

Electronic Supplementary Information

Highly Efficient Electrochemiluminescence from Iridium(III) Complexes with 2-Phenylquinoline Ligand

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China.

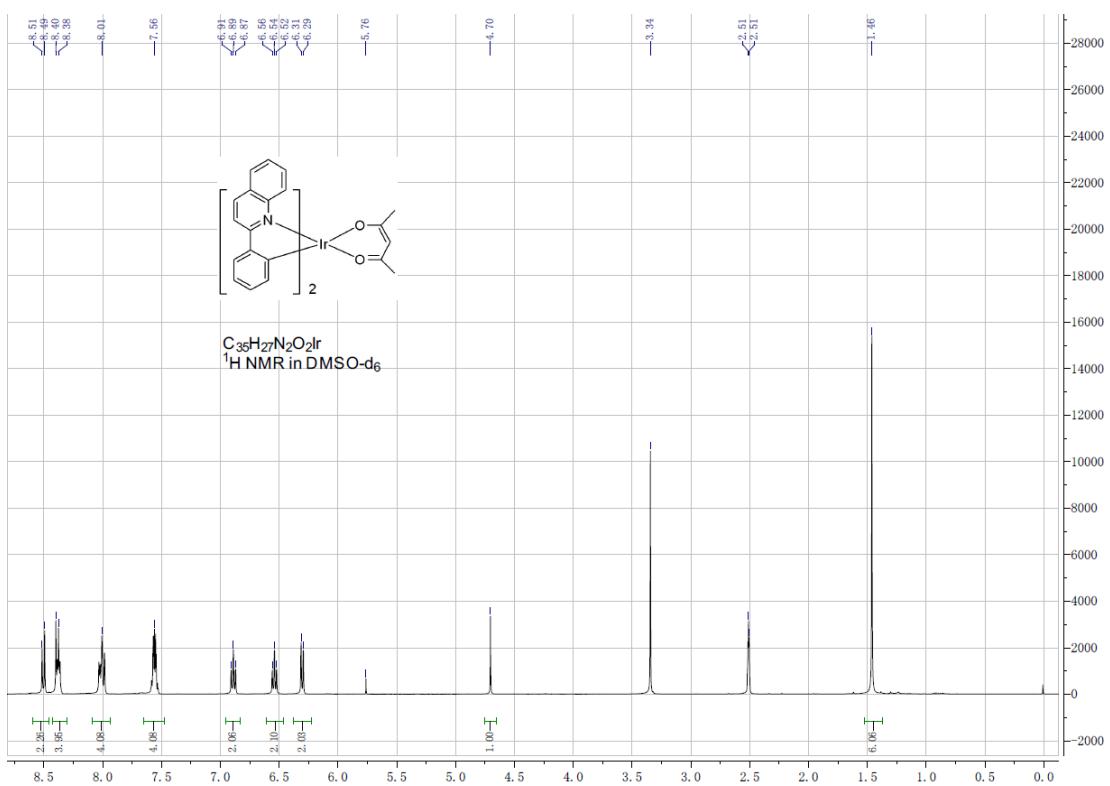


Figure S1. ^1H NMR spectrum of complex **1** in DMSO-d_6 .

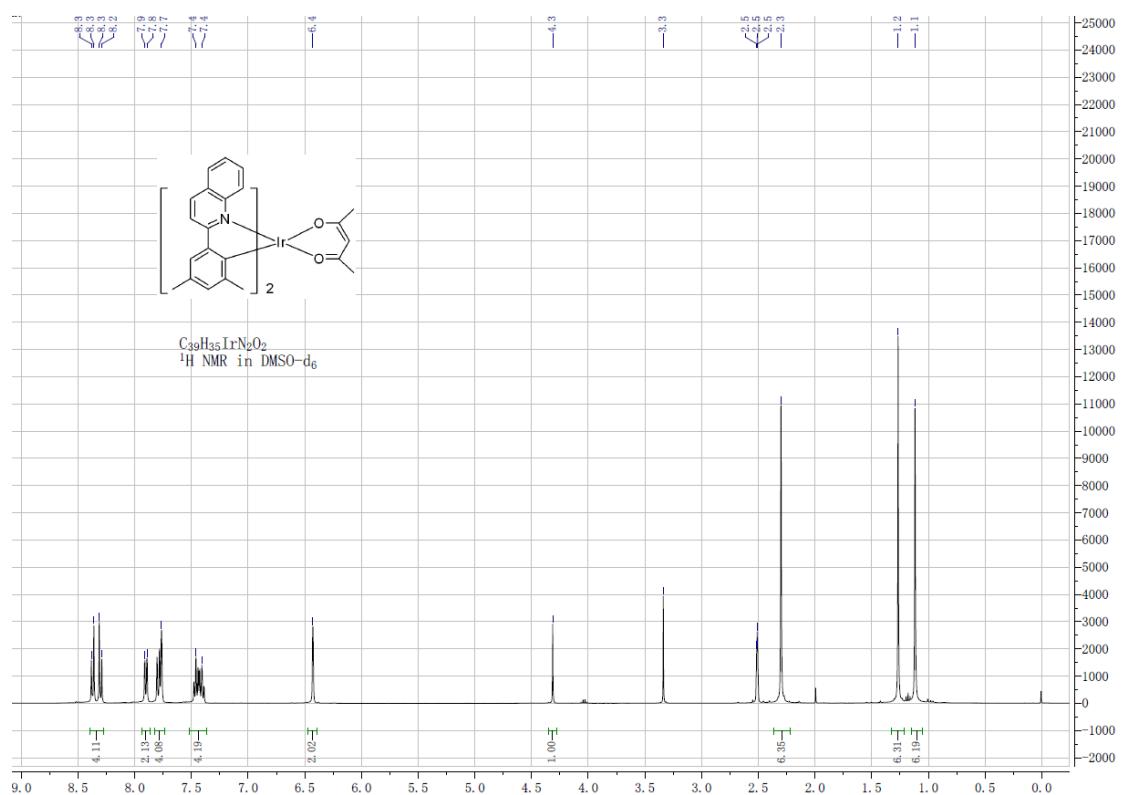


Figure S2. ¹H NMR spectrum of complex **2** in DMSO-d₆.

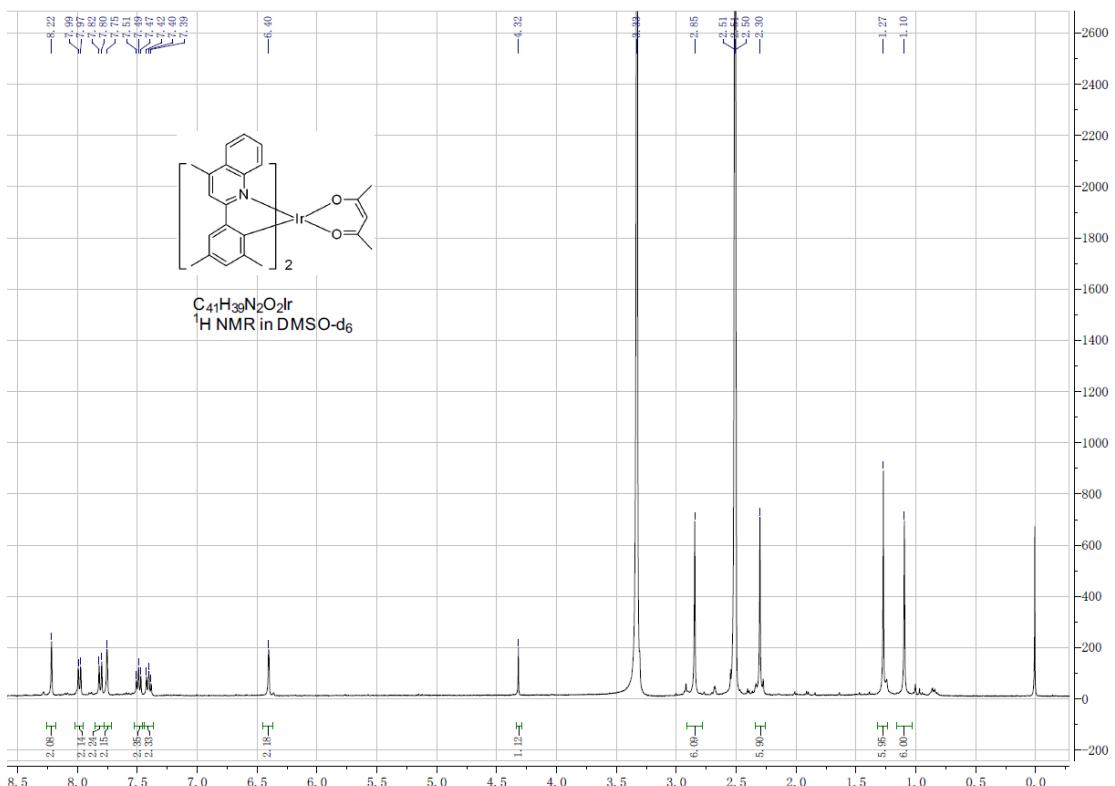


Figure S3. ^1H NMR spectrum of complex **3** in DMSO-d_6 .

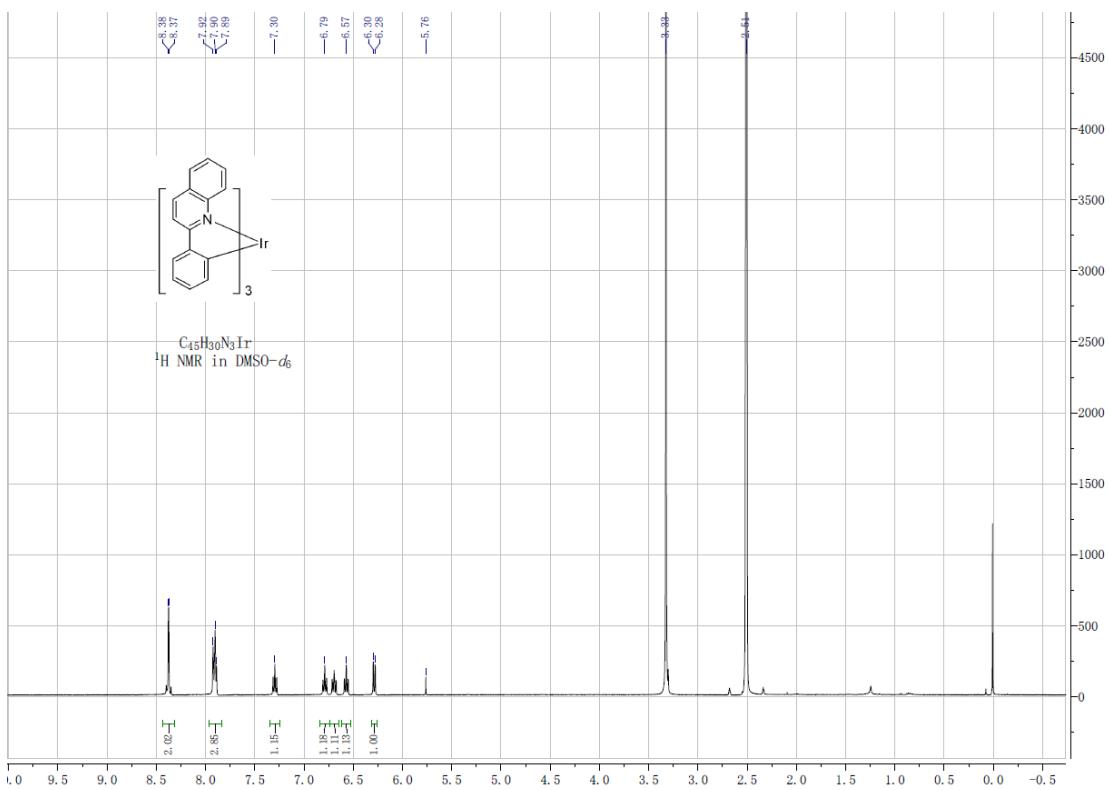


Figure S4. ^1H NMR spectrum of complex 4 in $\text{DMSO}-d_6$.

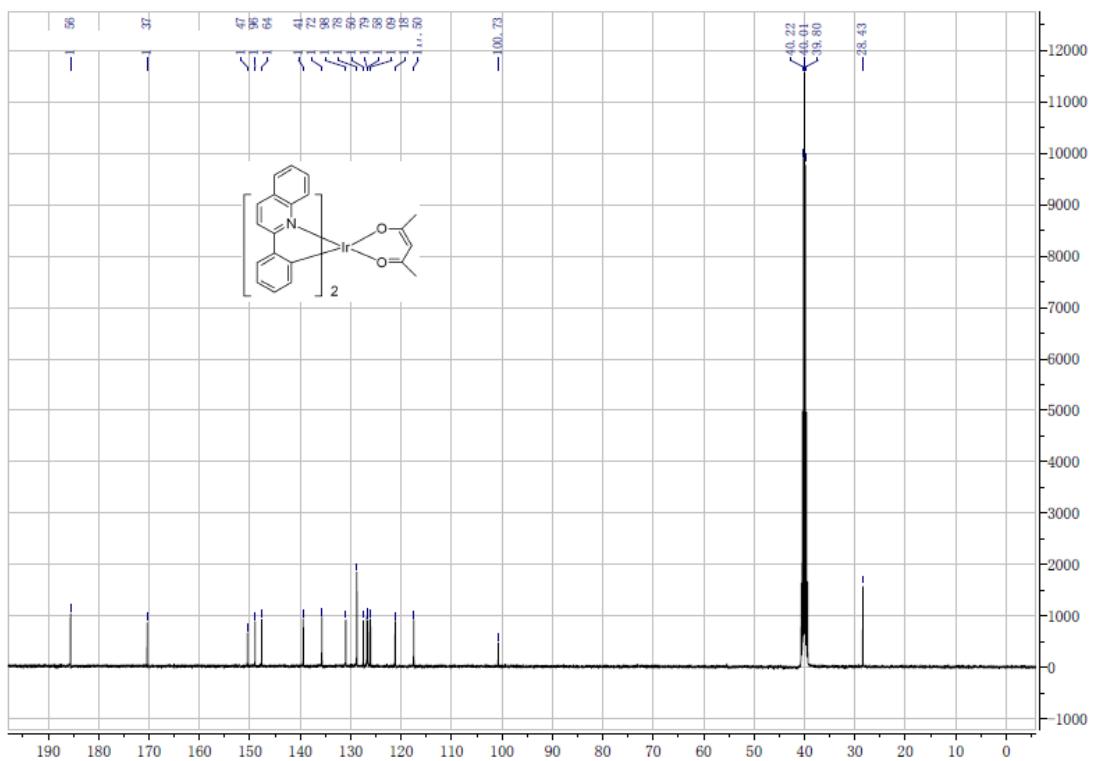


Figure S5. ^{13}C NMR spectrum of complex **1** in DMSO-d_6 .

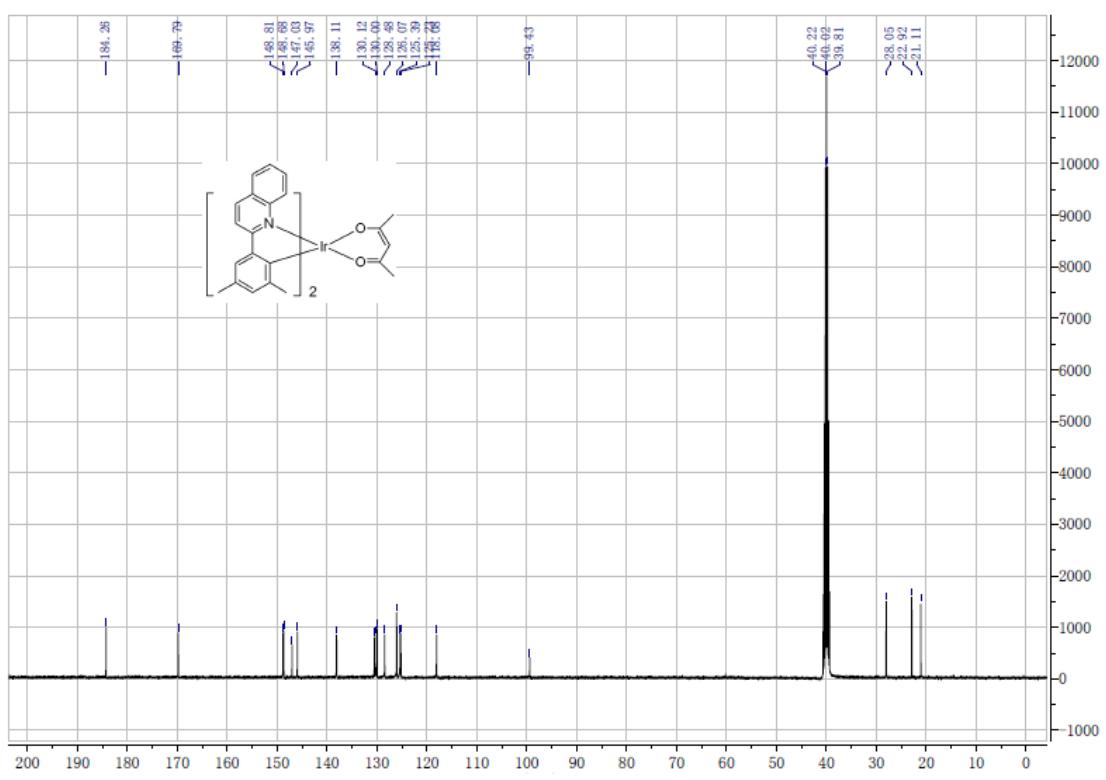


Figure S6. ^{13}C NMR spectrum of complex **2** in DMSO-d_6 .

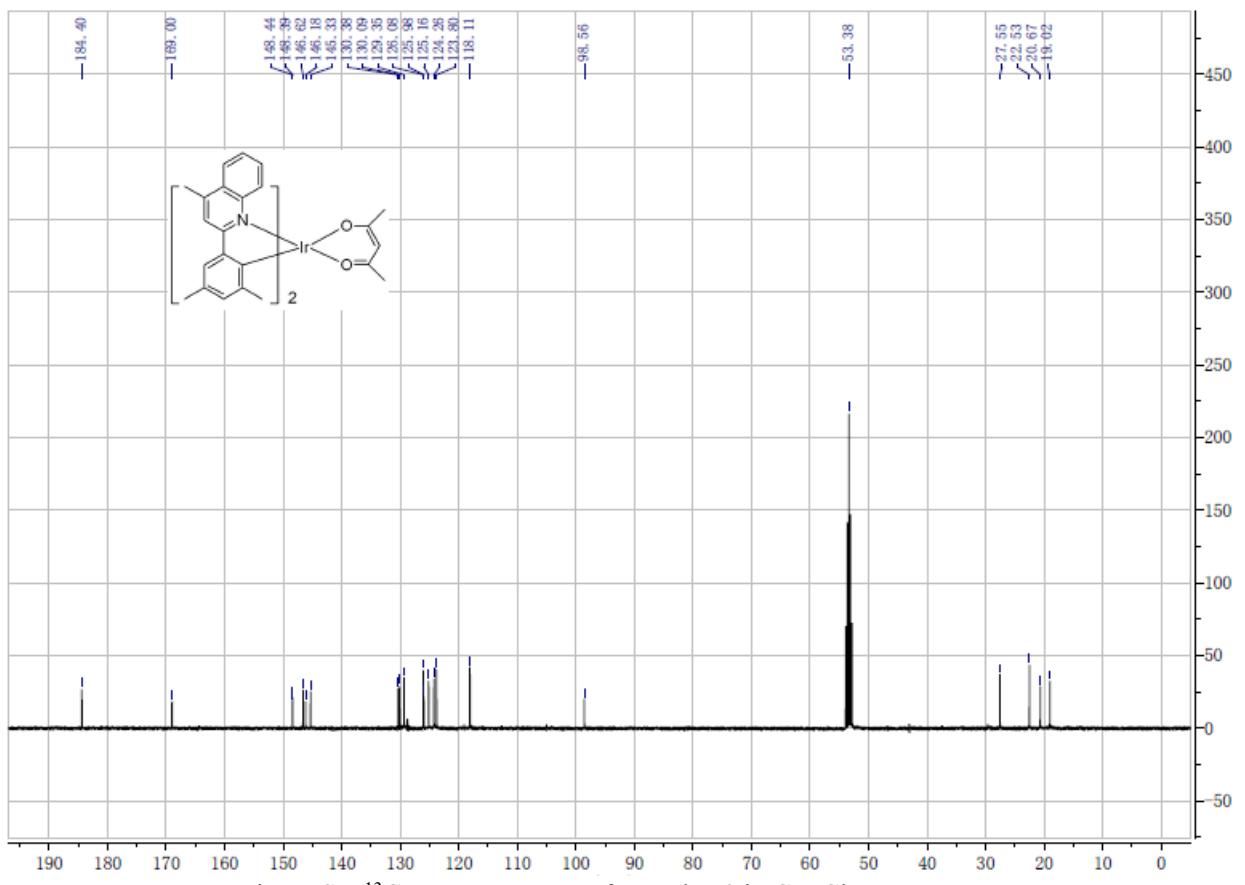


Figure S7. ^{13}C NMR spectrum of complex **3** in CD_2Cl_2 .

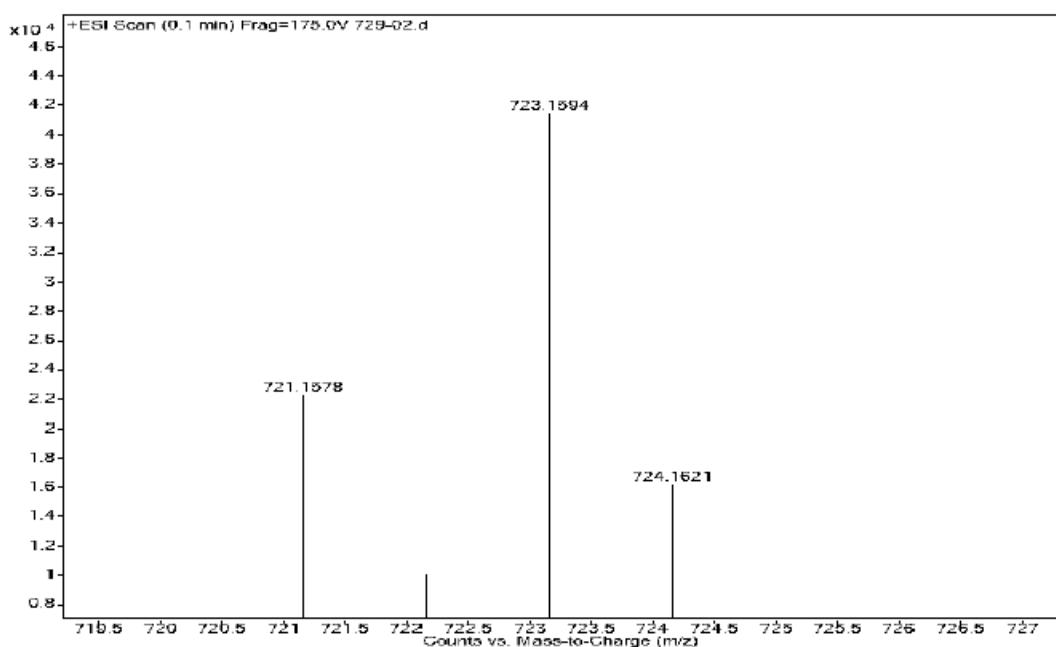


Figure S8. Tof-mass spectrum of complex **1**.

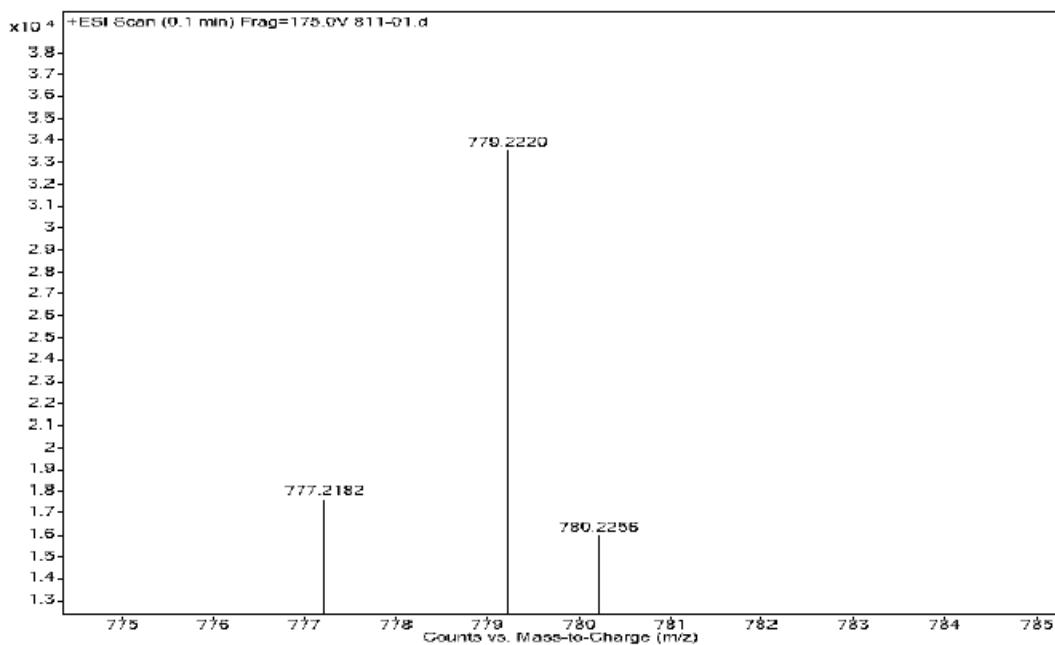


Figure S9. Tof-mass spectrum of complex **2**.

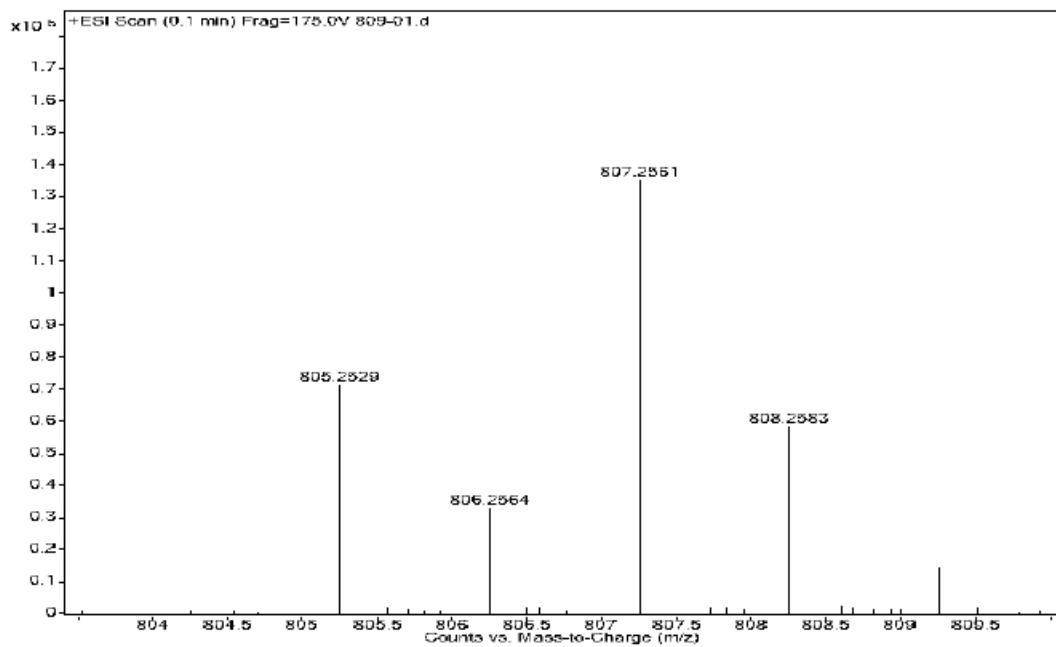


Figure S10. Tof-mass spectrum of complex 3.

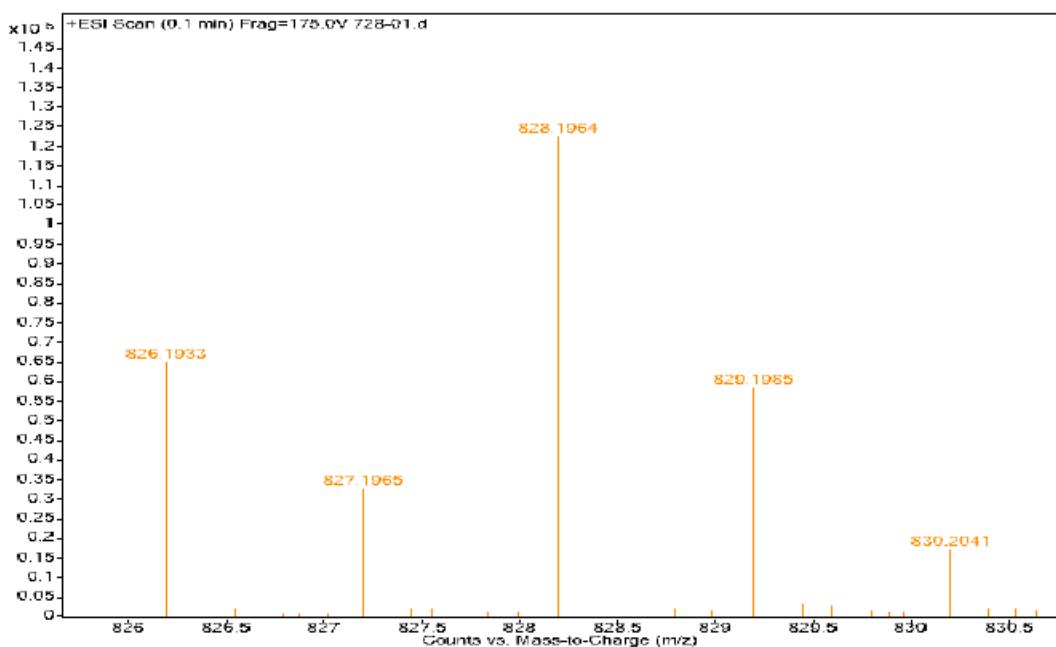


Figure S11. Tof-mass spectrum of complex 4.

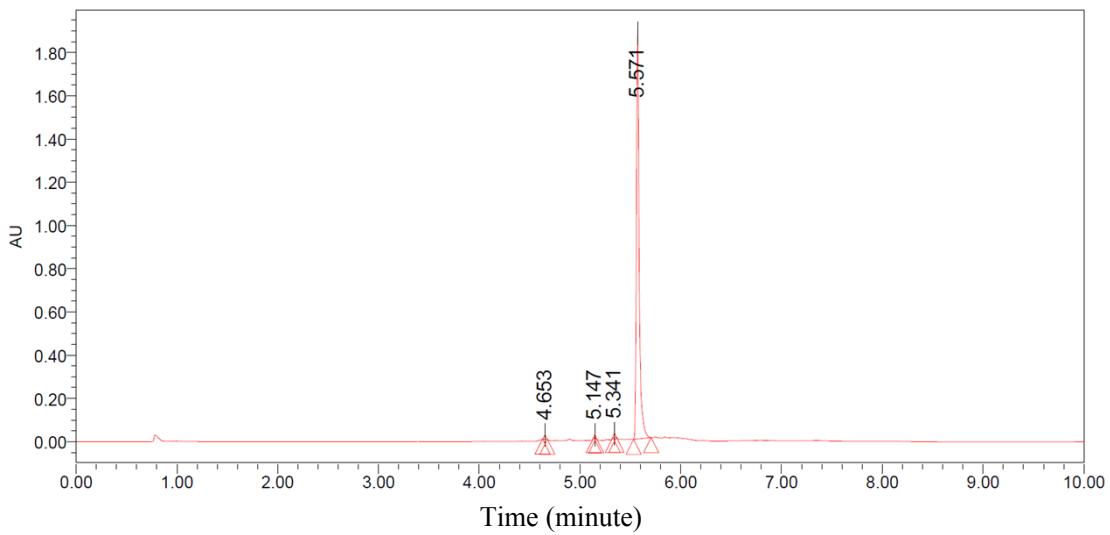


Figure S12. UPLC spectrum of complex **1**.

Table S1. The UPLC result of complex 1.

	Time (minute)	Area ($\mu\text{V}\times\text{s}$)	Height (μV)	% area
1	4.653	28101	19692	0.78
2	5.147	11867	14255	0.33
3	5.341	24266	17481	0.67
4	5.571	3559368	1890618	98.23

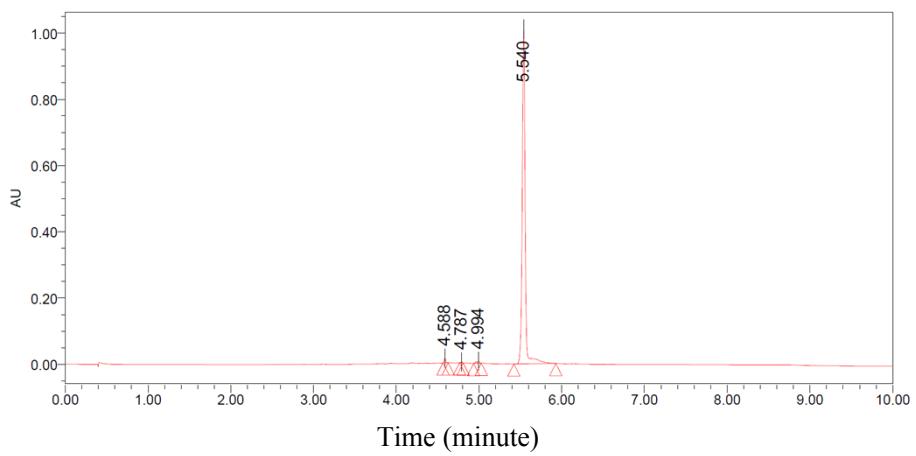


Figure S13. UPLC spectrum of complex **2**.

Table S2. The UPLC result of complex **2**.

	Time (minute)	Area ($\mu\text{V}\times\text{s}$)	Height (μV)	% area
1	4.588	17692	13426	0.70
2	4.787	3570	2738	0.14
3	4.994	16333	6311	0.65
4	5.540	2481385	1011216	98.51

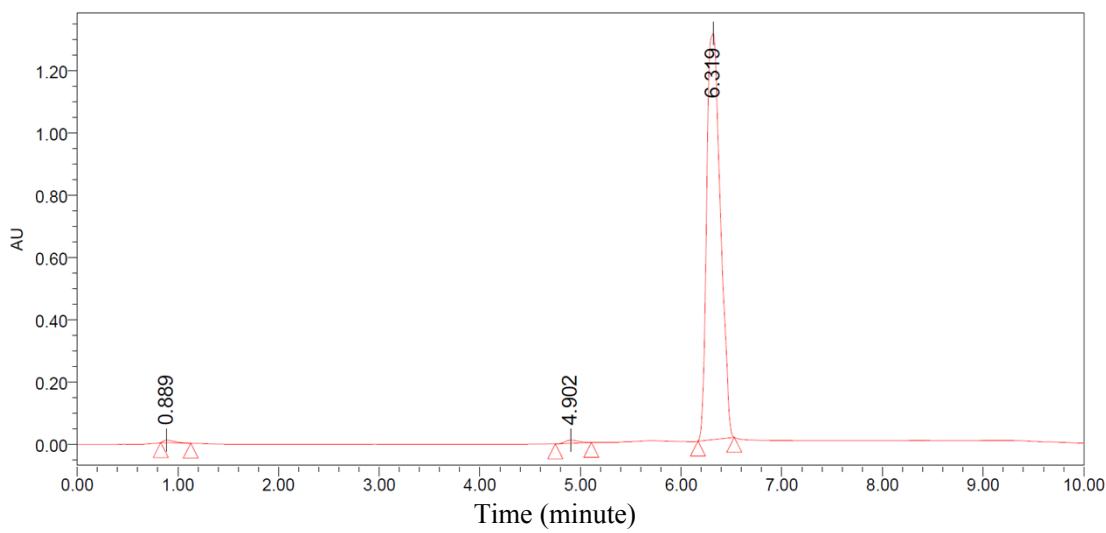


Figure S14. UPLC spectrum of complex 3.

Table S3. The UPLC result of complex 3.

	Time (minute)	Area ($\mu\text{V}\times\text{s}$)	Height (μV)	% area
1	0.889	62776	8409	0.50
2	4.902	92601	9969	0.74
3	6.319	12298873	1304041	98.75

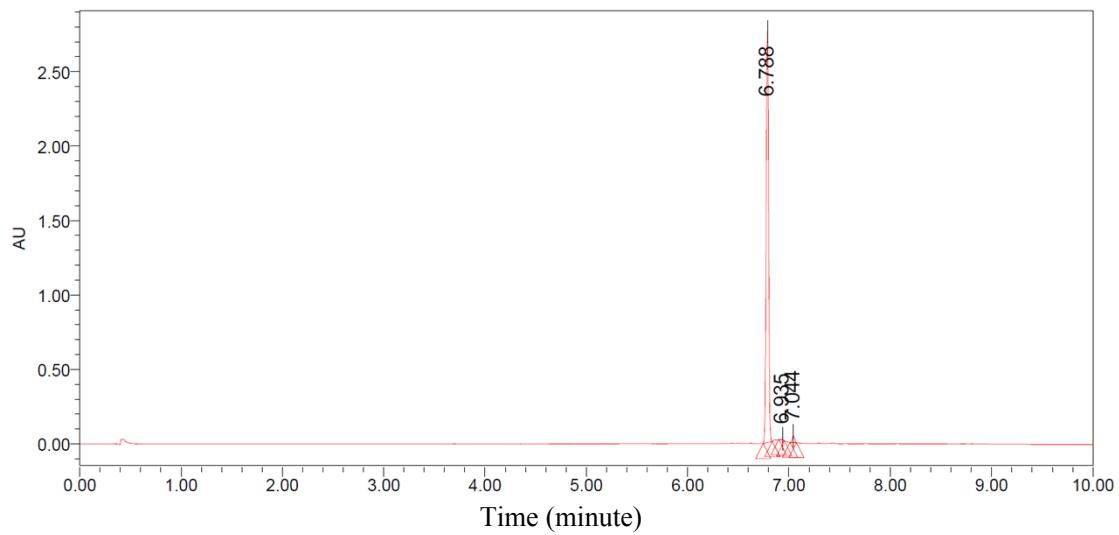


Figure S15. UPLC spectrum of complex 4.

Table S4. The UPLC result of complex 4.

	Time (minute)	Area ($\mu\text{V}\times\text{s}$)	Height (μV)	% area
1	6.788	4658738	2761596	98.13
2	6.935	12346	8471	0.26
3	7.044	76537	46796	1.61

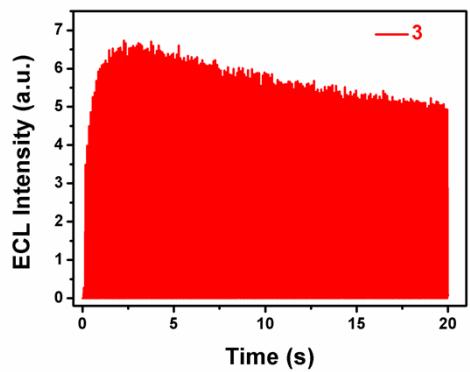


Figure S16. ECL intensity *vs* time plot of 0.1 mM complex **3** in acetonitrile solution through annihilation process. (oxidative potential= ($E_a^{\text{ox}}+0.2$) V, reductive potential= ($E_c^{\text{re}}-0.2$) V, 10 Hz, 20 s acquisition).