

Supplementary Information for

**Small Molecular Amine Mediated Synthesis of Hydrophilic
CdS Nanorods and their Photoelectrochemical Water
Splitting Performance**

Chunlin Bao,^a Guoxing Zhu,^{*, a,b} Jing Yang,^a Miaomiao Liu,^a Rongxian Zhang,^a and Xiaoping
Shen^{*, a}

^aSchool of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang, 212013, China,

Fax: (+86)511-84401889; Tel: (+86)511-84401889; E-mail: zhuguoxing@ujs.edu.cn;

xiaopingshen@163.com

^bState Key Laboratory of Coordination Chemistry, Nanjing National Laboratory of
Microstructures, Nanjing University, Nanjing, 210093, China

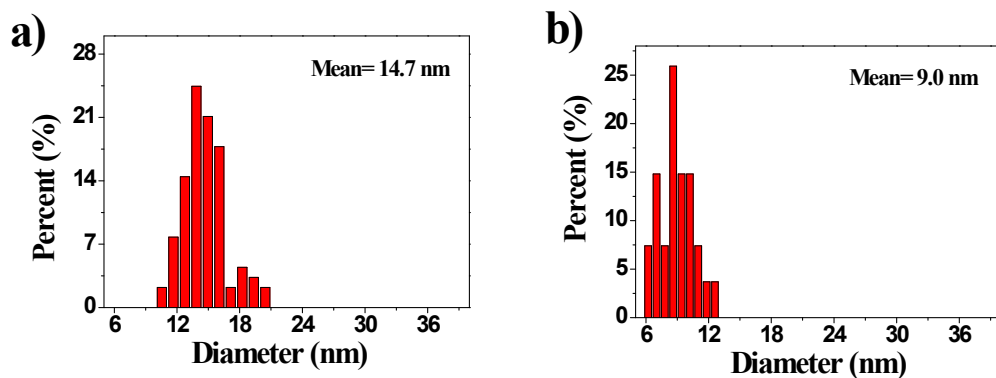


Fig. SI-1. Diameter distribution of the CdS-190 and CdS-130 product.

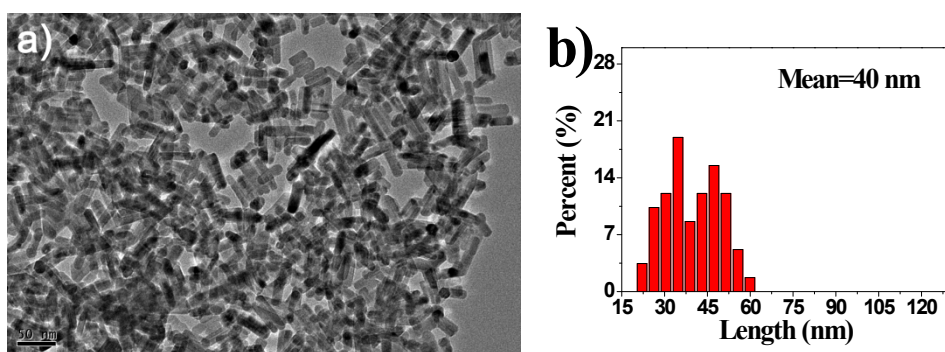


Fig. SI-2. a) TEM image and b) length distribution of CdS-150 product.

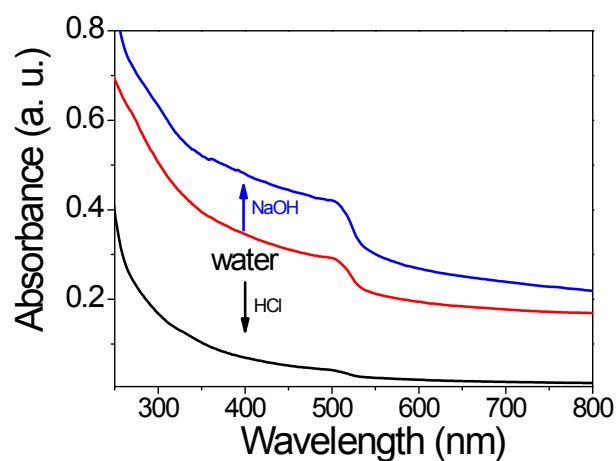


Fig. SI-3. The absorption spectra of CdS nanorods (CdS-190) dispersion (0.25 mg/mL) in aqueous solution with different pH values.

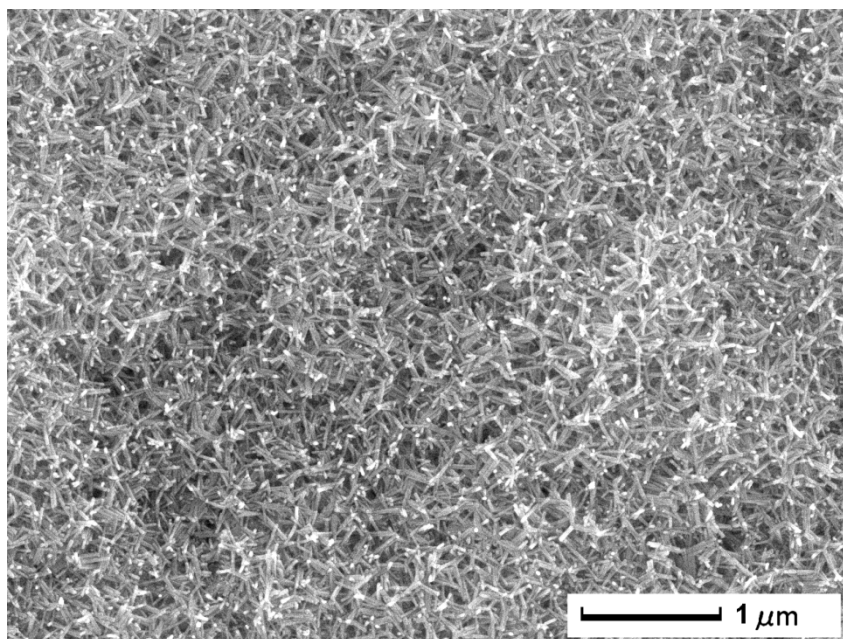


Fig. SI-4. SEM image of the CdS-190 film.

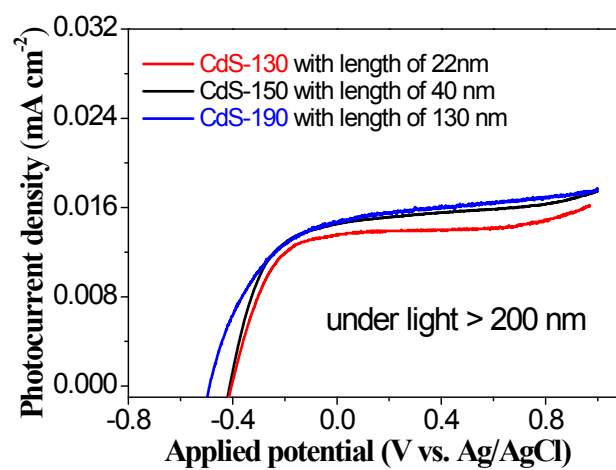


Fig. SI-5. The current-voltage characteristic of CdS photoelectrodes with different length under illumination of light > 200 nm.

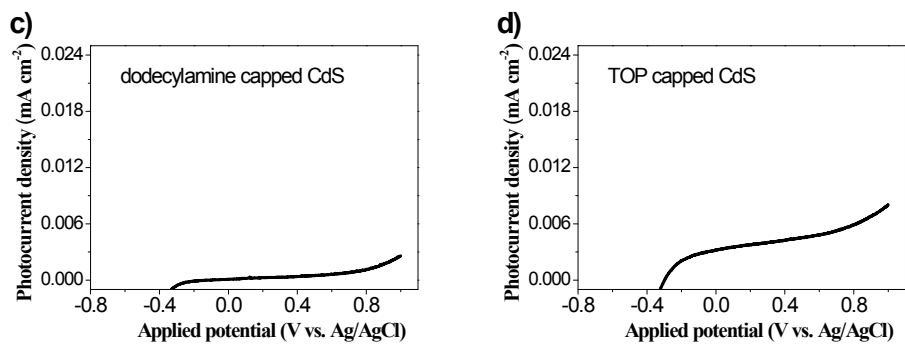
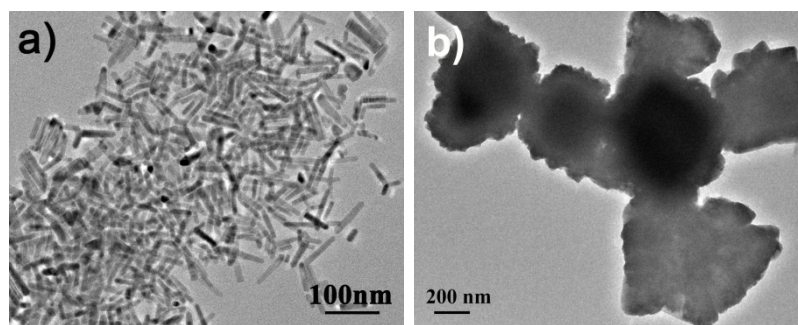


Fig. SI-6. TEM images of a) dodecylamine capped CdS nanorods and b) TOP capped CdS. c) The corresponding current-voltage characteristic of hydrophobic CdS thin films under illumination of light > 200 nm.

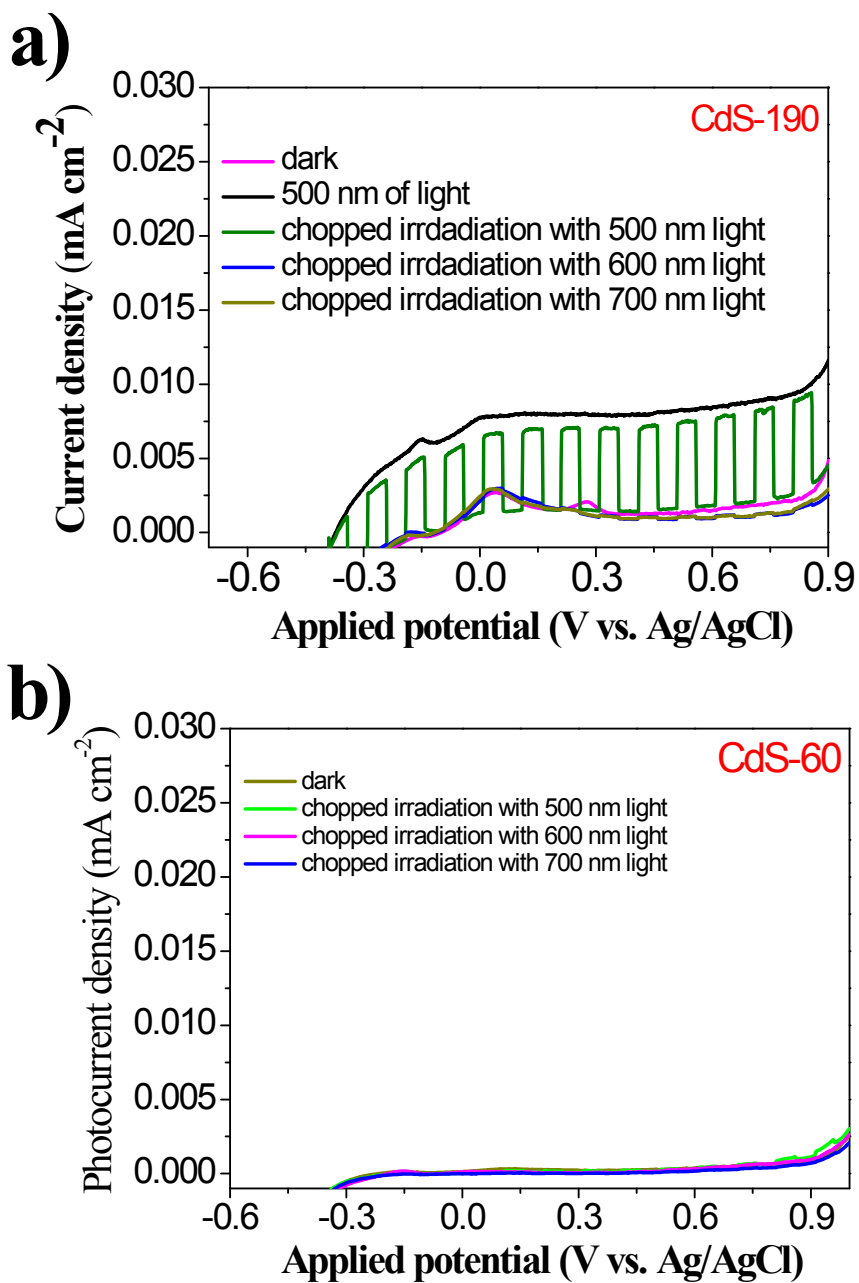


Fig. SI-7. The current-voltage characteristics for (a) the CdS nanorod sample (CdS-190) and (b) the CdS particle (CdS-60) thin films under chopped monochrome light illumination with wavelengths of 500, 600, 700 nm, respectively.