

## Electronic Supporting Information

for

### Iron(II) complexes featuring $\kappa^3$ - and $\kappa^2$ -bound PNP pincer ligands - The significance of sterics

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#### Synthesis and Characterization of Ligands **1e** and **1f**

**N,N'-Bis(*n*-propylphosphino)-2,6-diaminopyridine (PNP-*n*Pr) (**1e**)**. **1e** was prepared analogously to **1d** with 2,6-diaminopyridine (1.1 g, 10.0 mmol) and *n*Pr<sub>2</sub>PCl (3.0 g, 20 mmol) as starting materials. After removal of the solvent under reduced pressure **1e** was obtained as colorless oil. Yield: 3.37 g (98 %). Anal. Calcd. for C<sub>17</sub>H<sub>33</sub>N<sub>3</sub>P<sub>2</sub> (341.41): C, 59.81; H, 9.74; N, 12.31. Found: C, 59.83; H, 9.61; N, 12.60. <sup>1</sup>H NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 7.15 (t, J<sub>HH</sub> = 7.0 Hz, 1H, py<sup>4</sup>), 6.32 (d, J<sub>HH</sub> = 7.8 Hz, 2H, py<sup>3,5</sup>), 4.23 (d, J<sub>HP</sub> = 8,5 Hz, 2H, NH), 1.49-1.46 (m, 16H, CH<sub>2</sub>), 0.96 (m, 12H, CH<sub>3</sub>). <sup>13</sup>C{<sup>1</sup>H} NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 158.7 (d, J<sub>CP</sub> = 18.3 Hz, py), 139.2 (d, J<sub>CP</sub> = 10.7 Hz, py), 97.9 (d, J<sub>CP</sub> = 18.0 Hz, py), 34.3 (d, J<sub>CP</sub> = 11.1 Hz, CH<sub>2</sub>), 18.1 (d, J<sub>CP</sub> = 13.6 Hz, CH<sub>2</sub>), 15.7 (d, J<sub>CP</sub> = 11.8 Hz, CH<sub>3</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 26.2.

**N,N'-Bis(*n*-butylphosphino)-2,6-diaminopyridine (PNP-*n*Bu) (**1f**)**. **1f** was prepared analogously to **1d** with 2,6-diaminopyridine (450 mg, 4.4 mmol) and *n*Bu<sub>2</sub>PCl (1.6 g, 8.8 mmol) as starting materials. After removal of the solvent under reduced pressure a colorless oil was obtained. Yield: 1.68 g (96 %). Anal. Calcd. for C<sub>21</sub>H<sub>41</sub>N<sub>3</sub>P<sub>2</sub> (397.52): C, 63.45; H, 10.40; N, 10.57. Found: C, 64.23; H, 10.21; N, 10.71. <sup>1</sup>H NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 7.18 (t, J<sub>HH</sub> = 7.6 Hz, 1H, py<sup>4</sup>), 6.29 (d, J<sub>HH</sub> = 7.7 Hz, 2H, py<sup>3,5</sup>), 4.14 (d, J<sub>HP</sub> = 9.0 Hz, 2H, NH), 1.4-1.31 (m, 24H, CH<sub>2</sub>), 0.80 (m, 12H, CH<sub>3</sub>). <sup>13</sup>C{<sup>1</sup>H} NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 158.6 (d, J<sub>CP</sub> = 18.4 Hz, py), 139.3 (d, J<sub>CP</sub> = 14.3 Hz, py), 97.9 (d, J<sub>CP</sub> = 18.1 Hz, py), 31.7 (d, J<sub>CP</sub> = 11.2 Hz, CH<sub>2</sub>), 26.8 (d, J<sub>CP</sub> = 13.1 Hz, CH<sub>2</sub>), 24.2 (d, J<sub>CP</sub> = 11.6 Hz, CH<sub>2</sub>), 18.8 (CH<sub>3</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR ( $\delta$ , CDCl<sub>3</sub>, 20°C): 27.0.

**Synthesis and Characterization of Complexes **2fBPh<sup>Me</sup><sub>4</sub>, 2gBF<sub>4</sub>, 2hBF<sub>4</sub> 5dB<sub>4</sub>, 5eBF<sub>4</sub>, and 5fBF<sub>4</sub>.****

**[Fe( $\kappa^3$ -P,N,P-PNP-Et)( $\kappa^2$ -P,N-PNP-Et)Br]BPh<sup>Me</sup><sub>4</sub> (**2fBPh<sup>Me</sup><sub>4</sub>**).** The synthesis of **2fBPh<sup>Me</sup><sub>4</sub>** was performed analogously to **2eBPh<sup>Me</sup><sub>4</sub>** with PNP-Et (**1d**) (200 mg, 0.70 mmol) and anhydrous FeBr<sub>2</sub> (151 mg, 0.35 mmol) as starting materials. Yield 232 mg (95%) of [Fe( $\kappa^3$ -P,N,P-PNP-Et)( $\kappa^2$ -P,N-PNP-Et)Br]Br. Anal. Calcd. for C<sub>26</sub>H<sub>50</sub>Br<sub>2</sub>FeN<sub>6</sub>P<sub>4</sub> (786.26): C, 39.72; H, 6.41; N, 10.69. Found: C, 39.10; H, 6.29; N, 10.05. The green precipitate was suspended in 8 mL of THF and NaBPh<sup>Me</sup><sub>4</sub> was added (139 mg, 0.35 mmol). Workup was done analogously to the procedure of **2c**. Crystals of **2fBPh<sup>Me</sup><sub>4</sub>** were grown from a solution of THF by slow diffusion of diethyl ether. Yield: 352 mg (93%). Anal. Calcd. for C<sub>54</sub>H<sub>78</sub>BBrFeN<sub>6</sub>P<sub>4</sub>·C<sub>12</sub>H<sub>24</sub>O<sub>3</sub> (1298.04): C, 61.07; H, 7.92; N, 6.47. Found: C, 60.85; H, 7.61; N, 6.51. <sup>1</sup>H NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 8.25 (s, 2H, NH), 7.98 (s, 1H, NH), 7.50 (t, J<sub>HP</sub> = 8.0 Hz, 1H, py), 7.23 (m, 8H, Ph), 7.09 (t, J<sub>HP</sub> = 7.8 Hz, 1H, py), 6.74 (m, 8H, Ph), 6.65 (d, J<sub>HP</sub> = 8.0 Hz, 2H, py), 6.40 (dd, J<sub>HP</sub> = 4.0 Hz, 1H, py), 6.09 (d, J<sub>HP</sub> = 7.0 Hz, 1H, py), 4.81 (d, J<sub>HP</sub> = 10.0 Hz, 1H, NH), 2.85 - 2.54 (m, 8H, CH<sub>2</sub>), 2.51 - 2.34 (m, 4H, CH<sub>2</sub>), 2.15 (s, 12H, CH<sub>3</sub>), 2.07 - 2.00 (m, 18H, CH<sub>3</sub>), 1.38 - 1.31 (m, CH<sub>2</sub>), 1.10 - 0.85 (m, CH<sub>3</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): A<sub>2</sub>B spin system,  $\delta_A$  = 114.8 (2P),  $\delta_B$  = 114.7 (1P), J<sub>PP</sub> = 50 Hz (shifts and J<sub>PP</sub> determined from simulation), 36.5 (1P).

**[Fe( $\kappa^3$ -P,N,P-PNP-nPr)( $\kappa^2$ -P,N-PNP-nPr)Cl]BF<sub>4</sub> (**2gBF<sub>4</sub>**).** To a solution of PNP-nPr (**1e**) (200 mg, 0.59 mmol) in acetone (7 mL) anhydrous FeCl<sub>2</sub> (37 mg, 0.29 mmol) and NaBF<sub>4</sub> (32 mg, 0.29 mmol) was added and the mixture was stirred for 3 h. The green suspension was then filtrated over Celite and the solution was evaporated to dryness. The remaining solid was washed with diethyl ether (5 mL) and n-hexane (10 mL). The remaining green powder is dried in vacuum. Yield: 240 mg (95 %). Anal. Calcd. for C<sub>34</sub>H<sub>66</sub>BClF<sub>4</sub>FeN<sub>6</sub>P<sub>4</sub> (860.93): C, 47.43; H, 7.73; N, 9.76. Found: C, 46.99; H, 7.53; N, 9.88. <sup>1</sup>H NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 8.18 (s, 2H, NH), 7.97 (d, J<sub>HP</sub> = 6.1 Hz, 1H, NH), 7.49 (t, J<sub>HH</sub> = 8.0 Hz, 1H, py), 7.06 (t, J<sub>HH</sub> = 7.5 Hz, 1H, py<sup>4</sup>), 6.64 (d, J<sub>HH</sub> = 8.0 Hz, 2H, py), 6.31 (m, 1H, py), 6.07 (d, J<sub>HH</sub> = 7.8 Hz, 1H, py), 4.86 (d, J<sub>HP</sub> = 10.1 Hz, 1H, NH), 2.61-2.18 (m, 4H, CH<sub>2</sub>), 1.89-1.53 (m, 4H, CH<sub>2</sub>), 1.23-0.84 (m, 15H, CH<sub>2</sub>, CH<sub>3</sub>), 0.64 (t, J<sub>HP</sub> = 6.8 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C{<sup>1</sup>H} NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 165.5 (d, J<sub>CP</sub> = 18 Hz, py), 164.3-163.8 (m, py), 140.0 (py), 138.1 (py), 99.4 (py), 97.9 (py), 97.4 (py), 37.5 (d, J<sub>CP</sub> = 24.5 Hz, CH<sub>2</sub>), 32.7 (d, J<sub>CP</sub> = 11.0 Hz, CH<sub>2</sub>), 31.2 (m, CH<sub>2</sub>), 22.3 (m, CH<sub>2</sub>), 18.5 (d, J<sub>CP</sub> = 16.1 Hz, CH<sub>2</sub>), 17.1 (CH<sub>3</sub>) 16.7 (d, J<sub>CP</sub> = 3.7 Hz, CH<sub>3</sub>), 15.4-14.7 (m, CH<sub>2</sub>, CH<sub>3</sub>), 13.4 (CH<sub>3</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): A<sub>2</sub>B spin-system,  $\delta_A$  = 110.1 (2P),  $\delta_B$  = 107.5 (1P), J<sub>PP</sub> = 50 Hz (shifts and J<sub>PP</sub> determined from simulation), 27.5 (1P).

**[Fe( $\kappa^3$ -P,N,P-PNP-nBu)( $\kappa^2$ -P,N-PNP-nBu)Cl]BF<sub>4</sub> (**2hBF<sub>4</sub>**).** This compound was prepared analogously to **2gBF<sub>4</sub>** using PNP-nBu (**1f**) (200 mg, 0.50 mmol) in acetone (7 mL), anhydrous FeCl<sub>2</sub> (32 mg, 0.25 mmol) and NaBF<sub>4</sub> (28 mg, 0.25 mmol) as reactants. Yield: 235 mg (96 %). Anal. Calcd. for C<sub>42</sub>H<sub>82</sub>BClF<sub>4</sub>FeN<sub>6</sub>P<sub>4</sub> (973.15): C, 51.84; H, 8.49; N, 8.64. Found: C, 52.03; H, 8.23; N, 8.76. <sup>1</sup>H NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 8.42 (s, 2H, NH<sup>3</sup>), 8.31 (d, J<sub>HP</sub> = 5.3 Hz, 1H, NH<sup>2</sup>), 7.47 (t, J<sub>HH</sub> = 7.5 Hz, 1H, py<sup>4</sup>), 7.04 (t, J<sub>HH</sub> = 7.9 Hz, 2H, py<sup>4</sup>), 6.63 (d, J<sub>HH</sub> = 7.7 Hz, 1H, py<sup>3,5</sup>), 6.29 (m, 1H, py<sup>3</sup>), 6.09 (d, J<sub>HH</sub> = 7.4 Hz, 1H, py<sup>5</sup>), 4.85 (d, J<sub>HP</sub> = 10.2 Hz, 1H, NH), 2.71-2.20 (m, 4H, CH<sub>2</sub>), 1.87-1.14 (m, 20H, CH<sub>2</sub>), 0.88 (m, 9H, CH<sub>3</sub>), 0.65 (m, 3H, CH<sub>3</sub>). <sup>13</sup>C{<sup>1</sup>H} NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 165.7 (d, J<sub>CP</sub> = 21 Hz, py), 165.3-164.3 (m, py), 139.8 (py), 138.0 (py), 99.2 (py), 97.8 (py), 97.3 (py), 39.6 (m, CH<sub>2</sub>), 35.0 (m, CH<sub>2</sub>), 30.2 (d, J<sub>HP</sub> = 11.0 Hz, CH<sub>2</sub>), 27.2 (d, J<sub>CP</sub> = 15.5 Hz, CH<sub>2</sub>), 25.6-25.2 (m, CH<sub>2</sub>), 24.2-23.5 (m, CH<sub>2</sub>, CH<sub>3</sub>), 13.2 (d, J<sub>CP</sub> = 17.5 Hz, CH<sub>3</sub>), 12.9 (d, J<sub>CP</sub> = 22.9 Hz, CH<sub>3</sub>). <sup>31</sup>P{<sup>1</sup>H} NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): A<sub>2</sub>B spin-system,  $\delta_A$  = 110.3 (2P),  $\delta_B$  = 107.6 (1P), J<sub>PP</sub> = 50 Hz (shifts and J<sub>PP</sub> determined from simulation), 28.0 (1P).

**[Fe( $\kappa^3$ -P,N,P-PNP-Et)<sub>2</sub>](BF<sub>4</sub>)<sub>2</sub> (**5dB<sub>4</sub>**).** This complex was prepared in analogous fashion to **5cBF<sub>4</sub>** with **1d** (200 mg, 0.70 mmol), anhydrous FeCl<sub>2</sub> (44 mg, 0.35 mmol), and AgBF<sub>4</sub> (136 mg, 0.70 mmol) as starting materials. Yield: 265 mg (95%). Anal. Calcd. for C<sub>26</sub>H<sub>50</sub>B<sub>2</sub>F<sub>8</sub>FeN<sub>6</sub>P<sub>4</sub> (800.08): C, 39.03; H,

6.30; N, 10.50. Found: C, 39.18; H, 6.17; N, 10.21.  $^1\text{H}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 7.86 (s, 4H, NH), 7.34 (m, 18H, py<sup>4</sup>, Ph), 6.89 (m, 16H, Ph), 6.77 (m, 8H, Ph), 6.37 (d,  $J_{\text{HP}} = 7.8$  Hz, 4H, py<sup>3,5</sup>), 2.87 (m, 16H, CH<sub>2</sub>), 1.03 (m, 24H, CH<sub>3</sub>).  $^{13}\text{C}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 163.3 (py), 141.7 (py), 100.6 (py), 25.6 (CH<sub>2</sub>), 7.71 (CH<sub>3</sub>).  $^{31}\text{P}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 119.0. Crystals suitable for X-ray crystallography were grown with BPh<sub>4</sub><sup>-</sup> as counterion (analogously prepared with NaBPh<sub>4</sub> as halide scavenger) from an acetone solution by slow diffusion of diethyl ether.

**[Fe( $\kappa^3$ -P,N,P-PNP-*n*Pr)<sub>2</sub>](BF<sub>4</sub>)<sub>2</sub> (**5e**BF<sub>4</sub>).** This compound was prepared analogously to **5c**BF<sub>4</sub> using PNP-*n*Pr (200 mg, 0.59 mmol) in acetone (10 mL), anhydrous FeCl<sub>2</sub> (37 mg, 0.29 mmol) and AgBF<sub>4</sub> (114 mg, 0.59 mmol) as reactants. Yield: 254 mg (95 %). Anal. Calcd. for C<sub>34</sub>H<sub>66</sub>B<sub>2</sub>F<sub>8</sub>FeN<sub>6</sub>P<sub>4</sub> (912.29): C, 44.76; H, 7.29; N, 9.20. Found: C, 44.56; H, 7.32; N, 9.50.  $^1\text{H}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 7.74 (s, 4H, NH), 7.34 (t,  $J_{\text{HH}} = 7.8$  Hz, 2H, py<sup>4</sup>), 6.36 (d,  $J_{\text{HH}} = 7.9$  Hz, 4H, py<sup>3,5</sup>), 2.65 (m, 8H, CH<sub>2</sub>), 2.34 (m, 8H, CH<sub>2</sub>), 0.98 (t,  $J_{\text{HH}} = 6.4$  Hz, 24H, CH<sub>3</sub>).  $^{13}\text{C}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 162.5 (py), 140.7 (py), 99.5 (py), 27.5 (CH<sub>2</sub>), 16.6 (CH<sub>2</sub>), 14.9 (CH<sub>3</sub>).  $^{31}\text{P}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 114.2. Crystals suitable for X-ray crystallography were grown from a THF solution by slow diffusion of *n*-pentane.

**[Fe( $\kappa^3$ -P,N,P-PNP-*n*Bu)<sub>2</sub>](BF<sub>4</sub>)<sub>2</sub> (**5f**BF<sub>4</sub>).** This compound was prepared analogously to **5c**BF<sub>4</sub> using PNP-*n*Pr (200 mg, 0.50 mmol) in acetone (10 mL), anhydrous FeCl<sub>2</sub> (32 mg, 0.25 mmol) and AgBF<sub>4</sub> (98 mg, 0.50 mmol) as reactants. Yield: 211 mg (92 %). Anal. Calcd. for C<sub>42</sub>H<sub>82</sub>B<sub>2</sub>F<sub>8</sub>FeN<sub>6</sub>P<sub>4</sub> (1024.50): C, 49.24; H, 8.07; N, 8.20. Found: C, 49.43; H, 8.12; N, 8.08.  $^1\text{H}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 7.69 (s, 4H, NH), 7.32 (s, 2H, py<sup>4</sup>), 6.33 (s, 4H, py<sup>3,5</sup>), 2.59 (m, 16H, CH<sub>2</sub>), 1.33 (m, 32H, CH<sub>2</sub>), 0.76 (m, 24H, CH<sub>3</sub>).  $^{13}\text{C}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 162.5 (py), 140.5 (py), 99.6 (py), 27.7 (CH<sub>2</sub>), 25.1.6 (CH<sub>2</sub>), 23.8 (CH<sub>2</sub>), 13.0 (CH<sub>3</sub>).  $^{31}\text{P}\{\text{H}\}$  NMR ( $\delta$ , acetone-d<sub>6</sub>, 20°C): 115.0.

**Table S1.** Details for the crystal structure determinations of complexes **2e**, **2f**, **3**, **5a**, **5b**, **5c**, **5d**, and **5e**. -- Part 1

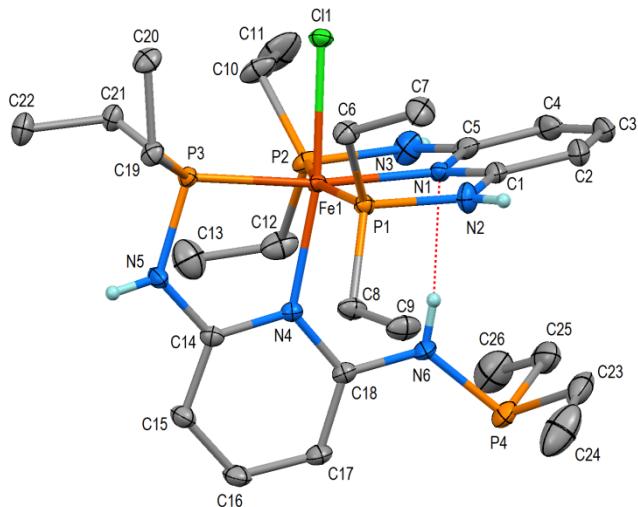
Compound	<b>2eBPh<sup>Me</sup>4</b> .3THF – [Fe( $\kappa^3$ -P,N,P-PNP-Et)( $\kappa^2$ -P,N-PNP-Et)Cl]BPh <sup>Me</sup> 4.3THF	<b>2fBPh<sup>Me</sup>4</b> .3THF -- [Fe( $\kappa^3$ -P,N,P-PNP-Me-Ph)( $\kappa^2$ -P,N-PNP-Et)Br]BPh <sup>Me</sup> 4.3THF	<b>3BF<sub>4</sub></b> -- [Fe( $\kappa^3$ -P,N,P-PNP-Me-Ph)( $\kappa^2$ -P,N-PNP-NHMe-Ph)Cl]BF <sub>4</sub>	<b>5aCl.solv</b> – [Fe( $\kappa^3$ -P,N,P-PNP-Ph) <sub>2</sub> ]Cl <sub>2</sub> ·solv
CCDC number	1005380	1005381	1005382	1005385
formula	C <sub>66</sub> H <sub>102</sub> B <sub>2</sub> Cl <sub>2</sub> FeN <sub>6</sub> O <sub>3.16</sub> P <sub>4</sub>	C <sub>66</sub> H <sub>102</sub> B <sub>2</sub> Br <sub>2</sub> FeN <sub>6</sub> O <sub>3</sub> P <sub>4</sub>	C <sub>50</sub> H <sub>49</sub> B <sub>2</sub> Cl <sub>2</sub> FeN <sub>6</sub> P <sub>3</sub>	C <sub>58</sub> H <sub>50</sub> Cl <sub>2</sub> FeN <sub>6</sub> P <sub>4</sub> <sup>b)</sup>
fw	1256.1	1298.0	1005.0	1081.67
cryst.size, mm	0.72 x 0.35 x 0.25	0.50 x 0.32 x 0.06	0.46 x 0.40 x 0.19	0.55 x 0.42 x 0.35
color, shape	green rod	green plate	green prism	red block
crystal system	triclinic	triclinic	monoclinic	tetragonal
space group	<i>P</i> – <i>1</i> (no. 2)	<i>P</i> – <i>1</i> (no. 2)	<i>P</i> 2 <sub>1</sub> / <i>n</i> (no. 14)	<i>I</i> –42d (no. 122)
<i>a</i> , Å	12.725(2)	12.778(5)	14.5198(17)	15.1224(2)
<i>b</i> , Å	15.628(3)	15.546(6)	21.308(3)	15.1224(2)
<i>c</i> , Å	18.643(3)	18.790(7)	15.1462(18)	30.8856(6)
$\alpha$ , deg	75.440(6)	75.954(7)	90	90
$\beta$ , deg	70.622(6)	70.741(7)	99.710(4)	90
$\gamma$ , deg	86.483(6)	86.697(8)	90	90
<i>V</i> , Å <sup>3</sup>	3384.0(11)	3417(2)	4618.9(10)	7063.1(2)
<i>T</i> , K	100	100	100	100
<i>Z</i>	2	2	4	4
$\rho_{\text{calc}}$ , g cm <sup>-3</sup>	1.233	1.2611	1.4448	1.017
$\mu$ , mm <sup>-1</sup> (MoKα)	0.405	0.947	0.549	0.414
<i>F</i> (000)	1346.6	1380	2080	2240
absorption corrections	multi-scan, 0.76–0.91	multi-scan, 0.70–0.95	multi-scan, 0.77–0.90	multi-scan, 0.87–0.77
$\theta$ range, deg	1.70–30.17	2.03–28.34	1.67–32.61	2.39–30.00
no. of rflns measd	155879	27496	275623	31615
<i>R</i> <sub>int</sub>	0.037	0.023	0.047	0.031
no. of rflns unique	19859	16613	16814	5139
no. of rflns $>2\sigma(I)$	16152	12623	13277	4943
no. of params / restraints	772 / 0	755 / 0	599 / 0	162 / 0
<i>R</i> <sub>1</sub> ( $I > 2\sigma(I)$ ) <sup>a)</sup>	0.0459	0.0436 ( $I > 3\sigma(I)$ )	0.0319 ( $I > 3\sigma(I)$ )	0.0259
<i>R</i> <sub>1</sub> (all data)	0.0620	0.0599	0.0487	0.0278
<i>wR</i> <sub>2</sub> ( $I > 2\sigma(I)$ )	0.1112	0.1036 ( $I > 3\sigma(I)$ )	0.0868 ( $I > 3\sigma(I)$ )	0.0692
<i>wR</i> <sub>2</sub> (all data)	0.1236	0.1110	0.0976	0.0707
GooF	1.02	1.42	1.63	1.06
Diff.Four.peaks min/max, eÅ <sup>-3</sup>	-1.19 / 1.56	-0.45 / 0.91	-0.29 / 0.42	-0.22 / 0.34

<sup>a</sup>  $R1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ ,  $wR2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$ , GooF =  $\{\sum [w(F_o^2 - F_c^2)^2] / (n-p)\}^{1/2}$

<sup>b</sup> Disordered solvent (methanol, diethyl ether) squeezed, solvent amount unknown and therefore not given in chemical formula or derived quantities.

**Table S1.** Details for the crystal structure determinations of complexes **2e**, **2f**, **3**, **5a**, **5b**, **5c**, **5d**, and **5e**. -- Part 2

Compound	<b>5b</b> CF <sub>3</sub> SO <sub>3</sub> ·6.5THF [Fe(κ <sup>3</sup> -P,N,P-PNP- bipol) <sub>2</sub> ](CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub> · 6.5THF	<b>5c</b> CF <sub>3</sub> SO <sub>3</sub> ·2(Me <sub>2</sub> CO) -- [Fe(κ <sup>3</sup> -P,N,P-PNP-Me) <sub>2</sub> ] (CF <sub>3</sub> SO <sub>3</sub> ) <sub>2</sub> ·(Me <sub>2</sub> CO) <sub>2</sub>	<b>5d</b> BPh <sub>4</sub> ·0.5Et <sub>2</sub> O -- [Fe(κ <sup>3</sup> -P,N,P-PNP-Et) <sub>2</sub> ] (BPh <sub>4</sub> ) <sub>2</sub> ·0.5Et <sub>2</sub> O	<b>5e</b> BF <sub>4</sub> ·2(Me <sub>2</sub> CO) -- [Fe(κ <sup>3</sup> -P,N,P-PNP- nPr) <sub>2</sub> ](BF <sub>4</sub> ) <sub>2</sub> ·(Me <sub>2</sub> CO) <sub>2</sub>
CCDC number	1005384	1005387	1005386	1024076
formula	C <sub>86</sub> H <sub>94</sub> F <sub>6</sub> FeN <sub>6</sub> O <sub>20.5</sub> P <sub>4</sub> S <sub>2</sub>	C <sub>26</sub> H <sub>46</sub> F <sub>6</sub> FeN <sub>6</sub> O <sub>8</sub> P <sub>4</sub> S <sub>2</sub>	C <sub>76</sub> H <sub>95</sub> B <sub>2</sub> FeN <sub>6</sub> O <sub>5</sub> P <sub>4</sub>	C <sub>40</sub> H <sub>78</sub> B <sub>2</sub> F <sub>8</sub> FeN <sub>6</sub> O <sub>2</sub> P <sub>4</sub>
fw	1897.52	928.54	1302.0	1028.5
cryst.size, mm	0.40 x 0.25 x 0.06	0.35 x 0.25 x 0.15	0.42 x 0.32 x 0.22	0.68 x 0.60 x 0.10
color, shape	red plate	red block	red block	red plate
crystal system	triclinic	monoclinic	triclinic	orthorhombic
space group	<i>P</i> ī (no. 2)	<i>C</i> 2/c (no. 15)	<i>P</i> ī (no. 2)	<i>Cc</i> c <sub>e</sub> (no. 68)
<i>a</i> , Å	13.4039(8)	13.6764(17)	13.3977(13)	15.1780(6)
<i>b</i> , Å	16.9467(9)	18.291(2)	14.1994(14)	21.4529(9)
<i>c</i> , Å	20.9385(13)	16.390(2)	21.6176(19)	16.2363(6)
$\alpha$ , deg	78.058(4)	90	93.677(3)	90
$\beta$ , deg	80.833(4)	98.886(2)	105.861(3)	90
$\gamma$ , deg	68.818(3)	90	117.314(3)	90
<i>V</i> , Å <sup>3</sup>	5283.5(3)	4051.0(9)	3427.3(6)	5286.7(4)
<i>T</i> , K	100	100	100	100
<i>Z</i>	2	4	2	4
$\rho_{\text{calc}}$ , g cm <sup>-3</sup>	1.459	1.522	1.261	1.292
$\mu$ , mm <sup>-1</sup> (MoKα)	0.386	0.712	0.362	0.473
<i>F</i> (000)	1976	1920	1386	2176
absorption corrections	multi-scan, 0.97–0.83	multi-scan, 0.79–0.90	multi-scan, 0.87–0.92	multi-scan, 0.73–0.95
$\theta$ range, deg	1.71–23.82	1.87–32.00	1.00–27.54	1.90–34.93
no. of rflns measd	41700	33718	134214	36808
<i>R</i> <sub>int</sub>	0.049	0.026	0.036	0.038
no. of rflns unique	12816	6952	15798	5804
no. of rflns $> 2\sigma(I)$	10115	6060	12425	4420
no. of params / restraints	1141 / 358	261 / 2	836 / 0	150 / 1
<i>R</i> <sub>1</sub> ( $I > 2\sigma(I)$ ) <sup>a</sup>	0.1130	0.0460	0.0328 ( $I > 3\sigma(I)$ )	0.0400 ( $I > 3\sigma(I)$ )
<i>R</i> <sub>1</sub> (all data)	0.1353	0.0525	0.0489	0.0588
<i>wR</i> <sub>2</sub> ( $I > 2\sigma(I)$ )	0.2895	0.1237	0.0870 ( $I > 3\sigma(I)$ )	0.1036 ( $I > 3\sigma(I)$ )
<i>wR</i> <sub>2</sub> (all data)	0.3093	0.1277	0.0939	0.1117
<i>GooF</i>	1.10	1.03	1.53	1.63
Diff.Four.peaks min/max, eÅ <sup>-3</sup>	-1.38 / 3.22	-0.91 / 1.20	-0.34 / 0.37	-0.29 / 0.64



**Figure S1.** Structural view of complex **2e** in **2eBPh<sup>Me</sup><sub>4</sub>·3THF** -  $[\text{Fe}(\kappa^3\text{-P},\text{N},\text{P}-\text{PNP-Et})(\kappa^2\text{-P},\text{N}-\text{PNP-Et})\text{Cl}] \text{BPh}^{\text{Me}}_4 \cdot 3\text{THF}$  - showing 50% thermal ellipsoids (H-atoms, solvents and  $\text{BPh}^{\text{Me}}_4^-$  omitted for clarity). Selected bond lengths ( $\text{\AA}$ ) and bond angles (deg): Fe(1)-P(1) 2.2406(6), Fe(1)-P(2) 2.2461(7), Fe(1)-P(3) 2.1858(6), Fe(1)-N(1) 2.059(1), Fe(1)-N(4) 2.117(1), Fe(1)-Cl(1) 2.3504(6), P(1)-N(2) 1.706(2), P(2)-N(3) 1.711(2), P(3)-N(5) 1.687(1), P(4)-N(6) 1.722(2), N(2)-C(1) 1.362(2), N(3)-C(5) 1.355(3), N(5)-C(14) 1.375(2), N(6)-C(18) 1.383(2), P(1)-Fe(1)-P(2) 163.85(2), N(1)-Fe(1)-P(3) 171.59(4), Cl(1)-Fe(1)-N(4) 170.50(4). **2eBPh<sup>Me</sup><sub>4</sub>·3THF** is isostructural with the corresponding Br complex **2fBPh<sup>Me</sup><sub>4</sub>·3THF**. In **2eBPh<sup>Me</sup><sub>4</sub>·3THF** about 15% of the phosphorus P4 is oxidized to the phosphinoxide.