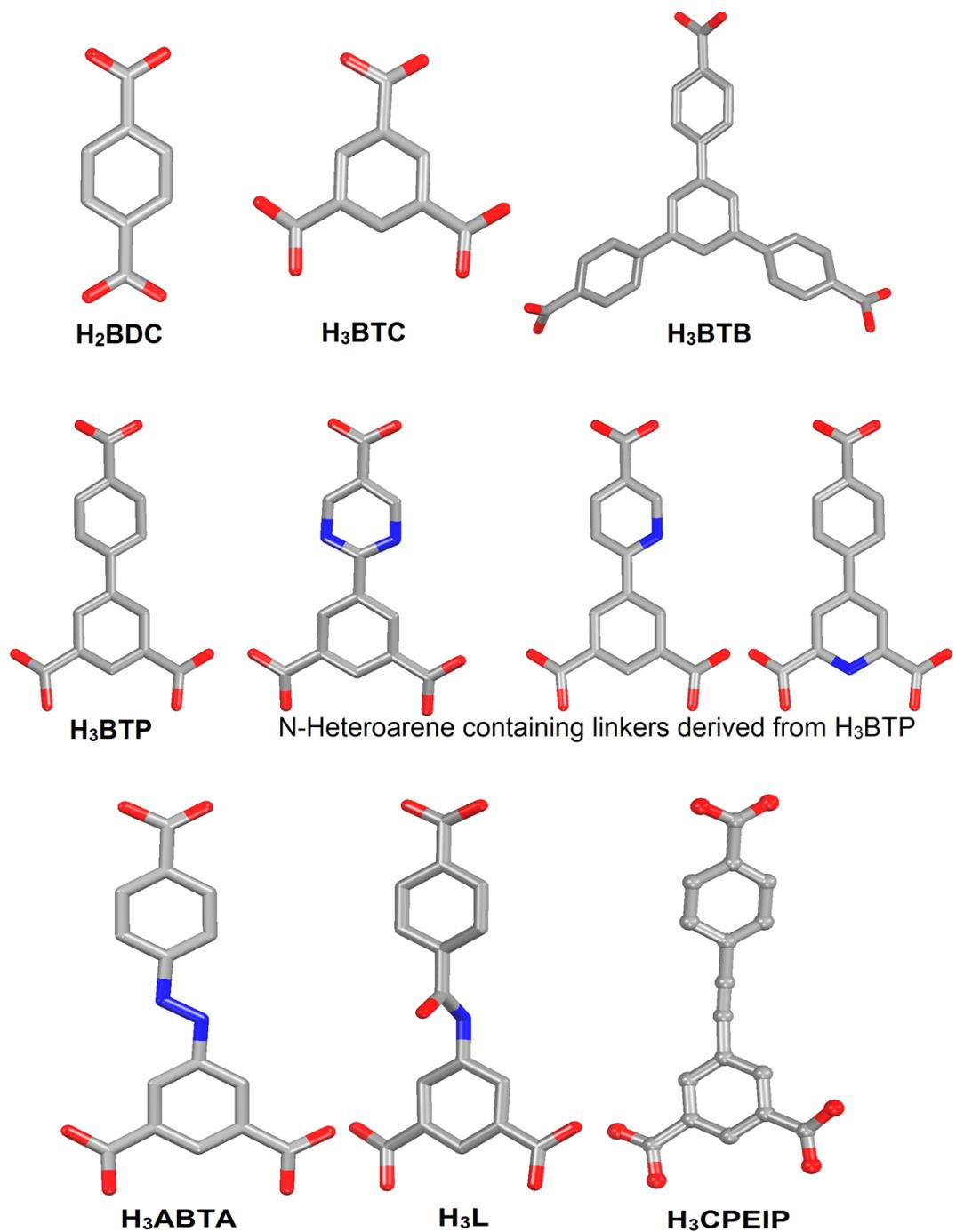
## Supporting Information

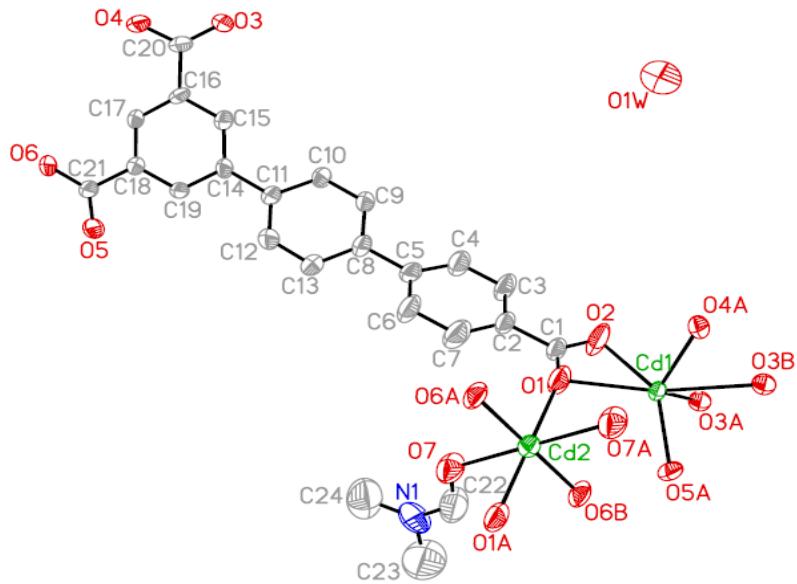
### A luminescent cadmium metal-organic frameworks for sensing of nitroaromatic explosives

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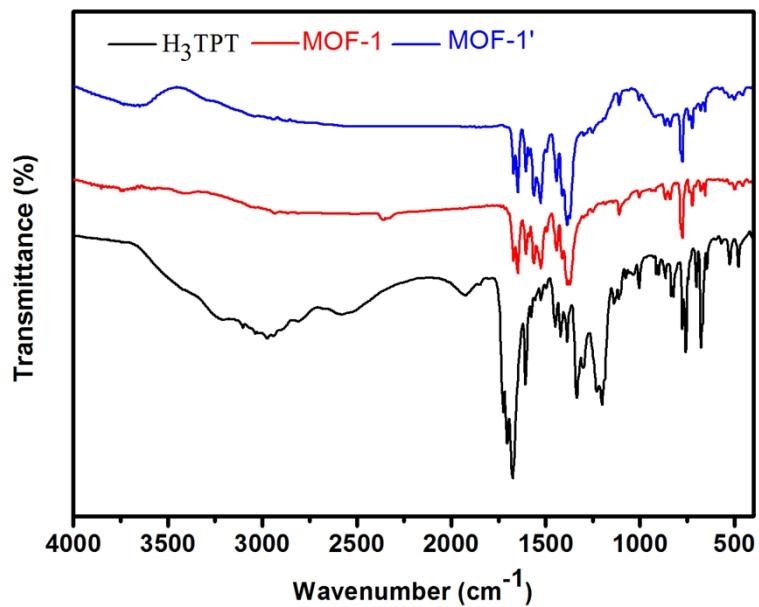
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**Fig. S1** Reported high symmetric aromatic polycarboxylates ( $H_2BDC^1$ ,  $H_3BTC^2$ ,  $H_3BTB^3$ ) and unsymmetrical aromatic polycarboxylate ( $H_3BTP^4$ , N-Heteroarene containing linkers derived from  $H_3BTP^5$ ,  $H_3ABTA^6$ , 5-(4-carboxybenzoylamino)-isophthalic acid ( $H_3L^7$ ), and  $H_3CPEIP^8$ .

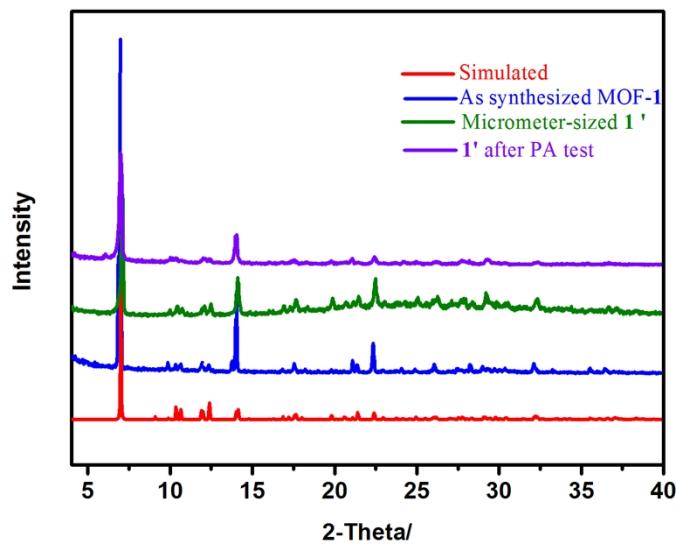




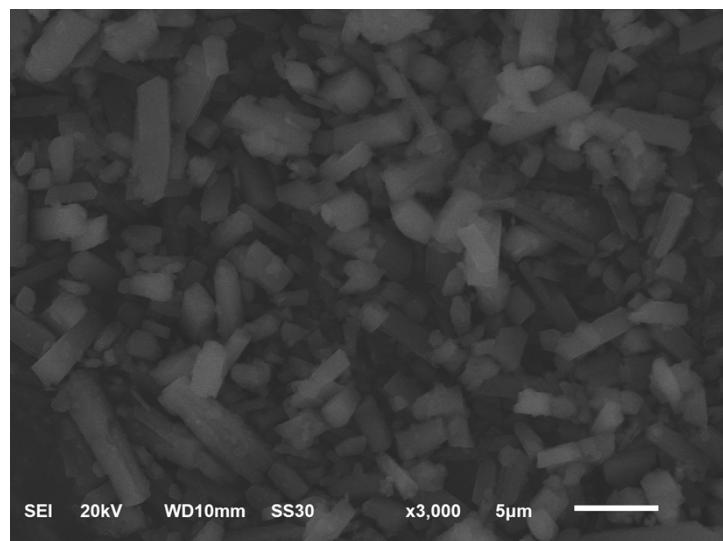
**Fig. S2** Representation of the asymmetric unit of **1** showing ellipsoid at the 50% probability level.



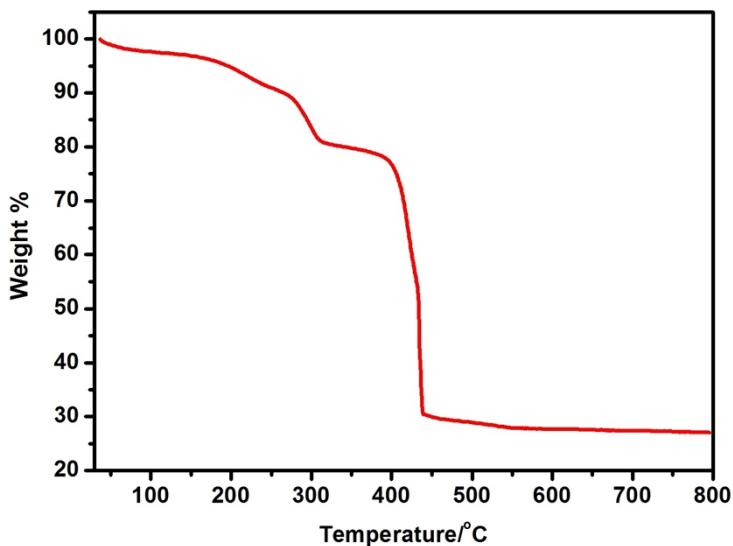
**Fig. S3** Infra-red spectra of the H<sub>3</sub>TPT, MOF-1 and MOF-1'.



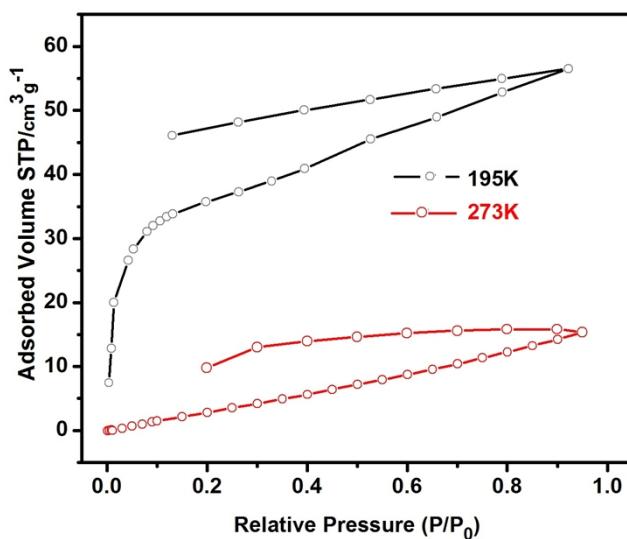
**Fig. S4** PXRD patterns of the simulated pattern calculated from the single crystal X-ray data, the as-synthesized sample **1**, micrometer-sized **1'** and sample of **1'** after PA quenching and recovery test for five times.



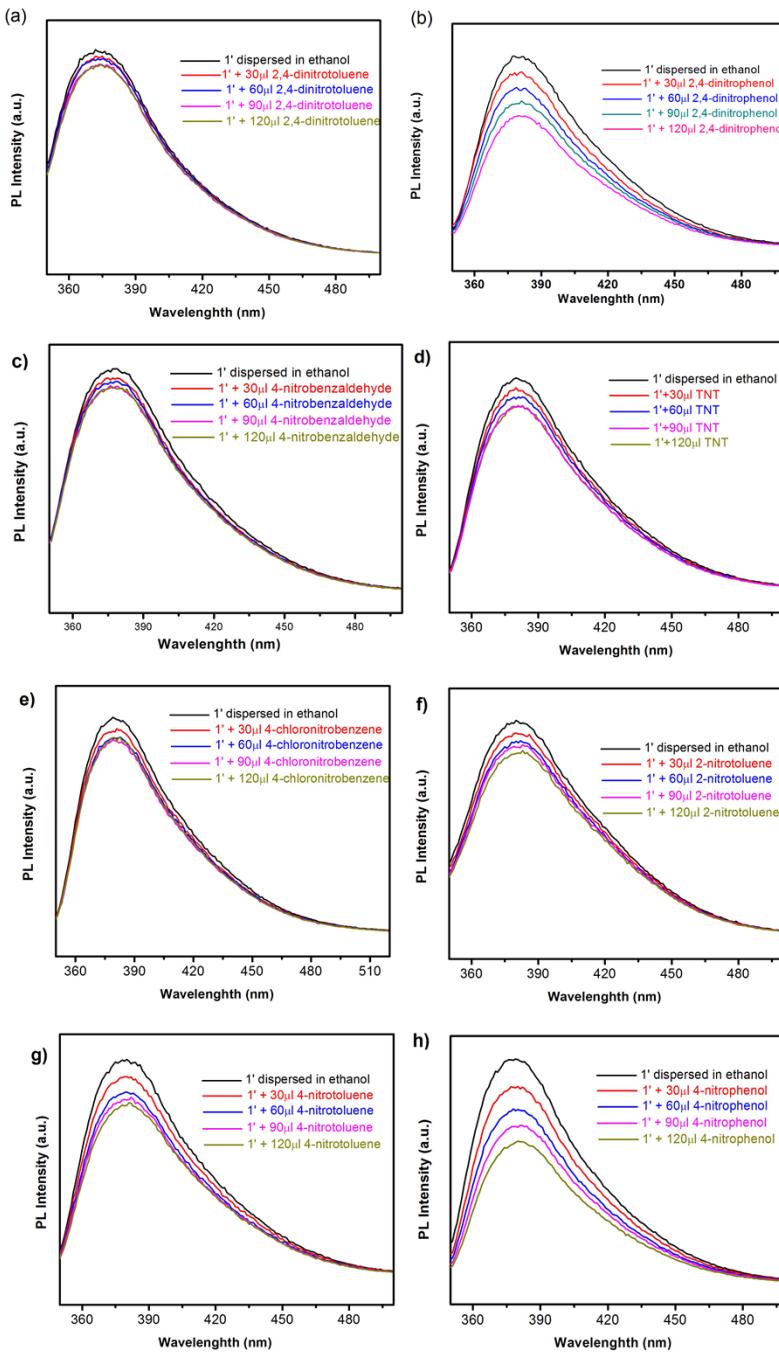
**Fig. S5** The SEM image of micrometer-sized **1'**



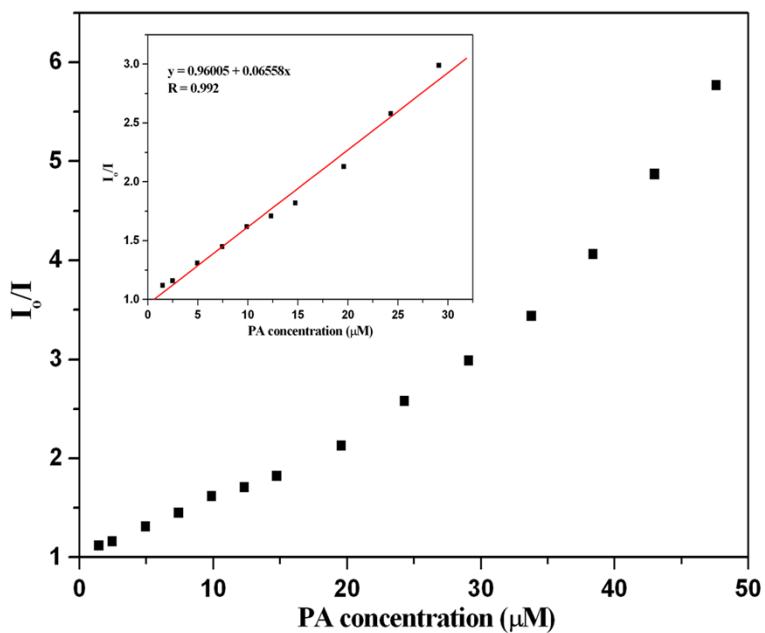
**Fig. S6** The TGA curve of **1**



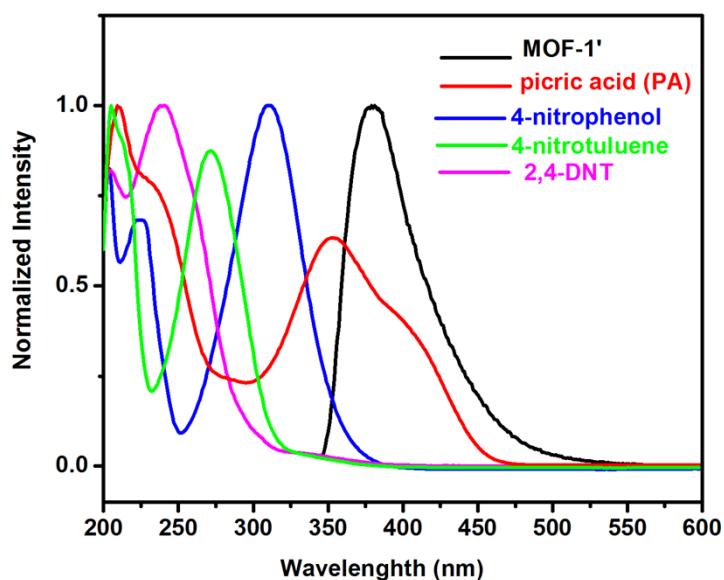
**Fig. S7** CO<sub>2</sub> adsorption-desorption isotherms measured at 195 K and 273 K.



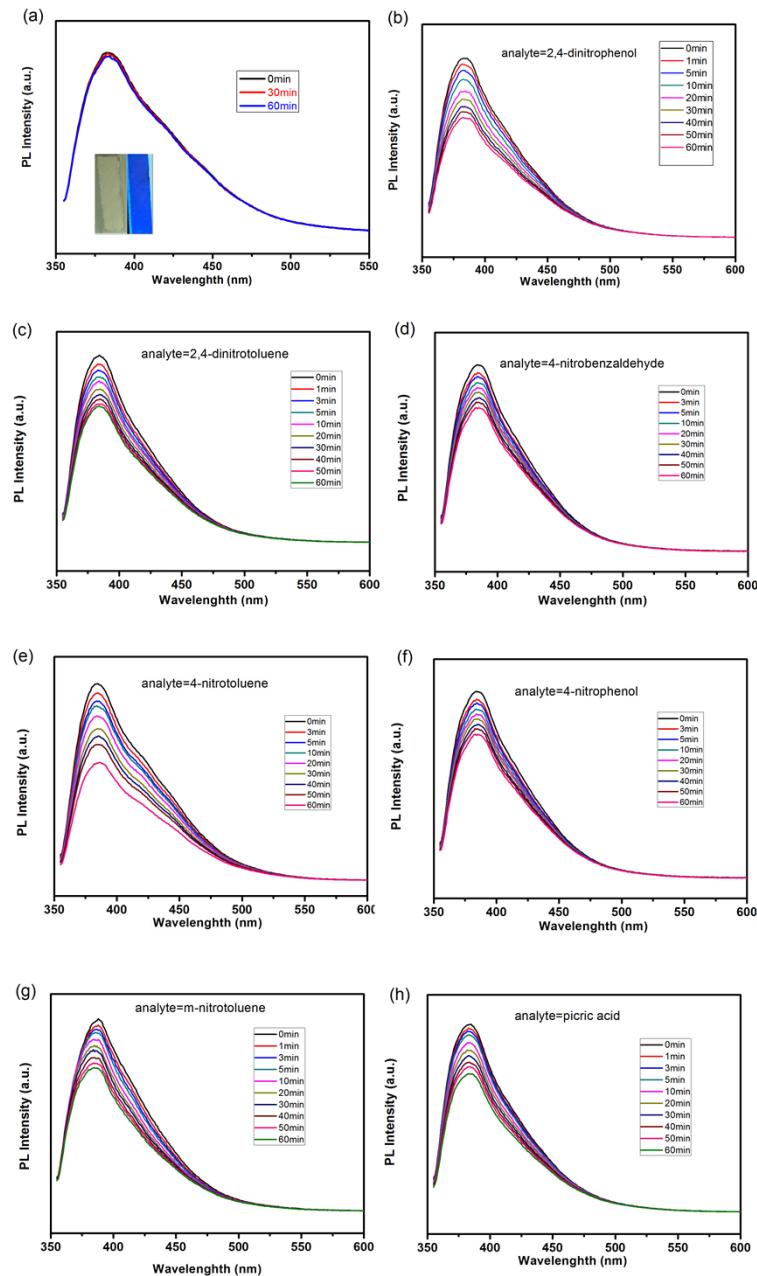
**Fig. S8** Fluorescence titrations of 0.5mg **1'** dispersed in 2mL ethanol solution with the addition of different volume of 0.001M solution of nitromethane (a), 2, 4-dinitrophenol (DNP) (b), 4- nitrobenzaldehyde (c), 2, 4, 6-trinitrotoluene (TNT) (d), 4-chloronitrobenzene (e), 2-nitrotoluene (f), 4-nitrotoluene (g), and 4- nitrophenol (h) in ethanol. Excited at 334 nm and fluorescence emission was recorded from 350 nm to 500 nm. The slit width for both excitation and emission was 1.5 nm.



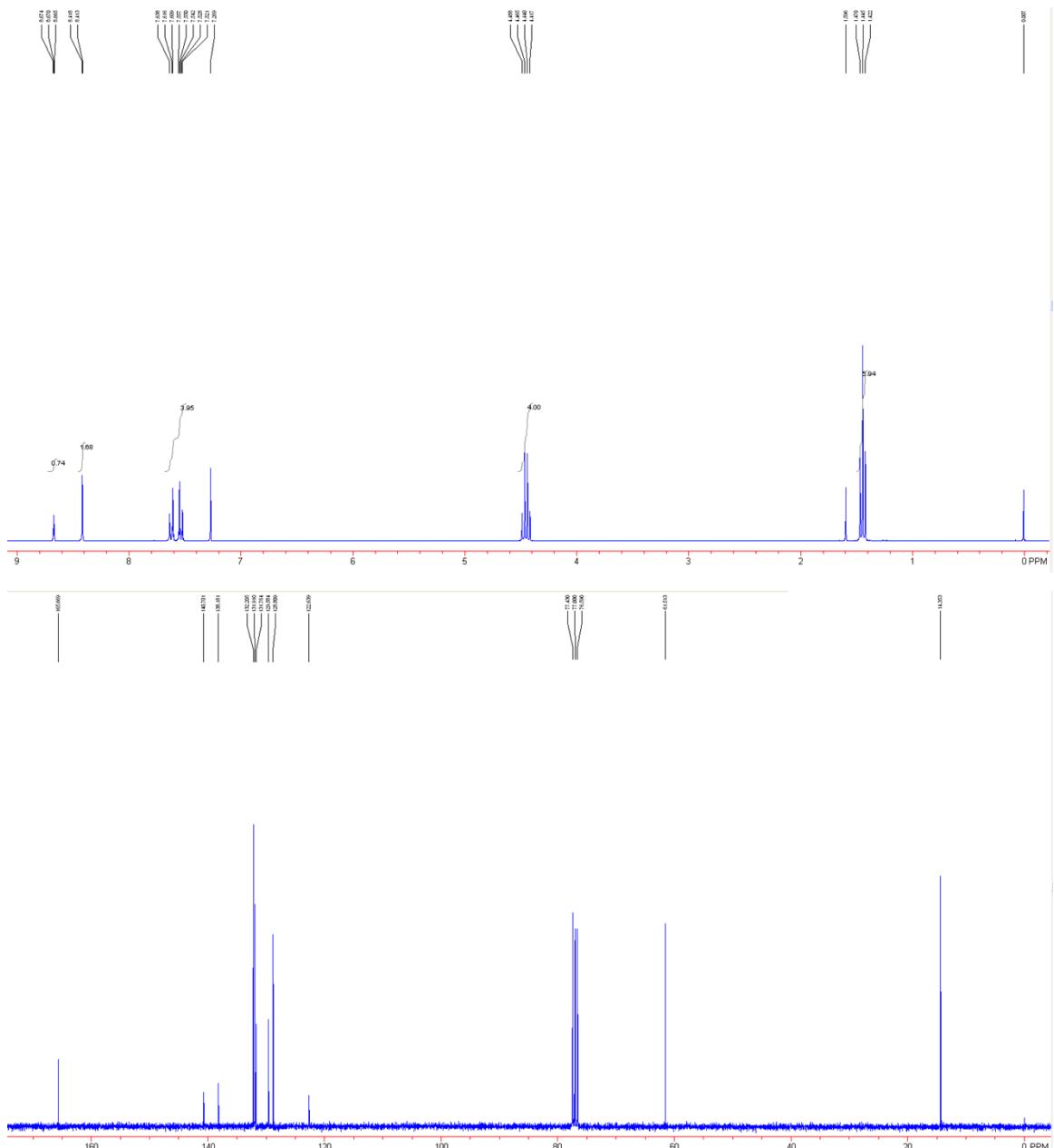
**Fig. S9** Plot of  $I_0/I$  versus PA concentration in ethanol for **1'**.



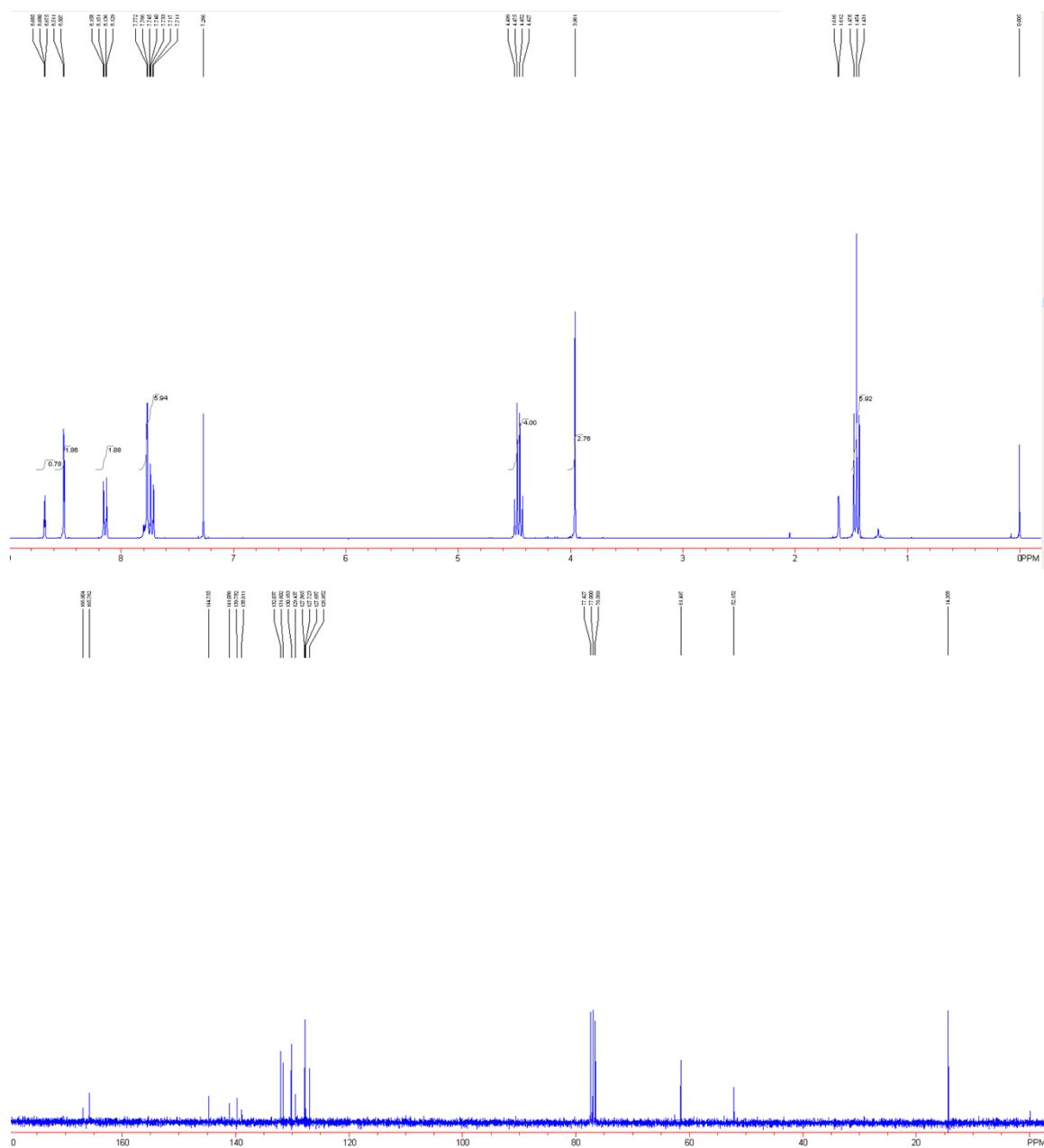
**Fig. S10** Spectral overlaps between the absorption spectra of analytes with the concentration of  $1.0 \times 10^{-3}$  M and the emission spectra of **1'** in ethanol ( $2.06 \times 10^{-4}$  M).



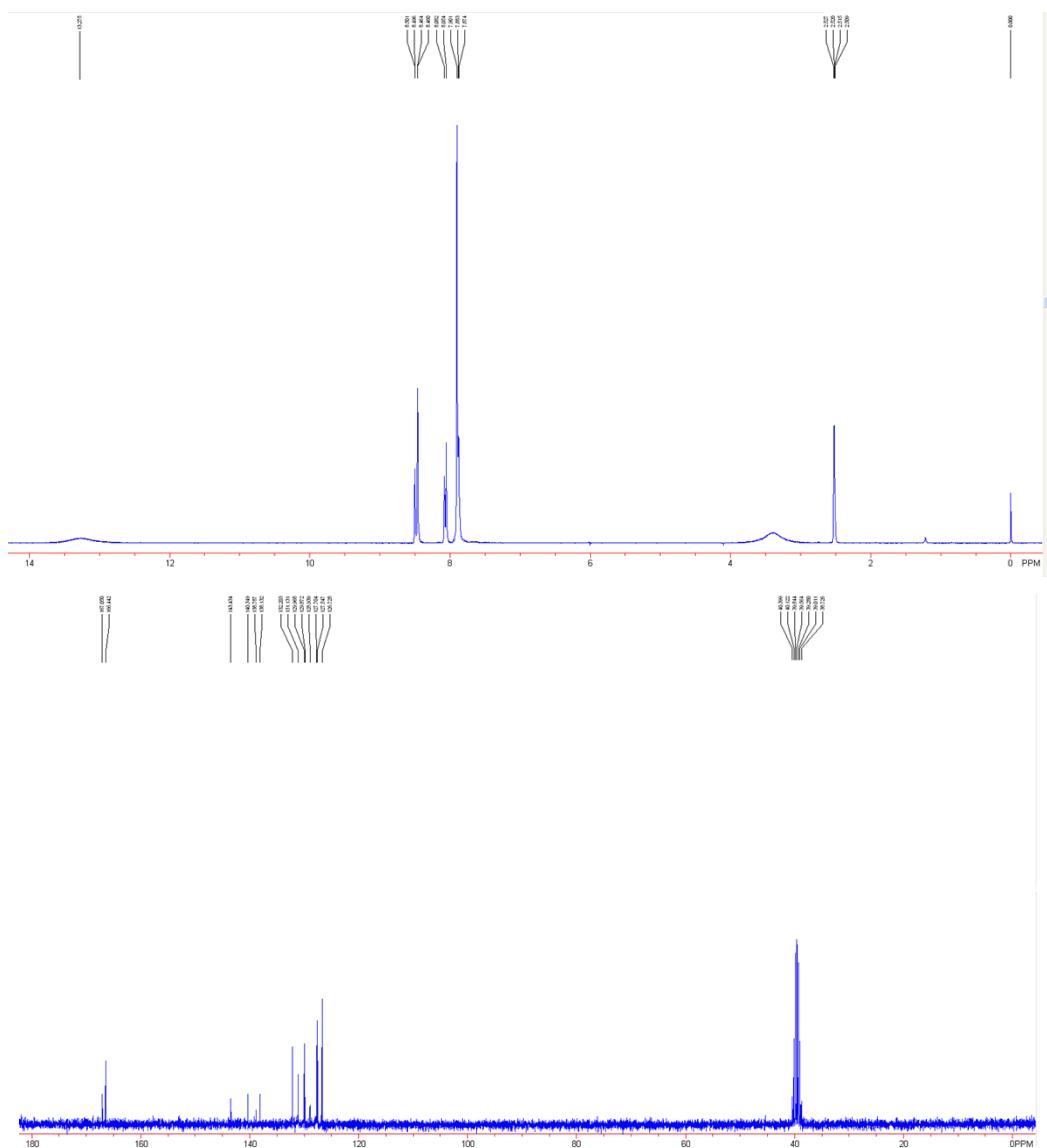
**Fig. S11** (a) The luminescent spectra of **1'** on double-sided tape at different time, and the photographs of the layer of **1'** under the sunlight and the UV light (inset). The vapor sensing of **1'** for 2,4-dinitrophenol (b), 2,4-dinitrotoluene (c), 4-nitrobenzaldehyde (d), 4-nitrotoluene (e), 4- nitrophenol (f), m-nitrotoluene (g), and picric acid (h).



**Fig. 12** The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of **M1**



**Fig. 13** The <sup>1</sup>H and <sup>13</sup>C NMR spectra of M2



**Fig. S14** The <sup>1</sup>H and <sup>13</sup>C NMR spectra of H<sub>3</sub>TPT

**Table S1.** Crystal Data and Structure Refinement for **1**

Empirical formula	C <sub>48</sub> H <sub>37</sub> Cd <sub>3</sub> N <sub>2</sub> O <sub>14.5</sub>		
Formula weight	1211.04		
Temperature	293(2) K		
Wavelength	0.71073 Å		
Crystal system, space group	Monoclinic, C2/c a = 19.524(4) Å      α = 90° b = 10.049(2) Å      β = 95.409(4)° c = 25.359(5) Å      γ = 90°		
Unit cell dimensions			
Volume	4953.2(17) Å <sup>3</sup>		
Z	2		
Absorption coefficient	1.343 mm <sup>-1</sup>		
F (000)	2392		
Theta range for data collection	1.61 to 28.21°		
Limiting indices	-25 ≤ h ≤ 25, -13 ≤ k ≤ 9, -29 ≤ l ≤ 33		
Reflections collected / unique	17595 / 6090		
R <sub>int</sub>	0.1021		
Completeness to theta = 28.21	99.6 %		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	0.7751 and 0.7476		
Refinement method	Full-matrix least-squares on F <sup>2</sup>		
Data / restraints / parameters	6090 / 7 / 311		
Goodness-of-fit on F <sup>2</sup>	1.013		
Final R indices [I>2σ(I)]	R <sub>1</sub> = 0.0704, wR <sub>2</sub> = 0.1698		
R indices (all data)	R <sub>1</sub> = 0.1487, wR <sub>2</sub> = 0.2050		
Largest diff. peak and hole	1.612 and -1.183 e. Å <sup>-3</sup>		

$$^bR_1=\sum||F_o|-|F_c||/\sum|F_o|. \quad wR_2=[\sum[w(F_o^2-F_c^2)^2]/\sum[w(F_o^2)^2]]^{1/2}.$$

**Table. S2** Selected bond lengths [Å] and angles [°] for **1**.

## Bond lengths / Å

Cd(1)-O(5)#1	2.180(3)	C(6)-H(6)	0.93
Cd(1)-O(4)#2	2.270(3)	C(7)-H(7)	0.93
Cd(1)-O(3)#3	2.308(3)	C(8)-C(13)	1.376(7)
Cd(1)-O(2)	2.354(4)	C(8)-C(9)	1.394(7)
Cd(1)-O(1)	2.363(4)	C(9)-C(10)	1.371(7)
Cd(1)-O(3)#2	2.529(3)	C(9)-H(9)	0.93
Cd(2)-O(6)#4	2.215(3)	C(10)-C(11)	1.403(7)
Cd(2)-O(6)#1	2.215(3)	C(10)-H(10)	0.93
Cd(2)-O(7)	2.289(4)	C(11)-C(12)	1.373(7)
Cd(2)-O(7)#5	2.289(4)	C(11)-C(14)	1.484(6)
Cd(2)-O(1)	2.293(4)	C(12)-C(13)	1.374(7)
Cd(2)-O(1)#5	2.294(4)	C(12)-H(12)	0.93
O(1)-C(1)	1.244(6)	C(13)-H(13)	0.93
O(2)-C(1)	1.245(6)	C(14)-C(19)	1.396(6)

O(3)-C(20)	1.255(6)	C(14)-C(15)	1.402(7)
O(3)-Cd(1)#6	2.308(3)	C(15)-C(16)	1.391(6)
O(3)-Cd(1)#7	2.529(3)	C(15)-H(15)	0.93
O(4)-C(20)	1.253(6)	C(16)-C(17)	1.370(6)
O(4)-Cd(1)#7	2.270(3)	C(16)-C(20)	1.499(6)
O(5)-C(21)	1.233(6)	C(17)-C(18)	1.385(6)
O(5)-Cd(1)#8	2.180(3)	C(17)-H(17)	0.93
O(6)-C(21)	1.256(5)	C(18)-C(19)	1.387(6)
O(6)-Cd(2)#4	2.215(3)	C(18)-C(21)	1.506(6)
O(7)-C(22)	1.244(8)	C(19)-H(19)	0.93
C(1)-C(2)	1.488(6)	C(22)-N(1)	1.317(8)
C(2)-C(3)	1.377(7)	C(22)-H(22)	0.93
C(2)-C(7)	1.378(8)	N(1)-C(24)	1.369(6)
C(3)-C(4)	1.379(7)	N(1)-C(23)	1.433(10)
C(3)-H(3)	0.93	C(23)-H(23A)	0.96
C(4)-C(5)	1.396(7)	C(23)-H(23B)	0.96
C(4)-H(4)	0.93	C(23)-H(23C)	0.96
C(5)-C(6)	1.381(7)	C(24)-H(24A)	0.96
C(5)-C(8)	1.499(7)	C(24)-H(24B)	0.96
C(6)-C(7)	1.402(7)	C(24)-H(24C)	0.96

#### Bond angles / °

O(5)#1-Cd(1)-O(4)#2	128.21(12)	C(4)-C(5)-C(8)	121.6(4)
O(5)#1-Cd(1)-O(3)#3	85.13(12)	C(5)-C(6)-C(7)	119.9(5)
O(4)#2-Cd(1)-O(3)#3	111.10(12)	C(5)-C(6)-H(6)	120
O(5)#1-Cd(1)-O(2)	137.65(15)	C(7)-C(6)-H(6)	120
O(4)#2-Cd(1)-O(2)	93.66(15)	C(2)-C(7)-C(6)	121.0(5)
O(3)#3-Cd(1)-O(2)	84.98(12)	C(2)-C(7)-H(7)	119.5
O(5)#1-Cd(1)-O(1)	103.27(14)	C(6)-C(7)-H(7)	119.5
O(4)#2-Cd(1)-O(1)	103.26(14)	C(13)-C(8)-C(9)	117.9(5)
O(3)#3-Cd(1)-O(1)	127.89(12)	C(13)-C(8)-C(5)	120.7(5)
O(2)-Cd(1)-O(1)	54.24(13)	C(9)-C(8)-C(5)	121.4(5)
O(5)#1-Cd(1)-O(3)#2	85.43(11)	C(10)-C(9)-C(8)	121.6(5)
O(4)#2-Cd(1)-O(3)#2	54.11(11)	C(10)-C(9)-H(9)	119.2
O(3)#3-Cd(1)-O(3)#2	77.04(12)	C(8)-C(9)-H(9)	119.2
O(2)-Cd(1)-O(3)#2	131.65(14)	C(9)-C(10)-C(11)	121.1(5)
O(1)-Cd(1)-O(3)#2	153.68(13)	C(9)-C(10)-H(10)	119.4
O(5)#1-Cd(1)-C(1)	123.53(14)	C(11)-C(10)-H(10)	119.4
O(4)#2-Cd(1)-C(1)	98.73(13)	C(12)-C(11)-C(10)	115.4(4)
O(3)#3-Cd(1)-C(1)	107.67(13)	C(12)-C(11)-C(14)	123.0(4)
O(2)-Cd(1)-C(1)	27.14(14)	C(10)-C(11)-C(14)	121.6(4)
O(1)-Cd(1)-C(1)	27.12(13)	C(11)-C(12)-C(13)	124.7(5)
O(3)#2-Cd(1)-C(1)	150.62(13)	C(11)-C(12)-H(12)	117.7

O(5)#1-Cd(1)-C(20)#2	108.73(13)	C(13)-C(12)-H(12)	117.7
O(4)#2-Cd(1)-C(20)#2	27.00(13)	C(12)-C(13)-C(8)	119.2(5)
O(3)#3-Cd(1)-C(20)#2	93.10(13)	C(12)-C(13)-H(13)	120.4
O(2)-Cd(1)-C(20)#2	112.85(16)	C(8)-C(13)-H(13)	120.4
O(1)-Cd(1)-C(20)#2	129.65(15)	C(19)-C(14)-C(15)	118.1(4)
O(3)#2-Cd(1)-C(20)#2	27.22(12)	C(19)-C(14)-C(11)	120.2(4)
C(1)-Cd(1)-C(20)#2	124.49(14)	C(15)-C(14)-C(11)	121.7(4)
O(6)#4-Cd(2)-O(6)#1	180.00(18)	C(16)-C(15)-C(14)	121.1(4)
O(6)#4-Cd(2)-O(7)	91.84(13)	C(16)-C(15)-H(15)	119.5
O(6)#1-Cd(2)-O(7)	88.15(13)	C(14)-C(15)-H(15)	119.5
O(6)#4-Cd(2)-O(7)#5	88.15(13)	C(17)-C(16)-C(15)	119.1(4)
O(6)#1-Cd(2)-O(7)#5	91.85(13)	C(17)-C(16)-C(20)	121.2(4)
O(7)-Cd(2)-O(7)#5	180.000(1)	C(15)-C(16)-C(20)	119.8(4)
O(6)#4-Cd(2)-O(1)	92.98(14)	C(16)-C(17)-C(18)	121.5(4)
O(6)#1-Cd(2)-O(1)	87.02(14)	C(16)-C(17)-H(17)	119.2
O(7)-Cd(2)-O(1)	100.63(14)	C(18)-C(17)-H(17)	119.2
O(7)#5-Cd(2)-O(1)	79.37(14)	C(17)-C(18)-C(19)	119.1(4)
O(6)#4-Cd(2)-O(1)#5	87.02(14)	C(17)-C(18)-C(21)	119.7(4)
O(6)#1-Cd(2)-O(1)#5	92.98(14)	C(19)-C(18)-C(21)	121.2(4)
O(7)-Cd(2)-O(1)#5	79.37(14)	C(18)-C(19)-C(14)	121.0(4)
O(7)#5-Cd(2)-O(1)#5	100.63(14)	C(18)-C(19)-H(19)	119.5
O(1)-Cd(2)-O(1)#5	180.00(13)	C(14)-C(19)-H(19)	119.5
C(1)-O(1)-Cd(2)	157.2(4)	O(4)-C(20)-O(3)	122.2(4)
C(1)-O(1)-Cd(1)	92.9(3)	O(4)-C(20)-C(16)	117.1(4)
Cd(2)-O(1)-Cd(1)	106.18(14)	O(3)-C(20)-C(16)	120.6(4)
C(1)-O(2)-Cd(1)	93.3(3)	O(4)-C(20)-Cd(1)#7	55.4(2)
C(20)-O(3)-Cd(1)#6	124.7(3)	O(3)-C(20)-Cd(1)#7	67.2(2)
C(20)-O(3)-Cd(1)#7	85.6(3)	C(16)-C(20)-Cd(1)#7	170.1(3)
Cd(1)#6-O(3)-Cd(1)#7	102.96(12)	O(5)-C(21)-O(6)	126.4(4)
C(20)-O(4)-Cd(1)#7	97.6(3)	O(5)-C(21)-C(18)	117.8(4)
C(21)-O(5)-Cd(1)#8	124.4(3)	O(6)-C(21)-C(18)	115.9(4)
C(21)-O(6)-Cd(2)#4	133.1(3)	O(7)-C(22)-N(1)	123.6(6)
C(22)-O(7)-Cd(2)	119.1(4)	O(7)-C(22)-H(22)	118.2
O(1)-C(1)-O(2)	119.5(4)	N(1)-C(22)-H(22)	118.2
O(1)-C(1)-C(2)	121.4(4)	C(22)-N(1)-C(24)	130.8(7)
O(2)-C(1)-C(2)	119.1(4)	C(22)-N(1)-C(23)	116.6(6)
O(1)-C(1)-Cd(1)	60.0(3)	C(24)-N(1)-C(23)	112.5(7)
O(2)-C(1)-Cd(1)	59.6(2)	N(1)-C(23)-H(23A)	109.5
C(2)-C(1)-Cd(1)	174.5(3)	N(1)-C(23)-H(23B)	109.5
C(3)-C(2)-C(7)	119.6(5)	H(23A)-C(23)-H(23B)	109.5
C(3)-C(2)-C(1)	121.1(4)	N(1)-C(23)-H(23C)	109.5
C(7)-C(2)-C(1)	119.3(4)	H(23A)-C(23)-H(23C)	109.5
C(2)-C(3)-C(4)	119.3(5)	H(23B)-C(23)-H(23C)	109.5
C(2)-C(3)-H(3)	120.4	N(1)-C(24)-H(24A)	109.5

C(4)-C(3)-H(3)	120.4	N(1)-C(24)-H(24B)	109.5
C(3)-C(4)-C(5)	122.4(5)	H(24A)-C(24)-H(24B)	109.5
C(3)-C(4)-H(4)	118.8	N(1)-C(24)-H(24C)	109.5
C(5)-C(4)-H(4)	118.8	H(24A)-C(24)-H(24C)	109.5
C(6)-C(5)-C(4)	117.8(5)	H(24B)-C(24)-H(24C)	109.5
C(6)-C(5)-C(8)	120.6(5)		

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