

**Electronic Supplementary Information (ESI)**

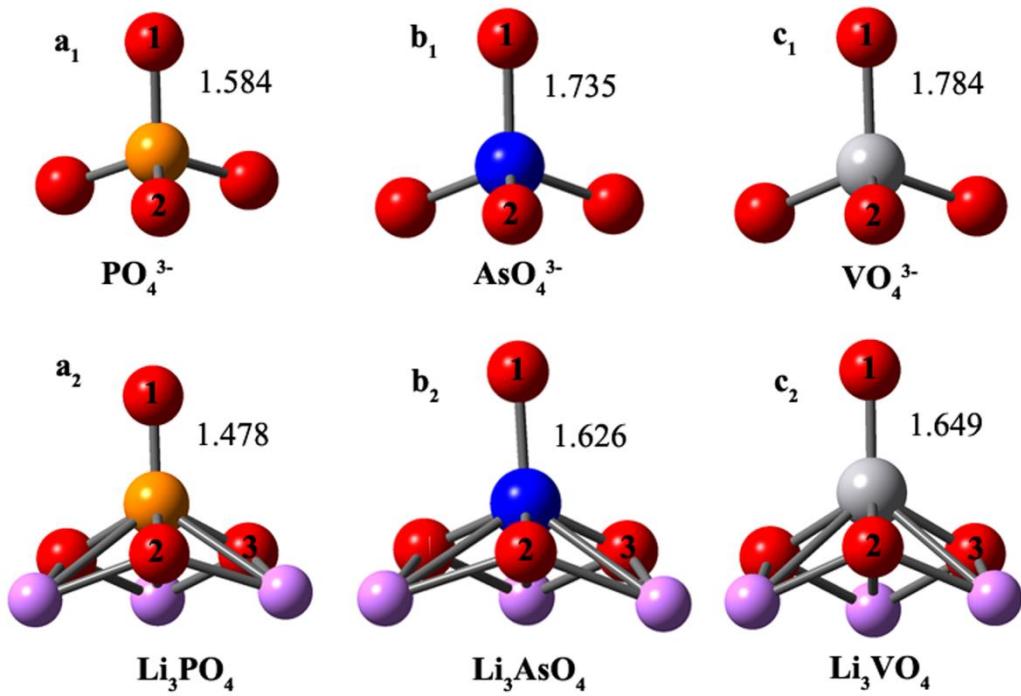
**Trivalent acid radical-centered  $\text{YLi}_4^+$  ( $\text{Y} = \text{PO}_4, \text{AsO}_4, \text{VO}_4$ ) cations: new polynuclear species designed to enrich the superalkali family**

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**Figure S1.** Optimized structures of isolated (a<sub>1</sub>)  $\text{PO}_4^{3-}$  (b<sub>1</sub>)  $\text{AsO}_4^{3-}$  (c<sub>1</sub>)  $\text{VO}_4^{3-}$ ; (a<sub>2</sub>)  $\text{Li}_3\text{PO}_4$  (b<sub>2</sub>)  $\text{Li}_3\text{AsO}_4$  (c<sub>2</sub>)  $\text{Li}_3\text{VO}_4$ . Color legend: P, orange-yellow; O, red; Li, purple; As, blue; V, grey.

**Table S1.** The bond length (Å) and the bond angle (°) of (**a<sub>1</sub>**) PO<sub>4</sub><sup>3-</sup> (**b<sub>1</sub>**) AsO<sub>4</sub><sup>3-</sup> (**c<sub>1</sub>**) VO<sub>4</sub><sup>3-</sup>; (**a<sub>2</sub>**) Li<sub>3</sub>PO<sub>4</sub> (**b<sub>2</sub>**) Li<sub>3</sub>AsO<sub>4</sub> (**c<sub>2</sub>**) Li<sub>3</sub>VO<sub>4</sub>. Nx = P, As, V for **a**, **b** and **c**, respectively.

structure	Nx-O1	Nx-O2	∠O1NxO2	∠O2NxO3	Nx-Li	O2-Li
<b>PO</b> <sub>4</sub> <sup>3-</sup>	1.584		109.5			
<b>AsO</b> <sub>4</sub> <sup>3-</sup>	1.735		109.5			
<b>VO</b> <sub>4</sub> <sup>3-</sup>	1.784		109.5			
<b>Li<sub>3</sub>PO<sub>4</sub></b>	1.478	1.594	115.9	102.3	2.348	1.858
<b>Li<sub>3</sub>AsO<sub>4</sub></b>	1.626	1.749	118.9	98.6	2.447	1.884
<b>Li<sub>3</sub>VO<sub>4</sub></b>	1.649	1.788	116.9	101.1	2.403	1.885

**Table S2.** The NBO charges of the  $\text{PO}_4\text{Li}_4^+$  cations.

isomer	P	O				Li			
		1	2	3	4	1	2	3	4
<b>P1</b>	2.443	-1.266	-1.266	-1.266	-1.266	0.906	0.906	0.906	0.906
<b>P2</b>	2.467	-1.282	-1.279	-1.279	-1.279	0.948	0.901	0.901	0.901
<b>P3</b>	2.417	-1.179	-1.269	-1.269	-1.331	0.907	0.906	0.912	0.906
<b>P4</b>	2.618	-1.331	-1.331	-1.331	-1.331	0.927	0.927	0.927	0.927
<b>P5</b>	2.455	-1.116	-1.116	-1.116	-1.636	0.867	0.867	0.867	0.927
<b>P6</b>	2.495	-1.214	-1.073	-1.073	-1.661	0.966	0.831	0.831	0.899
<b>P7</b>	2.440	-1.089	-1.160	-1.039	-1.664	0.938	0.831	0.839	0.904

**Table S3.** The NBO charges of the  $\text{AsO}_4\text{Li}_4^+$  cations.

isomer	As	O				Li			
		1	2	3	4	1	2	3	4
<b>A1</b>	2.476	-1.279	-1.279	-1.279	-1.279	0.909	0.909	0.909	0.909
<b>A2</b>	2.531	-1.301	-1.296	-1.296	-1.296	0.952	0.902	0.902	0.902
<b>A3</b>	2.461	-1.172	-1.281	-1.281	-1.359	0.909	0.906	0.912	0.906
<b>A4</b>	2.487	-1.205	-1.205	-1.365	-1.375	0.903	0.903	0.958	0.898
<b>A5</b>	2.725	-1.362	-1.362	-1.362	-1.362	0.931	0.931	0.931	0.931
<b>A6</b>	1.614	-1.180	-1.180	-0.818	-0.789	0.819	0.819	0.812	0.903
<b>A7</b>	1.625	-1.306	-1.301	-0.725	-0.789	0.858	0.839	0.883	0.916
<b>A8</b>	1.582	-1.273	-1.096	-0.819	-0.819	0.803	0.860	0.860	0.901
<b>A9</b>	1.580	-1.393	-1.244	-0.652	-0.747	0.839	0.854	0.816	0.948
<b>A10</b>	2.514	-1.239	-1.069	-1.069	-1.660	0.965	0.838	0.838	0.899
<b>A11</b>	2.442	-1.079	-1.176	-1.028	-1.662	0.935	0.829	0.837	0.904
<b>A12</b>	1.558	-1.077	-1.274	-0.816	-0.835	0.774	0.872	0.899	0.899
<b>A13</b>	1.560	-1.087	-1.274	-0.810	-0.806	0.904	0.809	0.809	0.894

**Table S4.** The NBO charges of the  $\text{VO}_4\text{Li}_4^+$  cations.

isomer	V	O				Li			
		1	2	3	4	1	2	3	4
<b>V1</b>	0.677	-0.820	-0.820	-0.820	-0.820	0.901	0.901	0.901	0.901
<b>V2</b>	0.774	-0.839	-0.846	-0.846	-0.846	0.942	0.887	0.887	0.887
<b>V3</b>	0.678	-0.552	-0.840	-1.036	-0.840	0.902	0.901	0.893	0.893
<b>V4</b>	0.740	-0.570	-0.570	-1.080	-1.120	0.889	0.944	0.889	0.877
<b>V5</b>	0.772	-0.605	-1.101	-0.605	-1.101	0.879	0.875	0.943	0.943
<b>V6</b>	0.810	-0.651	-0.651	-0.651	-1.393	0.869	0.869	0.869	0.929
<b>V7</b>	1.071	-0.592	-0.592	-0.592	-1.631	0.805	0.805	0.805	0.919
<b>V8</b>	1.210	-0.878	-0.574	-0.574	-1.654	0.946	0.812	0.812	0.899

**Figure S2.** HOMOs of the  $\text{PO}_4\text{Li}_4^+$  cations.

